

COMPUTER TEACHING-AIDS FOR AN  
UNDERGRADUATE COURSE IN DISTILLATION

BY

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in fulfilment of the requirements for the  
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## ABSTRACT

Four computer programs dealing with the following aspects of distillation are developed as an integrated teaching package.

- (1) Binary batch distillation in tray columns.
- (2) Binary continuous distillation in tray columns.
- (3) Binary continuous distillation in packed columns.
- (4) Multicomponent distillation in tray columns.

A plotting program utilizing CALCOMP plotting software and hardware is developed for graphical representation of the results generated by the above routines. The following diagrams are included:

- (1) McCabe-Thiele  $x, y$  diagrams for batch and continuous distillation.
- (2) Ponchon-Savarit diagram for continuous distillation.
- (3) Temperature, flowrate, and composition profiles for multicomponent distillation.

The programs are designed to supplement an undergraduate course in distillation at U.C.T. Mathematical and computational techniques used in the programs correspond to material presented in the course. A mainline-subroutine program structure is used throughout to facilitate future additions and alterations to the program.

The routines compare favourably with similar published teaching programs, with respect both to computer-time and to core-storage requirements. Program solutions to tutorial problems agree well with manual textbook calculations, in the design of tray columns. The program for designing packed columns has revealed discrepancies in the application of certain mass transfer correlations to distillation systems.

The CACHE format for describing programs is adopted in this text.



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## CHAPTER 1

### INTRODUCTION AND LITERATURE SURVEY

#### 1.1 INTRODUCTION

This study is concerned with the development of computer programs to be used as teaching aids in an undergraduate course on Distillation at the University of Cape Town.

##### 1.11 Historical Background

At the turn of the century distillation was more an art than a science, being associated mainly with the French wine industry. The basis of modern distillation theory was presented by Ernest Sorel (1893) in his book "La Rectification de l'alcool" and the first English text on the subject appeared in 1903. There was little progress in distillation technology, however, until the 1920's, when the distillation of crude oil attracted the interest of a number of research workers. Two algebraic techniques for solving steady-state multicomponent distillation problems were developed by Lewis and Matheson [1], and Thiele and Geddes [2]. These procedures form the basis of most modern computer algorithms. Robinson (1930) published the first teaching text on distillation, which was later to be co-edited by Gilliland [3].

Unfortunately the manual calculation methods of Lewis and Thiele proved to be tedious and time-consuming. Simplifying assumptions, that were necessary in most cases, reduced the precision of the techniques and engineers tended to rely on experience and rule-of-thumb design methods. In 1950, when the fourth edition of Robinson's book was published, Gilliland noted that for a number of years the "law of diminishing returns" had been applying to distillation research.



The late 1950's saw the advent of large, high-speed digital computers and interest in rigorous design methods was revived. The first computer algorithm for solving distillation problems was published by Amundson and Pontinen [4] in 1958. Several more followed soon afterwards, and throughout the 1960's considerable research effort was directed towards either improving the stability or reducing the computer-time requirements of the various procedures.

More recently it has been recognised that the accuracy of equilibrium, enthalpy and stage efficiency data is often the limiting factor in design or simulation of distillation systems. Ellis and Boyes [5] have recently reviewed the present extent of knowledge in distillation, indicating fields in which research is most urgently required.

Distillation is now a vital part of many industrial processes. Modern computers make possible the use of mathematical models that were previously too complex for convenient use. Zuiderweg [6] has estimated that research in distillation has saved industry a possible \$2000 million over the past 20 years.

Despite these advances the content of texts dealing with the subject at the undergraduate level has not changed appreciably since 1950. This study is concerned with the needs of students rather than qualified engineers.

#### 1.12 Objectives

The chemical engineer is expected to be familiar with all the separation processes commonly used in industry. Consequently undergraduate courses on these topics are invariably congested, and there is little time for more than a general treatment of each process. At U.C.T. a third-year course in mass-transfer operations presently allows approximately nine lectures and three tutorial sessions for distillation processes.

In this time students are taught the basic principles of both binary and multicomponent distillation, and must apply these principles to textbook problems. As even elementary problems can be time-consuming, solution procedures are restricted to the graphical techniques of McCabe and Thiele [7] and Ponchon and Savarit [8] for binary systems, and to various short-cut methods for multicomponent problems.

Due to the limited time available, problems are usually solved for a single fixed set of conditions. Although the effects of reflux ratio, feed condition, stage efficiency and other variables are discussed theoretically, they are seldom demonstrated in tutorials. As a result, the average student's understanding of distillation is often inadequate.

It was proposed that a set of computer programs be developed, capable of solving the problems normally set in tutorials. The programs would be made available to students who would then be able to:

- 1) test manual solutions,
- 2) investigate the effect of changes in key variables, and
- 3) analyse the results and comment on the operating conditions specified in a problem.

The objective of this study was to develop such a set of programs for solving distillation problems. The following requirements and programming guidelines were specified:

1. Type of distillation problem considered:
  - (a) Binary Batch distillation in Tray towers
  - (b) Binary Continuous distillation in Tray towers
  - (c) Binary Continuous distillation in Packed towers
  - (d) Multicomponent Continuous distillation in Tray towers.

2. Structure of Programs:

Modular, mainline-subroutine structure to be adopted, with generalised subroutines for standard

calculations such as bubble point or dew point. Programs to be structured to minimise core storage requirements.

3. Operation of Program:

Both Demand- and Batch-mode operating facilities to be provided, with emphasis on user-program interaction in Demand-mode.

4. Computer Algorithms:

Mathematical correlations and solution procedures to correspond to those discussed in the course lectures.

5. Data Collation:

Optional data input facilities to be included to minimise the time required to set up problem data tables.

6. Error messages:

Common solution errors, such as pinch points, to be searched for, and remedies suggested through program error messages.

7. Graphical solution procedures:

A plotting program to be developed for construction of McCabe-Thiele or Ponchon-Savarit diagrams for binary systems, and temperature, composition and flow profiles for multicomponent systems.

Further requirements were:

- (a) Thorough testing of programs using textbook examples;
- (b) A comprehensive User Manual for the program package.

## 1.2 LITERATURE SURVEY

This survey is divided into four sections. The first deals with availability and utilization of computer programs at the undergraduate level. Section 1.22 reviews distillation processes with specific reference to the undergraduate-course content at U.C.T. The selection and adaptation of solution procedures for the various programs are discussed in section 1.23. The last part is a detailed review of the numerous computer algorithms that are available for multi-component distillation design.

### 1.21 Use of computers for teaching purposes

Although computers have been used in industry for many years, for both design and control of equipment, their application in undergraduate courses has been fairly limited. In the past the only contact many students had with a computer was a basic course in FORTRAN programming, often given by a department outside the Engineering faculty. Wellknown texts on distillation, and other separation processes, such as Treybal [9] and McCabe and Smith [10] mention computerised solution procedures briefly, but do not consider them in tutorial examples. Buford Smith [11] gives a concise description of four computer algorithms for solving multi-component problems, but states that his book is written primarily for practising engineers.

Recent cost escalations have created a demand for more efficient design and control of industrial equipment. Computers are being used more extensively and engineers consequently require a better understanding of their application and capabilities. Since 1970 several universities in the U.S.A. have included computer programming as an integral part of courses in their Chemical Engineering curricula, rather than as a separate subject. The CACHE\* committee has compiled a comprehensive set of programs covering a wide range

\* CACHE - Computer Aids for Chemical Engineering Education.

of subjects. Many of the contributors are Chemical Engineering lecturers. All the programs are written specifically for undergraduate use and are readily available at a reasonable fee, in contrast to commercial programs which are often extremely expensive.

A recent text on distillation by Thibault-Brian [12] of M.I.T. deals extensively with computerised solution procedures. In common with the approach of several CACHE contributors, the tutorial examples require students to write their own programs to solve problems. With the limited computer facilities available at U.C.T., and the short duration of the course, it would not be practical to adopt a similar system here.

On examining the CACHE programs on distillation [13] it was found that they were, in general, unsuitable for the type of facility required at U.C.T. Many of them were merely examples of the type of program required of students as a tutorial solution. As all the programs were contributed by different authors they did not form an integrated package. Solution procedures used in some of the programs did not correspond to the U.C.T. course-content. Enthalpy effects were not considered at all in programs dealing with binary systems. A multicomponent algorithm required equilibrium and enthalpy data to be supplied as polynomials in temperature. Data for the sample problem shown with this program had been extracted from Amundson's paper [4]. To supply data in this form for new examples is usually inconvenient.

It was decided that a unique set of programs should be developed as an integrated package. The programs would be designed to supplement a course on distillation given at U.C.T.

#### 1.22 Distillation processes

Separation processes have as their common aim the isolation of a desired component, or group of components, from a mixture of the desired product and impurities. Distillation

is the separation, by vaporization, of a liquid mixture of miscible, volatile substances. The degree of separation achieved depends upon the distribution of the components between the two phases.

Specifically, the word "distillation" applies to a single vaporization process in which a liquid mixture is heated and the evolved vapour recovered, usually by condensation. If the vapour is kept in continuous, counter-current contact with a condensate, the process is termed "rectification". Usually, however, the two words are considered synonymous.

The principle of separation by successive vapour-liquid phase-distribution operations gives rise to the concept of an "equilibrium stage". The basic assumption of an equilibrium stage is that contact between the initial vapour and liquid phases within the stage is sufficient for the system to reach thermodynamic equilibrium. This concept is used to design a system of contact devices that will achieve the required degree of separation.

Counter-current contacting of the vapour and liquid phases is usually achieved in a column containing either trays or a packing material. In tray columns each theoretical tray represents an equilibrium stage whereas in packed column the stage is represented by a fixed height of packing. This height equivalent to a theoretical plate (HETP) is partly dependent on the characteristics of the packing material. Theoretical trays, or plates, are trays on which a true equilibrium is achieved. The performance of a real tray is related to an ideal equilibrium stage by a proportionality factor termed a stage-efficiency.

Modelling real trays in terms of ideal trays and efficiency factors is a far easier task than the direct solution of the complex heat and mass transfer operations associated

with a real tray. All solution procedures considered at the undergraduate level are based on ideal models that include efficiency factors.

In industry, distillation processes are run as either batch or continuous operations. Either operation can be carried out in tray or packed columns, to separate either binary or multicomponent solutions. The incidence of true binary solutions in practice is small, however, and consequently most of the research in distillation has involved multicomponent systems. The processes considered here have been listed in section 1.12.

Traditionally, in an introductory course on distillation binary systems are discussed first. The multicomponent process is introduced later as a complex extension of the binary case. The four basic equations describing the generalized equilibrium stage (Figure 1.1) may be expressed as follows:

1. The equilibrium relationship on stage  $j$

$$y_{i,j} = K_{i,j} x_{i,j} \quad (1.1)$$

2. The component material balance around stage  $j$

$$l_{i,j} + v_{i,j} - l_{i,j+1} - v_{i,j-1} - f_{i,j} = 0 \quad (1.2)$$

3. The enthalpy balance around stage  $j$

$$L_j h_j + V_j H_j - L_{j+1} h_{j+1} - V_{j-1} H_{j-1} - F_j h_{fj} + q_j' = 0 \quad (1.3)$$

4. The restriction on fractional concentrations on stage  $j$

$$\sum x_{i,j} = 1.0 \quad \text{and} \quad \sum y_{i,j} = 1.0 \quad (1.4)$$

The terms in the above equations are defined in the nomenclature. (Appendix B)

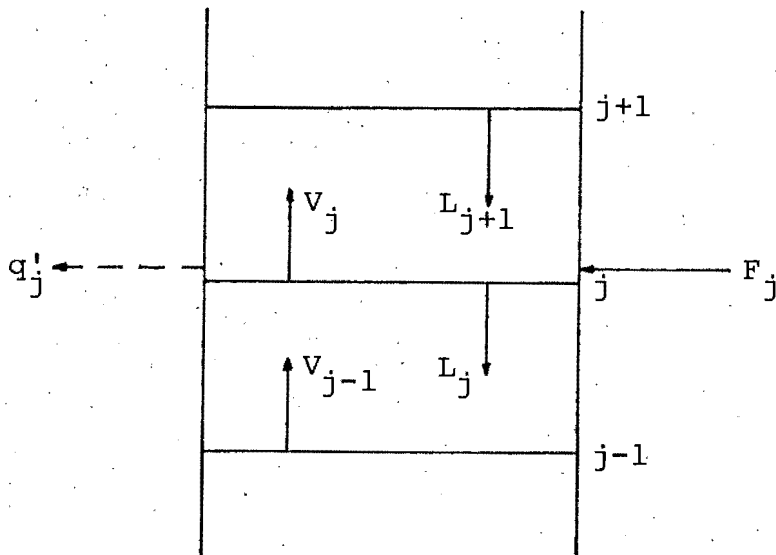


Fig. 1.1

Generalised equilibrium stage

Friday and Smith [14] note that there are many ways of combining and solving these equations. This has resulted in the development of a large number of solution procedures for multicomponent systems. For binary systems the solution of the above equations can be conveniently represented graphically in two dimensions.

1.23 Solution Procedures

The calculations required to solve any distillation problem can be divided into three parts.

1. Prediction of phase equilibria.
2. (a) Calculation of the number of equilibrium stages required for a given product distribution, OR  
(b) Calculation of the separation achieved in a column containing a specified number of equilibrium stages.
3. Conversion of the number of equilibrium stages to real stages.



The second part of the solution procedure is the simplest and is the only aspect discussed in any detail in the undergraduate course. The first and third parts are extremely complex, usually involving empirical or simplified mathematical models. It is in these fields that further research is required [5].

#### A. Binary distillation in Tray towers.

All distillation problems considered in the course involve ideal mixtures that obey the equilibrium expressions shown previously in Equation (1.1). The equilibrium constant  $K_i$  for binary systems is derived from Raoult's Law and Antoine's equations (see Chapter 2). The graphical solution procedures of McCabe and Thiele [7] and Ponchon and Savarit [8] are considered in this part of the course.

Batch distillation is introduced first and standard McCabe-Thiele assumptions are applied to all systems considered. The problems are thus solved manually by constructing McCabe-Thiele diagrams, with the necessary integration performed graphically as well. These procedures are well documented in a number of texts [9,10,11] and a review of the literature did not reveal any new solution procedures. A recent paper by Koppel [15] describes a computerised McCabe-Thiele solution procedure, but does not suggest any significant changes in method. Batch distillation systems operating under both constant and variable reflux ratio are considered. Chapter 2 describes the program devised for binary batch distillation. The approach is similar to that of Koppel except that a Simpson's Rule numerical integration technique is used, instead of a graphical one. Stage-efficiencies are not considered in batch distillation.

Continuous binary distillation is considered next in the course. McCabe-Thiele systems are discussed first, with the McCabe-Thiele diagram used to demonstrate the overall effect of feed condition, reflux ratio, and other variables.

Enthalpy effects are then considered and the Ponchon-Savarit diagram is described.

Very little new work has been done in binary distillation. Strangio and Treybal [16] have developed a series of exact relationships, in algebraic and graphical form, that may be used for "short-cut" design of both binary and multi-component systems. The applications of these methods are not relevant to the course, however. The programs published by the CACHE committee ignore enthalpy effects and columns with side streams or more than one feed. They were thus unsuitable for this section of the course.

Stage-efficiency is discussed in the course in terms of Murphree tray efficiencies. The effect of efficiency is considered in the program, and an efficiency factor may be specified. The program contains both the McCabe-Thiele and the Ponchon-Savarit procedures and is discussed in Chapter 3. The output from either the batch or the continuous distillation programs may be used directly by the plotting program to construct McCabe-Thiele or Ponchon-Savarit diagrams. The plotting program is discussed in Chapter 6. Reference to plotting programs of this type could not be found in the literature.

#### B. Binary distillation in Packed towers.

Packed columns for distillation operations are usually designed manually, using empirical Height of Transfer Unit (HTU) or Height-Equivalent-to-Theoretical Plate (HETP) relationships. Empirical data are available for many different types of packing and the various design equations have been well tested over the years.

The concepts of HTU assumes that the vapour and liquid flowrates within each section of a column are constant. In a recent paper Eckert [17] states that vapour-liquid equilibrium is the only factor significantly affecting a binary distillation process, and thus justifies continued use of HTU relation-

ship to design packed columns. The more rigorous procedure, which includes enthalpy considerations and hence does not assume constant molal overflow conditions, requires the calculation of mass transfer coefficients at intervals through the column. The height of packing required is then determined by integrating the following expression between appropriate limits.

$$Z = \int_{V_{Y1}}^{V_{Y2}} \frac{d(Vy)}{k_y a (y_1 - y)} = \int_{L_{X1}}^{L_{X2}} \frac{d(Lx)}{k_x a (x - x_1)} \quad (1.5)$$

In manual calculations the integration is performed graphically and the accuracy depends on the integration interval considered. The procedure is tedious and extremely time-consuming with the result that any improvement in accuracy over an HTU method is not justified, especially as it has become standard practice to add extra packing length, in an almost arbitrary fashion, as a safety factor.

In terms of a computer teaching-package, however, the rigorous design procedure is a useful means of demonstrating both the concepts of mass transfer, and the possible variations in design obtained using the different techniques. Students can design a packed column manually using an HTU relationship, and then compare their results with a computer solution.

The effect of packing type and size can be demonstrated more easily using a computer routine. The use of mass transfer coefficients also enables the effect of physical properties of a system to be illustrated and shows how mass transfer theory is applied to process operations.

CACHE have not published a program dealing with packed column design for distillation purposes. Their program for

packed absorption tower design uses a mass transfer correlation procedure similar to that used here (see Chapter 8).

The program developed for packed tower design is discussed in Chapter 4. A Ponchon-Savarit operating diagram is used by the program whereas manual calculation procedures using an HETP method require a McCabe-Thiele diagram.

### C. Multicomponent Distillation

Continuous distillation of multicomponent solutions is the most common distillation process encountered in industry. Design methods are more complex because the process cannot be represented graphically. There are a number of solution procedures available of which one was chosen to meet the requirements of this study. Friday [14] was the first to analyse the possible combination of the four equations stated previously, and to characterise the corresponding solution procedure. He proposed a series of six "decisions" that would lead to the correct choice for a given problem. Seppala [18] applied sixteen different solution procedures to nine problems and showed that no single technique was superior in all cases. This supported Friday's contention that the nature of the problem should determine the choice of solution procedure.

The theory and application of the trial and error methods of Lewis [1] and Thiele [2] are discussed in the course. Due to the time factor, however, short-cut procedures, such as that of van Winkle and Todd [19], are usually used for problem solving. The short-cut method of van Winkle and Todd was shown by Guerreri [20] to produce results that agree well with those calculated by the Thiele-Geddes procedure. An extensive literature survey was conducted to determine the most suitable solution procedure for the multicomponent program. The survey is summarised in section 1.24.

The combination of the Thiele-Geddes calculation procedure and Holland's [21] ' $\theta$ ' convergence technique was finally chosen for the following reasons:

- (1) Results agree well with short-cut method of van Winkle and Todd which will facilitate comparison of short-cut and computer solutions.
- (2) The Thiele-Geddes- $\theta$  combination is computationally fast and stable.
- (3) It is a tray-to-tray calculation procedure that does not require an understanding of matrix algebra in order to follow the programming procedure.
- (4) The Thiele-Geddes method is discussed in the course and is recognised as being more stable than the Lewis-Matheson procedure.

Ideal equilibrium relationships are assumed to hold for all problems considered because the scope of the course does not include non-ideal solution thermodynamics. Stage-efficiency is considered in terms of component vaporization efficiency. This procedure was first modified for computer application by Holland [21]. There is no realistic method for determining component efficiency factors analytically, but fairly representative results have been obtained experimentally [22]. Miskin [23] demonstrated that vaporization efficiency factors were more consistent than the more widely used Murphree efficiency factors. Vaporization efficiency is useful for demonstrating the effect of inefficiency on a system even if true data are unavailable. As yet there is no satisfactory way of either measuring or calculating reliable efficiency data [22].

The multicomponent distillation program is discussed in Chapter 5.

### 1.24 Survey of Multicomponent design procedures

Computerised calculation procedures for the rigorous solution of distillation problems may be divided into four main categories as follows:

Class 1: Programs based on solving the four equations listed in section 1.22 simultaneously for the whole column using matrix algebra.

Class 2: Programs that consider the dynamic problem of the column settling down to steady-state using a relaxation method.

Class 3: Programs based on the Lewis-Matheson [1] procedure.

Class 4: Programs based on the Thiele-Geddes [2] procedure.

Class 3 and 4 programs are similar in that they are both tray-to-tray methods. They involve different initial assumptions, however, and traditional class 3 programs are design programs whereas class 4 are used for rating columns containing a fixed number of trays. This, however, is not an essential feature of either type.

Class 1 programs involve the simultaneous solution of a set of equations describing a distillation system. The equations are set up in a matrix of order  $n$ , where  $n$  is the number of trays in the column including the condenser and reboiler. The matrix is reduced to a tridiagonal form that can be solved using either matrix-inversion or Gauss-elimination techniques. Class 1, 3 and 4 programs all consider columns operating at steady-state conditions.

Class 2 programs involve an entirely different approach, as they consider the dynamic problem of column's achieving steady-state conditions from start-up. A relaxation method is used to achieve a solution.

Each class of program will now be discussed in detail. They are listed in order of publication of the original papers. The reasons for selecting the procedure used in this study are then summarised.

(a) Class 1 Programs

Amundson and Pontinen [4] were the first to investigate the use of large digital computers for distillation calculations. They observed that the basic equations describing an equilibrium stage process, listed in section 1.22, could be grouped as a set of simultaneous algebraic equations in composition, that describe a complete distillation column.

Their proposed calculation procedure involves the simultaneous solution of this set of equations, component by component, for a fixed temperature profile. Successive iterations correct the temperature distribution until a convergence is achieved. The procedure is strictly valid only for ideal systems. In such systems, the equilibrium constant is not a function of composition and thus the fixing of the temperature profile is sufficient to ensure that the overall material balance is not violated during the calculation. Most low molecular weight hydrocarbon systems may be considered ideal.

Amundson and Pontinen employed a matrix inversion procedure, and noted that this procedure is practical only on a very fast computer with a large memory capacity. The method proposed by these authors does not include any forcing or convergence procedures. A bubble-point routine, based on a Newton approximation procedure and normalized stage concentrations, is used to calculate new stage temperatures. The authors state that "the number of iterations depends on the precision required, but generally four or five are adequate". Sargent [24] has found, however, that convergence is often very slow using this procedure, and that sometimes no convergence is achieved at all.

In 1964 Friday and Smith [14] published a paper in which they analysed the reasons for this diverse convergence behaviour. They proposed a set of six decisions, based on physical and numerical criteria, to assist designers in selecting an appropriate procedure for a specific problem. Their work is particularly useful in that it summarizes the work done in the field of computerised solutions to distillation problems, up to that time. The main points to emerge from their study of the situation can be summarised as follows:

(1) No completely general solution procedure had been developed up to that time. They suggested that, owing to the importance of the physical characteristic of different processes, it was unlikely that such a procedure was feasible. A more recent survey has supported this suggestion [18].

(2) The appropriate matching of variables to equations is an important feature of a design procedure.

(3) Procedures involving simultaneous solution of the system equations are superior to stage-to-stage methods in that they can deal with complex columns easily, and are not susceptible to truncation errors.

(4) Stage-to-stage methods are, in general, faster than matrix-type methods and do not require large amounts of computer storage.

The next significant contribution to simultaneous matrix-type solution methods was made by Wang and Henke [25]. They note the work of Friday and Smith, and conclude that the practical advantages of matrix methods outweighed the disadvantages. They criticized previous algorithms on the following points:

(1) They were computationally slow and hence expensive to run.



(2) They required large amounts of core storage which meant they could be run only on big machines.

(3) The use of Newton-Raphson convergence procedures by the algorithms did not guarantee a converged solution. Their work attempts to make matrix-type procedures more attractive.

Wang and Henke adopted the basic method of Amundson and Pontinen, and proposed two alterations in the calculation procedure. These are the use of Gauss-elimination, rather than matrix inversion, to solve the tridiagonal matrix of equations, and the incorporation of Muller's convergence method. Muller's method is described as a generalization of Newton's method of secants, commonly referred to as "reguli falsi", and it is shown to achieve convergence for situations where the Newton-Raphson method diverges.

The Gauss-elimination algorithm has several advantages over the matrix inversion procedure. For large matrices, inversion requires large amounts of both computer storage and time. In addition, it is susceptible to round-off errors, although these are not as significant as in stage-to-stage techniques. The Wang and Henke algorithm is shown to be faster than stage-to-stage methods for complex columns, as well as being inherently more stable for such applications [24]. This algorithm is used in a program published by the CACHE committee. (See Chapter 8.)

Sargent and Murtagh [24] were the first to consider the problem of real stages in connection with a matrix-type solution procedure. They note that Holland [21] had proposed an extension of his stage-to-stage procedure to include efficiency considerations, but dismiss the technique as not being sufficiently rigorous.

Sargent describes a general purpose calculation procedure based on Amundson's original method. The main extension proposed to Amundson's method is a technique for

considering stage performance in terms both of Murphree tray efficiencies and of vapour- and liquid-phase thermal efficiencies. A specially developed relaxation technique is used to achieve convergence, and this is compared to the Newton-Raphson method. The relative stability of the relaxation convergence technique is not discussed, and no mention is made of the Wang-Henke algorithm. A comparison of the computer-time required by the methods of Amundson, Wang, and Sargent indicates that the Wang-Henke procedure is superior in this respect.

A combination of the Amundson matrix method and the stage-to-stage method of Greenstadt et al. [26] was proposed by Naphtali et al. [27]. The algorithm employs a Newton-Raphson convergence technique and can be described briefly as grouping the basic equations by stages, linearizing the equations, and solving simultaneously in tri-diagonal matrix form using Gauss elimination. The method is shown to solve problems in which other techniques have failed, although no mention is made of the computer-time required. The grouping of the equations by stages is not recommended by Friday's analysis, as it leads to a build-up of truncation errors, and requires the initial assumption of end compositions.

Greenstadt was the first and only worker to use this grouping method before Naphtali adopted it. The fact that Naphtali's algorithm includes this procedure and the potentially unstable Newton-Raphson convergence technique, to produce an algorithm that will solve both conventional and complex columns, adds support to the theory that specific techniques should be developed for specific problems.

The chief disadvantages quoted by Naphtali are the large core-storage requirements and the need to estimate end compositions. If this estimate is not fairly accurate the method tends to become unstable.

The most recent article surveyed dealing with matrix method is one by Ishii and Otto [28]. They propose a multivariate Newton procedure in which all equations are linearized and solved simultaneously. It is similar to Naphtali's procedure in that it can handle non-ideal solutions. The equations are grouped by type, and the computer storage requirement is reduced by ordering the linearized equations so as to minimise the number of partial derivatives to be evaluated in each iteration.

In conclusion, the following general statements can be made in connection with matrix methods.

- (1) They are numerically stable and their accuracy is not significantly affected by round-off errors.
- (2) They work equally well for conventional and complex columns.
- (3) They require relatively large computer-storage facilities.
- (4) The rate and stability of convergence depend largely on the accuracy of initial assumptions and the matching of variables with particular equations.

(b) Class 2 Programs

Programs in this category represent a completely different approach to the solution of distillation problems. A relaxation method is used to consider the dynamic problem of a system settling down to steady-state from an arbitrary initial state, for fixed operating conditions. The method seems attractive because it is equivalent to collecting results from an actual pilot-plant operation, and one might expect the physical stability of such a column to improve the computational convergence characteristics. It has, however, been found that the difficulties encountered in integrating the relevant differential equations are even

greater than those associated with steady-state solution procedures [23].

Rose et al. [29] were the first to publish work in this field. They used a particularly poor numerical procedure, and the rate of convergence was extremely slow. Subsequent investigators used a very stable algorithm and they reported excellent convergence characteristics. The example chosen was later shown to be a particularly favourable one, and it was demonstrated that even with a stable integration technique, convergence was often slow [30]. As a result of these findings the method has been generally ignored by research workers.

Petryschuk and Johnson [31] published the most recent paper on the relaxation method. They compared Rose's technique with Holland's stage-to-stage method, and concluded that the relaxation method was favourable for use in the development of process models and in particular for handling multi-column systems interconnected by recycle streams.

In general, it must be concluded that the dynamic approach inherently requires more computer time, and thus cannot compete with the steady-state methods in solving standard problems.

#### (c) Class 3 and 4 Programs

Although apparently very different, the Lewis-Matheson [1] and Thiele-Geddes [2] calculation procedures, on which class 3 and 4 programs are respectively based, have much in common. A traditional, though not essential, feature is that the number of trays in the Thiele-Geddes application is kept fixed whereas in the Lewis-Matheson application the number is varied. Thus class 3 programs are usually design-orientated and class 4 are used as rating programs.

(i) The Lewis-Matheson method: Lewis and Matheson [1] were the first to advance a method for solving multi-component systems. Their procedure requires the specification of feed and reflux conditions, the number of plates in each section of the column and an initial estimate of the product distribution. A stage-to-stage calculation procedure is then used to calculate from both ends of the column towards the feed tray. The resulting mis-match at the feed tray is used to re-estimate the product distribution with the procedure being repeated until a match is obtained.

It was not until the appearance of computers that the matching procedure was systematized. Greenstadt et al. [26] were the first to publish a computer technique using the Lewis-Matheson solution method.

They used a Newton-Raphson procedure for the simultaneous solution of the basic equations, which were grouped by stage as mentioned previously. The authors claimed that their method worked both for conventional and for complex columns, but the method was found to be potentially unstable [14]. In addition to Friday's analysis, it was found that unless the initial estimate of the product compositions was reasonable the Newton-Raphson convergence procedure became unstable. This method was not considered again until Naphtali incorporated it in his matrix procedure.

Shortly after Greenstadt's paper, another computerized solution procedure, based on the Lewis-Matheson method, was proposed by Bonner [32]. This algorithm determined the number of trays required for a specified product distribution and reflux ratio. Bonner's work lead to the traditional design orientation of the Lewis-Matheson method. The number of stages in the rectifying and stripping section is adjusted alternately, and thus the feed plate position is finally chosen to minimise the total number of trays required in the column.

The Lewis-Matheson method has, however, not been generally accepted owing to convergence failure when applied to many standard distillation problems.

(ii) The Thiele-Geddes method: Thiele and Geddes proposed a stage-to-stage calculation procedure for multi-component systems that is similar in many ways to the Lewis-Matheson method. An important difference is that the Thiele-Geddes procedure replaces the initial assumption of product distribution, required in the Lewis-Matheson method, with an estimate of the equilibrium ratios for all stages in the column. For systems involving ideal solutions this is equivalent to estimating a temperature profile through the column. The initial estimation of tray conditions required in this method is probably the reason for its traditional role as a rating procedure [24]. It is this factor, as well, that makes the Thiele-Geddes method more stable than that of Lewis and Matheson.

Although the constant equilibrium-ratio profile enhances the stability of the Thiele-Geddes method it makes the procedure susceptible to truncation errors. These errors accumulate in the sections of column between feed and side streams, making the stability of the method unreliable for complex columns [24].

Thiele and Geddes did not indicate how to re-estimate the temperature profile between iterations, and stated that "correction requires much judgment, but one or two repetitions of the calculation will generally give a fairly correct solution to one familiar with the process" [2,24]. However, such a trial-and-error procedure is, in general, very time-consuming. The two most commonly used convergence techniques are Newton's method, and the procedure known as Holland's ' $\theta$ ' method.

The  $\theta$  (theta) method was first proposed by Lyster et al. [33], and was successfully applied both to the Thiele-

Geddes and to the Lewis-Matheson procedures. Holland [21] and Petryschuk [31] further developed its application with the Thiele-Geddes solution method.

Holland [21] attempted to extend the Thiele-Geddes- $\theta$  procedure to include absorbers and strippers, with limited success. The method was found to be unstable for many absorption problems.

Billingsley [34] was the first to place the  $\theta$  method on a sound mathematical basis. Seppala and Luus [18] have published a comprehensive comparison of the Thiele-Geddes- $\theta$  method and most of the other existing procedures, as mentioned previously.

A recent paper by Holland [35] proposed a modification to the basic  $\theta$  method that places the technique on a better theoretical basis by using it with a tridiagonal matrix procedure. The new algorithm is shown to be stable for absorber, stripper and reboiled absorber application. This new method is termed the "Multi- $\theta$ " procedure. The use of a simultaneous solution procedure rather than the original stage-to-stage method, enables complex columns to be considered.

The application of Newton's method to the Thiele-Geddes calculation procedure was investigated by several workers, including Newman [36], Boynton [37] and Goldstein [38]. Although these methods can solve both distillation and absorber problems more easily than the standard  $\theta$  method, they have been shown to be considerably slower than the simpler Thiele-Geddes- $\theta$  combination [18].

In view of the requirements of the undergraduate distillation course the original combination of the Thiele-Geddes stage-to-stage solution procedure and the Holland- $\theta$  convergence technique, proposed by Lyster et al. [33], has been chosen. The reasons for this choice can be summarized as follows:

Summary of the Thiele-Geddes- $\theta$  method

(1) The calculation procedure is a stage-to-stage type and can thus be more easily explained to undergraduates. It corresponds directly with the material presented in the course, and requires no knowledge of matrix algebra, or other complex mathematical procedures, in order to follow the computational procedure.

(2) The Thiele-Geddes method, combined with the  $\theta$  convergence technique, is recognized as one of the fastest computational methods [35]. The Lewis-Matheson procedure, although found to be faster for simple problems, was not considered owing to its unfavourable convergence characteristics.

(3) Non-Ideal mixtures are not considered in the distillation course and are thus ignored here. In addition the data-bank required by a rigorous procedure, such as the Chao-Seader correlation, is not available on the computer system at U.C.T. The Thiele-Geddes- $\theta$  procedure is adequate for handling thermodynamically ideal systems.

(4) Matrix procedures, in general, require both more computer storage and more time for execution than do stage-to-stage methods [18,24,25].

(5) As the usual undergraduate tutorial problem seldom involves systems containing more than 5 components and 20 stages, the type of limitation of the Thiele-Geddes method, suggested by Petryschuk [31], is not significant. Modifications to the basic procedure are kept to a minimum, consistent with stability requirements, so that the algorithm can be easily understood at the undergraduate level.



### 1.25 Recent research trends

In the past, a primary objective of research into process design methods was to reduce capital costs. Since 1973, however, the increased price of fuel has brought the question of operating costs to the fore. In distillation, operating costs can be considerably reduced by running a column at the lower reflux ratios. This requires a very accurate design because product quality is more sensitive to variations in operating conditions at low reflux ratios. Representative system-data are essential to accurate design, with vapour-liquid equilibria and stage-efficiency being two important aspects. Much of the current research on distillation is in these two fields.

In connection with vapour-liquid equilibria, Ellis [5] notes that the limiting factor in many practical designs is the shortage of some binary parameters, necessary for prediction of multicomponent phase equilibria. The most recent treatise in this field, with specific reference to distillation, is Rowlinson's application of the "principle of corresponding states" [39]. Although such advances are of practical value, they are beyond the scope of an undergraduate course. For the purposes of this study Equation (1.1) provides an adequate representation of vapour-liquid equilibria.

Over the years a number of workers have proposed methods for including stage-efficiency in distillation calculations. Most of these use a modification of either Murphree's expression for tray efficiency [40] or Holland's modification of vaporization efficiency. According to Holland [41] vaporization efficiency was first defined by McAdams [42] and Carey [43].

In his first treatment of the subject, Holland [21], observed that vaporization efficiencies were computationally more convenient than the Murphree efficiency factors. Recently Holland and McMahon [41] showed that the Murphree expression

has limitations in that apparently reasonable values of the Murphree tray efficiencies yield negative flowrates within the column. Experimental work by Miskin et al. [23] showed that Murphree tray efficiencies can be negative whereas vaporization efficiencies are always positive. Negative efficiency factors do not necessarily invalidate the Murphree analysis because they are usually associated with large positive values on adjacent trays [5].

Recently a number of models have been proposed that attempt to predict the behaviour of real trays, based on the physical parameters of a system. The main contributors have been Yamada [44,45,46], Holland [44,46], Onda [47] and MacFarland et al. [48]. Asbjørnsen [49] has proposed a dynamic model in which vaporization efficiency is re-defined as "equilibrium efficiency". An inefficient stage is considered to be in a state of pseudo-equilibrium with the efficiency factor describing the state of equilibrium. This work has been ignored, probably owing to the inherent difficulties in dynamic solution procedures.

A survey of the available models has been carried out by Standart [50], who is developing yet another model\*. He claims that previous models are unrealistic in that they either do not consider the effects of entrainment or attempt to derive overall efficiencies from point efficiency data, rather than from direct mass-balance considerations. A recent paper by Biddulph [51] describes the simulation of an air-distillation column. He uses experimental point efficiency data with an eddy-diffusion model to achieve a good comparison with actual plant experiences. This model assumes, however, that the point efficiency factors at any point in the column are equal for all components [22]. This is true of the argon-nitrogen-oxygen system. For systems not obeying this condition, the direct approach of

\*Standart's model still in preparation at time of writing [50].

Standart would be more realistic. At present there is no general method for describing stage efficiency satisfactorily [22].

In this study, vaporization efficiencies are used in the manner prescribed by Holland [21]. It is assumed that experimental data are available to the user in the form described by Miskin [23]. This relatively simple procedure is easily explained at an undergraduate level and provides a fairly realistic description of the behaviour of a real column [22].

#### 1.26 Summary

The literature survey revealed that although a considerable amount of work has been done in the field of multicomponent distillation, the original graphical procedures for binary distillation have not been significantly improved upon.

The programs dealing with binary systems will be developed using the standard graphical solution procedures. The theoretical background will be extracted from textbooks currently prescribed for the course. This will ensure a good correspondence between the course work and the computer package.

The multicomponent program will be based on the Thiele-Geddes calculation procedure, incorporating Holland's  $\theta$  convergence technique. Only ideal systems will be considered both in the binary and the multicomponent programs. Stage efficiency will be considered in terms of Murphree tray efficiency for binary systems, and vaporization efficiency for multicomponent systems.

Four separate mainline programs will be developed, dealing respectively with:

- (a) Binary batch distillation in tray towers - Chapter 2.
- (b) Binary continuous distillation in tray towers - Chapter 3.
- (c) Binary continuous distillation in packed towers - Chapter 4.
- (d) Multicomponent continuous distillation in tray towers - Chapter 5.

Subroutines will be common to all mainline programs requiring the same calculation, thus reducing the overall number of subroutines. The programs will be designed primarily for operation in demand mode with data requests and error messages inserted to guide users.

A program for graphical representation of results will be developed (Chapter 6). This program will incorporate standard "CALCOMP" plotting software and will require a CALCOMP plotter for constructing the various diagrams.

The whole package is designed to supplement the undergraduate course on distillation at U.C.T.

## CHAPTER 2

### BINARY BATCH DISTILLATION

#### 2.1 INTRODUCTION

Batch distillation is usually employed in industry to treat relatively small quantities of solutions for which the cost of a continuous operation is not justified. It is an unsteady state process, often carried out in a column that is used to distil a variety of solutions. Thus batch distillation problems considered in the undergraduate course entail the rating of a known column for a given feed stock.

A typical batch still with its rectifying column is illustrated schematically in Figure 2.1. A representative McCabe-Thiele diagram is shown in Figure 2.2. The column acts entirely as an enriching section and is therefore described by a single operating line on the x-y diagram. There are two basic modes of operation for batch distillation columns. In one the reflux ratio is increased as the distillation proceeds, thus maintaining a constant distillate product composition. In the other the reflux ratio is held constant and the final distillate composition is determined as the average for the run. The variation in reflux ratio necessary to achieve a constant distillate composition is non-linear and hence sophisticated control equipment is required. Industrial operations are more often carried out under constant reflux.

The theoretical discussion of each mode of operation, given below, assumes McCabe-Thiele conditions of constant molal overflow, and neglects the effects of liquid hold-up within the column. These assumptions are in keeping with

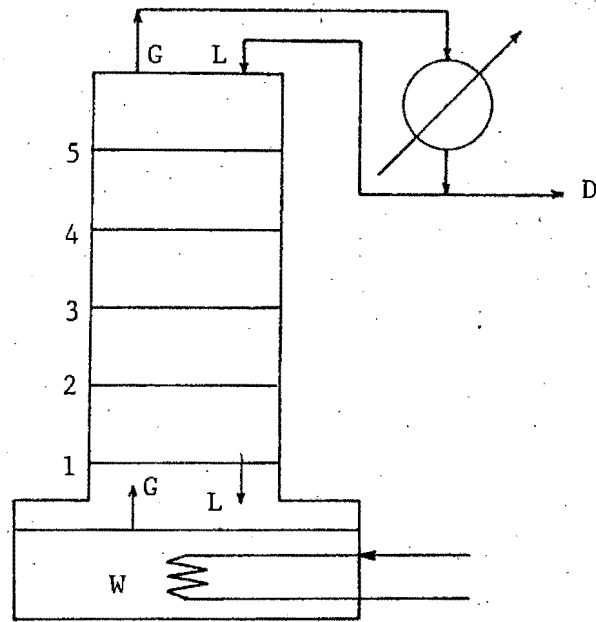


FIGURE 2.1 BATCH DISTILLATION COLUMN

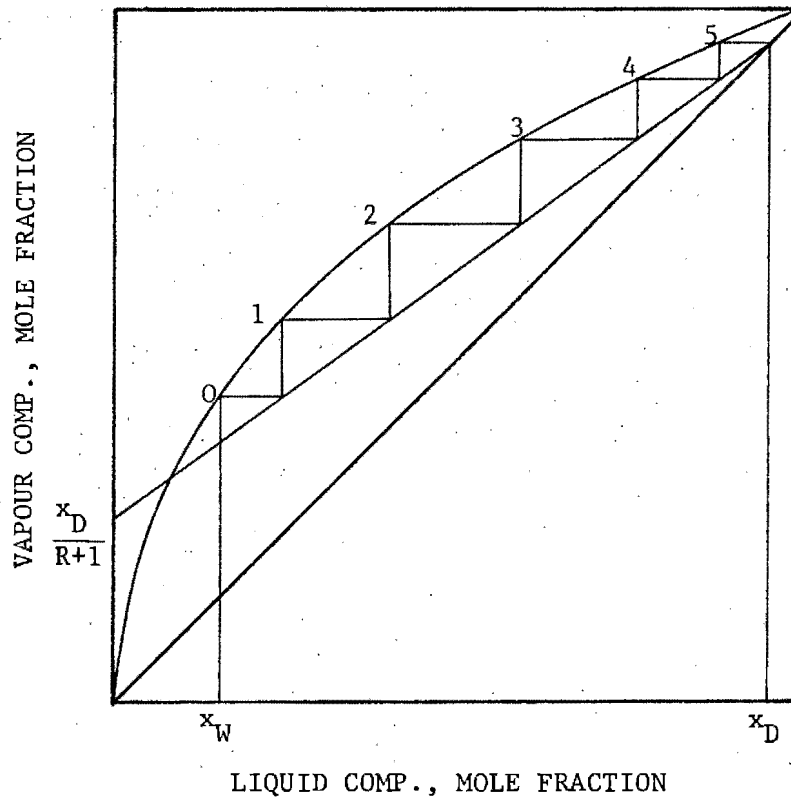


FIGURE 2.2 MCCABE-THIELE DIAGRAM OF BATCH DISTILLATION SYSTEM

the material presented in lectures. In practice this treatment is applicable to systems in which the amount of any component in the batch is at least an order of magnitude larger than the hold-up [52].

## 2.2 THEORY OF BINARY BATCH DISTILLATION

### 2.21 Vapour-liquid equilibria

These remarks apply to all the binary programs in the package. The general equilibrium relationship is described by Equation (1.1). For binary systems

$$y_i = K_i x_i \quad i = 1, 2 \quad (2.1)$$

In binary systems the vapour-liquid equilibrium can be described by a curve on an x,y diagram as shown in Figure 2.2. The greater the distance between the curve and the diagonal, the easier is separation by distillation. A useful measure of this distance is the relative volatility,  $\alpha$ , defined as

$$\alpha_{1,2} = \frac{y_1}{y_2} \cdot \frac{x_2}{x_1} \quad (2.2)$$

where  $y_2 = 1 - y_1$ , and  $x_2 = 1 - x_1$ .

Ordinarily the value of  $\alpha$  will change with composition, but the simplifying assumption of constant relative volatility is sometimes made.

Combining Equations (2.1) and (2.2)

$$\alpha_{1,2} = \frac{K_1}{K_2} \quad (2.3)$$

Ideal solution behaviour is assumed throughout the course, and for binary systems Raoult's Law can be expressed, for fixed temperature, as

$$p_i = P_i x_i \quad i=1,2 \quad (2.4)$$

Also, Dalton's law for ideal vapours is:

$$P_T = p_1 + p_2 \quad (2.5)$$

The equilibrium vapour composition, at the fixed temperature, is thus

$$y_i = \frac{p_i}{P_T} = \frac{P_i x_i}{P_T} \quad i=1,2 \quad (2.6)$$

Hence

$$K_i = \frac{P_i}{P_T} \quad i=1,2 \quad (2.7)$$

and

$$\alpha_{1,2} = \frac{P_1}{P_2} \quad (2.8)$$

The vapour pressure of a pure component is expressed as a function of temperature by Antoine's equation

$$\log_{10} P_i = A_i - \frac{B_i}{C_i + T} \quad i=1,2 \quad (2.9)$$

The user has a choice of three equilibrium data options in the binary programs:

1. The user may specify Antoine constants for each component. The program calculates a set of x,y equilibrium data according to the following scheme: For values of x between 0 and 1,0 in steps of 0,02, the bubble point of liquid mixture  $x_1, x_2$  is computed (see discussion of subroutine BINBPT). The vapour pressure of the light component at the bubble point is then determined using Equation (2.9). The equilibrium vapour composition is then calculated using



Equation (2.6) and the heavy component value determined by difference ( $y_2 = 1 - y_1$ ). The relative volatility at each point is computed through Equation (2.2). A set of  $x, y, \alpha$  data is then stored for use in the solution procedure.

2. A constant relative volatility may be specified. This is acceptable only for McCabe-Thiele systems as it implies a constant temperature through the column. If this option is used,  $x$  and  $y$  data are computed as required in the program, using Equation (2.2).

3. A set of  $x, y$  data, extracted from the literature, or determined experimentally, may be supplied directly. The number of data points must be specified, and to ensure reasonable accuracy at least twenty data pairs should be provided. End points 0,0 and 1,1 are assumed by the program. The relative volatility at each point is computed according to Equation (2.2). If a Ponchon-Savarit system is considered equilibrium temperature data must also be provided under this option. This is discussed in detail in Chapter 3.

#### 2.22 Varying reflux ratio operation

Assuming McCabe-Thiele conditions and negligible hold-up the following equations describe a binary distillation system:

Mass balance

$$\text{Overall} \quad F = D + W \quad (2.10)$$

$$\text{Component} \quad Fx_F = Dx_D + Wx_W \quad (2.11)$$

Operating line as shown in Figure 2.2

$$y_n = \frac{R}{R+1} x_{n+1} + \frac{x_D}{R+1} \quad (2.12)$$

where  $R/R+1$  is the slope of the line and  $R$  is the reflux ratio defined as

$$R = \frac{L}{D} \quad (2.13)$$

Equation (2.12) represents the material balance across tray  $n$  in the column. For McCabe-Thiele assumptions  $L$  is constant through the column and the operating line is straight.

The following data are usually known when rating a column for a constant distillate composition under varying reflux conditions.

- (a) Feed composition  $x_F$
- (b) Required distillate product composition  $x_D$
- (c) Number of trays in the column  $N$
- (d) Final residue concentration  $x_{Wf}$

Assuming perfect trays and referring to Figure 2.2, the first step is to choose a reflux ratio such that the number of steps constructed from  $x_F$  to  $x_D$  equals the number of equilibrium stages in the system.

Note: The still-pot (a partial reboiler) and a partial condenser, if specified, both represent an equilibrium stage. Thus a system with 10 trays and a partial condenser will be described by 12 steps on the McCabe-Thiele diagram.

As product is withdrawn and  $x_F$  tends to  $x_{Wf}$ , the reflux ratio must be increased to maintain a constant  $x_D$ . The time required for a batch distillation run can be expressed as follows:

$$\theta_T = F \frac{(x_D - x_F)}{G} \int_{x_{Wf}}^{x_F} \frac{dx_W}{(1 - \frac{L}{G})(x_D - x_W)^2} \quad (2.14)$$

This equation was developed by Bogart [53] for systems with negligible hold-up. As the vapour flowrate  $G$  is not necessarily known, it is often more convenient to rewrite Equation (2.14) in terms of the total vapour boiled up during a run,  $V$ ;

$$\text{i.e.} \quad V = \theta_T G \quad (2.15)$$

$$\text{also} \quad G = L + D \quad (2.16)$$

$$\text{hence} \quad V = F(x_D - x_F) \int_{x_{Wf}}^{x_F} \frac{dx_W}{\left(1 - \frac{R}{R+1}\right)(x_D - x_W)^2} \quad (2.17)$$

G is related to the reboiler heat duty and is often limited by the heating device or the column size. V, however, may be determined by graphical or numerical integration and then used to compute either  $\theta_T$  or G, whichever is unknown.

When solving problems manually, students are required to integrate Equation (2.17) graphically. The program uses a numerical procedure based on Simpson's Rule.

### 2.23 Constant reflux ratio operation

Varying the reflux ratio in order to maintain a constant distillate product requires a fairly sophisticated control system. It is possible to achieve the same result by holding the reflux ratio constant and collecting a distillate product of varying composition until the average composition reaches a specified value. This requires a higher initial purity than ultimately required, which is not always feasible.

The reflux ratio is fixed throughout the run and the operating lines are parallel to each other on the x-y diagram. A mass balance at anytime can be written as follows:

$$Wx_W = x_D dW + (W-dW)(x_W - dx_W) \quad (2.18)$$

Neglecting second order differentials, Smoker and Rose [54] have shown that for negligible hold-up the quantities of feed and residue may be related by the expression

$$\ln \frac{F}{W} = \int_W^F \frac{dW}{W} = \int_{x_{Wf}}^{x_F} \frac{dx_W}{x_D - x_W} \quad (2.19)$$

The total moles of vapour boiled up, for constant reflux operation is given by

$$V = D(R+1) \quad (2.20)$$

Equation (2.15) can be applied as before to determine either  $\theta_T$  or  $G$ . Equation (2.19) is solved by graphical or numerical integration.

The stepping procedure is performed within the program by alternate solution of the equilibrium and material balance equations. These are expressed in the following form:

Equilibrium (from Equation (2.2))

$$y = \alpha x / [1 + x(\alpha - 1)]$$

Material balance (Equation 2.12)

$$y_n = \frac{R}{R+1} x_{n+1} + \frac{x_D}{R+1}$$

The development of the computer program for batch distillation is now discussed. The quoted program text line numbers refer to the printout of the program attached in the Appendices. The variable names extracted from the program are shown in the text in capitals and are defined in the nomenclature (Appendix B).

## 2.3 DESCRIPTION OF PROGRAM BATCH

### 2.31 General structure of the program

BATCH is the mainline routine for the solution of batch distillation problems, considering either variable or constant reflux conditions. It shares a common block with only one other subroutine, XDSTEP, although it uses several other routines available to the binary system programs. This

program is different from others in the package in that the mainline is used to read in all data and to output all the results, in addition to performing all the general computation required. Figure 2.3 is an overall flow diagram of the program. Sections requiring further explanation are discussed under the headings of particular sub-routines.

### 2.32 Limitations of the program

BATCH is a rating program in that the number of theoretical stages in the rectifying column is specified by the user.

BATCH is limited to binary systems for which the normal McCabe-Thiele assumptions hold, as discussed in section 2.1. The program is set up to accommodate a maximum of 25 theoretical stages plus a partial reboiler and partial condenser, or 26 stages if a total condenser is specified. A maximum of 55 x,y equilibrium-data points may be specified. The number of data points that may be computed for use in the integration procedure is at present limited to 150. These limits can be adjusted by changing the dimensions of the relevant arrays.

A check on the feasibility of the final reflux ratio must be done by the user. If this is too high the final residue composition should be increased and the problem solved again.

### 2.33 Operating procedure and data input specifications

BATCH is designed to run on a conversational basis in demand mode. For sample runstreams for solving variable reflux and constant reflux problems refer to Chapter 7. The user may suppress the individual data requests and add a previously prepared data element to save input-output time on the terminal. The data element option should be used for batch mode operation. The conversational responses of

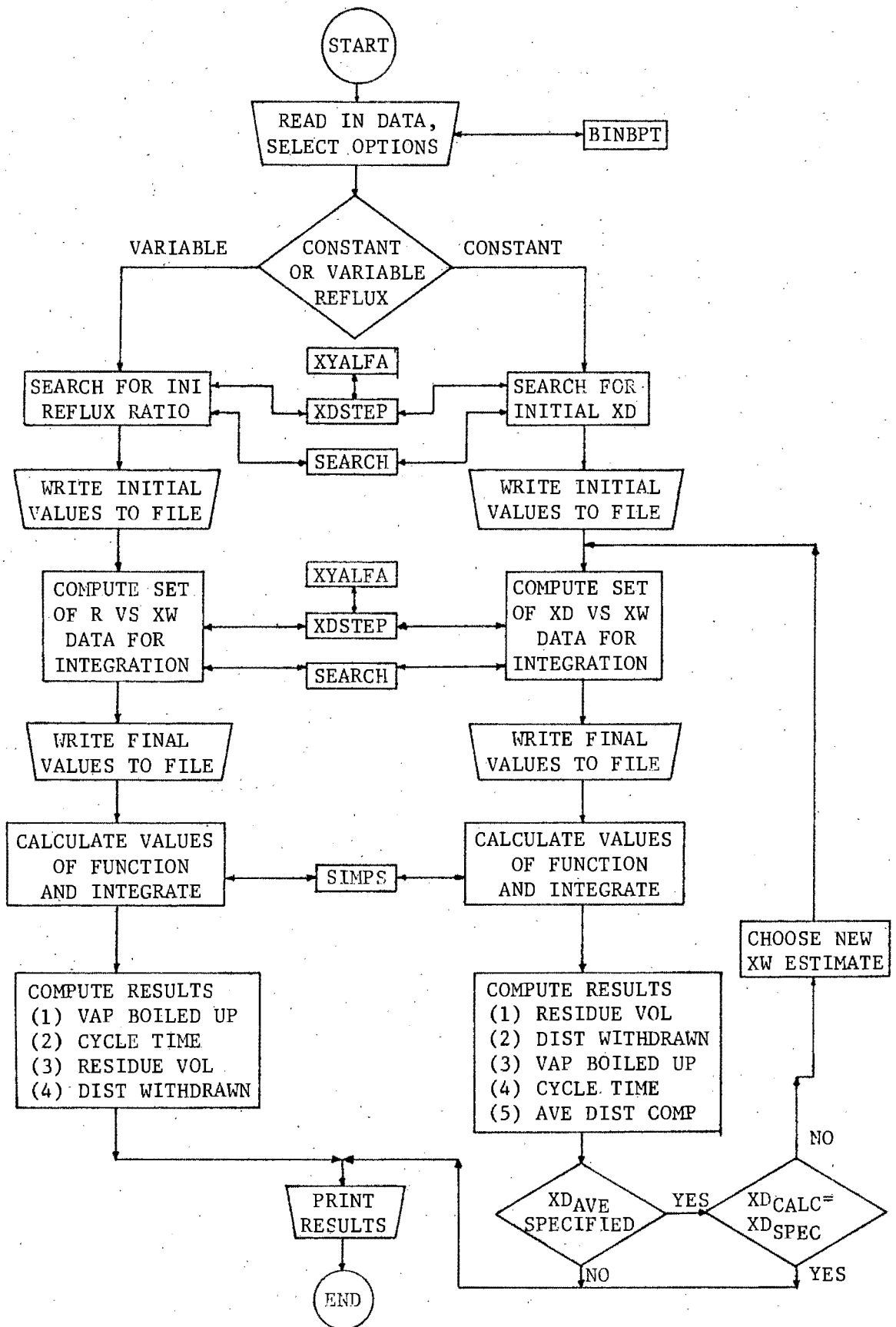


FIGURE 2.3 FLOWCHART OF PROGRAM BATCH

the program in demand mode are shown in the User Manual (Chapter 7). The various options available to the user are also discussed in the Manual.

### 2.34 Calculation Procedures

2.341 Varying reflux ratio: Once all the necessary data has been accepted, the iterative calculation procedure can be summarised as follows:

<u>Line No.</u> *	<u>Program Action</u>
175,176	Estimate an initial reflux ratio $R=1,1$ and specify an incremental value to be used in the iterative sequence.
180	Call XDSTEP to determine the residue concentration corresponding to this value of $R$ .
181	Compare resultant residue composition with the specified feed composition. If they are equal, the estimate of $R$ is correct.
183-185	Should the equality not hold $R$ must be adjusted and subroutine SEARCH is used for this purpose. In general, for a fixed distillate composition and number of stages in the column, the greater the reflux ratio the lower the residue composition. Thus counter NCALL causes SEARCH to decrease the value of $R$ if $XNTC < XF$ (see discussion of SEARCH). The procedure is repeated until the convergence limit is met.
(182)	A check on the size of DELR prevents the loop continuing, if for some reason the convergence criterion is not met.
187-192	Resultant initial conditions of reflux ratio and composition are written to a file for use by the plotting routine.

\*Note: Line numbers refer to the FORTRAN text shown in Appendix C.

- 193        The interval  $(X_F - X_W)$  is divided into NINT increments of size H. Present dimension sizes allow a maximum value  $NINT = 100$ .
- 198-218    Computation of data for numerical integration of Equation (2.17). Starting at  $X_W$  and taking an initial estimate of the final reflux ratio,  $R=10$ , calculations between lines 180 and 185 above are repeated for each value of residue concentration from  $X_W$  to  $X_F$  in steps of H. The values of  $X_W$  and  $R$  are stored in XDATA and YDATA respectively.
- (211-214)   Final conditions of reflux ratio are stored for use by the plotting routine.
- 220-222    Values of the function are calculated at every value of residue composition.
- 223        Integration of the function from  $X_F$  to  $X_W$  is by Simpson's Rule\*.

\*Note: Numerical integration by Simpson's Rule requires equal intervals along the X-axis. This condition is satisfied by using increment H along the X-axis and fitting a corresponding R to each X value. An alternative procedure would involve finding the initial and final reflux ratios and then dividing the interval  $R_{\text{final}} - R_{\text{initial}}$  into increments. The residue composition corresponding to each value of R would then be determined and a polynomial subsequently fitted to the data. Values of the reflux ratio at equal increments of  $X_W$  would then be generated from this function. The saving in computer-time involved in reducing the number of search iterations was found to be negligible, and accuracy was lost through the curve-fitting procedure.



224-227 The following data are computed.

- (i) Total moles of vapour boiled up during batch cycle - VAP.
- (ii) Batch cycle time or vapour boil-up rate, whichever value is not specified initially - TIME or G.

Note: Internally TIME and G are interchangeable. The specific meaning is set by the data input specification and the output record is adjusted accordingly.

- (iii) Quantity of residue by material balance - W.
- (iv) Total distillate withdrawn - DTOT.

305-345 Results are printed out. For sample printouts see Chapter 8.

2.342 Constant reflux ratio: The basic procedure is similar to that used for varying reflux problems. In these problems, however, either a final residue composition or a final average distillate composition may be specified as an end-point.

<u>Line No.</u>	<u>Program Action</u>
230	Initial estimate of XD taken as 0.99.
234-241	Search for correct initial XD based on specific reflux ratio.
243-248	Initial operating diagram data written to file for use by plotting program.

(a) Specification of final residue composition

The procedure is the same as for varying reflux problems except that a set of XD values are searched for using the constant value of R.

258-282 'DO' loop 21 computes data for numerical integration of Equation (2.19).

(265,266) At constant R, XD decreases with decrease in residue concentration; hence NCALL = 1 and

SEARCH reduces XD if XNTC > XSTEP.

(267,268) Check for XD > 1,0.

284-286 Values of the function to be integrated are calculated. The Simpson Rule routine is used to integrate numerically.

288-292 The following data are computed.

(i) Quantity of residue - W.

(ii) Total distillate withdrawn - DTOT.

(iii) Total moles of vapour boiled up during the cycle - VAP.

(iv) Batch cycle time or vapour boil-up rate - TIME or G.

(v) Average distillate product composition - XDAVE.

Result tables are then printed out as shown in Chapter 8.

(b) Specification of average distillate composition.

If this option is used an initial estimate of the final residue composition is set on line 253 ( $XW = 0,1$ ). The calculation procedure is then identical up to (v) above, whereupon the calculated average and the specified average are compared (line 294). The SEARCH routine is used to select a better estimate of XW and the procedure is repeated. An error message prevents final residue composition falling below 0,001 (line 297).

#### 2.4 GENERAL COMMENTS ON BATCH

The program is written in standard FORTRAN and is set up for use on the UNIVAC 1106 - EXEC 8 system at U.C.T. Adaptation to other FORTRAN compilers should be accomplished by changing read and write keys, contained in a DATA statement, in the mainline routine.

Units of output variables conform to SI and all flow-rates are on a molar basis. Where the program will accept units other than SI, this is indicated by the input instruction. This is discussed in detail in the User Manual (Chapter 7).

Stages are numbered up from the bottom with the reboiler designated zero. An extra stage is automatically included for partial condenser operation.

Liquid or vapour hold-up and enthalpy considerations are not taken into account.

Error messages include a check on the final residue composition during the adjustment procedure for the average distillate composition specification option, and several data input checks. If XW becomes negative or an input-question reply is incorrect, the program is terminated.

The full operating procedure is outlined in the User Manual (Chapter 7) and test examples are discussed in Chapter 8.

## 2.5 SUBROUTINES ASSOCIATED WITH MAINLINE BATCH

### 2.5.1 Subroutine XDSTEP

This is the only subroutine sharing a common block with 'BATCH'. It is specific to the binary batch distillation program and is used to construct McCabe-Thiele stages on the x-y diagram for batch distillation problems. Although a similar technique is used in the routine STEPD, which performs the stepping operation for the other binary system programs, a separate routine was needed in this case. This is because the batch program is essentially a rating program in which the number of theoretical stages is known, while the other systems are design programs in that they determine the number of stages required. The subroutine is flowcharted in Figure 2.4.

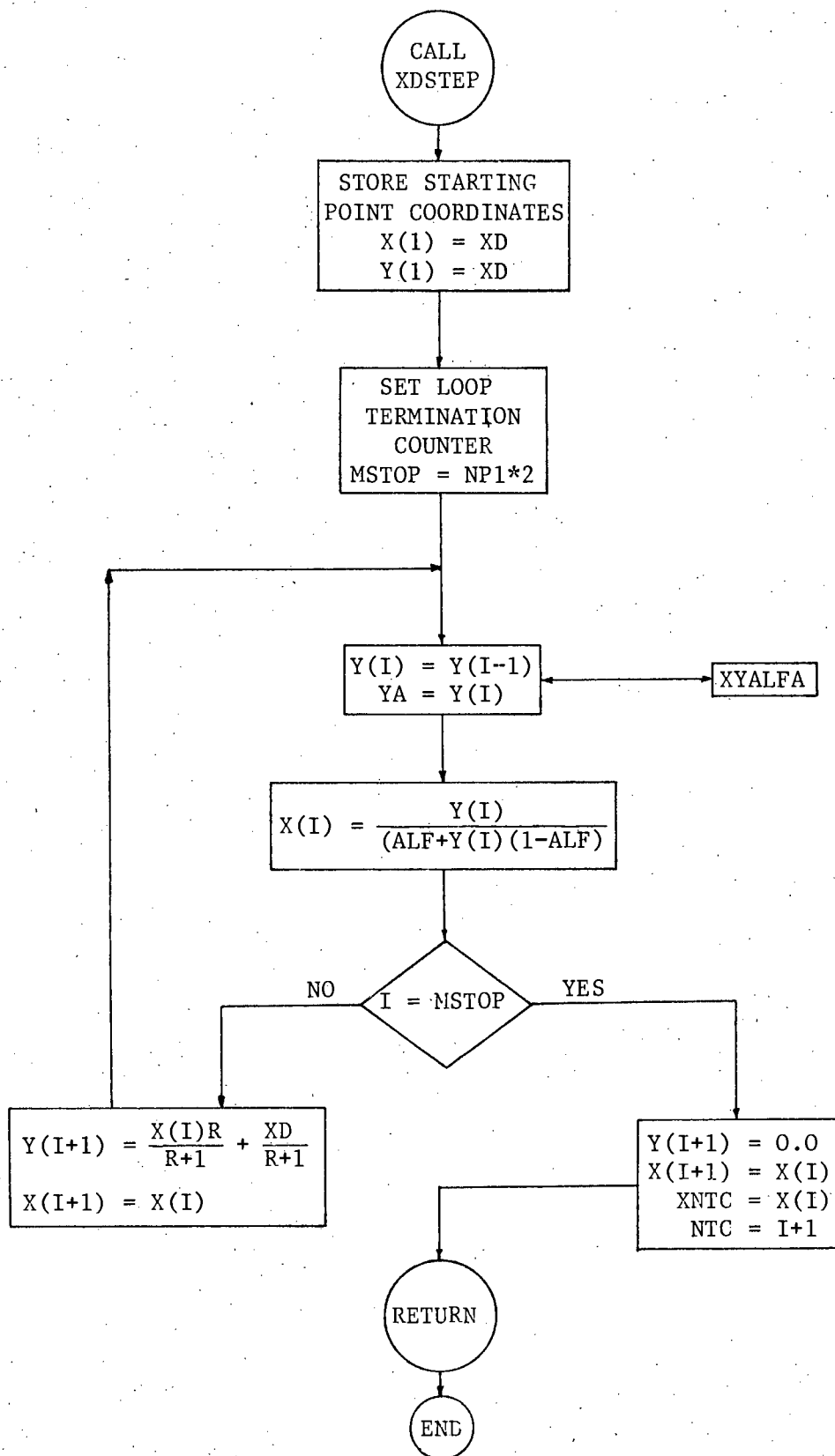


FIGURE 2.4 FLOWCHART OF SUBROUTINE XDSTEP

The computational procedure is equivalent to successive equilibrium - material balance calculations. Starting at point XD on the 45° diagonal an operating line of slope  $R/R+1$  is assumed, and the stages are stepped off as on a McCabe-Thiele diagram. The coordinates of each point determined in the equilibrium curve and operating line are stored in arrays X and Y. Counter NTC records the total number of such points for use in the plotting routine. The sequence of calculations is shown diagrammatically in Figure 2.5 below.

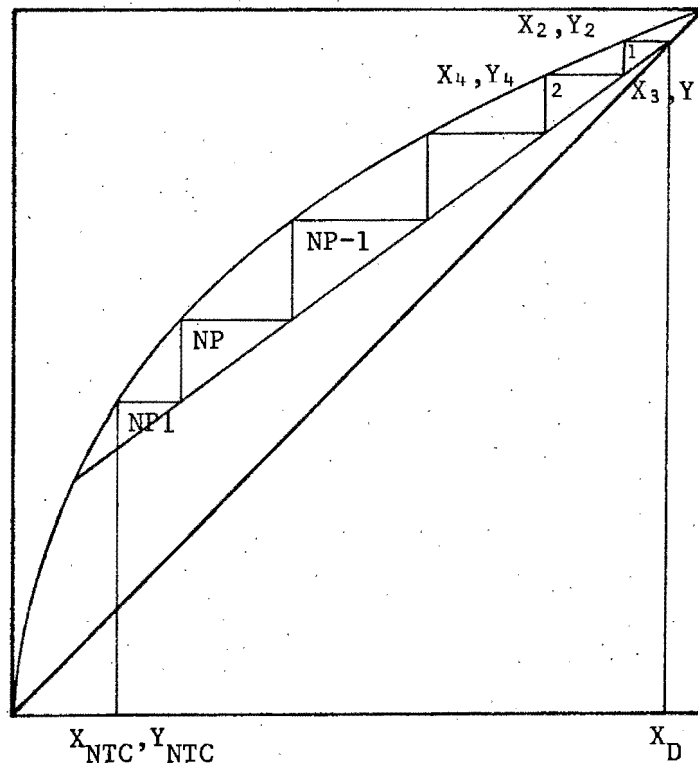


FIGURE 2.5 STEPPING OFF STAGES BY SUBROUTINE XDSTEP

Stage numbers shown in Figure 2.5, 1, 2, ..., NP1 are those used in the routine. The resulting printout and x-y diagram plotting data are converted to the conventional numbering system which designates the reboiler as zero and numbers from the bottom upwards.

Calculation procedure:

<u>Line No.</u>	<u>Program Action</u>
7,8	Store starting point XD as X(1),Y(1).
11-13	Store Y(2) = Y(1) = XD and call subroutine XYALFA to determine corresponding relative volatility ALF.
14	Compute X(2) using variation of Equation (2.2).
16	Compute Y(3) from material balance expression (operating line equation).
17	Store X(3) = X(2).

Counter MSTOP terminates loop after NP1 stages have been stepped off. The final material balance calculation, representing the reboiler is stored as X(NTC), Y(NTC) with Y(NTC) = 0 for convenient x-y diagram construction by the plotting program.

XNTC  $\equiv$  X(NTC) is returned to the mainline for comparison with the set value of the residue composition.

## 2.52 Bubble point and Dew point calculations

2.521 Subroutine BINBPT: This subroutine has no common block and is employed by all the binary programs. Its main function is to return to the calling program the bubble point of the liquid solution with composition X1. It is also used in the computation of equilibrium data when Antoine constants are specified (see section 2.21). For this purpose the vapour pressure, P1, of the light component is returned.

Two procedures for computing the bubble point of a solution are included. The choice is determined by the type of equilibrium data specified.

(a) Antoine constant specifications

For such systems the routine employs a standard Newton procedure to converge on a bubble point  $T_1$  at composition  $X_1$ . The Antoine constants for each component, the total pressure and an initial estimate of  $T_1$ , are provided by the calling program through the calling list. The specified pure component boiling point is used as an initial estimate of  $T_1$ . Newton's method is used to converge Raoult's law to find  $T$  such that  $\sum y_i = 1$ .

From Raoult's Law  $y_i P_T = P_i x_i \quad i = 1, 2$

$$P_T \sum y_i = \sum P_i x_i$$

At bubble-point  $\sum y_i = 1, 0 \quad i = 1, 2$

hence  $P_T = \sum P_i x_i$

or  $P_T - \sum P_i x_i = 0$

where  $P_i$  is given by Antoine's function of temperature.

In Newton's method, a better estimate of the temperature is related to the previous estimate by the equation

$$T_{n+1} = T_n - \frac{f(T_n)}{f'(T_n)}$$

where  $f(T) = \text{SUM1} = P_T - \sum P_i x_i$  and for  $P_i = 10^{(A_i - \frac{B_i}{C_i + T})}$

the first derivative  $f'(T) = \text{SUM2} = -2,303 \frac{B_i}{(C_i + T)^2} \cdot P_i x_i$

Convergence is reached as  $\frac{\text{SUM1}}{\text{SUM2}} \rightarrow 0$

Once the bubble point has been determined in 'DO' loop 10, the vapour pressure of the light component,  $P_1$ , is calculated at the bubble point (line 24).

(b) x-y-T data specification

This procedure is used in Ponchon-Savarit designs when equilibrium composition and temperature data have been supplied. The bubble point is determined by searching through the available data. A straight line interpolation is used between data points and for  $N_{DATA} \geq 20$  this procedure is satisfactory for most systems. For systems that deviate from Raoult's law, such as acetone-water, more points should be provided.

2.522 Subroutine BINDPT: A variation of BINBPT is contained in BINDPT for computing dew points for binary systems. The only difference between the two is the structure of the equations used in the Newton convergence procedure. The two methods were not included in one subroutine because both calculations are not always required. Developed as separate routines, they are available for use in other mainline programs requiring either bubble or dew point calculations.

2.53 Subroutine XYALFA

This subroutine has no common block and is used by all the binary system programs. The routine is used to search for a value of the relative volatility ALPHA corresponding to the vapour or liquid composition value  $XY_1$  supplied by the calling program. For systems with constant relative volatility the routine returns the constant value ALFCON. Linear interpolation is used between actual data points, stored in ALFA, which means that accuracy is better for a greater number of points. It has been found that for most reasonably ideal systems a minimum of 20 data points is acceptable. The calling program determines whether



XYALF and XY1 represent vapour or liquid compositions and the corresponding relative volatility value is returned as ALPHA. NDATA is the total number of data points available and is the size to which ALFA and XYALF are automatically expanded. Counter MALF is set equal to 2 if a constant relative volatility is specified. A DATA statement in the mainline sets MALF = 1 initially.

#### 2.54 Subroutine SIMPS

This subroutine is used by both BATCH and PACKED for numerical integration of various expressions. The technique used is that known as Simpson's Rule.

Consider a function  $f(x)$  divided into  $n$  increments of size  $h$  between limits  $a$  and  $b$ . For an even number of increments,  $f(x)$  is expanded by Simpson's analysis, to give:

$$\begin{aligned} \text{SUM} &= \int_b^a f(x) dx = \frac{h}{3} [(f_1 + 4f_2 + f_3) + (f_3 + 4f_4 + f_5) + \dots \\ &\quad + (f_{n-1} + 4f_n + f_{n+1})] \\ &= \frac{h}{3} [f_1 + 4f_2 + 2f_3 + 4f_4 + 2f_5 + \dots + 2f_{n-1} + 4f_n + f_{n+1}] \end{aligned}$$

Hence

$$\text{SUM} = \frac{h}{3} \sum_{i=1,3,5,\dots}^{n-1} [f_i + 4f_{i+1} + f_{i+2}]$$

Subroutine SIMPS applies this formula to function values  $f_i$ , stored in FUNC, over  $M$  intervals of size  $H$  between limits ALIM and BLIM. The value of the resulting integral is returned to the calling routine as SUM.

#### 2.55 Subroutine SEARCH

This routine is a standard iterative calculation technique, and is used in most of the iterative procedures in the package.

In general  $FA1 \neq FA2$  is to be converged to  $FA1 = FA2$  where FACTOR is a variable that directly or indirectly affects the equality. DEL is the change in FACTOR per iteration that is reduced as convergence is reached. Counter MM causes DEL to be reduced as  $FA1 > FA2$  switches to  $FA1 < FA2$  and vice versa. N defines the overall effect of FACTOR on the inequality.

Example of use of SEARCH.

Consider a batch distillation system operating under variable reflux.

Distillate composition XD is fixed and the initial reflux ratio required for a feed of composition XF is unknown.

An estimate of reflux ratio R is made and a stepping procedure is used to step off the given number of stages from XD. The resultant residue composition will not in general equal XF. Let FA2 equal XF and FA1 equal the resultant residue composition. FACTOR is the reflux ratio and DEL, an incremental value. If the initial trial results in  $FA1 < FA2$ , the reflux ratio must be reduced to improve the result, hence  $N = 2$  for this procedure.

The variables used in this subroutine are not defined in the nomenclature. All the variables are contained in the calling list and initial values are provided by the calling program.

2.6 SUMMARY OF BINARY BATCH DISTILLATION PROGRAM

Program name: BATCH

Type: Mainline with subroutines

Operation mode: Demand or Batch on UNIVAC 1106 - Exec 8 system.

1. Title: Batch Distillation of Binary Systems by McCabe-Thiele method.

2. Solution type: Rating program.

3. Application: BATCH can be used to solve problems involving batch distillation systems operating under either variable or constant reflux conditions. The program is limited to systems for which standard McCabe-Thiele assumptions hold. It is essentially a system rating program in that it requires the number of theoretical stages in the rectifying column to be specified. The program is limited to 25 trays, although this number can be increased if required.

The user must provide equilibrium data in the form of  $x,y$  data, Antoine constants for each component, from which  $x,y$  data are then computed, or a constant relative volatility value.

Important features of BATCH include a Newton method bubble point routine for computing equilibrium data from Antoine constants, an iterative search technique used to generate data for the necessary numerical integrations, and a Simpson's Rule numerical integration routine.

4. Program Development:

(a) Computer used UNIVAC 1106 - Exec 8

(b) Minimum Core requirements: 10 K

(c) Language - Standard FORTRAN IV

(d) Typical run time: Variable reflux: 7 secs

Constant reflux: 9 secs

## CHAPTER 3

### BINARY CONTINUOUS DISTILLATION - TRAY COLUMNS

#### 3.1 INTRODUCTION

True binary distillation problems do not occur very often in practice, but despite this they are very useful for demonstrating the basic principles of the process. Their usefulness derives from the fact that such problems are conveniently solved graphically, in two dimensions. The diagrammatic representation provides the student with a means of visualizing the design and performance of a distillation column.

All three aspects of distillation column design mentioned in Chapter 1 are considered for binary continuous distillation systems. They are:

- (1) Vapour-liquid phase equilibrium on each stage.
- (2) Material and energy balances around each stage.
- (3) Stage efficiency.

The physical dimensions of the column are not considered.

Unlike batch distillation, continuous distillation may be regarded as a steady-state operation. The two graphical design procedures commonly used were proposed by

- (a) McCabe and Thiele, and
- (b) Ponchon and Savarit.

The McCabe-Thiele procedure assumes:

- (a) that the molar latent heats of the two components are equal;
- (b) the heat capacities are constant over the temperature range, and
- (c) that there is no heat of solution.

Enthalpy effects are thus ignored, constant molal overflow is implied and the operating lines on the x-y diagram are straight. This design procedure, although not rigorous, is adequate for systems of chemically similar components.

The Ponchon-Savarit procedure, by comparison, is rigorous and can be applied to all binary systems considered in the undergraduate course. Detailed enthalpy data must, however, be supplied.

The computer program developed here includes both techniques. The user specifies the procedure required and enters data accordingly. Each procedure will be discussed separately.

### 3.2 THEORY OF BINARY CONTINUOUS DISTILLATION

#### 3.2.1 McCabe-Thiele design

The basic simplifying assumption that characterises the McCabe-Thiele procedure is that of equimolal overflow, as defined above. A schematic diagram of a continuous distillation column and the corresponding McCabe-Thiele diagram are shown in Figure 3.1(a) and (b).

The column shows one feed stream, no side streams, a partial reboiler and a total condenser. For systems in which the heavy component is water, the partial reboiler may be replaced by open steam. The total condenser may be replaced by a partial condenser with either a single phase or a two phase distillate product withdrawn. The McCabe-Thiele diagrams for these options are shown in Figures 3.2(a) and (b).

Finally, the column may include several feed and side streams. A McCabe-Thiele diagram for a column with two feeds and a side stream positioned between the feeds is shown in Figure 3.2(c).

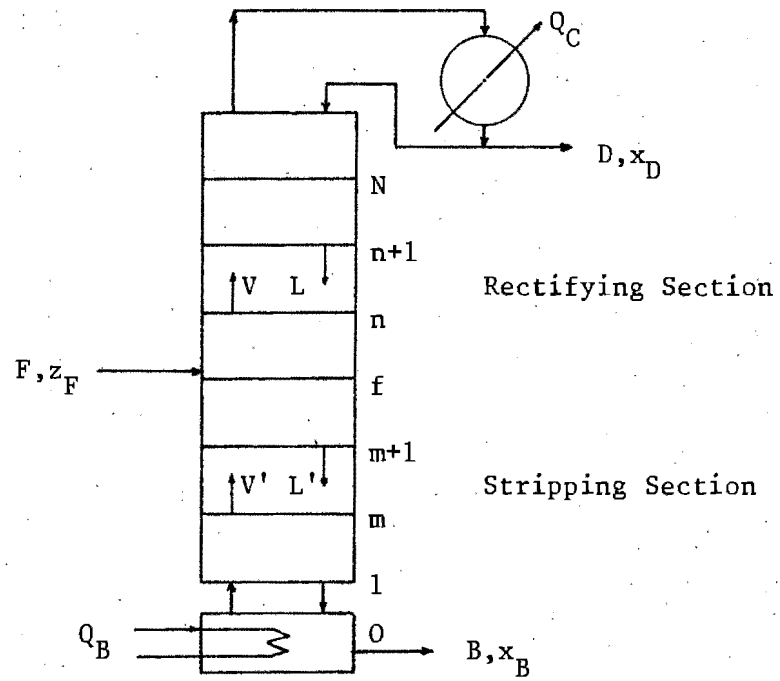


FIGURE 3.1(a) CONTINUOUS DISTILLATION COLUMN

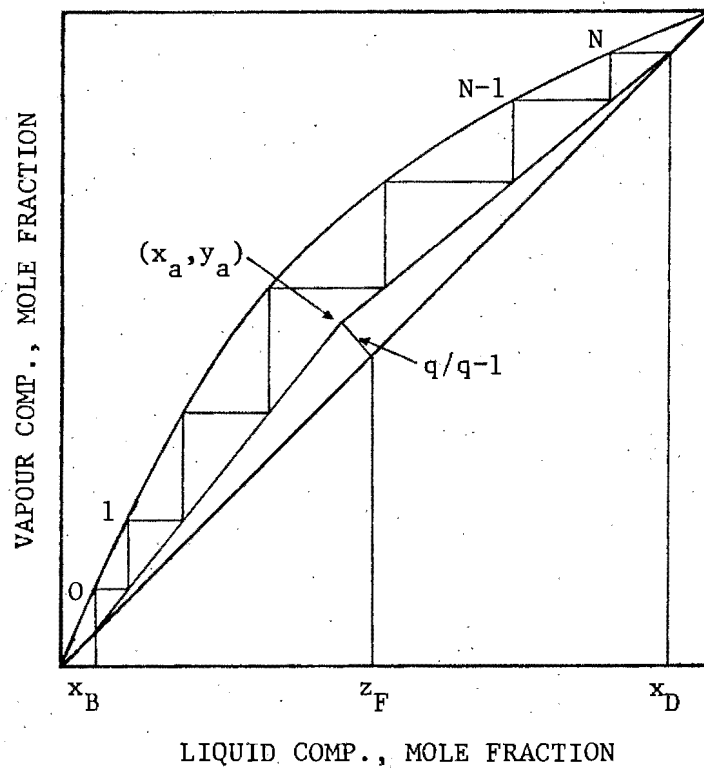


FIGURE 3.2(b) MCCABE-THIELE DIAGRAM OF CONTINUOUS  
DISTILLATION SYSTEM

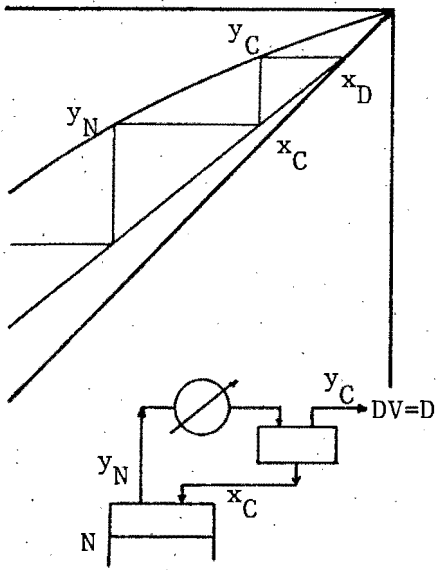


FIGURE 3.2(a) PARTIAL CONDENSER  
ALL VAPOUR DISTILLATE

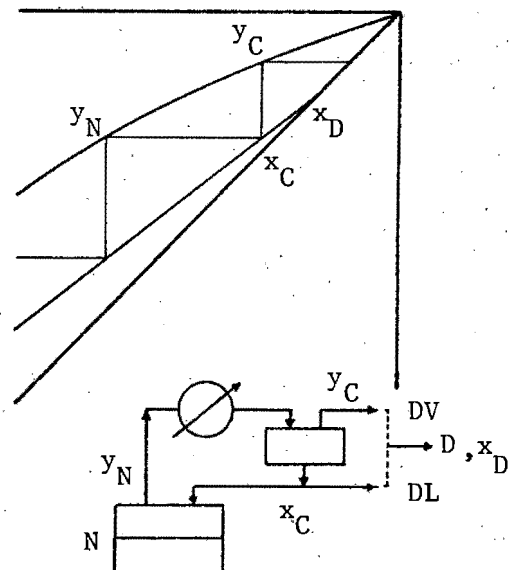


FIGURE 3.2(b) PARTIAL CONDENSER  
VAPOUR & LIQUID DISTILLATE

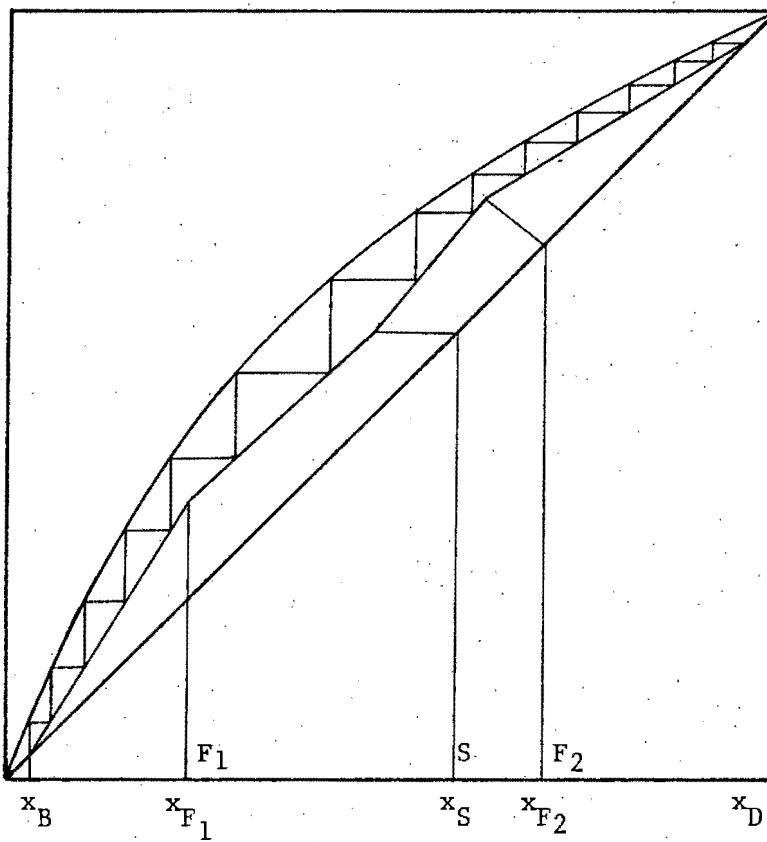


FIGURE 3.2(c) COMPLEX COLUMN  
2 FEEDS AND 1 SIDE-STREAM

The McCabe-Thiele graphical design of a simple column as shown in Figure 3.1 is described by the following equations.

1. Vapour-liquid equilibrium.

The equations listed in section 2.21 apply to continuous systems.

2. Material Balance.

On applying the McCabe-Thiele assumptions the general material balance equation (1.2) can be written around each section of the column.

Rectifying Section:

$$\text{Overall balance } V = L + D \quad (3.1)$$

$$\text{Component balance } Vy_n = Lx_{n+1} + Dx_D \quad (3.2)$$

Resulting operating line,

$$y_n = \frac{R}{R+1} x_{n+1} + \frac{x_D}{R+1} \quad (3.3)$$

Stripping section:

$$\text{Overall balance } L' = V' + B \quad (3.4)$$

$$\text{Component balance } L'x_{m+1} = V'y_m + Bx_B \quad (3.5)$$

Overall column material balance

$$F = D + B \quad (3.6a)$$

Component balance

$$Fz_F = Dx_D + Bx_B \quad (3.6b)$$

For complex columns the overall feed is computed as the difference between the sum of all feeds and the sum of all side streams. The Bottoms flowrate can then be calculated using Equation (3.6).

3. Stage efficiency in terms of Murphree tray efficiency factors.

$$E_M = \frac{y_{n+1} - y_n}{y_{n+1}^* - y_n} \quad (3.7)$$



The thermal condition of the feed is not generally known and should be determined by a flash calculation. For McCabe-Thiele systems, enthalpy and equilibrium-temperature data are not specified and hence a flash calculation is not possible. The feed condition is thus defined as the fraction of liquid in the feed,  $q$ , such that  $qF$  are the moles of liquid in the feed. The  $q$ -value must be known in order to determine the point of intersection of the operating lines. The locus of these points of intersection can be expressed as

$$y_q = \frac{q}{q-1} x_q - \frac{z_F}{q-1} \quad (3.8)$$

Equations (3.3) and (3.8) can be solved simultaneously to define the coordinates of the point of intersection of the operating lines (see Figure 3.1(b)).

$$x_a = [z_F(R+1) + x_D(q-1)]/(R+q) \quad (3.9)$$

$$y_a = [z_F R + x_D q]/(R+q)$$

The manual design procedure involves constructing the  $x$ - $y$  equilibrium curve, inserting the operating lines and stepping off the number of stages as shown in Figure 3.1. The stepping procedure will not in general, link  $x_B$  and  $x_D$  exactly. The fraction of a stage represented by a mismatch is computed as

$$\text{Stage fraction} = (x_D - y_{n-1})/(y_n - y_{n-1}) \quad (3.10)$$

The first step on the  $x, y$  diagram represents the reboiler and this is subtracted from the total to determine the number of theoretical trays in the column. Another step must be subtracted if a partial condenser is specified, see Figure 3.2(a & b).

### 3.22 Ponchon-Savarit design

The detailed development of the graphical design procedure first proposed by Ponchon and Savarit [8] is set out in Treybal [9], Smith [11] and several other texts on binary distillation.

The basic equations defining a binary system have been shown in section 3.21. When McCabe-Thiele assumptions are not made, an energy balance must be considered. In addition, flowrate subscripts must be included in the material balance equations to identify the individual stage positions.

The general enthalpy balance Equation (1.3) can be written around each section of the column as follows. Heat losses from the column are assumed to be negligible.

Rectifying section:

$$V_n H_n = L_{n+1} h_{n+1} + D H_D + Q_C \quad (3.11)$$

Stripping section:

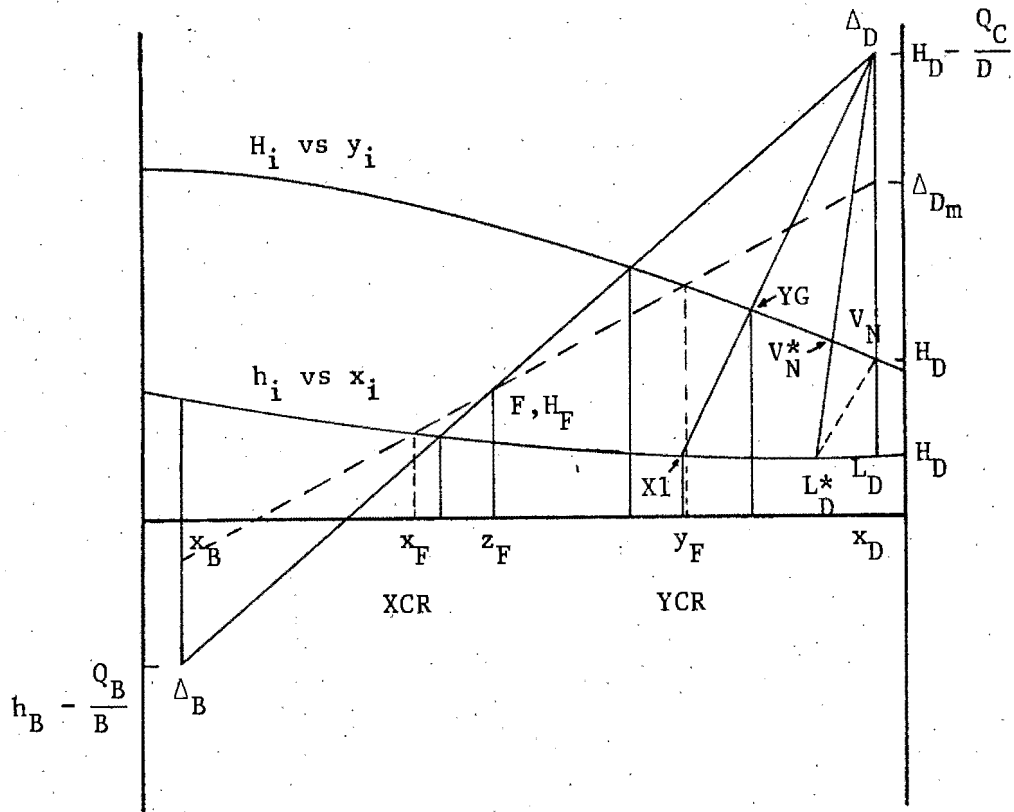
$$L_{m+1} h_{m+1} = V_m H_m + B h_B - Q_B \quad (3.12)$$

Equations (3.2) and (3.4) are rewritten as follows to account for varying flows through the column

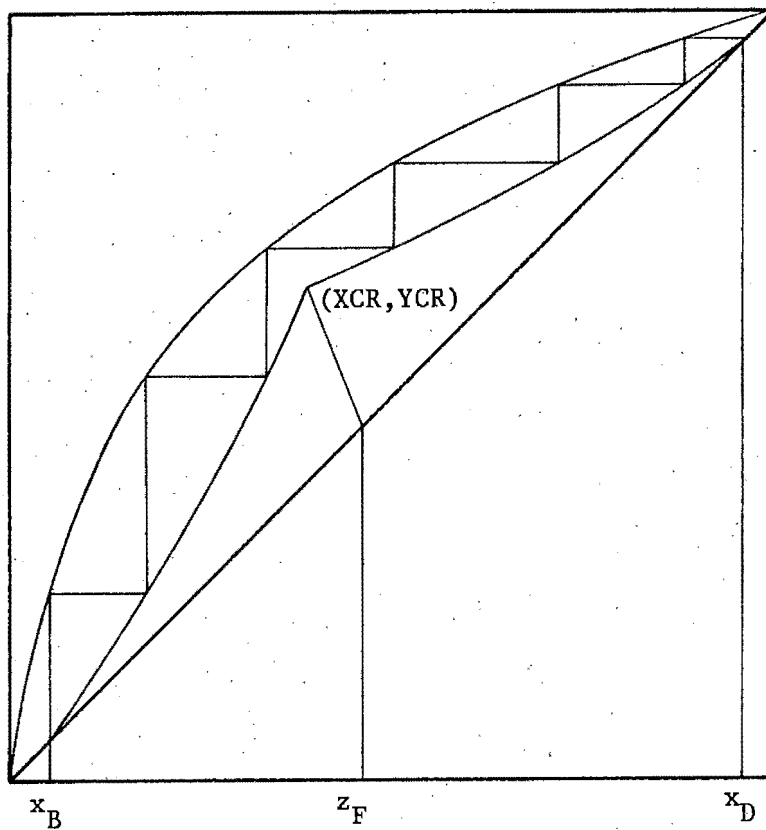
$$V_n y_n = L_{n+1} x_{n+1} + D x_D \quad (3.13)$$

$$L_{m+1} x_{m+1} = V_m y_m + B x_B \quad (3.14)$$

The operating lines are not straight and the material and energy balances must be solved simultaneously with the equilibrium expression. The Ponchon-Savarit procedure solves the equations graphically. Figure 3.3 shows an enthalpy-concentration diagram and the corresponding x,y diagram for a binary system. The first step in a Ponchon-



Enthalpy-Concentration Diagram



x,y diagram

FIGURE 3.3 PONCHON-SAVARIT DIAGRAM

Savarit design is the development of the enthalpy-concentration diagram which is then used to construct the operating lines on the x,y diagram.

The Ponchon-Savarit operating diagram is used in the design both of tray columns and of packed columns in this study. The program for designing packed columns (see Chapter 4) is thus referred to in the description of the computer techniques used to construct this operating diagram.

(a) Enthalpy concentration diagram.

The construction of the enthalpy-concentration diagram as described below is shown in Figure 3.3.

The enthalpy-concentration curves are established, assuming ideal solution behaviour using the following relationships.

Enthalpies of streams leaving stage j

$$\text{liquid enthalpy } h_j = x_1 h_1 + x_2 h_2 + H_S \quad (3.15)$$

$$\text{vapour enthalpy } H_j = y_1 H_1 + y_2 H_2 \quad (3.16)$$

Pure component enthalpies  $h_1$  and  $H_1$  are computed at the solution bubble point and dew point temperatures respectively.  $H_S$ , the heat of solution, is included only if such data are available. It is usually convenient to adopt the temperature, at which heat of solution data are quoted, as the base temperature for enthalpy calculations.

Pure component enthalpies are computed as follows:

$$h_1 = \int_{T_b}^{T_B} C_{pd} T \quad (3.17)$$

$$H_1 = \int_{T_b}^{T_D} C_{pd} T + L_{H_1, T_D} \quad (3.18)$$

where  $LH_{i,T_D}$  is the latent heat of vaporization of the component, calculated at the dew point temperature of the solution.

The heat capacity,  $C_p$ , is expressed as some function of temperature for which coefficients are supplied. The following function, developed by Kobe [55] is used in the programs in this study.

$$C_{pi} = a_i + b_i T + c_i T^2 + d_i T^3 \quad T \text{ in } ^\circ K \quad (3.19)$$

The latent heat of vaporization for each component at its boiling point under column pressure must be supplied by the user. These values are extrapolated to other temperatures using Watson's correlation [56], which is

$$LH_{i,T} = LH_{i,T_{BP}} \cdot \left[ \frac{T_C - T}{T_C - T_{BP}} \right]^{0.38} \quad (3.20)$$

This correlation requires component critical temperature data.

The overall heat balance around the column can be expressed as follows

$$FH_F = DQ' + BQ'' \quad (3.21a)$$

$$\text{or} \quad F = \Delta_D + \Delta_B \quad (3.21b)$$

where  $\Delta_D$  and  $\Delta_B$  represent imaginary streams defined as

$$\Delta_D = DH_D + Q_C \quad (3.22)$$

$$\Delta_B = BH_B - Q_B$$

Thus the "enthalpy" of these streams includes the condenser and reboiler heat duties respectively. The overall heat

and material balance is represented on the diagram by line  $\Delta_D - F - \Delta_B$  (from Equations (3.21b)).

Lines  $\Delta_D L_D$  and  $\Delta_D L_D^*$  represent the material and enthalpy balance around a total and partial condenser respectively. By the lever-arm rule, the reflux ratio for a total condenser is given as:

$$R = \frac{\text{line } \Delta_D V_N}{\text{line } V_N L_D} \quad (3.24)$$

and similarly for a partial condenser.

The tie line passing through point F on the diagram is defined by the equilibrium expression for the system. If this line is extended to points  $\Delta_{DM}$  and  $\Delta_{BM}$  the separation thus defined requires an infinite number of stages in both sections of the column. This pinch point condition at the feed position enables the minimum reflux ratio to be calculated as:

$$R_M = \frac{\text{line } \Delta_{DM} V_N}{\text{line } V_N L_D} \quad (\text{for total condensers}) \quad (3.25)$$

The program uses this technique for computing  $R_M$ , whereupon it checks that the specified value is greater than the minimum. If  $R_{\text{spec}} < R_M$  a value of  $R = 1.5 R_M$  is assumed and the calculations are continued. Treybal [9] notes that for many systems the optimal reflux ratio lies between  $1.2 R_M$  and  $1.5 R_M$ . This procedure avoids pinch points at the intersection of the operating lines. Pinch points resulting from Raoult's law deviations are determined during the stepping procedure.

To calculate  $R_M$  as described above, the tie line through F must be known. For Ponchon-Savarit designs it is assumed that the feed is at the same pressure as the column and that the temperature is known. The feed condition can be expressed

in terms of  $q$  which is defined as the heat required to convert 1 mole of feed from its condition  $H_F$  to a saturated vapour [9]. The  $q$ -value is thus defined by the equation

$$q = \frac{H_{V_F} - H_F}{H_{V_F} - H_{L_F}} \quad (3.26)$$

where  $H_{V_F}$  and  $H_{L_F}$  are the enthalpies of the feed as a saturated vapour and liquid respectively.

For superheated or supercooled feed the above equation is used in the program to calculate  $q$ . If a two phase feed is considered, Equation (3.8) is written as

$$q = \frac{y_q - z_F}{y_q - x_q}, \quad (3.27)$$

and is used as described in section 3.53. At this stage sufficient information is available to construct the operating lines on the  $x,y$  diagram.

(b)  $x,y$  diagram.

The equilibrium curve on the  $x,y$  diagram is determined as described in section 2.21. The manual procedure described in textbooks constructs the operating lines and the stepped stages simultaneously. It is more convenient to perform these constructions separately on a computer because, if the operating lines are defined, the same basic stepping procedure may be used both for McCabe-Thiele and for Ponchon-Savarit designs. In addition, the design of a packed column requires only the operating lines. Thus the same computer technique can be used for both types of column if stepping is not done simultaneously.

The program computes the coordinates of a number of points defining the operating lines for each column section. These data are stored in arrays for use in design of either tray or packed columns. The same stage-stepping subroutine is used in both design procedures for determining the number

of trays in the column. A different calculation technique is used in each case, however, because the operating line is defined by a series of points in Ponchon-Savarit design, and by a straight line equation in McCabe-Thiele designs. (See sections 3.516 and 3.517.)

### 3.3 DESCRIPTION OF PROGRAM BINARY

#### 3.31 General structure

BINARY is the mainline program for solving binary continuous distillation problems using either McCabe-Thiele or the Ponchon-Savarit design procedure. The mainline program is used primarily to link the various subroutines together for a specified design procedure. All the calculations are performed in these subroutines with the mainline providing the relevant data and determining the calculation sequence. The overall operation of BINARY is flowcharted in Figure 3.4.

#### 3.32 Limitations

The program is a design facility in that it computes the number of trays required to achieve a specified product purity. A Murphree tray efficiency factor may be included to account for real tray behaviour. McCabe-Thiele systems involving simple or complex columns may be considered whereas Ponchon-Savarit designs are restricted to single columns only. It is assumed that all binary systems considered are ideal and that, for Ponchon-Savarit designs, enthalpy function coefficients are available in the correct form. Entrainment, weeping and heat losses are not considered in either case.

Either partial or total condenser may be specified for either design procedure. Open stream operation may replace the partial reboiler in McCabe-Thiele problems. A flash calculation determines feed condition for Ponchon-Savarit designs. The feed  $q$ -value must be specified directly for



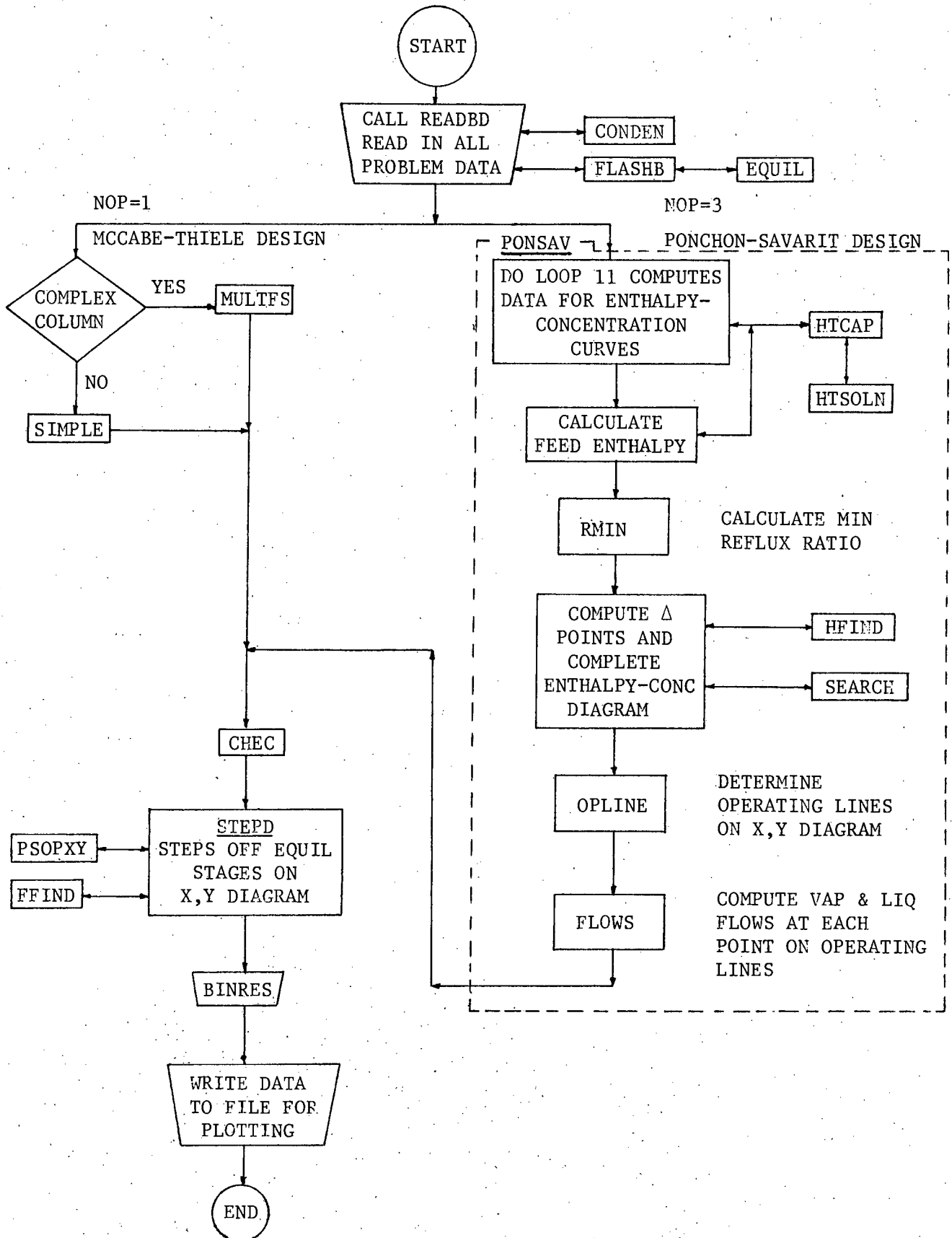


FIGURE 3.4 FLOWCHART OF PROGRAM BINARY

McCabe-Thiele designs. The dimensions of the column are not computed in either case (see Chapter 8). A full description of the options available to the user is given in the User Manual, Chapter 7.

### 3.33 Operating procedure

The program is designed for operation from either a terminal in demand mode or via a card deck in batch mode. A conversational facility is available in demand mode but, because of the type of enthalpy data required, the user is expected to insert this in a data element when using the Ponchon-Savarit design option (User Manual). The conversational output may be suppressed and a complete data element used.

Data for plotting operating diagrams for either design procedure are written to file by BINARY for subsequent use by the CALCOMP plotting program. Data files must be assigned for this purpose before BINARY is executed, if plotting is required. Subroutine TEACH provides a brief description of the program's capabilities and then terminates the run because users who are not familiar with the program are unlikely to operate it successfully without preparation.

### 3.34 Calculation procedure in mainline program BINARY

<u>Line No.</u>	<u>Program Action</u>
5,6	DATA statement for defining constants and counters.
7,8	NPPSC and NPTCC are limits used to prevent dimension overflows in various calculation sequences. NPPS and NPTC are parameters defining the size of arrays (see COMMON block BINCOM).
9-32	Program determines the design method required by the user and the data input procedure to be used.

\*See FORTRAN text, Appendix C.

33           Program calls subroutine READBD which requests data conversationally (MCAN=1) or reads a data element (MCAN=2). These options are described in the User Manual, Chapter 7.

(a) McCabe-Thiele design procedure.

36-39       If a complex column is specified, subroutine MULTFS is called to set up operating lines and compute flowrates in each section of the column (see section 3.55). Control is then transferred to line 49 for stepping off stages on the x-y diagram.

40-44       Subroutine SIMPLE determines the operating lines and section flowrates for simple columns. (See section 3.56.) Control is then transferred to line 49 as above.

(b) Ponchon-Savarit design procedure

46-48       Subroutine PONSAV constructs the complete operating diagram consisting of an enthalpy concentration diagram and an x,y diagram. The heat duties of the condenser and reboiler are also computed (see section 3.57). On return to the mainline the run continues as for McCabe-Thiele designs.

(c) Stepping off equilibrium stages on the x,y diagram

49           302 CONTINUE

50,51       Subroutine CHEC ensures that no pinch points exist on the x,y diagram (see section 3.515).

52           Subroutine STEPD steps off the stages and computes the total number of theoretical trays required in the column (see section 3.516).

- 53 Subroutine BINRES prints out the table of results. The user may truncate the table if the full set of results is not required. (See section 3.519.)
- 54-85 Data for use by the plotting program is written directly into previously assigned files. This is described in the User Manual.
- 87-89 An error message is printed out if errors are detected in the user's conversational inputs. Other error messages are contained in the associated subroutines.

### 3.4 GENERAL COMMENTS ON BINARY

The linear interpolation search procedure described in section 2.5 (subroutine XYALFA) is used extensively in BINARY and its subroutines. All data tables searched in this way involve composition as the independent variable. It has been found that a good comparison between computer and manual solution is achieved with a minimum of twenty data points (composition increment 0,05 mole fraction). The user may increase the size of the arrays, by changing a parameter in the COMMON block BINCOM, to accommodate a greater number of data for greater interpolation accuracy. For more than fifty points, however, the improvement in accuracy for ideal systems will not necessarily justify the increased core storage requirement. (See Chapter 8.)

Enthalpy data may be supplied in any convenient set of units. Conversion factors relating the data to the internally used British FPS system must also be supplied. Results are converted to the SI before printing to conform to the metrification regulations. This procedure has been adopted because many of the data and correlations quoted in the literature are in the FPS system. The use of conversion factors by the program is discussed fully in the User Manual.

Error messages result if material balances fail or if a pinch point is detected. Errors detected in subroutines causes MALF=1000 and this, on return to the mainline, causes termination of the run. (See User Manual, Chapter 7.)

This program will enable users to investigate the effects of parameters such as reflux ratio and stage efficiency. The relative merits of the two design procedures can also be compared.

### 3.5 SUBROUTINES ASSOCIATED WITH MAINLINE BINARY

Only those subroutines not discussed in Chapter 2 appear here.

#### 3.51 Subroutine READBD

This subroutine is used to read all the data required to construct either McCabe-Thiele or Ponchon-Savarit diagrams for binary continuous distillation systems. It is used both by BINARY and by PACKED, which uses a Ponchon-Savarit diagram in the design of packed columns. Counter NOP defines the design procedure required by the user.

NOP=1 Tray column McCabe-Thiele diagram.

NOP=2 Packed column Ponchon-Savarit diagram

NOP=3 Tray column Ponchon-Savarit diagram

In addition to accepting data READBD computes equilibrium data as described in section 2.21 and calls subroutine FLASHB to determine the feed condition for Ponchon-Savarit designs. A sample runstream showing the conversation data-input system is given in sets MCAN=2 which suppresses the conversational output. The output formats in the text serve to document the routine. Intermediate calculations are described below. Details of the data required and the input formats are given in the User Manual (section 7.2).

<u>Line No.</u>	<u>Program Action</u>
15-19	Conversion of supplied pressure units to mm Hg. The system pressure is required for computing bubble points and dew points using Antoine's equation. Antoine constants are usually related to vapour pressure in mm Hg and hence it is convenient to convert the total pressure to the same units.
29-98	Calculation of vapour-liquid equilibrium data as described in Chapter 2. Values are stored in arrays XEQUIL and YEQUIL for use by plotting routines.
37-54	Calculation of x-y and $\alpha$ data using Antoine constants (Equation (2.6) and (2.2)).
67-74	Calculation of x-y data using a constant relative volatility. (Equation (2.2)). This option is not available to Ponchon-Savarit designs.
(85-98)	Calculation of $\alpha$ data using supplied x-y data (Equation (2.2)).
Note:	The calculation or specification of equilibrium temperature data, TALF, is required for Ponchon-Savarit designs. If x-y data are supplied for a McCabe-Thiele design, array TALF may be filled with zeros (see User Manual).
113	Subroutine CONDEN is called to determine the composition and phase distribution of the distillate product. (See section 3.52.)
114	Complex columns may be specified only in McCabe-Thiele designs.
173-177	Correcting the enthalpy function coefficients. (See User Manual.)
185	Subroutine FLASHB called to determine the feed condition. (section 3.53)

### 3.52 Subroutine CONDEN

This subroutine determines the composition of the product stream leaving the condenser. Either a partial or a total condenser may be specified. When CONDEN is called the distillate product composition XD and the type of condenser are known.

#### Line No.

#### Program Action

##### (a) Total condenser (NCOND=1)

7-9        XD is the composition of the liquid distillate product. XC and YC are set equal to XD and control is returned to the calling routine. YC is the composition of the vapour in the condenser (see Figure 3.2 (a & b)) and is not equal to XD for total condensers. In the program, however, YC is not required as a composition, but only as value for terminating the stepping procedure in subroutine STEPD (see section 3.516, line 21). Thus setting YC=XD here is convenient and does not cause any errors in the final results.

##### (b) Partial condenser (NCOND=2)

12-15      Routine determines the fraction of each phase in the total distillate product stream.

##### (i) Single phase distillate product (all vapour) (Figure 3.2(a))

17-21      For vapour distillate, YC=YD and XC is determined from equilibrium data using XYALFA. This is the only partial condenser type allowed by the program in Ponchon-Savarit designs.

##### (ii) Two phase distillate product (Figure 3.2(b))

23-31      Routine determines the significance of the specified composition XD.

- 32-39 For specified XD referring to vapour phase, i.e.  $Y_C = X_D$ , the corresponding liquid phase composition  $X_C$  is determined using equilibrium data. The true overall composition is then computed as a new XD.
- 41-49 For specified XD referring to liquid phase the corresponding vapour phase and overall compositions are computed similarly.
- 50-63 If XD is specified as the overall composition, phase compositions  $Y_C$  and  $X_C$  are determined using the standard iterative search procedure . (See subroutine SEARCH, section 2.5).

### 3.53 Subroutine FLASHB

The feed condition is determined by this subroutine using the equations described in section 3.2. A flash calculation is unnecessary when considering ideal binary systems for which equilibrium temperature composition data are known.

<u>Line No.</u>	<u>Program Action</u>
5-7	Compute bubble and dew points of feed and test to determine whether or not it is superheated or supercooled.
(a) Two phase feed	
8	Call subroutine EQUIL to determine equilibrium vapour-liquid composition of feed based on known feed temperature.
9	Compute feed condition, Q (Equation 3.27 with $x_F, y_F$ substituted as $x_q, y_q$ ).
(b) Superheated or supercooled feed	
12-15	Determine enthalpy of feed at specified temperature (HVF or HLF) and at a saturated vapour (HV) and liquid (HL).



- 16            Compute feed condition Q (Equation 3.26).
- 18-21        Feed q-value and temperature data are printed  
             out for inspection.

### 3.54 Subroutine EQUIL

The standard linear interpolation search technique described in section 2.53 (subroutine XYALFA) is used here to determine the equilibrium liquid and vapour compositions of a two phase feed at the specified feed temperature, TF. Equilibrium temperature composition data stored in TALF, XALF and YALF are searched for the required values. The selected compositions are returned through the calling list as X1 and Y1.

### 3.55 Subroutine MULTFS

This subroutine is called by BINARY for McCabe-Thiele design of complex columns. When control is returned to the mainline, the x,y diagram is ready for stepping off equilibrium stages.

<u>Line No.</u>	<u>Program Action</u>
3-34	Overall material balance calculation. In the event of failure an error message is printed out and the program is terminated. (Equation (3.6))
37-38	Coordinates of point of intersection of overall operating lines (Equation (3.9)).
39-44	Slope of overall bottom operating line is then calculated and hence the flowrates in the bottom section of the column.
46-78	Feed and side streams may be specified in any order by the user. 'DO' loop 21 arranges the streams in order of increasing composition of the light component. They are then positioned

in the column in this order to minimise the number of trays required. The effect of inserting a more concentrated feed below a less concentrated feed may be investigated manually by the student, but is not considered in the program.

- 79-85 Bottom operating line equation slope and intercept stored.
- 86-94 'DO' loop 24 computes the points of intersection, slopes and intercepts of the operating lines defined by the order of the feed and side streams. The flowrates in each section of the column are also calculated.
- 98-107 Error messages are printed out if the material balance fails or if two streams with the same composition are found. MALF is set to 1000 to terminate the mainline on return.

### 3.56 Subroutine SIMPLE

This subroutine is called by BINARY for McCabe-Thiele design of simple columns.

<u>Line No.</u>	<u>Program Action</u>
6,7	Point of intersection of operating lines for a simple column is computed as XQM,YQM (Equation (3.9)).
9-19	Calculation of bottoms flowrate and composition by material balance for either partial reboiler or open steam operation (Equation (3.6)).
11 and 19	Material balance failure checks.
22-33	Development of operating line equations and calculation of vapour and liquid flowrates in each section of column.

- 34-39 Points of intersection of operating lines with the diagonal and with each other are stored for use in subroutine STEPD.
- 42-46 Error message is printed out and program is terminated if the material balance fails. MALF=1000 causes termination on return to main-line.

### 3.57 Subroutine PONSAV

This subroutine constructs a Ponchon-Savarit operating diagram. The enthalpy concentration diagram is set up first and the operating lines on the x,y diagram are defined from this. Condenser and reboiler heat duties are computed. The subroutine is called by BINARY and by PACKED (see Chapter 4).

<u>Line No.</u>	<u>Program Action</u>
3	Heat duty and stream enthalpy conversion factors.
4-6	Calculation of bottoms flowrate and composition by material balance, with check for failure. Only partial reboiler operation is considered (Equation (3.6)).
13-29	'DO' loop 10 computes data required for construction of enthalpy-concentration diagram. Procedure as follows.
(14-23)	(i) At vapour and liquid composition increments of 0,05 mole fraction, dew point and bubble point are calculated in subroutines BINBPT and BINDPT respectively (see section 2.52).
(24)	(ii) Subroutine HTCAP computes vapour and liquid enthalpies for the current composition and temperature data. (See section 3.58.)
(25,26)	(iii) Enthalpy data converted to SI units.

- 31-36      Feed enthalpy calculated similarly using HTCAP.
- 37          Subroutine RMIN computes the minimum reflux ratio (section 3.511).
- 39          This value is compared with the specified reflux ratio.
- 41          If the specified value is too low a value of 1,5 times the minimum is assumed. The user is informed of this action by error message. See line 106 below.
- 43,44      Distillate product enthalpy determined by either total or partial condenser specification. ( $H_D$  on Figure 3.3.)
- 45-51      Development of ' $\Delta$ ' points on enthalpy-concentration diagram (Equation (3.21-3.23)).
- 52,53      The equation of the line passing through these points and the feed enthalpy is defined. (Line  $\Delta_B-F-\Delta_D$ .)
- 57-77      Search for point of intersection of the operating lines, which in general are not straight. Subroutine SEARCH is used in the iterative search for the correct composition values as described in Chapter 2.
- 78          Subroutine OPLINE computes the coordinates of a set of points defining each operating line. (See section 3.512.)
- 80          Subroutine FLOWS calculates the vapour and liquid flowrates at each point in the column defined by OPLINE. (See section 3.513.)
- 82,83      Conversion of heat units from FPS system to the SI.

- 85-97      The points of intersection of the operating lines with the diagonal and each other, are stored for use in the stepping procedure. These calculations are not required in packed column designs and are therefore avoided (NOP=2).
- 99-110     Error messages are printed out if the material balance fails or if the specified reflux ratio is less than the calculated minimum. The former error terminates the execution.

### 3.58 Subroutine HTCAP

This subroutine calculates vapour and liquid enthalpies at specified points through the column, thus defining the enthalpy-concentration diagram for Ponchon-Savarit designs. The calling list supplies the temperature and composition data and returns the calculated enthalpies to the calling program as HV and HL. The calculations are performed using the equations discussed previously.

<u>Line No.</u>	<u>Program Action</u>
5-7	Convert temperature values to units required by function coefficients (see User Manual).
8-14	'DO' loop 10 computes the molar enthalpy of each component in the vapour over the temperature range TD-T where T is some base temperature (integral of Equation (3.19)).
15-16	Calculation of enthalpy contribution of each component based on composition YS, (1-YS). Conversion factors CPMW and CPCONV ensure consistency of units (Equation (3.16)).
17-20	'DO' loop 11 computes latent heat of solution at vapour dew point (Equation (3.20)).
21	Total enthalpy of saturated vapour at temperature TD (Equation (3.18)).

- 22-28. 'DO' loop 20 performs the same calculations as 'DO' loop 10, for the liquid phase. (Equation (3.17).)
- 29 Heat of solution may be included if data are available. Subroutine HTSOLN is called to calculate this value.
- 34 Total enthalpy of saturated liquid at temperature TB. (Equation (2.15).)

### 3.59 Subroutine HTSOLN

This subroutine is called by HTCAP to compute liquid phase heats of solution if data are supplied. It is assumed that the base temperature taken for the enthalpy calculations is the temperature at which heat of solution data are quoted.

The standard linear interpolation search procedure is used to determine a value H for a solution of composition XA. HEVABS is equal to either +1 or -1 for endothermic or exothermic mixing respectively.

### 3.510 Subroutine HFIND

The vapour and liquid enthalpy data calculated in HTCAP is stored in arrays HVAP and HLIQ respectively, with corresponding composition arrays YH and XH. Subroutine HFIND accepts enthalpy-composition data for either phase through HENTH and XYH. The standard linear interpolation search procedure is used to determine the enthalpy H of a solution of composition XA using these data. NENTH data points are contained in the data arrays. The search procedure is discussed in subroutine XYALFA, section 2.53.

The dimension statement expands automatically to accommodate the particular arrays carried through in the calling list. This reduces the total core requirements during execution. (See Chapter 8.)

### 3.511 Subroutine RMIN

This subroutine calculates the minimum reflux ratio allowed by the specifications for a Ponchon-Savarit design problem. The routine is called by subroutine PONSAV as part of the enthalpy-concentration diagram construction. The procedure followed is discussed in section 3.22. The programming of this procedure is described below.

The standard iterative search technique, using subroutine SEARCH, is used to determine the tie line passing through the feed point on the diagram. The steps are as follows. (Refer to Figure 3.3.)

<u>Line No.</u>	<u>Program Action</u>
5	Initial vapour concentration chosen $Y_1 = X_D$ .
11,12	XYALFA called to determine relative volatility ALF corresponding to $Y_1$ and equilibrium liquid composition $X_1$ is calculated.
16,17	HFIND uses linear interpolation search procedure to determine enthalpies of vapour and liquid of composition $Y_1$ and $X_1$ .
20,21	Test if feed point (enthalpy HF) lies on the tie line.
23	SEARCH called for new estimate of $Y_1$ .
24	Return to repeat lines 11-23 above.
25	When HF falls on line within limit ACC (line 3), search is complete. Point $X_1, Y_1$ is the point of intersection of the two operating lines and the equilibrium curve.
28	Determine enthalpy of reflux liquid using HFIND. For a total condenser this is equal to the distillate product enthalpy ( $X_C = X_D$ ).

- 30 Determine enthalpy of top vapour for a total condenser operation.
- 32-46 Determine enthalpy of top vapour for a partial condenser operation.
- 48 Compute minimum reflux ratio RM. (Equation (3.25))
- 50-71 Error messages prevent excessive looping in search procedures, all of which cause termination of the run.

### 3.512 Subroutine OPLINE

This subroutine uses the standard Ponchon-Savarit graphical projection technique to determine the coordinates of points on the x,y diagram defining the operating lines. OPLINE is called after the relevant data on the enthalpy-concentration diagram has been computed. The projection technique is programmed as follows (refer to Figure 3.3):

<u>Line No.</u>	<u>Program Action</u>
7	Divide interval (XD-XCR) into NINT(50) increments of size DELX.
11-35	For each value of X1 between these limits determine the ordinate YG defining a point on the operating line as follows. [This ordinate is given as the point of intersection of the vapour-enthalpy curve and the external reflux line passing through X1 (line X1AD on Figure 3.3).]
(15)	Use HFIND to determine enthalpy of liquid X1.
(16-17)	Define equation of external reflux line passing through X1.
(23-31)	Use iterative search procedure SEARCH to find point of intersection of vapour enthalpy curve and reflux line, hence YG. Repeat 11-35 for each X1.



- 33,34 Values are stored in arrays XROP, YROP.
- 37-72 Above procedure repeated for each point defined in the stripping section interval (XCR-XB). Coordinates are stored in XSOP, YSOP. The increment size in this section is the same as in the rectifying section (see below). Lines 37-40 determine the number of increments and check this against the dimension parameter NPPSC.  
Note: NPPSC=NPPS-2, NPPS defined in PARAMETER statement.
- 73-83 Error messages prevent excessive looping in search procedures. This could be avoided by increasing the error limit ACC (line 3) which is presently set at 0.01%.

The data computed here is used by PACKED as well as BINARY. PACKED calculates mass transfer coefficients and uses a Simpson's Rule numerical integration technique to calculate the height of packing required in each section of the column. The size of increment DELX above is thus kept constant throughout the column to maintain a consistent level of accuracy. It is possible that the size of the interval XCR-XB could be large enough relative to (XD-XCR) to cause the number of intervals to exceed the dimensions of XSOP and YSOP. Dimension parameter NPPS could be increased to avoid this if necessary. A situation such as this is unlikely, however, as it would require a high feed concentration, or a highly supercooled liquid, or a combination of both to cause it. The use of NINT=50 with NPPS=150 proved to be sufficient for all examples attempted, and results agreed well with manual solutions (see Chapter 8).

### 3.513 Subroutine FLOWS

This subroutine determines the vapour and liquid flowrates at each point in the column defined by OPLINE. Standard mass balance equations are used and both mass and molar flowrates are computed.

The mass flowrates are required in the design of packed columns only. Subroutine AVEMW is called to calculate solution molecular weights for both phases. The bubble point or dew point of each stream is also computed.

<u>Line No.</u>	<u>Program Action</u>
9-17	Calculation of data for reflux liquid and top tray vapour streams.
18-34	'DO' loop 11 calculates the data corresponding to all streams in the rectifying section of the column.
35-50	'DO' loop 12 duplicates 'DO' loop 11 for the stripping section.
51-52	Reboiler stage is stepped off in the normal way to determine the degree of separation achieved. For some systems such as Acetone-Water, this step indicates that the feed should be entered below the first tray, which means that the column has no stripping section.
54-57	Message printed out to inform the user if no stripping section is considered.
58 and 67	At this point the calculations required in tray column designs are complete and control is returned to the calling routine.
59-65, 75-89	For packed columns the reboiler step sets the conditions at the bottom of the packing. The relevant data for packed columns that have no stripping section are calculated here.

- 68-74, 90-103      Conditions at bottom of the stripping section in a packed column are calculated here.
- 105-123      Error messages indicate detected inconsistencies such as negative flowrates.

#### 3.514 Subroutine AVEMW

This subroutine calculates the molecular weight of a vapour or liquid mixture of composition  $Y_1$  or  $X_1$  respectively. It is used by Subroutine FLOWS to determine mass flowrates, through the column for use in packed column designs. (See Chapter 4.)

#### 3.515 Subroutine CHEC

This subroutine checks that no pinch points occur on the  $x,y$  diagram before an attempt is made to step off the equilibrium stages. This is done by checking that all points of intersection of pairs of operating lines lie below the equilibrium curve. If a pinch point is detected an error message is printed out and the execution is terminated.

<u>Line No.</u>	<u>Program Action</u>
5	Selection of abscissa of operating line intersection points, $X_A$ .
6-7	Calculation of equilibrium vapour composition corresponding to $X_A$ .
8	Check to ensure that the point of intersection does not lie either on or above the equilibrium curve.
10-17	Error message indicates position of any pinch point found. Error results in termination of execution.

### 3.516 Subroutine STEPD

This subroutine steps off equilibrium stages on an x-y diagram for either McCabe-Thiele or Ponchon-Savarit designs. A different procedure is used for each design method with counter NOP selecting the appropriate calculations. A Murphree tray efficiency factor is included and the change from one operating line to the next is done so as to minimise the total number of stages.

<u>Line No.</u>	<u>Program Action</u>
5-9	Stepping procedure is started from the bottom of the column. Starting position is set at XB,YB on the bottom operating line.
11	'IF' statement causes change from one operating line to the next.
14	Check to ensure that array dimension size is not exceeded.
15-18	Determination of point on equilibrium curve above point on operating line. (Equation (2.2) - vertical line of step.)
19	Adjustment of step size using tray efficiency factor (Equation (3.7)). The reboiler is assumed to be a perfect stage and hence MT = 1 is not included.
20	Same as line 11. Check within inner 'DO' loop 11.
21	Procedure terminated when the point of intersection of the top operating line and the diagonal is reached. YC set in subroutine CONDEN.
22-23	Determination of operating line abscissa corresponding to point found by lines 15-19 (horizontal line of step), for McCabe-Thiele design (Equation 3.3 as written for each section of column).

- 24-26 Subroutine PSOPXY is called to find abscissa for Ponchon-Savarit design.
- 28 'DO' loop 11 returns to repeat calculations to define each stage in the column.
- 31 Determination of starting point on following operating line after a change is indicated by line 11 or 20 above, for McCabe-Thiele designs. (Equation (3.3)).
- 34,35 Same calculation for Ponchon Savarit design using PSOPXY.
- 37-41 Storing of feed and side stream tray positions defined as those trays straddling points of intersection of the appropriate operating lines.
- 44-45 Calculation of the fraction of a theoretical stage required to complete an ideal column. (Equation (3.10).) TRAYNO is the total number of theoretical stages required.
- 48-68 Determination of vapour and liquid flows leaving each tray in a Ponchon-Savarit design. Subroutine FFIND uses the standard linear interpolation search procedure to extract data from arrays set up by FLOWS. (See note below.)
- 70-77 Error message. If the number of stages stepped off exceeds the array dimension limits the procedure is terminated and operating diagram data is printed out. This will reveal the types of pinch point caused by deviations from Raoult's law. If a pinch point is not indicated, the array size should be increased.

Note: Arrays LRWT, VRWT and TRV are used to store flow and temperature data for Ponchon-Savarit designs of tray columns. Their definition in the nomenclature applies only to their

application in packed column designs. This is done to reduce the total number of dimensioned arrays.

### 3.517 Subroutine PSOPXY

This subroutine is called by STEPD in Ponchon-Savarit designs. Its purpose is to return the abscissa of a point on an operating line corresponding to the ordinate  $Y_1$ , which has been set by a point on the equilibrium curve. This determines the horizontal line of the stepped stage on the x,y diagram.

The routine uses the standard linear interpolation search procedure to extract a value from either XROP or XSOP. These arrays contain x-coordinates of points on the top and bottom operating lines respectively previously determined by subroutine OPLINE.

<u>Line No.</u>	<u>Program Action</u>
3-15	Calculations for points lying on the bottom operating line.
17-29	Calculation for points lying on the top operating line.

### 3.518 Subroutine FFIND

This subroutine is called by STEPD to determine the molar vapour and liquid flows leaving each stage, and the temperature on each stage, defined on a Ponchon-Savarit x,y diagram. The linear interpolation search procedure is used. A COMMON block is not included and the data arrays are provided through the calling list. The internal arrays FVAP, FLIQ, FTEMP and FX are expanded to accommodate the data arrays VAPR, LIQR, TRL and XROP or VAPS, LIQS, TSL and XSOP respectively, depending on the section of column referred to. These data arrays are compiled in FLOWS.

Using a calling list and dimension statement in this way reduces the total core requirements for running the program. (See Chapter 8.)

### 3.519 Subroutine BINRES

This subroutine prints out the results of either a McCabe-Thiele or a Ponchon-Savarit design of a tray column. The format statements are self-explanatory and counter NOP selects the appropriate commands. Tables of data describing the equilibrium curve, operating lines and equilibrium stage positions may be printed out on request. These data could then be used to construct an operating diagram, or to check a previously constructed diagram. The tables are lengthy and it is thus often convenient to suppress them. The data in the tables is available on file should the user wish to plot a diagram on the CALCOMP plotter (see User Manual). Sample printouts are shown in Chapter 8.

## 3.6 SUMMARY OF BINARY CONTINUOUS DISTILLATION PROGRAM

Program name: BINARY

Type: Mainline with subroutines.

Operation mode: Demand or Batch on UNIVAC 1106 - EXEC 8 system.

1. Title: Design of binary continuous distillation tray towers by methods of McCabe and Thiele, and Ponchon and Savarit.

2. Solution type: Design program.

3. Application: BINARY can be used to design binary continuous distillation systems by McCabe-Thiele or Ponchon-Savarit procedures.

a. McCabe-Thiele:

The program is limited to the following,

- 1) Standard McCabe-Thiele assumptions.
- 2) Thermodynamically ideal systems.
- 3) Simple or complex columns.
- 4) Partial or total condenser operation.
- 5) Partial reboiler or open steam operation.
- 6) Single Murphree tray efficiency factor for all trays.
- 7) Assumed zero pressure drop across trays.

b. Ponchon-Savarit:

The program is limited to the following,

- 1) (2), (4), (6) and (7) above.
- 2) Simple columns (one feed, no side streams).
- 3) Partial reboiler operation.

4. Program Development:

- a) Computer system UNIVAC 1106 - EXEC 8.
- b) Minimum core requirements: 14,5 K
- c) Language: FORTRAN IV with modifications permitted on above system.
- d) Typical run time: McCabe-Thiele: 0,6 secs  
Ponchon-Savarit: 4,6 secs



## CHAPTER 4

### BINARY CONTINUOUS DISTILLATION - PACKED COLUMN DESIGN

#### 4.1 INTRODUCTION

Although most industrial rectifying columns are tray devices, packed columns are better suited for the following applications [58]:

- (1) Handling of corrosive fluids;
- (2) Food and pharmaceuticals, where contact with metal could cause harmful contamination;
- (3) Low pressure distillation, which requires a low pressure drop;
- (4) Distillation of heat sensitive materials, where high vapour-low liquid flowrates are maintained;
- (5) Small-scale application, where small cross-sectional area requirements reduce the efficiency and increase the relative cost of tray towers, e.g. pilot plant and laboratory studies.

Peters and Timmerhaus [59] and King [60] tabulate a number of points to be considered in selecting either a tray or a packed column for a given service. These confirm the suitability of packed columns for the applications listed above. In particular:

(a) Random-packed columns seldom exceed 1,2 metres in diameter while the diameters of industrial tray columns are usually greater than 0,6 m.

(b) Packed columns are cheaper and easier to construct in the following instances:

- i) Corrosive fluid handling
- ii) Vacuum operation - the packing reduces the

possibility of tower wall collapse and in general, provides a lower pressure drop than a tray column designed for the same duty.

(c) Liquid hold-up is considerably less in packed columns.

## 4.2 THEORY OF BINARY DISTILLATION IN PACKED COLUMNS

### 4.2.1 General Design procedure

In packed columns there are no discrete identifiable points at which the vapour and liquid streams are considered to be in equilibrium. Thus, rather than the stepwise change in concentration with height achieved in tray towers, the concentration changes in packed columns are continuous and differential. This differential change must be accounted for in the computational procedure.

Figure 4.1 shows a schematic diagram of a packed column, a corresponding operating diagram, and a differential section of packing. Lengths  $Z_R$  and  $Z_S$  represent sections of packing (see Figure 4.1 (a)), contained between support plates. Plug flow is assumed for each phase, thus implying no axial mixing. This permits the use of standard material and energy balance equations. Operating lines may therefore be constructed on an  $x, y$  diagram, as for a tray column design. The procedure for stepping off stages does not apply in packed column design however.

The height of packing required must be determined from a consideration of the rate of mass transfer of material across the vapour-liquid interface.

The two primary mechanisms of mass transfer considered are molecular diffusion and eddy diffusion. Molecular diffusion is the dominant mechanism under laminar flow conditions whereas eddy diffusion is rate controlling in the

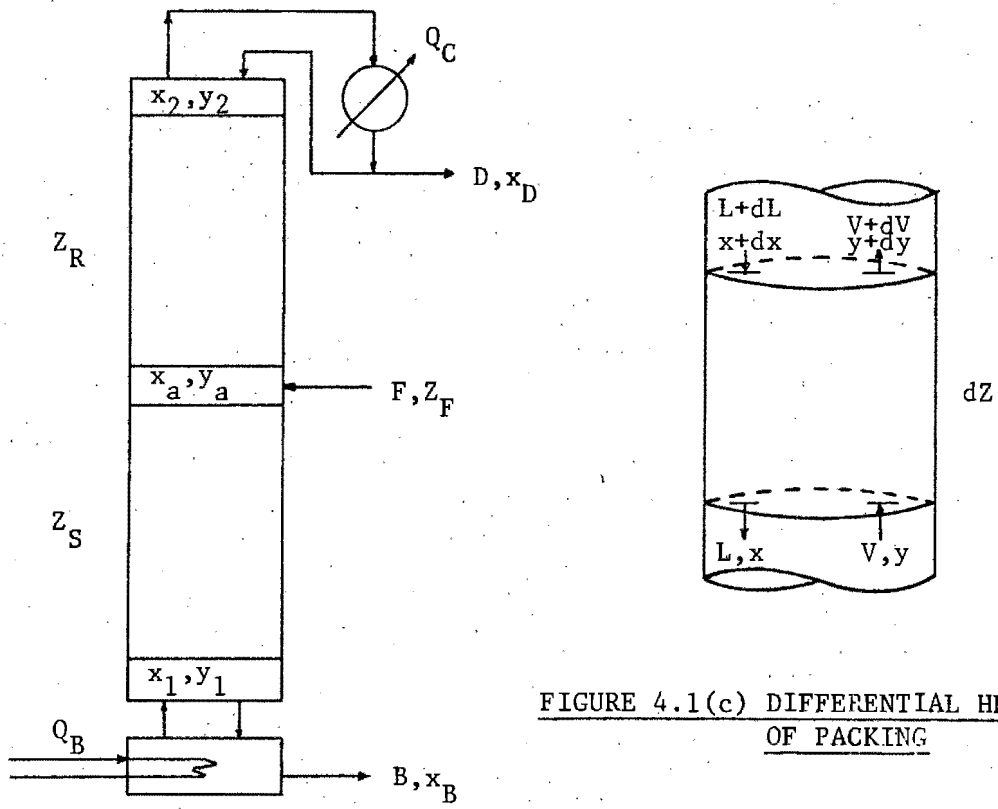


FIGURE 4.1(a) PACKED COLUMN

FIGURE 4.1(c) DIFFERENTIAL HEIGHT OF PACKING

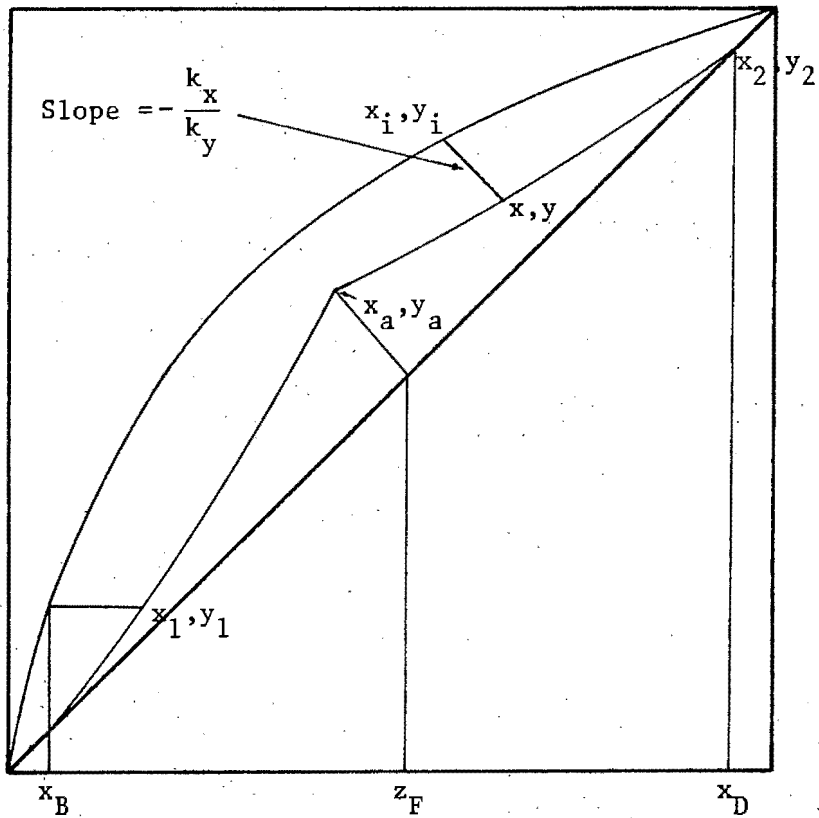


FIGURE 4.1(c)  $x, y$  DIAGRAM - PACKED DISTILLATION COLUMN

turbulent regime. In between the two flow regimes both mechanisms contribute to mass transfer.

In this study the vapour-liquid equilibrium curve and the operating lines are determined as in a Ponchon-Savarit design of a tray column (see Chapter 3). Mass transfer coefficients are calculated directly using the empirical relationship proposed by Shulman et al. [61]. The height of packing required is computed using the procedure suggested by Treybal [9]. A numerical integration technique is used instead of Treybal's graphical procedure.

#### 4.22 Mass transfer analysis for packed column design

The theoretical analysis of mass transfer between phases is discussed in several of the texts referred to in the undergraduate course [9,58,62]. A discussion of the theory relating to the design of packed distillation columns is given here.

The rate of mass transfer across an interface is proportional to the difference between the bulk phase concentration and the concentration at the interface. The constant of proportionality is the phase film mass transfer coefficient. This rate of transfer can be expressed as follows:

$$N_A = k_x(x - x_i) = k_y(y_i - y) \quad (4.1)$$

The bulk phase concentrations are related by the operating line equation for the system (Equation 3.3). For a counter-current column Equation (4.1) can thus be solved graphically, as shown in Figure 4.1 (c), or analytically, by simultaneous solution of Equation (4.1) and the equilibrium relationship, Equation (2.2), provided that the ratios of the film coefficients  $k_x$  and  $k_y$  can be determined for sets of values of  $x$  and  $y$ .

The shortest length of packing in each section required to achieve the given separation at a given reflux ratio, is defined when the bulk phase compositions at the feed point are  $x_a$  and  $y_a$  (Figure 4.1 (c)). This corresponds to the stepping procedure resulting in the least number of stages in a tray column.

In packed columns, flowrates are based on unit column cross-sectional area for use in flooding and mass transfer coefficient correlations. The height of packing required in each section is determined as follows.

Consider the differential height element,  $dZ$ , in Figure 4.1 (b). The interface available for mass transfer within this section is given by  $a dZ$ , where  $a$  is the specific interfacial surface area for distillation (see section 4.3.4.3 (c)).

The amount of component A passing through the element of height  $dZ$  is  $V \cdot y_A$  moles A/hr  $m^2$  in the vapour phase and  $L \cdot x_A$  moles A/hr  $m^2$  in the liquid phase.

The rate of mass transfer is thus

$$N_A = \frac{dVy}{a dZ} = \frac{dLx}{a dZ} \quad (4.2)$$

Equation (1.5) is derived by combining Equation (4.1) and (4.2), and rearranging; thus for the rectifying section

$$\begin{aligned} Z_R &= \int_0^{Z_R} dZ = \int_{Vy_a}^{Vy_2} \frac{dVy}{k_y a (y_1 - y)} \\ &= \int_{Lx_a}^{Lx_2} \frac{dLx}{k_x a (x - x_1)} \end{aligned} \quad (4.3)$$

A similar expression, with appropriate limits, may be written for the stripping section.

Equation 4.3 applies to the general case, but does not lend itself to manual calculation procedures. Most manual procedures reviewed in the literature involve at least one of two standard simplifications [17,58,62]. These are:

(a) McCabe-Thiele assumptions:

Assuming constant molal overflow, the operating lines are straight and the concept of a Height of Transfer Unit (HTU), introduced by Chilton and Colburn [58], may be used. For a distillation process the HTU is related to the mass transfer coefficient by the equation

$$\begin{aligned} \text{Liquid phase} \quad H_{tL} &= \frac{L}{k_x a} \\ \text{Vapour phase} \quad H_{tV} &= \frac{V}{k_y a} \end{aligned} \quad (4.4)$$

Hence Equation (4.3) may be written as

$$\begin{aligned} Z_R &= H_{tV} \int_{Y_a}^{Y_2} \frac{dy}{y_i - y} = H_{tV} N_{tV} \\ \text{or} \quad &= H_{tL} \int_{x_a}^{x_2} \frac{dx}{x - x_i} = H_{tL} N_{tL} \end{aligned} \quad \begin{matrix} (4.5) \\ (4.5) \end{matrix}$$

A number of empirical relationships are available for calculating HTU's [58]. As mass transfer coefficients and interfacial area depend on mass flowrates rather than molar flowrates, McCabe-Thiele assumptions can not ensure that the HTU's are constant. This aspect must thus be checked before using the procedure.

(b) If, in addition to straight operating lines, the equilibrium curve can be considered as a straight line,

$$(4.6)$$

Equation (4.5) can be written in terms of "overall" values.

$$Z_R = H_{tOV} \int_{Y_a}^{Y_2} \frac{dy}{y^*-y} = H_{tOV} N_{tOV} \quad (4.6)$$

In such instances the number of transfer units may be determined without graphical integration, using empirical relationships. This method has limited application but is useful for computing a rough estimate manually.

The program developed here uses the general expression shown in Equation (4.3). The above manual procedures are used by students for comparative purposes.

Considering Equation (4.3), either integral may be used to determine the height of packing required. The choice depends on the relative magnitudes of  $k_x$  and  $k_y$ .

For  $k_x > k_y$  the principal resistance to mass transfer lies in the vapour phase. In Figure 4.1 (c)  $(y_i - y)$  is greater than  $(x - x_i)$  and hence can be determined more accurately. Therefore the expression involving  $(y_i - y)$  in Equation (4.3) would be used in the integration. Similarly for  $k_x < k_y$  the expression in terms of  $(x - x_i)$  would be used.

In general distillation processes are vapour-phase controlled with  $k_x > k_y$  [63].

#### 4.23 Design Calculations

4.231 Calculation procedure: The sequence of design calculations for a packed column used in this study runs as follows:

(1) Determine operating diagram for system using standard Ponchon-Savarit procedure. The reboiler, considered to be an ideal equilibrium stage, is stepped off

as shown in Figure 4.1 (c), thus defining point  $(x_1, y_1)$ .

(2) Choose a number of equally spaced points on each operating line from  $(x_1, y_1)$  to  $(x_a, y_a)$  and from  $(x_a, y_a)$  to  $(x_2, y_2)$  (see Figure 4.1 (c)). The greater the number of points, the more accurate will be the numerical integration of Equation (4.3). The interval between points is the same for both operating lines to ensure similar accuracy in the integration procedure for each section.

(3) At each chosen point determine the vapour and liquid mass and molar flowrates, temperatures and compositions.

(4) Select the point in the column at which the liquid flow is greatest. It is assumed that if the column is designed so as not to flood under the conditions of flow at this point, it will not flood anywhere else [9].

(5) A pressure drop is chosen such that the tower will operate at about 60% of the flooding velocity. This value is usually 0,75-1,0 in.  $H_2O$ /ft packed height for distillation at a total pressure of 1 atm and about 0,01-0,05 for vacuum distillation [17]. The generalized flooding and pressure drop correlation chart, shown in Treybal [9], see Appendix A, Figure 1, is used in the program.

Operation at 60% of the flooding value is usually adopted to ensure that the system will not become unstable [9,64], although some workers have reported a technique for operating a packed column under flooding conditions [65].

(6) The x-axis coordinate for the generalized flooding correlation chart is computed and the y-axis value is determined, corresponding to the chosen pressure drop curve. From this ordinate, the superficial vapour velocity at loading  $V'$  is calculated.



(7) The diameter of the column is then computed such that the vapour flowrate, at the selected point in the column, is less than or equal to the value at loading.

Alternatively, steps (5) to (7) are reversed. If a diameter is specified, the abscissa and ordinate values for the flooding correlation chart are computed and the resultant pressure drop is checked to ensure loading conditions. If necessary the diameter is adjusted and the calculations repeated until acceptable conditions are achieved.

(8) The height of packing required in each section is computed by numerical integration of Equation (4.3). The unknowns in the expression are determined as follows, for  $k_x > k_y$  (see Figure 4.1 (c)).

At each chosen point on the operating lines,  $V$  and  $y$  in this equation are known.  $k_y$ ,  $a$  and  $y_i$  must be evaluated. Composition  $y_i$  is found graphically by constructing a line of slope  $-k_x/k_y$  from the known point  $(x, y)$  to intersect the equilibrium curve at  $(x_i, y_i)$ .  $y_i$  is then read off the  $y$ -axis. The values of  $k_x$ ,  $k_y$  and  $a$  are determined using Shulman's correlations discussed below.

This completes the design of the packed column as defined in this study. The diameter of the column and the height of packing in each section are calculated. The number and position of redistribution devices within the column are not considered. The diameter of the column is limited to 2 metres in the program as industrial columns seldom exceed this size [9].

4.232 Mass Transfer and Flooding Correlations: The mass transfer coefficient expressions used in this study have been extracted from the original work of Shulman as it appears in Treybal [9]. Treybal shows how the expression should be adjusted for distillation applications. Shulman's

analysis applies to Raschig rings and Berl saddles, both of which are common packing materials.

(a) Liquid phase mass transfer coefficient

In the liquid phase, Shulman et al. [61] proposed that the values of  $k_x$  and  $a$  be treated separately. It was proposed that the interfacial area should be obtained from available packing data and that the mass transfer coefficient should be determined using the following relationship

$$\frac{k_L d_S}{D_L} = 25,1 \left( \frac{d_{SL}'}{\mu_L''} \right)^{0,45} (Sc)_L^{0,5} \quad (4.7)$$

For distillation the flowrates within any section of a column are usually sufficiently constant to permit the assumption of equimolar counter diffusion between the phases. Thus the local mass transfer coefficient at the phase interface,  $F$ , is equal to the film coefficients.

$$F_V = k_Y ; F_L = k_X \quad (4.8)$$

The film mass transfer coefficient  $k_L$  in Shulman's equation (4.7) was derived for the system, A diffusing into non-diffusing B. The conversion of  $k_L$  to  $F_L$  for such a system is given by

$$F_L = k_L x_{LM} c \quad (4.9)$$

where  $x_{LM}$  is the log mean concentration driving force across the interface.

For the very low solute concentrations used in the derivation,  $x_{LM}$  approaches 1 and the following relationship holds

$$k_X = F_L = k_L c \quad (4.10)$$

c is the molar density of the solvent liquid, or in the case of distillation, the heavy component.

(b) Vapour phase mass transfer coefficient

Shulman and Margolis [66] proposed the following expression for the vapour side mass transfer coefficient, based on the j-factor theory

$$F_V \frac{(Sc)_V^{2/3}}{V} = 1,195 \left[ \frac{d_S V'}{\mu_V'' (1-\epsilon_0)} \right]^{-0,36} \quad (4.11)$$

where  $\epsilon_0 = \epsilon - \phi_t$  is the operating void volume in the packing.  $\epsilon$  is the characteristic void volume of the dry packing and  $\phi_t$ , the total liquid hold-up, is defined in part (c) below. The mass transfer coefficient  $F_V$  included in the relationship is the local phase-interface value, and from Equation (4.8)  $k_y = F_V$ .

(c) Interfacial area for mass transfer

The interfacial area for distillation, as derived by Shulman is given by the following expression [9]

$$a = a_{AW} \frac{\phi_0}{\phi_{0W}} \quad (4.12)$$

where  $\phi_0$  and  $\phi_{0W}$  account for the effect of liquid hold-up in the packing. In the original work of Shulman et al. [61], three types of liquid hold-up are defined, namely total, static and operating hold-up. These are related by the expression

$$\phi_t = \phi_s + \phi_0 \quad (4.13)$$

Using water and air in columns containing various packings, Shulman derived empirical expressions for calculating the hold-up of water in a packing and hence the hold-up of a process liquid for which certain physical properties are known.

For water, or a very dilute aqueous solution

$$\phi_{tW} = \phi_{SW} + \phi_{0W} \quad (4.14)$$

Equations (4.13) and (4.14) are related by the expression

$$\phi_0 = \phi_{0W} H^* \quad (4.15)$$

where  $H^*$  is a dimensionless correction factor. Empirical expressions defining  $\phi_s$ ,  $\phi_{tW}$ ,  $\phi_{SW}$  and  $H^*$  are shown in Appendix A, Table 1. These expressions were derived by Shulman [61], and are tabulated in Treybal [9].

The interfacial surface area for mass transfer in the air-water system,  $a_{AW}$ , was defined by Shulman as

$$a_{AW} = m(V')^n (L')^P \quad (4.16)$$

for columns operating under loading conditions. For systems with vapour phase densities similar to that of air Equation (4.15) may be used directly in Equation (4.13). For other vapour densities  $V'$  in Equation (4.15) is replaced by  $V' \cdot (0.075/\rho_V)^{0.5+}$ . Values of the coefficient and exponents in Equation (4.16) are shown in Appendix A, Table 2.

Note: The Shulman correlations require data in a variety of units. Conversion factors are built into the program using a DATA statement. These have been chosen to convert the units most often associated with variables quoted in the literature to the form required by the correlations. These factors may be changed by the user if necessary (see User Manual).

#### (d) Flooding correlation chart

The generalised flooding and pressure drop correlation chart of the U.S. Stoneware Co. used in this study has been

---

† Units of  $\rho_V$  - lb/ft<sup>3</sup>.

published by a number of authors [9,17,62,64]. This chart is used to determine the superficial vapour velocity for a specified pressure drop through the column or vice versa.

The x-axis coordinate is expressed as

$$ABCIS = \frac{L'}{V'} \left( \frac{\rho_V}{\rho_L} \right)^{0.5} \quad (4.17)$$

and the y axis coordinate as

$$ORD = \left[ (V')^2 C_f \mu_L^{0.2} \left( \frac{\rho_W}{\rho_L} \right) \right] / (g_c \rho_V \rho_L) \quad (4.18)$$

The chart may be used in one of two ways in the program.

(i) In the program, if a pressure drop is specified, the value of ABCIS is computed, a polynomial is fitted to the specified pressure drop correlation curve and the value of ORD is determined. The superficial vapour velocity  $V'$  is then determined using Equation (4.14).

(ii) If a diameter is specified,  $V'$ , ABCIS and ORD are computed using the above expressions and the values are printed out. The user then determines whether the point (ABCIS,ORD) lies within the loading zone and if necessary adjusts the diameter. (See section 4.34.)

Characteristic constants for the packings considered in this study are shown in Appendix A, Table 3.

#### 4.24 Choice of column packing

The performance of a packed column is dependent primarily upon the type of packing used, and the material from which it is made. The choice of a particular packing is normally based on cost, type of fluids to be handled and efficiency considerations.

A number of packings were extensively studied by Shulman et al. [61] and two of these, namely ceramic Raschig

rings and ceramic Berl saddles, were considered by Treybal [9]. These two packings are commonly used in industrial columns because of their relatively low cost. They are well documented in the literature and are thus useful for teaching purposes. For these reasons they are considered in this study.

Data for various sizes of these two packings are included in the program. The user is able to insert data for other packings if necessary. (See User Manual, section 7.3.)

### 4.3 DESCRIPTION OF PROGRAM PACKED

#### 4.3.1 General Structure

The overall flowchart of the packed column design program is shown in Figure 4.2. The functions of the main-line routine are as follows.

- (1) Start execution and read in data via READBD.
- (2) Set up Ponchon-Savarit operating diagram using PONSAV.
- (3) Perform all column-sizing and height-of-packing calculations.

Various subroutines are called through the program, to perform repetitive calculations and to determine physical property data. Integration to find the height of packing required in each section is performed by the subroutine SIMPS (see Chapter 2).

The operating diagram is determined as described in Chapter 3 for Ponchon-Savarit designs. Only the mass transfer calculation section will be explained in detail here.

#### 4.3.2 Limitations

The same limitations concerning ideal-system behaviour, discussed in Chapters 2 and 3, apply to PACKED. There are

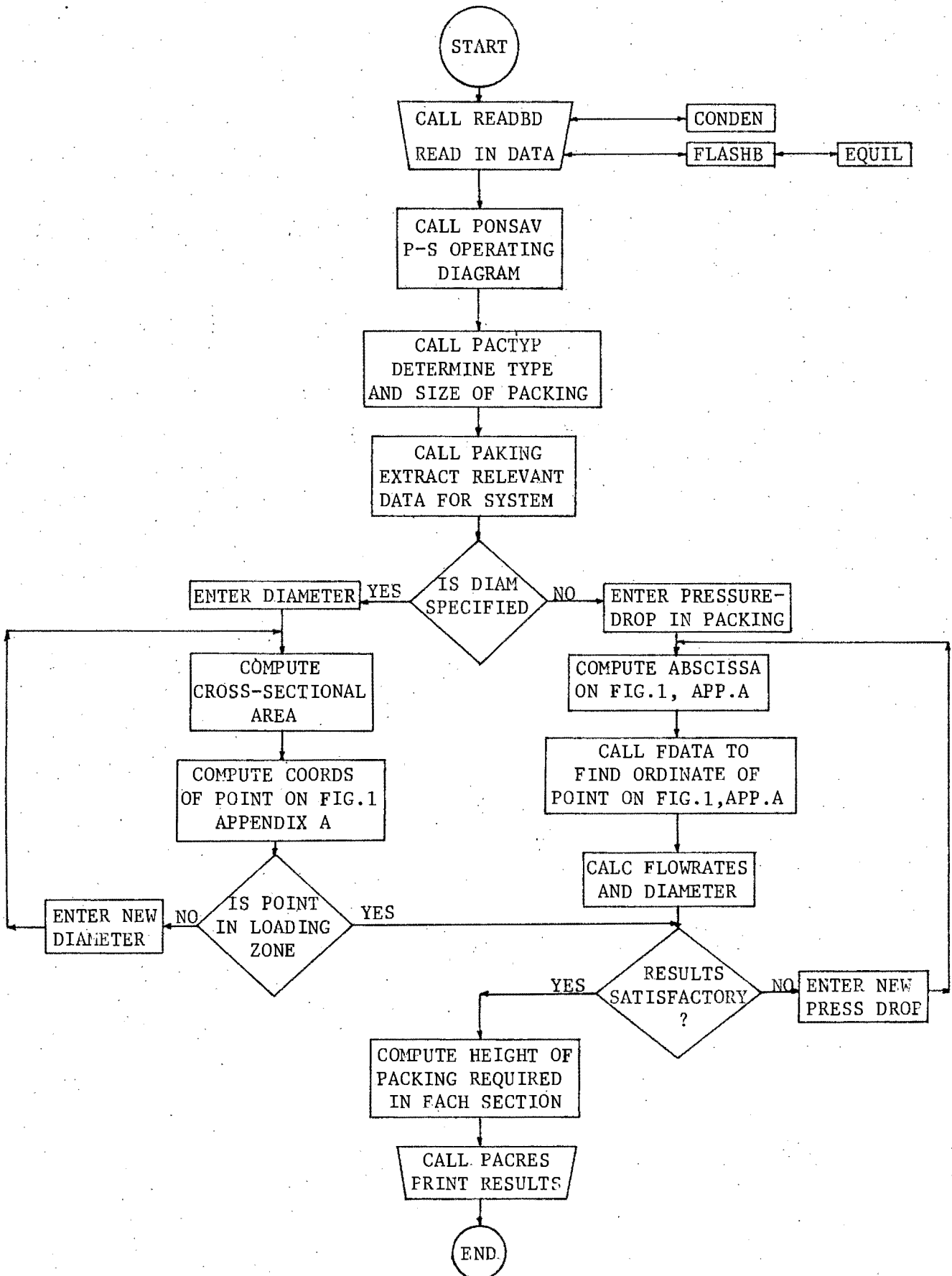


FIGURE 4.2 FLOWCHART OF PROGRAM PACKED

no plotting routines associated with PACKED and thus no data files need be assigned before the run. If a plot of the operating diagram is required, the problem should be solved using BINARY. This will also yield the number of theoretical trays equivalent to the designed height of packing. Data for four sizes of Raschig rings and three sizes of Berl saddles are included in the package. If other packings are specified the package data subroutine PAKING must be altered accordingly (see User Manual).

Internal conversion factors allow the user to supply data in any convenient system of units. Those factors currently built into the program apply to commonly encountered data tables. These are discussed individually in the User Manual and the user must ensure that the correct factor is available. Results are all returned in the SI.

#### 4.33 Operating Procedure

It is recommended that this program be run only in demand mode. The program calculations are divided into two parts, as listed in section 4.31, and a separate data element may be added for each section. The user is required to check certain results visually, and to suggest changes if necessary, during the course of a run. In batch mode the outcome of these checks would have to be anticipated, which would defeat the purpose.

#### 4.34 Program Calculations

<u>Line No.*</u>	<u>Program Action</u>
8,9	Conversion factors associated with units of physical property data and flowrates (see User Manual).
13-29	Start of program and reading of data through subroutine READBD as described in BINARY. The "first" data element contains all data pertaining to the operating diagram. (See User Manual.)

\*See FORTRAN text, Appendix C.



- 33 Construction of a Ponchon-Savarit operating diagram as described in BINARY. On return to PACKED the reboiler stage has been stepped off (see subroutine FLOWS, section 3.513).
- 40-46 A second data element may be added to specify packing data. Alternatively data may be inserted conversationally.
- 47 Subroutine PACTYP determines the type and size of packing through conversational or data-element input.
- 49,50 Subroutine PAKING contains all the relevant data for the seven packings included in the program. Additional data may be inserted by the user if required. (See User Manual.)
- 51-58 The program computes solution-densities using pure component density data. If a table of solution-density data is available counter NRHOM is set from 0 to 1. (See section 4.551.)
- 60-66 The class into which each component falls is required for component density calculations. (See section 4.551.)
- 67-100 Array LRWT and LSWT, compiled by subroutine FLOWS, are examined to determine the point in the column with the greatest liquid flowrate. Once found, temperature and composition data at this point are noted and the point is identified by counter NPOINT. The column diameter will be designed such that the column will not flood at this point.
- 102-107 x-axis coordinate of generalized flooding and loading chart is calculated. Liquid density through subroutine RHOLIQ and vapour density

- assuming ideal solution behaviour. (Equation (4.17).)
- 108-114 Optional specification of column diameter. If specified, the abscissa and ordinate on the flooding correlation chart are calculated. These values are pointed out for checking by the user to ensure that the point on the chart is within the loading zone. If not, a large diameter may be specified. (See section 4.232 (d) and the User Manual.)
- 117-121 Column cross-sectional area and superficial flowrates computed.
- 122 The calculation of ORD for pressure drop determination uses a higher superficial flowrate than the operating value. This represents a 5% safety factor for increases in operating vapour flowrate.
- 124-131 The alternative to specifying a diameter is to specify an allowable pressure drop across the packing at loading conditions.
- 135-137 VPD<sub>L</sub> is the pressure drop value displayed in the result table by subroutine PACRES. The choice of values corresponds to the curves on the chart, but the units have been changed to kPa/m to comply with metrication regulations.
- 138-142 Subroutine FDATA fits a polynomial to the selected curve and the resulting ordinate is printed out for checking by the user. The data fitted in FDATA is included in the program and was taken from the curves. The user may insert a set of data for a curve corresponding to a specific pressure drop if required. (See section 4.54.)

- 143-146      Viscosity and density calculations (values required in Equation (4.18).)
- 147-157      For specification of column diameter, ORD is computed (Equation (4.18)) and the coordinates of the point on the flooding correlation chart are printed out (see section 4.232 (d)(ii).) The user decides whether the specified diameter is satisfactory. If the point is in the flooding zone, a larger diameter should be specified.
- 159            Return to repeat calculations if necessary.
- 161,162      Determination of superficial vapour velocity from Equation (4.18) for specification of pressure drop (see section 4.232 (d)(i).)
- 163-177      Having selected a pressure drop, the column diameter is computed accordingly. The calculation increment is 0,01 m and the maximum allowable size is 2 m. The resulting flowrates and diameter are printed out for inspection. The user may investigate the effect of specified pressure drop on column diameter by specifying a new pressure drop value.

At this point the operating diagram is complete, the column diameter is calculated and the size and type of packing to be used are known. The final stage of the design is the computation of the height of packing required in each section of the column. The calculations are performed as shown in section 4.23.

- 195-289      'DO' loop 41 computes vapour and liquid mass transfer coefficients and interfacial surface area at each point in the stripping section of the column. The composition and flowrates at these points have been previously calculated in

OPLINE and FLOWS respectively. FLOWS has also stepped off an ideal reboiler stage. (See Chapter 3.) The check on line 195 results in a single pass through loop 41 if no stripping section exists. This sets all counters required in loop 42, and reads physical property data.

The mass transfer calculations discussed in section 4.232 are performed within loop 41 as follows. The calculations start at the feed point MPTS ( $x_a y_a$  on Figure 4.1 (c)) and move down the stripping section to point MFIN ( $x_1 y_1$ ). At each point:

- 197-201     Extract flow, temperature and composition data from relevant arrays. Units converted to FPS for use in Shulman's correlations.
- 203,204     Liquid density determined in RHOLIQ and converted to FPS units.
- 205         Vapour density computed.
- 206         Liquid viscosity determined in VISCL (units centipoise).
- 207-215     Choice of coefficient and exponents for Equation (4.16) based on liquid flowrate (see Appendix A, Table 2).
- 217,218     Check vapour density value for determining the form of Equation (4.15). See section 4.232 (c). ( $\pm 2\%$  variation allowed.)
- 220          $a_{AW}$  computed using Equation (4.16).
- 222,223     Surface tension determined in SURTEN and converted to units of dynes/cm (required in empirical expressions for  $\phi_s$  and  $H^*$ , see Appendix A, Table 1.)
- 224-231     Selection of empirical expression for  $\phi_s$ , based on type of packing used. (See Appendix A, Table 1.)

232	$\phi_s$ computed (see Appendix A, Table 1.)
233	$\phi_{sW}$ " " " " " "
234	$\phi_{tW}$ " " " " " "
235-252	Selection of empirical expressions for $H^*$ based on type of packing and liquid viscosity. (See Appendix A, Table 1.)
253	$H^*$ computed.
254	$\phi_{0W}$ computed using Equation (4.14).
255	$\phi_0$ " " " (4.15).
256	$\phi_t$ " " " (4.13).
257	$a$ " " " (4.12).
258	$\epsilon_0 = \epsilon - \phi_t$ (see section 3.232 (b)) for use in vapour phase mass transfer coefficient calculation.
259,260	Vapour diffusivity determined in DIFVAP and converted to FPS units $\text{ft}^2/\text{hr}$ .
262,263	Vapour viscosity determined in VISCV and converted to FPS units $\text{lbs}/\text{ft hr}$ from centipoise.
264	Calculation of vapour phase Schmidt number.
265	Superficial molar vapour flowrate $\text{kg moles}/\text{hr m}^2$ .
266-268	Calculation of $F_V$ using Equation (4.11). Note that VMOLE sets $F_V$ in SI units.
269	Vapour phase mass transfer coefficient $k_y = F_V$ (Equation (4.8)).
271-272	Liquid diffusivity determined in DIFLIQ and converted to FPS units $\text{ft}^2/\text{hr}$ .
274,275	Liquid density from RHOLIQ, units $\text{lbs}/\text{ft}$ .
276	Convert liquid viscosity (see line 205 above) to units of $\text{lbs}/\text{ft hr}$ .

- 277           Liquid phase Schmidt number.
- 278-280       Calculation of  $k_L$  using Equation (4.7).
- 282,283       Calculation of molar density of heavy component.
- 284           Equation (4.10) used to determine  $F_L$  (units FPS).
- 285            $k_x = F_L$  (converted to SI units).
- 286-288       Store  $V, L$ , and  $a$ , in SI units, for use in integration procedure.
- 290           Return to start of 'DO' loop for each point defined on the operating line.
- 292-384       'DO' loop 42 performs the same calculations as loop 41, for the rectifying section starting at the top and working down  $((x_2, y_2) \rightarrow (x_a, y_a))$ .

Referring to Equation (4.3) and Figure 4.1 (c), values of  $V, L, y, x, k_y, k_x$  and  $a$  are known at each point in the column at the end of loop 42. The remaining unknowns are  $y_i$  and  $x_i$ .

- 385-389       'DO' loop 43 defines the equations of the lines of slope  $-k_x/k_y$  joining points  $x, y$  and  $x_i, y_i$  in the rectifying section.
- 391-394       'DO' loop 44 duplicates 43 for stripping section.
- 396-415       'DO' loop 45 finds points of intersection of equilibrium curve and lines with slopes  $-k_x/k_y$ ,  $(x_i, y_i)$ , using the standard iterative search technique.
- 417-436       'DO' loop 46 duplicates 45 for the stripping section.
- 437           Determines which phase limits mass transfer ( $k_x > k_y$  or  $k_x < k_y$ ). The check is made at the point of maximum flowrate and the rest of the column is assumed to behave similarly. The results of this check determine the expression used in the integration.

- 439-451 'DO' loops 61 and 62 compute the values of the function selected by the check.
- 456 Subroutine SIMPS performs a numerical integration to determine the height of packing required in the rectifying section.
- 460-475 'DO' loop 63 and 64 duplicate 61 and 62 for the stripping section and the height of packing is computed through SIMPS.
- 477-478 Subroutine PACRES prints out a full set of results. Tables of mass transfer coefficients and other design data may be suppressed if they are not required.
- 480-500 If a pinch point is detected while computing  $x_i$  and  $y_i$  values in loop 45 and 46, a set of results is printed out that will enable the user to construct the  $x,y$  diagram to determine the position of the pinch.
- 501-504 A maximum diameter of 2 metres is allowed.

#### 4.4 GENERAL COMMENTS ON PACKED

In the numerical integration procedure using Simpson's rule, the limits of integration for each section of column are as follows (refer to Figure 4.1).

Rectifying section  $V_y$  at position  $y_2$  to  $V_y$  at  $y_a$ .

Stripping section  $V_y$  at  $y_a$  to  $V_y$  at  $y_1$ .

The interval between points  $V_y$  on the  $x$  axis is not necessarily constant because of the variation in flowrate caused by enthalpy considerations. It has been found, however, that for all systems encountered, the variation in flowrate is small over a short length of packing. In general, provided that at least 20 points are chosen on a particular operating line, the effect is negligible. This is consistent

with the assumption of equimolar counter diffusion in distillation systems.

#### 4.5 SUBROUTINES ASSOCIATED WITH MAINLINE PROGRAM PACKED

The subroutines dealing with the operating diagram are discussed in Chapter 3.

##### 4.51 Subroutine PACRES

This routine is used to print out the results of a packed column design. The format statements are self-explanatory. For further details refer to the sample printouts in Chapter 8.

Either or both of the following sets of data tables may be suppressed if the user does not require them.

- (1) Composition, temperature and flowrate profiles.
- (2) Mass-transfer data tables.

##### 4.52 Subroutine PACTYP

This subroutine determines the type and size of packing to be used in the column design. If the user is providing original packing data this must be inserted in PAKING with appropriate code numbers defining its position through the GO TO statements. These code numbers may then be entered through PACTYP and the data will be processed. This is discussed further in the User Manual (see section 7.3).

The conversational output from PACTYP can be suppressed and a data element used. The format statements preceding each input explain the subroutine's text.

##### 4.53 Subroutine PAKING

This subroutine provides the empirical data relating to the type of packing specified. The user may add data on other packings to this routine and assign them code numbers.



The first code number refers to the type of packing and the second number to the size. The required packing is selected and the data is returned through the calling list (see User Manual). The variables listed in PAKING are defined in the nomenclature.

#### 4.54 Subroutine FDATA

This subroutine uses a least-squares technique to fit a polynomial to given set of data. The routine is called by the mainline program 'PACKED' to determine the ordinate corresponding to a specified abscissa on the flooding and pressure drop correlation chart.

Data has been taken from the curves on the chart shown in Figure 1, Appendix A. These data tables are automatically added to the runstream through the ERTRAN facility. The user specifies the allowable working pressure drop and the routine selects the corresponding correlation curve. The user may supply data for other correlation curves, if required. Full details on how this may be done are given in the User's Manual, section 7.3.

The subroutine makes use of a UNIVAC library-routine 'GJR' to solve the simultaneous equations in the least-squares procedure. Polynomials of orders one through ten are tested by the routine and the best fitting polynomial is selected for the required calculations.

Subroutine FDATA applies the least-squares fit technique as follows. The calling list includes the abscissa, ABS, and a counter specifying the correlation curve to be used, NCURVE. ORD is the ordinate value returned to the mainline on completion of the calculations.

<u>Line No.</u>	<u>Program Action</u>
6-24	Correlation curve data extracted from flooding chart is stored in data elements. The required set is selected by NCURVE.
25-28	Data read in from element.
33-50	'DO' loops 10,11 and 12 calculate the terms in the matrices AMAT and CMAT.
51-53	CMAT inserted in row N+1 of AMAT, for nth order system, as required by subroutine GJR.
55	UNIVAC subroutine GJR called to solve matrix AMAT by Gauss-Jordan reduction.
57-65	Terms of the solution matrix are extracted from row N+1 of AMAT and the sum of squared errors is computed.
66	Resultant error compared with previous value (initial value = 1000; line 30).
68	System limited to a 10th order polynomial.
71-75	Coefficients of polynomial with the best fit are stored in COEFF.
78-85	Calculation of ORD using the best fitting polynomial.
86-88	The order of the polynomial fitted, and the resultant sum of squared errors are printed out for inspection.
90-92	Error message printed out if GJR fails. JC is defined in the UNIVAC manual describing GJR.

#### 4.55 Physical property calculations

The subroutines described in this section compute physical property data required in PACKED. The subroutines are designed for general use by programs requiring physical property data, and are not specific to PACKED. In each subroutine, program data are entered through the calling list. Any additional data required are read from a prepared data element using the ERTRAN routine. The subroutines may thus be incorporated in any program by defining the variables in the calling lists and setting up the necessary data elements.

General correlations for which data are readily available have been incorporated, and a choice of calculation procedures is offered in certain subroutines. This facilitates collation of required data tables. PACKED assumes that data returned from the various subroutines are expressed in specific units. Conversion factors are included in PACKED to adjust these units where necessary in the course of the calculations. The use of conversion factors is discussed in the User Manual.

The correlations used in each subroutine are discussed in this section and the calling list variables are defined. Details of the data elements required by the subroutines are given in the User Manual.

4.551 Subroutine RHOLIQ: This subroutine is used to compute liquid-phase pure-component and binary solution densities.

(a) Pure-component density:

Two procedures are included for determination of pure component densities at a given temperature.

(1) Extrapolation of a single value  $\rho_1$ , specified at temperature  $T_1$ , using the correlation [67]

$$\frac{(\rho_L - \rho_V)_2}{(\rho_L - \rho_V)_1} = \left[ \frac{T_C - T_2}{T_C - T_1} \right]^{\text{EXP}} \quad (4.19)$$

where the component critical temperature,  $T_C$ , must be specified, and exponent EXP depends on the type of compound considered [68]. The components are categorized by the user in the main-line program (lines 60-66 in PACKED), thus defining the appropriate exponents. Compound categories and corresponding exponents are tabulated in the User Manual.

Vapour density terms  $\rho_V$  are not included in the subroutine. This limits the subroutine to systems at total pressures below  $10^3$  kPa.

(2) Interpolation between points in a temperature vs. density data table.

The data table is provided by the user and the subroutine interpolates using the standard linear interpolation technique described previously.

Any combination of the above two procedures is permitted. If, for example, a set of temperature-density data is available for one component and the density of the second component is known at  $25^\circ\text{C}$ , the subroutine will accept these data and then select procedure (2) above for the first component and procedure (1) above for the second. Data are compiled in element as shown in the User Manual (section 7.341).

(b) Solution density:

Two procedures are included for determination of solution densities at a given temperature.

(1) Combination of individual component densities as calculated above, using the following expression. For components A and B:

$$\rho_s = \frac{\rho_A \rho_B (x_A \text{MW}_A + x_B \text{MW}_B)}{(x_A \text{MW}_A \rho_B + x_B \text{MW}_B \rho_A)} \quad (4.20)$$

(2) Interpolation between points in a composition vs. density data table with subsequent extrapolation to the correct temperature. The composition vs. density data must be quoted at a known base temperature. Critical temperatures of both components must also be supplied. The extrapolation correlation is adapted from the pure component correlation as follows [68].

For pressures below  $10^3$  kPa

$$\rho_s = \rho_{sb} \left[ \frac{T_c' - T_2}{T_c' - T_b} \right]^{EXP'} \quad (4.21)$$

where  $T_c' = x_A T_{cA} + x_B T_{cB}$  (Kay's equation [58])

$EXP' = (EXP_A + EXP_B)/2$  where  $EXP_A$  AND  $EXP_B$  refer to the light and heavy components respectively.

$T_b$  = base temperature

$\rho_{sb}$  = solution density obtained from composition-density data table by interpolation.

Refer to the User Manual, section 7.341 for a description of the data elements required.

The terms in the calling list are defined as follows:

(a) Supplied by calling program:

<u>Variable</u>	<u>Program</u>
NPHY	NPHY = 1 reads data through ERTRAN. NPHY > 1 bypasses read statements.
NR	NR = 1 computes solution density using method (2) above. NR ≠ 1 computes solution density using method (1) above.
NL,NH	Define exponents EXP for light and heavy components respectively in Equation (4.19) above.

TG,X            Temperature and composition (light component) of  
binary solution considered. (°C, mole fraction)

(b) Returned to calling program

<u>Variable</u>	<u>Program</u>
RHO1,RHO2	Pure component densities of light and heavy components respectively.
RHOM	Density of binary solution.

Temperature  $T_c$ ,  $T_1$  and  $T_2$  in the above equations are in K. Internal conversion factors assume that all supplied temperature data are in °C, so conversion by the user is unnecessary providing data are available in °C.

Density data must be supplied in  $\text{g ml}^{-1}$  and the subroutine returns densities to the calling program in the same units. Conversion to other units must be done by the calling program where necessary. In mainline PACKED, conversion factor RHOCON converts  $\text{g ml}^{-1}$  to  $\text{lb ft}^{-3}$ .

4.552 Subroutine SURTEN: This subroutine calculates the surface tension of binary solutions. Two calculation procedures are included.

(a) Calculation of individual component surface tensions and combination of these to yield the surface tension of the mixture by means of Stakhorsky's correlation [69].

$$\sigma_m = \sigma_A \sigma_B / [x_A \sigma_A + x_B \sigma_B] \quad (4.22)$$

Component surface tensions may be computed in one of three ways

(1) Sugden's equation

$$\sigma = \left[ \frac{[P](\rho_L - \rho_V)}{MW} \right]^4 \quad (4.23)$$

where [P] is the Sugden Parachor for the component [58] and vapour density,  $\rho_V$ , is ignored. As only liquid mixture surface tensions are required, and  $T < T_{bp}$ , this is permissible [58]. Liquid density values,  $\rho_L$ , are provided by subroutine RHOLIQ.

(2) Extrapolation of a single temperature-surface tension data pair using Lövgren's correlation [70]

$$\sigma_2 = \sigma_1 \left[ \frac{T_C - T_2}{T_C - T_1} \right]^{1,2} \quad (4.24)$$

(3) Interpolation between values in a temperature-surface tension data table.

(b) Interpolation between values in a composition-surface tension data table with subsequent extrapolation from the base temperature to the required temperature using the correlation in (2) above. A psuedo critical temperature for the mixture is calculated using Kay's equation [58] as in RHOLIQ.

The terms in the calling list are defined in section 4.551, except for SIGM which is the binary solution surface tension returned to the calling program. NL and NH are required because RHOLIQ is called by SURTEN.

Mainline PACKED requires surface tension data in units of dynes/cm. The conversion factor SIGCON in PACKED must be appropriately set if data is provided in other units. For further details refer to the User Manual (section 7.342).

4.553 Subroutine DIFLIQ: In binary distillation systems, steady state equimolar counter-diffusion may be assumed in both the liquid and the vapour phases [9]. Liquid phase molecular diffusivities are determined by this subroutine. These values are used in PACKED to compute Schmidt numbers and mass transfer coefficients (Equation (4.7)).

Although the diffusivity of a solute in a solvent varies appreciably with concentration, the following relationship holds [58],

$$\frac{D_{AB}\mu}{T} = \text{const.} \quad (4.25)$$

where  $D_{AB}$  refers to the diffusivity of the light component A (solute) in the heavy component B (solvent) at infinite dilution. Values of  $D_{AB}$  may be extracted from the literature or derived using an expression such as that of Wilke and Chang [71]. At infinite dilution  $\mu$  is the viscosity of the pure solvent (component B) and  $T$  is the temperature at which both  $D_{AB}$  and  $\mu$  are derived. Equation (4.25) does not hold for systems with high viscosities, but as such systems are seldom encountered in distillation this is not of great importance [58].

The diffusivity of a solution of A and B,  $D_L$ , at any temperature  $T'$  may be computed as

$$D_L = \frac{\text{const.}T'}{\mu_{mL}} \quad (4.26)$$

where  $\mu_{mL}$  is the viscosity of the solution.

Most liquid diffusivities  $D_{AB}$  fall within the range  $0,5 \times 10^{-5} - 5,0 \times 10^{-5} \text{ cm}^2\text{sec}^{-1}$ . The factor  $10^{-5}$  is contained in subroutine and the user must therefore supply values as  $D_{AB} \times 10^5 \text{ cm}^2\text{sec}^{-1}$ . The temperature at which  $D_{AB}$  is known, is supplied in  $^{\circ}\text{C}$ . This is converted to K by the routine where necessary. Subroutine VISCL (section 4.555) is called to determine  $\mu$  and  $\mu_{mL}$ , and the solution diffusivity is computed using Equations (4.25) and (4.26) above.

Calling list variable  $D_L$  is the liquid diffusivity, in  $\text{cm}^2\text{sec}^{-1}$ , returned to the calling program. Factor DLCON in PACKED converts the units to  $\text{ft}^2\text{hr}^{-1}$  where necessary. The other calling list variables are defined in section 4.551. Counters NL and NH are required because VISCL calls RHOLIQ. The necessary data element is described in the User Manual, section 7.343.



4.554 Subroutine DIFVAP: This subroutine computes vapour diffusivities using Arnold's correlation [72]. These values are used in PACKED to calculate vapour phase Schmidt numbers.

Assuming equimolar counter diffusion as discussed in section 4.553, Arnold's correlation is written as follows. For components A and B

$$(D_V)_{AB} = \left[ \frac{0,00837 b^{0,5}}{(V_A^{0,333} + V_B^{0,333})^2} \right] \left[ \frac{T^{2,5}}{(T + C_{A,B})} \right] \quad (4.27)$$

$$\text{where } b = \frac{MW_B + MW_A}{MW_A MW_B}$$

and Sutherland's constant  $C_{A,B} = 1,47 F (T_{bpA} \cdot T_{bpB})^{0,5}$

$$\text{where } F = \left[ \frac{2 (V_A^{1/3} V_B^{1/3})^{0,5}}{V_A^{1/3} + V_B^{1/3}} \right]^3 \quad [72]$$

P = absolute pressure of system in atmosphere.

$V_A, V_B$  = molar volume at normal boiling point  $\text{ml.g mole}^{-1}$

$T_{bpA}, T_{bpB}$  = normal boiling points in K.

T = system temperature in K.

MW = molecular weight of component.

Vapour diffusivity varies only slightly with composition unless the molecular-mass ratio for the system is large [58]. In normal distillation systems this is not so. The molar volume, boiling point and molecular weight of each component must be provided by the user (see User Manual, section 7.344). Boiling point temperature is supplied in  $^{\circ}\text{C}$  and converted internally. The diffusivity value is returned to the calling

program in units of  $\text{cm}^2\text{sec}^{-1}$ . Factor DVCON in PACKED converts these units to  $\text{ft}^2\text{hr}^{-1}$  where necessary.

Willee and Lee [73] showed that the average and maximum deviation obtained using this correlation are 8,4% and 20,5% respectively. The subject of vapour phase diffusivity has been more extensively investigated than liquid phase diffusivity and the correlations may be applied with more confidence [58]. Although the procedure described in section 4.553 achieves reasonable results, it can not be considered as rigorous as the above correlation. Nevertheless, both procedures are sufficiently accurate for the requirements of the program PACKED.

The calling list variables not defined in section 4.551 are PT, the total pressure of the system in mm Hg, and DV, the vapour diffusivity value returned to the calling program.

4.555 Subroutine VISCL: This subroutine calculates the solution, and individual component, liquid viscosities for binary systems. Solution viscosities are computed using the component viscosity values and the following correlation [74],

$$\mu_{mL} = [x_1\mu_1^{1/3} + x_2\mu_2^{1/3}]^3 \quad (4.28)$$

Component viscosities are calculated in one of two ways:

(1) Thomas's equation [75]:

$$\mu = 0,1167 \rho_L^{1/2} 10^\gamma \quad (4.29)$$

where  $\gamma = B\left(\frac{1-T_r}{T_r}\right)$

and B is a viscosity constant.

The viscosity constant and critical temperature for each component must be specified to use this option (see User Manual, section 7.345). Equation (4.29) holds for  $T_r \leq 0,7$  [75].

(2) Interpolation between values in a temperature-viscosity table.

The calling list is similar to that of RHOLIQ. AMUA, AMUB and AMUM are the pure component and binary solution liquid viscosities returned to the calling program. Data are returned to PACKED in units of centipoise. Factor AMLCON converts the units to lb/ft hr. where necessary. For further details refer to the User Manual, section 7.345.

4.556 Subroutine VISCV: This subroutine calculates pure component, and binary mixture, vapour-phase viscosities. The choice of calculation procedures is the same as in subroutine VISCL, with different correlations used.

The vapour mixture viscosity is computed using the following equation [76],

$$\mu_{mV} = \frac{\sum Y_i \mu_i MW_i^{0.5}}{\sum Y_i MW_i^{0.5}} \quad (4.30)$$

The component viscosities are calculated either by interpolation of temperature-viscosity data or by means of Arnold's correlation [77].

For component i

$$\mu_i = \frac{27.0 \cdot MW_i^{0.5} T^{1.5}}{V_i^{2/3} (T + 1.47 T_{bp_i})} \quad (4.31)$$

where  $V_i$  is the molar volume in ml g mole<sup>-1</sup>.

The calling list variables not defined in section 4.551 are Y, the vapour composition (light component) and AMUM, the binary mixture vapour viscosity. The viscosity value is computed in micropoise in the above correlations and these units are converted to lb/ft hr in PACKED by factor AMVCON, where required. (See User Manual, section 7.346.)

4.557 Subroutine RHOH2O: Density of water is well documented [58] and this subroutine requires a table of NDATA temperature-density data pairs. The data element is H2ORHO and the required value is found by linear interpolation. The larger the data table, over a set temperature range, the more accurate the value obtained.

The calling list variables are:

NPHY: defined in section 4.551.

T : system temperature in °C.

RHO : density of water at temperature T returned to the calling program in  $\text{g ml}^{-1}$ .

#### 4.6 SUMMARY OF BINARY PACKED COLUMN DESIGN PROGRAM

Program name: PACKED

Type: Mainline with subroutines.

Operation mode: Demand, on UNIVAC 1106 with EXEC 8.

1. Title: Rigorous design of packed columns for binary continuous distillation.

2. Solution type: Design program.

3. Application: PACKED can be used to design packed columns for binary continuous distillation application. A Ponchon-Savarit operating diagram is used and the design procedure of Treybal, using Shulman's mass transfer analysis, is adopted. The program will compute the column diameter for a given pressure drop and feed rate or will indicate the coordinates of a point on the flooding and loading correlation chart of the U.S. Stoneware Co. [9], if a diameter is specified. The height of packing required in the rectifying and stripping sections of the column is computed using a Simpson's rule numerical integration procedure.

4. Program Development:

- (a) Computer Used: UNIVAC 1106 with EXEC 8 control.
- (b) Minimum Core Requirements: 23 K
- (c) Language: FORTRAN IV with modifications allowed on the above system.
- (d) Typical run time: 8 secs

## CHAPTER 5

### MULTICOMPONENT DISTILLATION

#### 5.1 INTRODUCTION

Solving multicomponent distillation problems manually is tedious and time-consuming because of the large number of repetitive calculations required. As a result students are usually unable to do more than one or two worked examples in the time allotted to this section of the undergraduate course and emphasis is placed on calculation technique rather than the significance of the results obtained. The iterative calculations necessary for the rigorous design of multicomponent systems are best performed on a computer, thus reducing the time required to solve a problem from hours to seconds. With a computer program available for solving problems, students would have more time for analysing the results and would thus be able to investigate multicomponent distillation in more detail. This is discussed further in Chapter 8.

The Thiele-Geddes calculation procedure used in this study simulates the performance of known distillation columns. Simulation can be used for design by successive applications, with adjustment of variables until product specifications are met. A single solution will determine the top and bottom compositions achieved in a given column configuration. The program developed here will enable students to determine the effects of a number of parameters on a multicomponent system.

Only thermodynamically ideal systems are considered, as in the binary processes discussed previously. Equilibrium data are considered in terms of equilibrium constants rather than relative volatilities, and the program includes facilities for investigating enthalpy and tray efficiency effects. The

calculation procedures used in the program correspond to material presented in the undergraduate course.

## 5.2 THEORETICAL BACKGROUND TO THE THIELE-GEDDES METHOD

### 5.2.1 Multicomponent systems

The basic relationships describing an equilibrium stage (see Chapter 1) can be written for a multicomponent distillation column as follows: Referring to Figure 5.1, for each component  $i$  on tray  $j$ :

$$\text{Equilibrium constant } K_{i,j} = y_{i,j}/x_{i,j} \quad (5.1)$$

Overall balances:

$$\text{Material } FZ_{i,F} = Dx_{i,D} + Bx_{i,B} \quad (5.2)$$

$$\text{Enthalpy } FH_F = Q_C - Q_R + DH_D + Bh_B \quad (5.3)$$

$$\text{where } H_F = \frac{V_F}{F} \sum_{i=1}^{NC} H_{i,F} y_{i,F} + \frac{L_F}{F} \sum_{i=1}^{NC} h_{i,F} x_{i,F} \quad (5.4)$$

Rectifying section:

$$\text{Material } V_n y_{i,n} = L_{n+1} x_{i,n+1} + Dx_{i,D} \quad (5.5)$$

$$\text{Enthalpy } V_n H_n = L_{n+1} h_{n+1} + DH_D + Q_C \quad (5.6)$$

$$\text{where } H_n = \sum y_{i,n} H_{i,n} \quad (5.7)$$

$$h_n = \sum x_{i,n} h_{i,n} \quad (5.8)$$

Stripping section:

$$\text{Material } L_{m+1} x_{i,m+1} = V_m y_m + Bx_{i,B} \quad (5.9)$$

$$\text{Enthalpy } L_{m+1} h_{m+1} = V_m H_m + Bh_B - Q_R \quad (5.10)$$

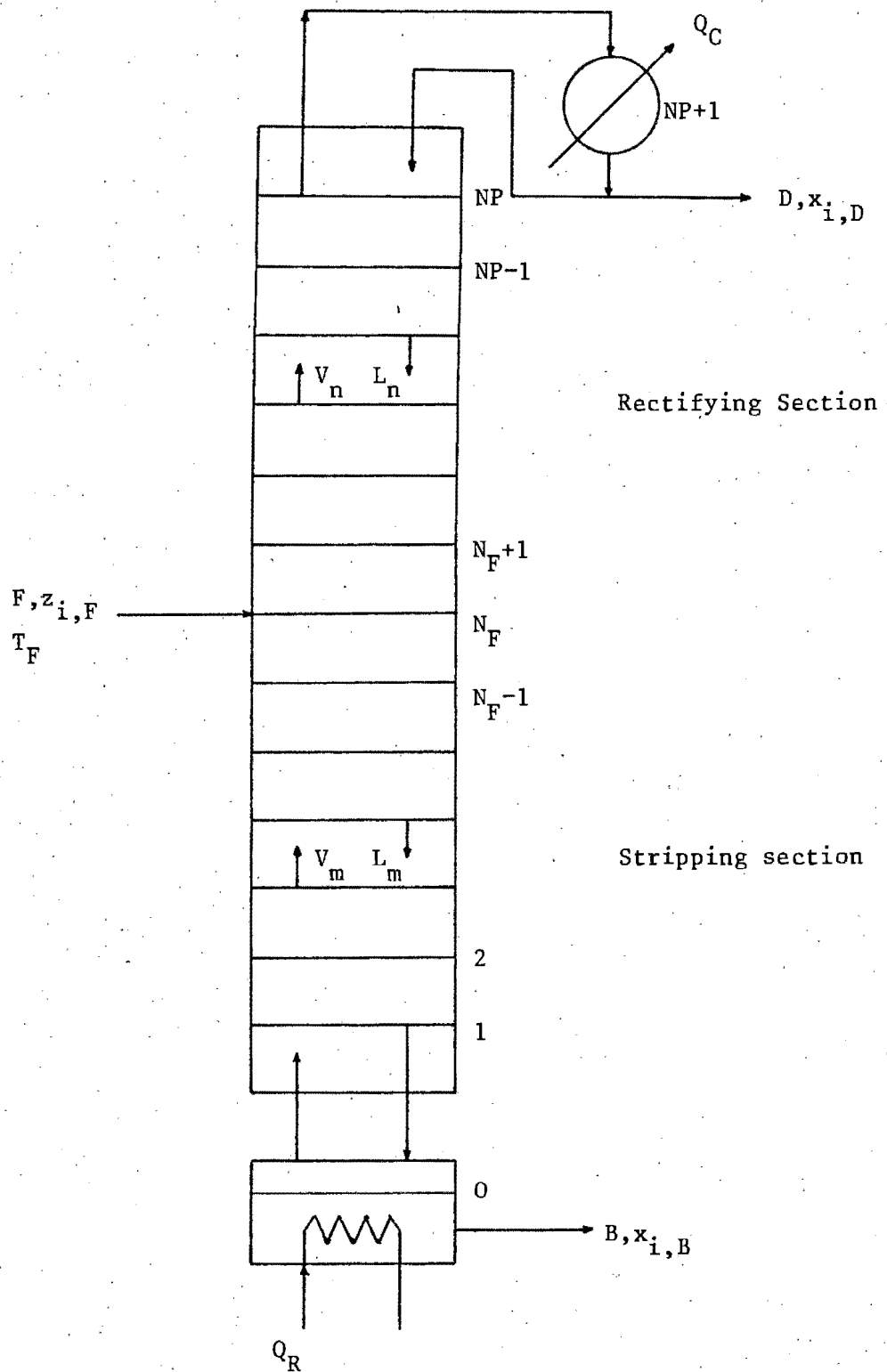


FIGURE 5.1 MULTICOMPONENT DISTILLATION COLUMN



Writing these equations for two components only, reduces them to the binary system equations shown in section 3.2.

### 5.22 General Specifications

The following parameters must be specified or calculated before the iterative calculations begin.

- (1) Number of theoretical trays in the column.
- (2) Position of feed tray.
- (3) Composition, condition and flowrate of feed.
- (4) Distillate withdrawal rate.
- (5) Reflux ratio.

The feed condition is computed from temperature or enthalpy specifications by means of a flash calculation routine. For problems in which constant molal overflow is assumed, a  $q$ -value must be specified if the feed is super-cooled or super-heated. (See section 5.54.)

### 5.23 Initial Assumption and Estimates

The Thiele-Geddes method adopts temperature on each tray as the independent variables. Thus an initial estimate of the temperature profile is required before calculations are begun. A linear profile, based on the condenser and reboiler temperatures, has been found to be a suitable estimate. If condenser and reboiler temperatures are not available the boiling points of the light and heavy key components, or some reasonable estimate of top and bottom temperatures, may be used.

The initial vapour and liquid profile through the column must also be determined. Many tutorial problems assume constant molal overflow, and in such instances the initial flowrate profile will be unaffected by subsequent calculation. For problems that include enthalpy consideration it is customary to use the assumptions of constant molal overflow to

find an initial flowrate profile. It is also advantageous to ignore enthalpy effects for the first two iterations as this allows the system to stabilize.

#### 5.24 Description of the Thiele-Geddes Algorithm

In describing the calculation procedure it is convenient to define several symbols that simplify the material balance and equilibrium expressions. The objective is to develop working equations that are not explicit functions of composition because, in a simulation procedure, stream compositions are in general, unknowns.

In the rectifying section the molal flowrates of component  $i$ , in the vapour and liquid leaving tray  $n$  and in the distillate product, are defined respectively by the three expressions (see Figure 5.1):

$$v_{i,n} = V_n y_{i,n} \quad n=f, f+1, \dots, N \quad (5.11)$$

$$l_{i,n} = L_n x_{i,n} \quad n=f+1, f+2, \dots, N \quad (5.12)$$

$$d_i = D x_{i,D} \quad (5.13)$$

Similarly, for the stripping section, where the bottom product replaces the distillate

$$v_{im} = V_m y_{i,m} \quad (m= 0, 1, 2, \dots, f-1) \quad (5.14)$$

$$l_{i,m} = L_m x_{i,m} \quad (m=1, 2, \dots, f) \quad (5.15)$$

$$b_i = B x_{i,B} \quad (5.16)$$

From the above definitions, the liquid leaving the feed tray,  $f$ , is included in the stripping section while the vapour

leaving  $f$  is included in the rectifying section. It is assumed that the streams leaving the feed tray are in equilibrium, as are the streams leaving all other trays.

The equilibrium relationship (Equation (5.1)) for tray  $n$  in the rectifying section may be rearranged, multiplied through by  $L_n$  and  $V_n/V_n$ , and written in terms of molal flowrates as

$$l_{i,n} = \frac{L_n}{V_n K_{i,n}} \cdot v_{i,n} \quad (5.17)$$

The term  $L_n/V_n K_{i,n}$  is defined as the "absorption factor",

$$A_{i,n} = \frac{L_n}{V_n K_{i,n}} \quad (5.18)$$

and is the molar ratio of component  $i$  leaving tray  $n$  in the liquid and vapour phase. For a partial condenser an equilibrium stage is assumed and

$$A_{i,N+1} = \frac{L_{N+1}}{D K_{i,N+1}} \quad (5.19)$$

For a total condenser

$$A_{i,N+1} = \frac{L_{N+1}}{D} \quad (5.20)$$

Similarly, for tray  $m$  in the stripping section, the equilibrium expression may be written in terms of molal flowrates as

$$v_{i,m} = \frac{V_m K_{i,m}}{L_m} \cdot l_{i,m} \quad (5.21)$$

where  $V_m K_{i,m}/L_m$  is the "stripping factor".

$$S_{i,m} = \frac{V_m K_{i,m}}{L_m} \quad (5.22)$$

For the reboiler, an equilibrium stage

$$S_{i,o} = \frac{V_o K_{i,o}}{B} \quad (5.23)$$

These definitions of molal flowrates, absorption factors and stripping factors can be used to modify the equilibrium expression for each tray (Equation (5.1)) and general material balance expressions for each section (Equation (5.5) and (5.9)) as follows:

Rectifying section

$$\text{Condenser Equilibrium } \frac{l_{i,N+1}}{d_i} = A_{i,N+1} \quad (5.24)$$

$$\text{Tray Equilibrium } \frac{l_{i,n}}{d_i} = A_{i,n} \frac{v_{i,n}}{d_i} \quad n=f+1, f+2, \dots, N \quad (5.25)$$

$$\text{Material balance } \frac{v_{i,n}}{d_i} = \frac{l_{i,n+1}}{d_i} + 1 \quad n=f, f+1, \dots, N \quad (5.26)$$

Stripping section

$$\text{Reboiler Equilibrium } \frac{v_{i,o}}{b_i} = S_{i,o} \quad (5.27)$$

$$\text{Tray Equilibrium } \frac{v_{i,m}}{b_i} = S_{i,m} \frac{l_{i,m}}{b_i} \quad m=1, 2, \dots, f-1 \quad (5.28)$$

$$\text{Material balance } \frac{l_{i,m+1}}{b_i} = \frac{v_{i,m}}{b_i} + 1 \quad m=0, 1, 2, \dots, f-1 \quad (5.29)$$

The unknowns  $d_i$  and  $b_i$  in Equations (5.5) and (5.9) have been eliminated by dividing through appropriately to yield the above expressions. The absorption and stripping factors are computed for each component on each tray using the initial temperature profile to compute equilibrium constants, and the initial vapour and liquid flowrate data. The tray-to-tray calculations then proceed as follows:

(i) Rectifying section

Calculations are initiated at the condenser. The type of condenser specified determines the form of  $A_{i,N+1}$  and Equation (5.24) computes the starting value  $l_{i,N+1}/d_i$ . This is substituted in Equation (5.26) to yield  $v_{i,n}/d_i$ . Equations (5.25) and (5.26) are then used alternately in working down the column to the feed tray until values  $l_{i,f+1}/d_i$  and  $v_{i,f}/d_i$  are determined.

(ii) Stripping section

The successive substitution procedure is the same as for the rectifying section. Calculations are initiated at the reboiler (Equation (5.27) and Equations (5.28) and (5.29) are used in working up the column to determine  $v_{i,f-1}/b_i$  and  $l_{i,f}/b_i$ .

At this point the first iteration is complete and the degree of matching achieved at the feed tray is determined.

5.241 Feed tray analysis: The above assumption of feed tray equilibrium conditions implies that the feed stream entering at the feed tray position achieves instantaneous equilibrium with the solution on the feed tray, resulting in equilibrium streams leaving the feed tray. This is an extremely idealized situation, but is one that has found wide acceptance as an approximation of a real system [12,21,31]. (If, in practice, the vapour fraction of the feed is sparged just below the actual feed tray so that it mixed well with the vapour rising from the stripping section, and the liquid portion of the feed is added to the liquid in the downcomer from tray  $f+1$ , the possibility of achieving equilibrium streams leaving the feed tray would be improved.)

The equilibrium expression for the feed tray may be written in terms of the previously defined parameters as

$$l_{i,f} = A_{i,f} v_{i,f} \quad (5.30)$$

$$\text{where } A_{i,f} = L_f / V_f K_{i,f} \quad (5.31)$$

$$\text{and } V_f = V_{f-1} + (1-q)F = V_{f-1} + V_F \quad (5.32)$$

$$L_f = L_{f+1} + qF = L_{f+1} + L_F \quad (5.33)$$

The feed condition,  $q$ , is determined by flash calculation (see section 2.263). In a constant molal overflow system,  $V_f$  is the rectifying section vapour rate and  $L_f$  is the liquid rate in the stripping section.

Multiplying and dividing Equation (5.30) by  $d_i$  and  $b_i$ , and rearranging, yields,

$$\frac{b_i}{d_i} = A_{i,f} \frac{b_{i,f}/d_i}{l_{i,f}/b_i} \quad (5.34)$$

The results of the tray-to-tray calculations are used to determine the ratios  $b_i/d_i$ . Values of  $b_i$  and  $d_i$  are then computed. The overall material balance, Equation (5.2), is rearranged and written in terms of molal flowrates to define  $d_i$  as

$$d_i = \frac{FZ_{i,F}}{1 + b_i/d_i} \quad (5.35)$$

Values of  $b_i$  are then computed using the expression

$$b_i = d_i \left( \frac{b_i}{d_i} \right) \quad (5.36)$$

The more obvious rearrangement of Equation (5.35),  $b_i = FZ_{i,F}d_i$  is to be avoided because of the inaccuracies involved in computed differences when  $d_i$  and  $FZ_{i,F}$  are almost equal, as

in the case of light components [12].

The computed values of  $b_i$  and  $d_i$  are then used to calculate the values of  $v_{i,n}$ ,  $l_{i,n}$ ,  $v_{i,m}$  and  $l_{i,m}$ , through the column, from the ratios defined by Equations (5.24)-(5.29). In general the sum of the component molal flowrates thus calculated will not equal the originally calculated total flowrate leaving a particular tray.

Similarly the summed values of  $b_i$  or  $d_i$  will not equal B or D respectively, unless the temperature profile originally chosen is the correct one. A new temperature profile is obtained by calculating the compositions of the streams leaving each tray and computing the bubble point of the liquid or the dew point of the vapour for these composition values.

Composition on any tray j can be computed using the following expression

$$x_{i,j} = \frac{l_{i,j}}{\sum l_{i,j}} \quad (5.37)$$

$$y_{i,j} = \frac{v_{i,j}}{\sum v_{i,j}} \quad (5.38)$$

Product compositions:

$$x_{i,D} = \frac{d_i}{\sum d_i} \quad (5.39)$$

$$x_{i,B} = \frac{b_i}{\sum b_i} \quad (5.40)$$

When convergence is obtained the following relationship will hold:

$$\sum d_i = D \quad (5.41)$$

$$\sum b_i = B \quad (5.42)$$

$$\sum l_{i,j} = L_j \quad (5.43)$$

$$\sum v_{i,j} = V_j \quad (5.44)$$

Thiele and Geddes did not propose a technique for improving the rate of convergence, and the method of direct iteration used by them can be extremely time-consuming. A convergence procedure is therefore required, and the Holland  $\theta$  method is used in this study.

#### 5.25 Holland's $\theta$ method for convergence

5.251 The  $\theta$  method: The procedure proposed by Lyster et al. [33] considerably increases the rate of convergence of the Thiele-Geddes algorithm. The primary function of a convergence technique is to improve the estimate of the independent variables used in each iteration, thus reducing the total number of iterations required to reach a solution. The  $\theta$  method is based on the fact that when the temperature profile is not correct, the  $b_i/d_i$  ratios (Equation (5.34)) will all generally be either too high or too low, with the result that, through Equation (5.35), Equation (5.41) will not hold.

Following Lyster's procedure, the calculated  $b_i/d_i$  ratios are corrected by some factor  $\theta$  such that the sum of the  $d_i$ 's, calculated using Equation (5.35), equals the specified distillate withdrawal rate (Equation (5.41)). These adjusted  $d_i$  values are then used, as described above, to calculate the molal flowrates, the liquid and vapour compositions, and finally, the new temperature profile.

This technique, combined with the Thiele-Geddes calculation procedure, is recognised as one of the fastest methods for solving multicomponent distillation problems [34,35].



Mathematically the  $\theta$  method can be described as follows:

The "calculated"  $b_i/d_i$  ratios are determined using Equation (5.34) and these ratios are "corrected" by the expression

$$\left[ \frac{b_i}{d_i} \right]_{co} = \left[ \frac{b_i}{d_i} \right]_{ca} \cdot \theta \quad (5.45)$$

Equation (5.35) is rewritten in terms of the corrected ratios as

$$(d_i)_{co} = \frac{FZ_{i,F}}{1 + \theta (b_i/d_i)_{ca}} \quad (5.46)$$

The method of computing  $\theta$  is discussed below.

The corrected value of  $b_i$  is calculated using the expression

$$(b_i)_{co} = (d_i)_{co} \left[ \frac{b_i}{d_i} \right]_{co} \quad (5.47)$$

These corrected values are used, as previously described, to compute molal flowrates through the column.

5.252 Calculation of  $\theta$ :  $\theta$  is to be evaluated such that

$$\sum (d_i)_{co} = D_{spec} \quad (5.48)$$

$$\text{i.e.} \quad \sum \left[ \frac{FZ_{i,F}}{1 + \theta (b_i/d_i)_{ca}} \right] - D_{spec} = 0 \quad (5.49)$$

Equation (5.49) is solved for  $\theta$  using Newton's method, previously described in section 2.52.

Written in terms of  $\theta$ , Newton's iterative equation is:

$$\theta_{k+1} = \theta_k - \frac{f(\theta_k)}{f'(\theta_k)} \quad (5.50)$$

where

$$f(\theta_k) = \Sigma \left[ \frac{FZ_{i,F}}{1 + \theta_k (b_i/d_i)_{ca}} \right] - D \quad (5.51)$$

$$f'(\theta_k) = - \Sigma \left[ \frac{(b_i/d_i)_{ca} FZ_{i,F}}{(1 + \theta_k (b_i/d_i)_{ca})^2} \right] \quad (5.52)$$

Convergence is achieved when  $f(\theta_k)$  approaches zero within a specified limit of accuracy.

Overall convergence of the solution procedure, indicating the choice of the correct temperature profile, is obtained when  $\theta$  approaches unity within a specified limit of accuracy.

The application of the convergence procedure is discussed in detail in subroutine THECAL (section 5.58). The convergence characteristics of the  $\theta$  method are discussed below.

5.253 Convergence Characteristics of the  $\theta$  method: It will be shown (1) that the  $\theta$  method adjusts the temperature profile in the correct direction for achieving convergence, and (2) that the correction achieved per iteration is greater than that achieved by direct iteration procedures.

(1) Consider a system in which constant molal overflow may be assumed. From Equation (5.26) the ratio  $v_{i,n}/d_i$  is proportional to the absorption factor and thus inversely proportional to the equilibrium constant  $K_{i,n}$  and hence temperature. Similarly from Equation (5.29)  $l_{i,m}/d_i$  is proportional to temperature. From Equation (5.34) therefore the  $b_i/d_i$  ratios will decrease with an increase in temperature.

Thus if the temperature profile is too high the  $b_i/d_i$  ratios will be too low and the corrected ratios must be larger than the calculated ratios. Hence, by Equation (5.45),  $\theta$  must be greater than unity in this instance.

On inspection of Equation (5.49) this equality will be satisfied for  $b_i/d_i$  ratios that are too low, if  $\theta$  is positive and greater than unity. Similarly, if the temperature profile is too low, the  $b_i/d_i$  ratios will be too high and  $\theta$  must be positive and less than unity in order to satisfy Equation (5.49).

Inspection of Equation (5.49) also shows that negative values of  $\theta$  can lead to  $f(\theta) \rightarrow \infty$  and that the overall material balance is destroyed. A qualitative plot of  $f(\theta)$  vs positive values of  $\theta$ , Figure 5.2, shows that  $f(\theta) = B$  at  $\theta = 0$  [12].

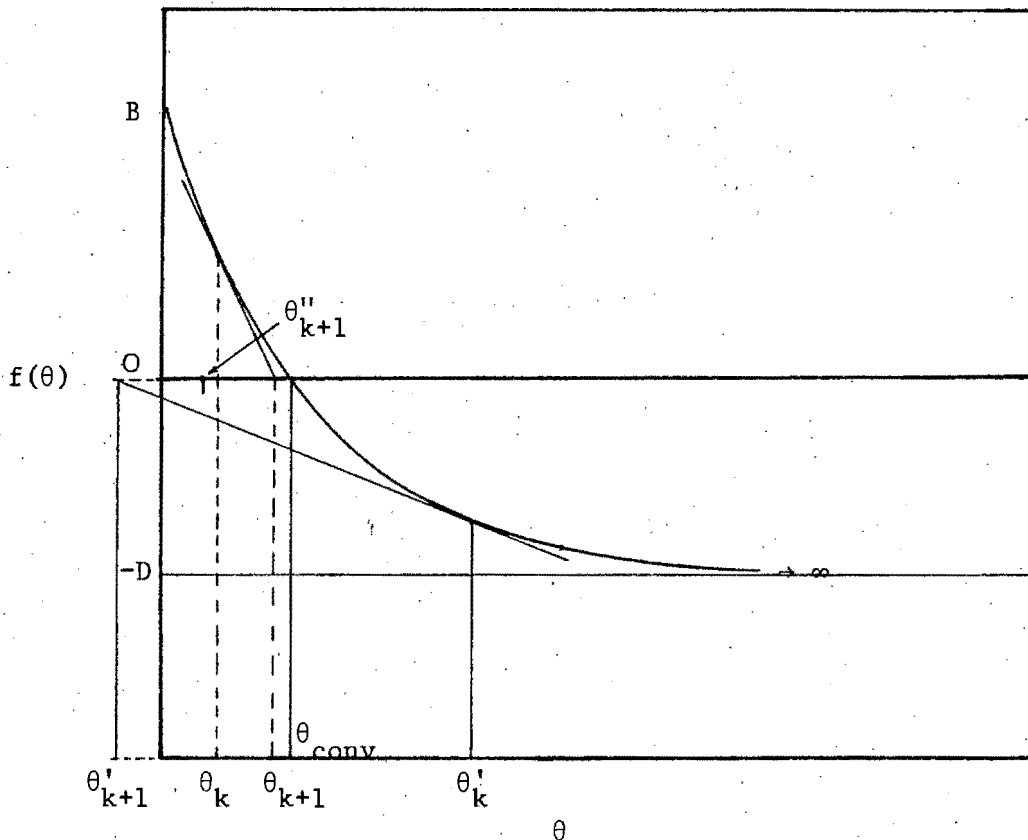


FIGURE 5.2 NEWTON ITERATION FOR  $\theta$

(from Brian [12], p.245)

The curve of  $f(\theta)$  vs  $\theta$  is a monotonically decreasing function for values of  $\theta$  greater than 0. In applying Newton's method of tangents, if the procedure is started at  $\theta = 0$  convergence will be achieved with no negative values of  $\theta$  occurring.  $\theta_{k+1}$  and  $\theta_k$  for the  $(k+1)$ th iteration are shown in Figure 5.2. The number of iterations required for convergence can be reduced, however, by starting with a value of  $\theta$  closer to 1,0. To this end the previous converged value of  $\theta$  is adopted as the starting value for all overall iterations other than the first. For most problems, providing that a reasonable initial temperature profile is chosen, the value of  $\theta$  seldom exceeds 2,0 and the above procedure is satisfactory. In certain situations, however,  $\theta$  occasionally becomes extremely large. In such cases it was found that the rate of convergence could be improved by restricting to a maximum value of 10 000. To guard against the possibility of negative values of  $\theta$  resulting during the application of Newton's method, a check is made on each iteration.

The manner in which a negative value could occur is shown in Figure 5.2, by  $\theta'_k$  and  $\theta'_{k+1}$ . Assume that the temperature profile and other factors caused the Newton procedure to converge at  $\theta = \theta'_k$ . In this case the  $f(\theta)$  vs  $\theta$  curve would be such as to intersect  $f(\theta) = 0$  at  $\theta = \theta'_k$ . The next overall iteration would use  $\theta'_k$  to adjust the whole system, which would result in a new  $f(\theta)$  vs  $\theta$  curve, such as that shown in Figure 5.2. If  $\theta'_k \leq 9999$  the  $\theta$  calculation routine will adopt  $\theta'_k$  as the starting value of  $\theta$ . The first iteration will result in  $\theta'_{k+1}$  which as shown in Figure 5.2 is negative. In this event, the negative value is multiplied by -1 (see section 5.58) and the calculations are continued. This effectively restarts the procedure from point  $\theta''_{k+1} = -1.\theta'_{k+1}$  shown in Figure 5.2.

By inspection, it is unlikely that a single overall iteration will cause the system to adjust itself sufficiently for negative values of  $\theta$  to occur. In fact no examples of negative values were encountered in any of the problems tested, although several systems caused values of  $\theta$  greater than 10 000. Thus, by restricting  $\theta$  to positive finite values, the temperature profile is adjusted in the correct direction for convergence.

(2) A convergence procedure must also, by definition, improve the rate of convergence.

Consider a system containing very light and very heavy components. For a very light component,  $b_i/d_i$  will be very small and for a very heavy component  $b_i/d_i$  will be very large. Thus Equation (5.46) reduces to the following forms for each case respectively.

$$\text{Very light component } (d_i)_{co} \approx FZ_{i,F} \quad (5.53)$$

$$\text{Very heavy component } (d_i)_{co} \approx \frac{FZ_{i,F}}{\theta(b_i/d_i)_{ca}} \quad (5.54)$$

Equations (5.37) and (5.38) can be written as follows when the  $\theta$  method of convergence is applied.

$$x_{i,n} = \frac{\frac{l_{i,n}}{(d_i)_{ca}} \cdot (d_i)_{co}}{\sum \left[ \frac{l_{i,n}}{(d_i)_{ca}} \cdot (d_i)_{co} \right]} \quad i = 1, NC \quad (5.55)$$

$$y_{i,n} = \frac{\frac{v_{i,n}}{(d_i)_{ca}} \cdot (d_i)_{co}}{\sum \left[ \frac{v_{i,n}}{(d_i)_{ca}} \cdot (d_i)_{co} \right]} \quad i = 1, NC \quad (5.56)$$

Considering Equation (5.54) above,  $\theta$  will result in a greater or smaller concentration of heavy component for values less than or greater than unity respectively. Consider the case where the temperature profile for the previous trial was too low. From part (1) above, for low temperature profiles  $\theta < 1.0$  and therefore the proportion of heavy component will be relatively higher (Equation (5.54)). Corrected compositions calculated from Equations (5.55) and (5.56) show a greater proportion of heavy component than uncorrected compositions from (5.37) and (5.38). Hence the temperature profile computed from these compositions will be higher than that computed from compositions calculated using Equations (5.37) and (5.38), which apply to the direct iteration procedure.

The control of overcorrection of the temperature and flowrate profiles is discussed in section 5.56.

#### 5.26 Supplementary Computation

5.261 Multicomponent vapour-liquid equilibrium: For ideal systems the vapour-liquid equilibrium relationships are expressed in terms of Henry's law as shown in Equation (5.1). Two calculation methods are available within the program for determining the equilibrium constants,  $K_{i,j}$ .

(i) Antoine's equation:

When ideal-solution behaviour is closely approached and the system pressure is less than  $10^3$  kPa, the K-values may be determined, using Raoult's Law, as the ratio of the component vapour pressure and the system total pressure (Equation (2.7)). The vapour pressure of each component is determined from Antoine's equation (Equation (2.9)).

(ii) Polynomials in temperature:

If K-value data are available in terms of temperature, at the system pressure, a polynomial can be fitted to these data and the coefficients supplied to the program. A number of authors have adopted this procedure and three different

polynomials have been included in the program. The polynomials were extracted from the work of Holland [21] and Seppala et al. [18] and are shown in the User Manual (section 7.433). These polynomials were chosen because corresponding coefficients for a number of systems suitable for tutorial problems are quoted in the literature (see Chapter 8).

The K-value calculation procedure selected by the user determines the form of the bubble point and dew point calculation. Newton's method is used to achieve convergence to the correct temperature in each case. The functions used in the calculations are discussed in section 5.42 and 5.44.

5.262 Enthalpy calculations: In the undergraduate course on distillation, the effect of enthalpy on multicomponent systems is often considered in theory only. Owing to the lengthy nature of most multicomponent problems, enthalpy balances are seldom included in manual solutions. Thus a student's practical experience of enthalpy effects is usually limited to the use of the Ponchon-Savarit procedure for solving binary problems. Enthalpy balances are included as an optional facility in the multicomponent program described here. The user is thus able to compare results obtained with and without these calculations.

The enthalpy balances are expressed by Equations (5.3), (5.6) and (5.10). These expressions are used to determine condenser and reboiler heat duties and stream enthalpies through the column. Enthalpy data are also required for an isenthalpic flash calculation (see section 5.263). The user may either provide data for the standard enthalpy subroutine or include a separate subroutine designed for a specific application.

(i) Standard enthalpy subroutine, ENERGY:

The standard enthalpy routine included in the package shares a COMMON block with the mainline line program and uses data read by subroutine READIN at the start of the run. (See section 5.41.) Vapour and liquid component enthalpies are determined from polynomials in temperature for which coefficients must be supplied. The polynomials were taken from the work of Holland [21] and Seppala [18] for reasons discussed in section 5.261. (See also Chapter 8.)

(ii) User subroutine, ENTHAL:

If the user provides an enthalpy routine it must conform to certain specifications set out in the User Manual. The calculation sequence is the same as that in the standard routine and any data required is read through the ERTRAN facility. Communication with the mainline program is through a calling list. This facility enables the user to adopt any convenient means of representing enthalpy data in the solution procedure.

A sample routine is included in the package for illustrative purposes. This routine uses enthalpy functions proposed by Kobel [55], as discussed in Chapter 3, for which coefficients for a number of compounds are available [57].

For the purposes of undergraduate tutorial problems the two enthalpy subroutines provided in the package will probably suffice. A user may wish to include a more rigorous procedure when considering real systems involving non ideal solutions, or other specific problems. It is for applications such as these that the User-routine option is included.

5.263 Flash calculation: In general, the proportions of liquid and vapour in a feed stream will not be known. At best, information on the flowrate, composition, temperature and pressure of the feed will be available. The proportions and compositions of each phase that will exist as the feed



enters the column, at the pressure of the feed tray, are determined using a flash calculation routine QFLASH. The specified feed temperature is taken as the flash temperature and the feed tray pressure is assumed to be the flash pressure. two cases are considered:

- (1) Pressure of feed = feed tray pressure
- (2) Pressure of feed > feed tray pressure.

For case (1) no physical flash operation is necessary, but the calculation is performed in order to determine the proportions of liquid and vapour in the feed and the corresponding composition of each phase.

Case (2) would in general require equilibrium and enthalpy data to be provided at the feed pressure. These data would be used to determine the proportion of vapour and liquid in the feed at the higher pressure, and hence the enthalpy of the feed. An isenthalpic flash calculation would then determine the vapour-liquid ratio at the flash temperature and pressure. As the details of sophisticated flash calculations are not relevant at the level of the undergraduate distillation course, it is assumed that the feed enthalpy is known at the elevated pressure.

An isenthalpic flash calculation then determines the  $q$ -value and temperature of the feed, at the feed tray pressure. Case (1) Having established that the feed temperature lies between the bubble and dew points the following material balance can be written:

$$FZ_{i,F} = V_F Y_{i,F} + L_F x_{i,F} \quad (5.57)$$

Substituting the equilibrium relationship and rearranging,

$$x_{i,F} = \frac{FZ_{i,F}}{V_F K_{i,F} + L_F} \quad (5.58)$$

where  $K_{i,F}$  is computed at the feed stream temperature. Substituting for  $V_F$  and  $L_F$  (Equations (5.32) and (5.33)),

$$x_{i,F} = \frac{z_{i,F}}{K_{i,F} - qK_{i,F} + q} \quad (5.59)$$

The flash calculation now involves finding a value of  $q$  such that:  $\sum x_{i,F} = 1.0$ . The calculation is performed by the standard iterative search technique described in section 2.55.

If the initial bubble point-dew point calculations show that the feed is in fact either superheated or supercooled at the temperature and pressure of the flash, the  $q$ -value is determined using the following expression

$$q = \frac{H_f - H_F}{H_f - h_f} \quad (5.60)$$

where  $H_f$  = enthalpy of a saturated vapour feed;

$h_f$  = " " " " liquid "

$H_F$  = " of feed stream at flash conditions.

Case (2): An isenthalpic flash calculation involves finding the temperature of the feed at the flash pressure and hence finding  $q$  by flash calculation. If a two phase equilibrium is found to exist at the flash conditions the procedure is as follows.

An estimate of the equilibrium vapour-liquid ratio is made, say 0.5, as an initial estimate. Using Equation (5.58) above and an iterative procedure, the temperature for this ratio is calculated at the flash pressure. Hence the phase compositions may be computed and the enthalpy at these conditions calculated. This enthalpy value is compared with the specified feed enthalpy and the estimate of the vapour-

liquid ratio is adjusted suitably. The process is then repeated until a match of enthalpies is obtained.

In the case of a superheated or a supercooled feed at flash conditions, the temperature is computed by direct iteration on the specified feed enthalpy.

5.264 Bubble Point, Dew Point and Tray Efficiency calculation: Subroutines BUBCAL and DEWCAL compute multi-component bubble points and dew points respectively, using Newton's method of convergence as described in section 2.52. Four calculation procedures are included in each subroutine corresponding to the four equilibrium data correlations described in section 5.261. The computational techniques used in each case are described in section 5.55.

In this study multicomponent tray efficiency is considered in terms of component vaporization efficiency factors as applied by Holland [21]. This has been discussed in Chapter 1.

A component vaporization efficiency factor is expressed by Holland as

$$E_{V_{i,j}} = \frac{y_{i,j}^*}{y_{i,j}} \quad (5.61)$$

where  $y_{i,j} = K_{i,j} \cdot x_{i,j}$  is the ideal equilibrium composition, and  $y_{i,j}^*$  is the composition of the vapour leaving tray  $j$ .

In the absence of any realistic procedure for calculating these efficiency factors mathematically, they must be determined from experimental data [22]. The efficiency factors are included in the tray-to-tray calculation procedure by means of modified absorption and stripping factors, defined as [21]

$$A_{i,n} = \frac{L_n}{E_{V_{i,n}} K_{i,n} V_n} \quad (5.62)$$

$$S_{i,m} = \frac{E_{V_{i,m}} K_{i,m} V_m}{L_m} \quad (5.63)$$

Computationally, this reduces to modifying the equilibrium constants defined by Equation (5.1). This is done in subroutine EQUCON where the equilibrium constants are calculated using the ideal-system data supplied by the user, and then multiplied by the specified efficiency factors. If efficiency data are not supplied these factors are set equal to 1.0 (see section 5.51).

In effect, the use of efficiency factors in this way imposes a state of pseudo-equilibrium of the system and the bubble point and dew point calculations must account for this. If the temperature profile is computed using ideal equilibrium data (Equation (5.1)) when the vapour-liquid equilibrium is expressed as

$$y_{i,j}^* = E_{V_{i,j}} K_{i,j} x_{i,j} \quad (5.64)$$

for  $E_{V_{i,j}} \neq 1.0$ , the tray temperatures will, in general, be either too high or too low. This results in  $\theta$  converging to a value other than 1.0. This inconsistency is avoided by incorporating efficiency factors in subroutines BUBCAL and DEWCAL. (See section 5.55.)

### 5.3 DESCRIPTION OF MAINLINE PROGRAM MULCOM

#### 5.31 General structure of MULCOM

The principal function of the mainline routine, in the multicomponent distillation program, is to link the various calculation steps together and to determine which options

the user intends to employ. Each stage of the Thiele-Geddes algorithm is contained in a separate subroutine. The input of initial data and the output of results are also performed through subroutines. This has been done to simplify the flowcharting and thus facilitate any future alterations.

### 5.32 Operation of the program

The program has been developed for operation in demand mode from a computer terminal. It can also be operated in batch mode, but this requires considerable care in setting up the runstream. In demand mode data may be supplied either conversationally or by means of a data element, as described for the binary programs.

A skeleton flowchart of MULCOM is shown in Figure 5.3. The user may solve a problem several times within a single execution for different values of certain variables (see User Manual). These values are entered conversationally at the start of each new run.

The variables that may be altered for the second and subsequent runs are

- (1) Reflux ratio
- (2) Distillate withdrawal rate
- (3) Feed condition
- (4) Feed tray position

Any number of runs may be attempted, providing the C.P.U. time limit for the execution is not exceeded, but only five detailed result tables are allowed (see section 5.33). The limit is imposed to encourage users to determine whether or not a detailed result table is warranted by analysing the abbreviated result table printed on the terminal, and to prevent the generation of excessive amounts of computer printout.

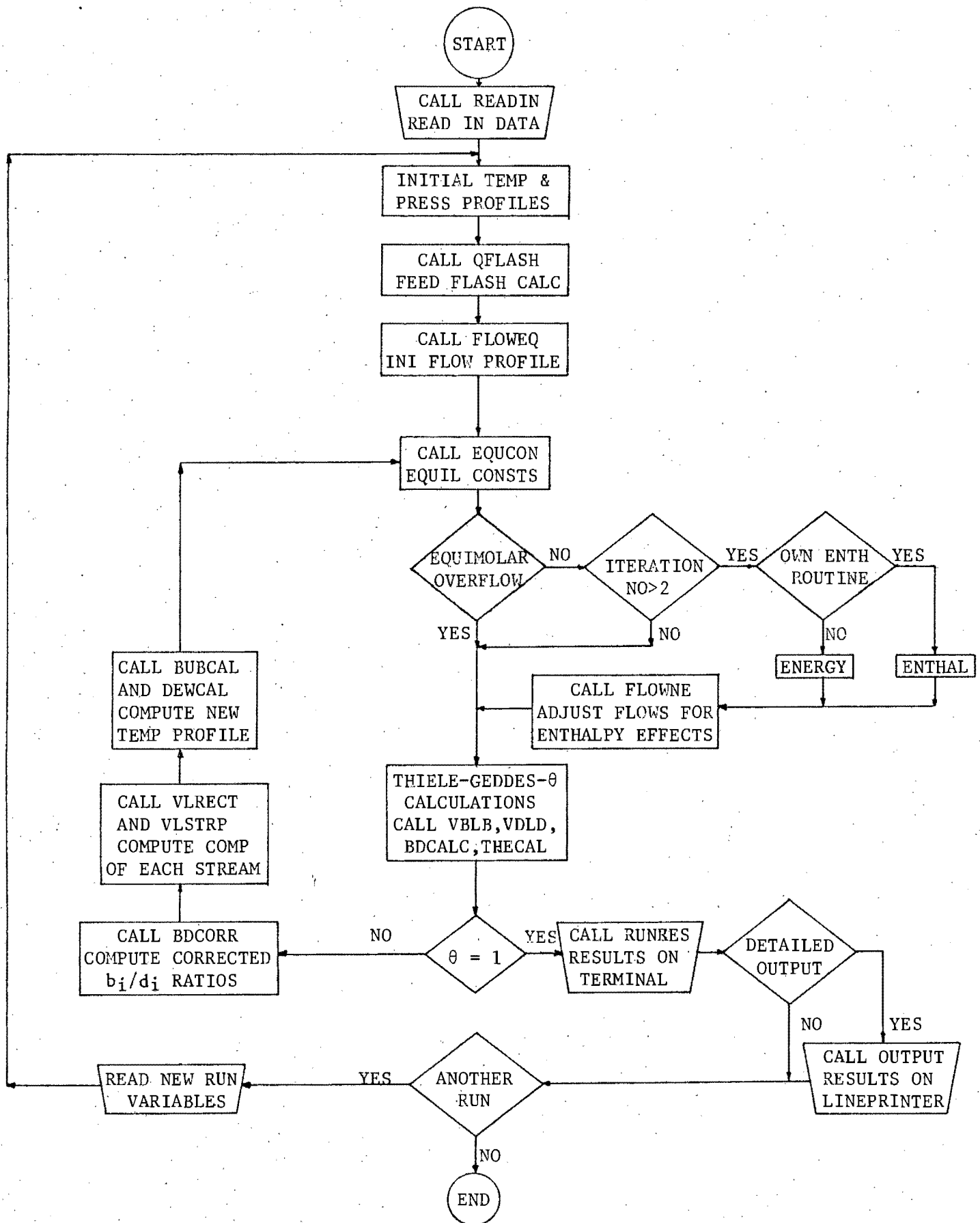


FIGURE 5.3 FLOWCHART OF PROGRAM MULCOM

### 5.33 Printout of results

Output of results is available in three forms:

- (1) Abbreviated result table for primary analysis of a run. See subroutine RUNRES.
- (2) Detailed result table showing composition and component flowrate data for the streams leaving each tray. See subroutine OUTPUT.
- (3) Tables of results for use by the plotting program in constructing the following diagrams:
  - (a) Temperature profile.
  - (b) Liquid and vapour flowrate profiles.
  - (c) Distillate compositions in the form of a bar diagram.
  - (d) Composition profiles through the column.

Output (1) is written directly to the terminal after every run. It includes all the results commonly required in tutorial problems and permits an overall analysis of a system. The user is able to select variables for subsequent runs on the basis of these results, and may request a detailed printout (Output (2) above) if this is required.

Output (2) is optional and should only be requested if detailed composition and flowrate data are required. These data would be required for column sizing or tray design calculations. An output file is assigned to store the data before the program is executed. On termination of the program the file is directed to the high-speed line printer, or may be deleted if not required.

Output (3) is activated at the same time as output (2), but the data are stored only if the user has assigned the necessary files (see User Manual, section 7.435). The plotting routine superimposes the temperature profiles for

all the runs on one diagram for comparative purposes, but will plot diagrams (b), (c) and (d) for only one run at a time. (See User Manual, section 7.53). The user selects the run for which diagrams (b), (c) and (d) would be most useful, on the basis of outputs (1) and (2).

#### 5.34 Termination procedure and error messages

When the execution is terminated the mainline routine will delete the file initially assigned for output purposes if it has not been used. If detailed result tables have been requested the mainline will inform the user of the commands necessary for sending the contents of the file to the line printer. The table is too lengthy for printing on the terminal.

Error messages are included to inform the user of any errors detected during the input of data. More detailed error messages are included in specific subroutines. Errors detected by the mainline routine result in termination of the program and deletion of the output file.

#### 5.35 Limitations of MULCOM

The program is limited to the simulation of simple columns with known configurations, processing ideal multi-component solutions. Dimension parameters may be adjusted to accommodate the number of components and trays in a system. These are set at present for a maximum of 30 components and 50 trays (see MC,MP in COMMON block MULTICOM), although MC=10 and MP=20 would be sufficient for most textbook tutorial problems. Computer core storage limitations must be considered when setting these parameters (see User Manual, section 7.47).



### 5.36 Description of Fortran text

The mainline program MULCOM directs the calculation procedure as follows:

<u>Line No.</u> <sup>*</sup>	<u>Program Action</u>
6-8	Set initial values of control counters.
9-24	Start of program.
25	Call subroutine READIN to read all data required for the first run.
27-36	Arrays TRAYNO and COMPNO are used by the plotting routines. They contain a list of tray numbers and component numbers respectively.
37	Control is returned to this point at the end of a run.
38-45	Start of run and run identification.
48-70	New values of variables that may be altered for second and subsequent runs.
74-77	Calculation of initial temperature profile.
78-85	Calculation of tray-pressure profile.
88	Subroutine QFLASH called to determine feed condition.
90	Subroutine FLOWEQ called to compute initial flows in each section of column, assuming constant molal overflow.
92	Start of iterative calculation procedure (Thiele-Geddes algorithm).
94	Call subroutine EQUCON to determine equilibrium constants for each component on each tray.

\*See FORTRAN text, Appendix C.

- 99-105      Enthalpy calculations through either ENERGY or ENTHAL if constant molal overflow conditions are not assumed.
- 106          FLOWNE computes flows leaving all trays when enthalpy effects are considered.
- 109          ABSTRP computes absorption and stripping factors for each component on each tray in the rectifying and stripping sections respectively (Equations (5.18 & 5.22)).
- 110,111      Subroutines VBLB and VDLD solve the stripping and reactifying sections respectively, working towards the feed tray. (VBLB - Equations (5.27, 5.28 and 5.29); VDLD - Equations (5.24, 5.25 and 5.26).)
- 112          Subroutine BDCALC matches the two sections at the feed tray (Equation (5.34)).
- 113          Equation (5.41) equality checked.
- 114          Subroutine THETA applies the Newton iterative method to determine a value of  $\theta$  in Holland's  $\theta$  convergence technique (Equations (5.50, 5.51 and 5.52).)
- 119,120      Convergence is achieved if  $\theta \rightarrow 1.0$  or Equation (5.41) holds within accuracy limit ACCLIM. This terminates the iterative process.
- 125          Subroutine BDCORR corrects the  $b_i/d_i$  ratios using the calculated  $\theta$  value. (Equations (5.45, 5.46 and 5.47).)
- 130,131      VLSTRP and VLRECT compute compositions of all streams in the stripping and rectifying sections respectively. (Equations (5.37 and 5.38).)

- 138-150 Subroutines DEWCAL and BUBCAL compute dew and bubble points on each tray, based on new composition values. These temperatures are averaged for each tray to yield a new temperature profile for the next iteration. (See Chapter 8.)
- 153 A maximum of 30 iterations is permitted to achieve convergence.
- 158 Subroutine RUNRES prints an abbreviated result table on the terminal. On the basis of these results a full set, detailing flows and compositions for each tray, may be requested.
- 167 OUTPUT writes a detailed set of results for the run to a previously assigned file, for printing on the line printer after execution is terminated.
- 168-195 Results for plotting of temperature, flowrate and composition profiles are written to file.
- 196 Return to line 36 above. If a further run is not required the program will then terminate.
- 197-204 Error messages are printed out if data are not supplied correctly. The program is then terminated.
- 206-217 Termination procedure. The user is told how to send the detailed result table to the line printer. If this facility is not required, the unwanted file is deleted through FINISH.

#### 5.4 GENERAL COMMENTS ON MULCOM

The program MULCOM has been designed as a teaching program and to this end many of the calculation techniques are highly idealized. A modular structure was adopted to facilitate incorporation of more sophisticated procedures

should these be required. The options included in the package should be sufficiently versatile to handle most standard tutorial problems.

The result table printed on the terminal in demand mode executions uses the standard tray numbering convention from the bottom of the column and provides data in SI units. In the internal numbering system the reboiler is stage 1 and not 0 to avoid using the zero address in arrays. The user is not normally concerned with this feature, however. Data supplied to the program may be in any convenient system of units providing that the conversion factors in the various subroutines are adjusted accordingly. These aspects are described in the User Manual (section 7.436).

Error messages are included in a number of subroutines. If negative flowrates are detected in the column the user is informed and corrective action is suggested (see subroutines FLOWEQ and FLOWNE). It is assumed that the user will have sufficient knowledge of distillation systems to understand the cause of such errors.

The application of the program to tutorial problems is discussed in Chapter 8. A number of forcing techniques, used to improve the stability of the iterative calculations and increase the rate of convergence, are discussed as they appear in the subroutines.

## 5.5 SUBROUTINES ASSOCIATED WITH MULCOM

All major input, output and computational operations within MULCOM are carried out by subroutines. Subroutines TEACH and SEARCH have been discussed in Chapter 2.

### 5.51 Subroutine READIN

This subroutine reads all the data required to solve a problem, excluding data required by a user's enthalpy

subroutine (see section 7.45). The formal statements in the FORTRAN text are self-explanatory and only intermediate calculations are described below.

<u>Line no.</u>	<u>Program Action</u>
4-11	DATA statements required for assigning files used to store plotting data. (See lines 196-210 below.)
16	Conversion factors for pressure units.
28-31	Feed tray position and condenser stage number are adjusted to internal numbering system. (See User Manual.)
33-63	Condenser pressure, tray pressure drop, condenser pressure drop, and condenser and reboiler temperatures are read in and the units converted to mm Hg and °C respectively.
96-103	Condenser type noted and counter NCOND set accordingly. NCOND = 1 - total condenser NCOND = 2 - partial condenser.
105-125	If the user is to provide an enthalpy routine this must be set up as described in the User Manual. Subroutine ENTHAL is an example of such a routine. Subroutine ENERGY includes three different enthalpy functions for which coefficients may be supplied. The required function is indicated by counter NENTH. The necessary coefficients are entered here for ENERGY whereas ENTHAL reads its own data as set out by the user.
126-144	Equilibrium data may be provided in any one of three forms. The function to be used is indicated by counter NEQUIL which is entered first,

followed by the appropriate data.

150 If a  $\theta$  convergence limit is not specified by the user a value of 0,00005 is adopted.

163-177 Vaporization efficiency data may be supplied in two ways.

MEFF = 1 efficiency factors for each component on each tray required;

MEFF = 2 efficiency factors for each component on one tray required.

These are assumed to be the same for all other trays.

178-183 If perfect trays are assumed the efficiency factors are all set to 1,0.

196-211 Automatic assigning of files if plotting is required. File names must be 6 characters long and must be entered as:

AFILE1,AFILE2,AFILE3,AFILE4 (entered on one line)

### 5.52 Subroutine EQUCON

This subroutine computes equilibrium constants,  $K_{i,j}$ , and  $K_{i,F}$  using one of four possible functions.

(1) Antoine's equation and  $K_{i,j} = P_{i,j}/P_T$  (Equation (2.7))

(2)  $K_{i,j} = \exp(a_i - (b_i/T_j) + c_i T_j)$   $T_j$  in  $^{\circ}R$  [18]

(3)  $K_{i,j} = \alpha_{i-b,j} \cdot K_{b,j}$

where  $\alpha_{i-b,j} = a_i + b_i T_j + c_i T_j^2$   $T_j$  in  $^{\circ}F$  [21]

and  $K_{b,j} = c_i + e_i T_j + f_i T_j^2 + g_i T_j^3$

(4)  $K_{i,j} = T_j [a_i + b_i T_j + c_i T_j^2 + d_i T_j^3]^3$   $T_j$  in  $^{\circ}R$  [21]

Tray efficiency is accounted for by multiplying the equilibrium constants by the vaporization efficiency factors  $E_{v,i,j}$  as described in section 5.264. The feed stream equilibrium constants  $K_{i,F}$  are calculated for use in the flash calculation when the subroutine is called by subroutine QFLASH. The feed flash equilibrium is assumed to be ideal. The procedure for selecting one of the above expressions and providing the necessary data is discussed in the User Manual (section 7.433).

<u>Line no.</u>	<u>Program Action</u>
3	Selection of equilibrium function.
5	NEQ = 1 when subroutine is called by QFLASH.
6-11	Calculation of $K_{i,j}$ using option (1) above.
12-17	'DO' loop 13 calculates $K_{i,F}$ values using option (1).
18-34	Similar calculation of $K_{i,j}$ and $K_{i,F}$ for option (2).
(21)	Tray temperature converted from °C to °R.
(22)	$T_j$ in the above correlations represents the tray temperature $\times 10^{-2}$ for (2) and (3), and tray temperature $\times 10^{-3}$ for (4). This improves the significance of the coefficient when working in single precision. (See User Manual, section 7.433.)
34-50	Calculations using option (3).
52-67	Calculations using option (4).

### 5.53 Subroutine ENERGY

This subroutine is called either by the mainline program to calculate stream enthalpies and the condenser and reboiler heat duties, or by QFLASH to determine the feed stream enthalpy. Component enthalpies are computed using one of three functions of temperature. For the vapour phase:

$$(a) \ H_i = (a_i + b_i T_j + c_i T_j^2)^2 \quad T_j \text{ in } ^\circ R \quad [21]$$

$$(b) \ H_i = a_i + b_i T_j + c_i T_j^2 \quad T_j \text{ in } ^\circ F \quad [21]$$

$$(c) \ H_i = a_i + b_i T_j + c_i T_j^2 \quad T_j \text{ in } ^\circ R \quad [18]$$

Similar equations define the liquid phase component enthalpy  $h_i$ . Coefficients for all three functions must be supplied for  $T_j$  = tray temperature  $\times 10^{-2}$ , as discussed in section 5.52. Stream enthalpies are computed using Equations (5.4, 5.7 and 5.8).

<u>Line no.</u>	<u>Program Action</u>												
3	Conversion factors (FPS to SI).												
4,7	TRANK and POWER are defined by the choice of enthalpy correlation.												
	<table><tr><th>NENTH</th><th>TRANK</th><th>POWER</th></tr><tr><td>1</td><td>459,4</td><td>2</td></tr><tr><td>2</td><td>0,0</td><td>1</td></tr><tr><td>3</td><td>459,4</td><td>1</td></tr></table>	NENTH	TRANK	POWER	1	459,4	2	2	0,0	1	3	459,4	1
NENTH	TRANK	POWER											
1	459,4	2											
2	0,0	1											
3	459,4	1											
8,9	NEN is set by QFLASH.												
10-23	'DO' loops 10 and 11 compute enthalpies of the vapour and liquid streams leaving each tray in the column.												
24-31	Condenser duty calculations (Equation (5.6)).												



- 32-42      Calculation of feed stream enthalpy (Equation (5.4)). This calculation is required by QFLASH for  $NEN = 2$  as well as for normal enthalpy balance purposes.
- 45         Reboiler heat duty by overall enthalpy balance (Equation (5.3)).
- 46,47      DM and BM are used in FLOWNE for determining flowrate profiles.
- 48-49      Unit conversion.
- 52-70      Calculations required only by QFLASH for  $NEN = 1$ . Computes enthalpy values for use in Equation (5.60).

Subroutine ENTHAL performs the same calculations shown above using the enthalpy correlation of Kobe [55] and the calculation technique described in Chapter 3 (subroutine HTCAP). ENTHAL is described fully in the User Manual, section 7.45.

#### 5.54 Subroutine QFLASH

This subroutine calculates the  $q$ -value and phase compositions of the feed to a distillation column by flash calculation. The flash operation may be either isobaric or isenthalpic, with counter NQVAL recording the user's choice.

The routine computes bubble point and dew point values by calling subroutines BUBCAL and DEWCAL respectively. Subroutine EQUCON is called to compute equilibrium constants for the flash calculations and either ENERGY or ENTHAL is called to calculate enthalpy values.

The sequence of calculations is as follows.

- (1) The bubble and dew points of the feed stream are calculated at the feed tray pressure.

(2) NQVAL, specified in READIN, selects calculation procedure for case (1) or (2) as described in section 5.263.

(a) Case (1):

If feed is superheated or supercooled the routine calculates the saturated liquid and saturated vapour enthalpy values and computes  $q$  according to Equation (5.60). For two phase feeds, an iterative flash calculation is performed, using Equation (5.59) above, to evaluate  $q$  and the composition of each phase.

(b) Case (2)

The same procedure used for isobaric systems is employed but the calculation techniques are as described in section 5.263. Error messages are included to prevent excessive looping of the iteration procedure.

(3) On completion of the calculations the computed value of  $q$ , the bubble and dew point temperatures and the actual feed temperature are printed out for examination by the user.

<u>Line no.</u>	<u>Program Action</u>
8-13	'DO' loop 10 sets initial values of $x_{i,F}$ and $y_{i,F}$ and stores these in arrays X and Y for use in subroutine BUBCAL and DEWCAL. Address IT is unused in the calculation procedure and it is for this purpose that the tray-dimension parameter MP must be equal to at least (NP+3) (see User Manual).
16-18	In BUBCAL and DEWCAL, MD = 2 indicates that a temperature is required for "tray" IT only. The bubble point and dew point for a stream of composition $z_i$ are computed.
19	NQVAL = 1 isobaric flash calculation. NQVAL = 2 isenthalpic flash calculation.

(1) Isobaric flash

- 21            If feed is superheated or supercooled calculations continue at line 50.
- 22-45        Iterative flash calculation using Equation (5.59) (line 30).
- 46-47        Moles of vapour and liquid in feed.
- 50            If enthalpy data are not supplied the feed q-value must be entered directly.
- 51-73        Determination of feed q-value using Equation (5.60) for superheated or supercooled feeds.
- 76-83        Feed q-value requested if enthalpy data are not available and feed is superheated or supercooled.
- 84            Return to line 56 to set feed compositions and flowrates.

(2) Isenthalpic flash

- 86-91        Check for superheated or supercooled feed.
- 92-135       Iterative search for  $V_F:L_F$  ratio that results in  $\sum x_{i,F} = 1.0$  with the calculated  $H_F$  equal to the specified feed enthalpy FENTH. Equation (5.58) is used to calculate the  $x_{i,F}$ 's.
- 137-164      Determination of temperature of superheated or supercooled feed such that the calculated  $H_F$  equals the specified enthalpy FENTH.
- 165-169      If the isenthalpic flash calculation for two phase feeds fails, an error message is printed out and the execution is terminated.
- 172-176      Feed q-value and temperature data printed out.

### 5.55 Subroutines BUBCAL and DEWCAL

The use of Newton's convergence method in these subroutines is similar to that described for the binary system routines in section 2.55. The two routines are different only in the form of the functions used and thus the following description of the calculation procedure applies to both. The four functions used to determine equilibrium constants must be used to compute corresponding bubble points and dew points. The function and its first derivative are required in Newton's method. Efficiency factors are included to account for pseudo-equilibrium conditions as discussed in section 5.264. Allowance is made for the assumption that a total condenser acts as perfect stage irrespective of the specified efficiency data [21].

An important feature of the routines BUBCAL and DEWCAL is the inclusion of an averaging procedure. It was found by Lyster et al. [33] that the combination of the Thiele-Geddes calculation procedure and the  $\theta$  method of convergence could lead to overcorrection of the independent variables from one iteration to the next.

In the case of temperature profiles Lyster observed that by averaging the new and old temperature values on each tray and adopting the mean as the correct value for the next iteration, the convergence characteristics were enhanced. In this study both the bubble point and dew point are calculated for each tray and averaged with the previous value as suggested by Lyster. The final temperature profile for the next iteration is then taken as the mean of the bubble and dew point values on each tray (lines 148 in main-line program). This is discussed in detail in Chapter 8.

The calculation procedure shown below applies to both routines.

<u>Line no.</u>	<u>Program Action</u>
3	Choice of equilibrium correlation as in EQUCON.
4-29	Calculation using Antoine's equation.
6-21	Newton's iterative convergence method.
(11)	MMD > 1 when called by QFLASH for IT = NP2+1 or when total condenser is considered a perfect stage. In both cases efficiency is not considered and EFF = 1,0.
(12,13)	Antoine's equation ( $f(T) = P_T - \sum P_{i,j} \cdot E_{V_{i,j}} \cdot x_{i,j}$ )
(14,15)	$F'(T) = -\sum x_{i,j} \cdot E_{V_{i,j}} \cdot P_{i,j} \cdot BSQ$ , hence TOLD = TOLD + DIV.
23-29	Averaging procedure.
30-37	Calculations using option (2) in EQUCON. The procedure is identical, with a different function used.
39	$f(T) = \sum E_{V_{i,j}} \cdot K_{i,j} \cdot x_{i,j} - 1.$
41	$f'(T) = \sum x_{i,j} \cdot DALFA.$
59-86	Calculations using option (3) in EQUCON.
73	$f(T) = E_{V_{i,j}} \cdot K_b - \frac{1}{\sum \alpha_{i,j} x_{i,j}}$
74	$f'(T) = E_{V_{i,j}} \frac{dK_b}{dT} + \frac{\sum \alpha_{i,j} \frac{d\alpha_{i,j}}{dT}}{(\sum \alpha_{i,j} x_{i,j})^2}$
87-115	Calculation using option (4) in EQUCON.
103	$f(T) = \sum E_{V_{i,j}} \cdot K_{i,j} \cdot x_{i,j} - 1.$
104	$f'(T) = \sum x_{i,j} \cdot E_{V_{i,j}} \frac{dK_{i,j}}{dT}$

Temperatures are converted back to °C before return to the calling program in option 2, 3 and 4 above. Antoine's equations use temperature in °C.

### 5.56 Vapour and Liquid flowrate profiles

5.561 Subroutine FLOWEQ: This subroutine computes the initial flowrate profiles through the column, assuming constant molal overflow. These profiles are maintained for two iterations when enthalpy effects are considered, to allow a degree of stability to be achieved. This procedure was suggested by Holland [21], who found that introducing enthalpy effects immediately could cause the convergence to fail for some systems.

Error messages are included to indicate any detected negative flows. As these are usually the result of a low reflux ratio specification the user is permitted to enter an increased value. To terminate the execution, a reflux ratio of 1 000 or greater is entered.

The standard material balance equations described in Chapter 3 for McCabe-Thiele assumptions are used in the routine. Comment cards in the FORTRAN text indicate the calculation sequence.

5.562 Subroutine FLOWNE: This subroutine is used to adjust the vapour and liquid flowrates according to the enthalpy balances. It is called by MULCOM only when enthalpy effects are considered. The conventional method for performing enthalpy balances is used in the routine [21].

The vapour and liquid flowrates leaving the trays in each section of the column are determined by alternate application of the material and enthalpy balance equations for each section.

An important feature of this subroutine is the "forcing" procedure that is included to restrict the magnitude of change in flowrate profiles, between successive iterations. This forcing technique was first proposed by Lyster et al.

[33], to counteract the instabilities caused by overcorrection of variables from one iteration to the next. It is similar in operation to the averaging procedure included in BUBCAL and DEWCAL.

Lyster observed that, when enthalpy balances were included, the deviation of the calculated value of  $D$ , from the specified value, was amplified. This appeared to be partly due to the fact that both the temperature and the flowrate profiles were changing with successive iterations, resulting in a general increase in perturbation within the system. This effect increased with increase in boiling point range of the feed components, to the point of causing instabilities and failure of the convergence procedure. Since the assumption of constant molal overflow enabled all the problems to be solved, Lyster reasoned that stability could be achieved by limiting the change in flowrate profiles for successive iterations.

The following procedure was proposed by Lyster, and has been adopted in this subroutine.

(a) The first two iterations are carried out assuming constant molal overflow conditions. The purpose of this step is to allow the temperature profile to adjust partially, before additional disturbance is imposed on the system.

(b) The maximum and minimum value of each new flowrate calculated in FLOWNE is limited to 1,2 and 1/1,2 times the previous value for the third, fourth, fifth, sixth and seventh iterations.

(c) These limits are reduced to 1,1 and 1/1,1, and 1,025 and 1/1,025 in the eighth and thirteenth cycle respectively.

(d) If the calculated flowrate falls within the limits imposed for any trial, the new value is averaged with the previous value to obtain the correct profile.

As most simple tutorial problems involve feeds with narrow boiling point ranges, this forcing procedure will often be unnecessary. It has, however, proved useful in stabilizing the solution of certain systems, as will be discussed in Chapter 8.

An error message indicates the occurrence of any negative flowrates. The run is then terminated, after the position of the calculated negative flow in the column has been printed out. Such errors are usually the result of low reflux ratios coupled with low feed  $q$ -values.

<u>Line no.</u>	<u>Program Action</u>
3-5	Iteration-number counter determines "forcing procedure" limit.
6-20	'DO' loop 11 computes flowrates leaving tray in the rectifying section.
(7-8)	Previous value stored and limits of new value calculated.
(10)	Enthalpy balance determines liquid flowrate (modified Equation (5.6) with DM computed in subroutine ENERGY or ENTHAL).
(12-18)	New value restricted within calculated limits.
(19)	Material balance around condenser and tray n.
21-35	'DO' loop 21 computes flowrates in stripping section. The calculation sequence is the same as for the rectifying section.
(25)	Enthalpy balance (modified Equation (5.10) with BM from ENERGY or ENTHAL).
(34)	Material balance around tray m and reboiler.
36-41	Flows leaving ends of each section of column are set. For a partial condenser the distillate



product is a vapour (line 39).

42-59 Error messages indicate detected negative flow-rates and terminates the program.

#### 5.57 Thiele-Geddes tray-to-tray calculations.

The Thiele-Geddes algorithm has been divided into the following sections, each of which is contained in a separate subroutine.

(a) Calculation of absorption and stripping factors: Subroutine ABSTRP.

(b) Calculation of  $v_{i,m}/b_i$  and  $l_{i,m}/b_i$  ratios in the stripping section of the column: Subroutine VBLR.

(c) Calculation of  $v_{i,n}/d_i$  and  $l_{i,n}/d_i$  ratios in the rectifying section of the column: Subroutine VDLD.

(d) Calculation of  $b_i/d_i$  ratios at feed tray position: Subroutine BDCALC.

(e) Calculation of the "corrected"  $b_i/d_i$  ratios using the computed value of  $\theta$ : Subroutine BDCORR.

(f) Calculation of component molar flowrates, total molar flowrates and component compositions of both phases on each tray in the stripping and rectifying sections: Subroutines VLSTRP - stripping section, and VLRECT - rectifying section.

The subroutines are called by the mainline in the order shown above. Subroutine THECAL is called between BDCALC and BDCORR to determine  $\theta$ , as described in the following section. Comment cards on the FORTRAN texts indicate the calculation sequence where necessary.

Subroutines ABSTRP, VLRECT and VLSTRP include a COMMON block whereas the other four use calling lists. FORTRAN

equations in the subroutines relate to the equations shown in section 5.2 as follows. Counter NCOND defines the type of condenser specified, which sets the starting values for calculations in the rectifying section.

<u>Line no.</u>	<u>Calculation performed</u>
Subroutine ABSTRP:	
5	Equation (5.20)
10	" (5.19)
15	" (5.18) including (5.31) (Flowrate values set in FLOWEQ or FLOWNE)
20	Equation (5.22) and (5.23)
Subroutine VBLB:	
5	" (5.27)
9	" (5.29)
10	" (5.28)
Subroutine VDLD:	
5	" (5.24)
9	" (5.25)
10	" (5.26)
Subroutine BDCALC:	
6	" (5.34)
7	" (5.35)
8	" (5.41) (yields $D_{calc}$ )
Subroutine BDCORR:	
6	" (5.45)
7	" (5.46)
8	" (5.47)
9	" (5.48)

<u>Line no.</u>	<u>Calculation performed</u>
Subroutine VLRECT:	
5	$v_{i,n} = (v_{i,n}/d_i)d_i$
6,8	$V_n = \sum v_{i,n}$
11,18	Equation (5.38)
20	Equation (5.5) (for partial condenser)
24	Composition normalization
39	Equation (5.39)
40	" (5.40)
Subroutine VLSTRP:	
7	$v_{i,m} = (v_{i,m}/b_i)b_i$
12,13	Equation (5.44) (5.43)
15	" (5.37)
16	" (5.38)

At the end of these calculations the vapour and liquid compositions on each tray are known. These data are then used in BUBCAL and DEWCAL to determine a new temperature profile.

#### 5.58 Subroutine THECAL

The theory of Holland's  $\theta$  method has been discussed previously. The application of the  $\theta$  method in this program is discussed here. Two convergence criteria are tested in the mainline program on return from THECAL. They are:

(1)  $|\theta - 1,0| \leq \text{ACCLIM}$

(2)  $|D_{\text{calc}} - D_{\text{spec}}| \leq \text{ACCLIM}$  and  $|\theta - 1,0| \leq 0,001$  where

$D_{\text{calc}}$  is computed in BDCALC and ACCLIM is the limit of accuracy, which is either specified by the user (see section 5.51) or taken as 0,00005. The value 0,00005 is suggested

by Petryschuk et al. [31] for computation in single-precision. The above two criteria have been used most often by workers investigating the Thiele-Geddes- $\theta$  solution procedure [21,31, 33,35] and although it is customary to consider only one criterion, both have been included in this application for the following reasons.

(a) As observed by Naphtali [27], it was found that for some systems the rate of convergence decreased as the solution was approached. The value of  $\theta$  converged very slowly or, in some cases, started to oscillate about a value close to 1.0 but not within the specified limit of accuracy. It was found however, that in most of these examples a satisfactory solution could be obtained by applying criterion (2) shown above. Allowing the program to run until criterion (1) was also met, in examples where  $\theta$  did not oscillate, produced no discernible change in the final results, and thus the extra computer time required was not justified. Systems exhibiting these characteristics usually involved (i) feeds with wide boiling point ranges, resulting in end-product composition of some components tending to zero, or (ii) columns with pinch zones extending across several trays. (See Chapter 8.)

(b) Multicomponent systems normally considered in tutorials involve a narrow boiling range feed and a column with little or no pinch zone. For systems such as these the  $\theta$  criterion was always met first.

Thus the use of two convergence checks ensured that a problem would be solved in the shortest possible time, if convergence could be achieved at all.

THECAL is called by the mainline routine after the  $b_i/d_i$  ratios for the particular iteration have been computed in BDCALC. The purpose of the routine is to compute a value of  $\theta$  such that Equation (5.49) holds.

Newton's iteration procedure is employed, using Equations (5.50) to (5.52), as discussed above. The previous value of  $\theta$  is taken as the starting point for the new calculation each time THECAL is called, except in the following instances:

(i) For the first iteration  $\theta$  has not been previously calculated, and a starting value of  $\theta = 0,1$  is used.

(ii) If the previous value of  $\theta$  is greater than 9999, the initial value is taken as 0,1.

If  $\theta$  becomes negative during the iterative procedure, it is multiplied by -1 and the calculations are continued. If the absolute value of  $\theta$  exceeds 10 000 at any stage during the iterative procedure, control is returned to the mainline where  $\theta$  is set equal to 10 000. This is done to avoid unnecessary computation for a system that is far from a converged solution. Extremely large values of  $\theta$  were found to be associated with the type of system described in (a) above. In such cases it was found that by constraining  $\theta$  as described above, the calculation procedure stabilized rapidly, and converged within ten iterations of  $\theta$  first being adjusted, with criterion (2) usually applying.

<u>Line no.</u>	<u>Program Action</u>
3	Check limits of $\theta$ (THETA).
4,5	Starting value $\theta = 0,1$
10	Equation (5.51).
11	Equation (5.52).
15-18	Equation (5.50).
19-21	Limits of $\theta$ allowed: $0 < \theta < 10\ 000$
24	Convergence limit on Newton's iterative procedure.

### 5.59 Output Procedures

The three types of output available from MULCOM have been described previously. Data for graphical representations are written to file by the mainline program. Two subroutines are used to point result tables for analysis by the user.

5.591 Subroutine RUNRES: This subroutine is designed to provide a compact but comprehensive result table on the terminal, during demand mode executions.

The results printed out by RUNRES include:

- (1) Overall material balance.
- (2) Compositions of distillate, bottoms and feed streams. The feed stream data include the overall values entered by the user and the liquid and vapour phase composition computed by flash calculation.
- (3) Tray Temperature and Pressure profiles. The temperature profile is shown in both  $^{\circ}\text{C}$  and  $^{\circ}\text{F}$  for easy comparison with solutions given in the literature.
- (4) Number of iterations required for convergence and final value of  $\theta$ .
- (5) Reflux ratio, distillate rate.
- (6) Feed conditions -  $Q$  value from flash calculation
  - Feed temperature
  - Feed enthalpy
- (9) Condenser and reboiler heat duties.
- (10) Type of condenser employed - partial or total.
- (11) Flowrates of vapour and liquid leaving each tray.

Sample printouts are shown in Chapter 8. The format statements in the FORTRAN text are self explanatory and will not be discussed here. The internal tray numbering system is converted to the conventional system through implied DO loops as the data are printed out. If complete composition

or molar flowrate profiles are required, the detailed result table must be called for.

The printout format is designed for terminals with a restricted carriage length of 70 characters. The detailed printout, discussed next, is designed for a full 130 character line printer output.

5.592 Subroutine OUTPUT: This subroutine produces a detailed table showing compositions and molar flowrates of streams leaving each tray in the column. These data would not be required for standard tutorial solutions, unless an analysis of the composition profiles is necessary. Detailed tray data are normally required in tray design calculation or for column sizing purposes. A sample printout is shown in Chapter 8 and on the UNIVAC system this is produced on the high-speed line printer.

The WRITE key NLP is set by the mainline routine. The file corresponding to this key is assigned to the run by the user before execution is initiated. The key can be adjusted by the user to refer to any convenient output device. The printout format assumes that the printer has a carriage width of at least 120 characters.

## 5.6 SUMMARY OF MULTICOMPONENT SIMULATION PROGRAM

Program name: MULCOM

Type: Mainline with subroutines.

Operation Mode: Demand or Batch on UNIVAC 1106.

1. Title: Simulation of multicomponent distillation systems.

2. Solution type: Rating program.

3. Application: MULCOM uses the Thiele-Geddes calculation procedure to simulate distillation operations in columns of known configuration. Holland's  $\theta$  method is used to improve the rate of convergence of the iterative calculations.

The program will consider simple columns producing a single phase distillate from either a partial or a total condenser. A partial reboiler is assumed with a liquid Bottoms product withdrawn. Ideal solution behaviour is assumed and a choice of four equilibrium constant correlations is included. Enthalpy effects may be considered if data are available. Provision is made for including enthalpy correlations of arbitrary complexity. Stage efficiency may be considered in terms of component vaporization efficiency factors.

The user may repeat a run using different values of certain variables without terminating the execution. In demand mode an abbreviated result table is printed on the terminal at the end of a run. A detailed result table may be directed to the line printer if required and a plotting program is available for graphical representation of temperature, flowrate and composition data.

#### 4. Program Development:

- a) Computer used: UNIVAC 1106 - EXEC 8 control.
- b) Minimum core requirements: 17 K
- c) Language: Standard FORTRAN IV.
- d) Typical run time: 5 secs



## CHAPTER 6

### GRAPHICAL REPRESENTATION OF RESULTS

#### 6.1 INTRODUCTION

The plotting routines included in this computer program package have been developed specifically to supplement the output from each program, excluding the packed column design routine PACKED.

The plotting software published by CALCOMP\* must be available, together with an integrated hardware plotting device.

The plotting routines are linked to a mainline program that operates conversationally, in demand mode. The mainline program and the associated subroutines call the standard 'CALCOMP' routines as required. A number of data elements are used to store the diagram axis titles. These are added to the runstream through the ERTRAN facility and can be changed by the user if necessary.

Data files are assigned by the user during the execution of a particular distillation program. These files are added to the plotting runstream as required through the ERTRAN facility.

The following diagrams may be called for by the user:

A. Multicomponent system.

- (1) Temperature profile through the column.
- (2) Vapour and liquid flowrate profiles through the column.
- (3) Distillate composition bar diagram.
- (4) Composition profile through the column.

B. Binary system.

- (1) McCabe-Thiele diagrams
  - (a) Batch distillation - Variable reflux ratio
  - (b) Batch distillation - Constant reflux ratio
  - (c) Continuous distillation.

\* CALCOMP - CALIFORNIA COMPUTER PRODUCTS INC.

(2) Ponchon-Savarit diagrams

(a) Continuous distillation - tray columns.

6.2 MAINLINE PROGRAM FOR PLOTTING ROUTINES - PLOTTER

PLOTTER serves to link the subroutines together and to write the identifying titles above each completed diagram. It initializes each plot by calling the standard CALCOMP routine PLOTS. Thereafter the relevant subroutine is called to construct the specified diagram.

The program is divided into three sections. The first deals with multicomponent systems, the second with binary batch systems and the last with binary continuous systems. These will be considered in order.

The operating runstream is fully discussed in the user manual, section 7.5. A description of the internal program organization is given here.

<u>Line No.</u> *	<u>Program Action</u>
23-28	Requests run identification number and date for reference purposes. This number is written at the top of the completed diagram (see lines 155-159).
29-31	Requests type of plot required. Code numbers 1-5 identify specific diagrams (see User Manual).
32	Assigned GO TO statement selects program section.
(a) Multicomponent systems.	
34-57	Input of execution data (see User Manual for details).
58-76	Add title data elements and read in diagram headings and axis titles (see User Manual).
77	Plotting initialized through CALCOMP subroutine PLOTS.

\*See FORTRAN text, Appendix C.

- 78 CALCOMP subroutine FACTOR scales overall size of diagram.
- 79-82 Calling of plotting subroutines for construction of each diagram.
- 83-93,  
155-159 Identifying heading and job number written above diagrams, using CALCOMP subroutines SYMBOL and NUMBER.

(b) Binary Batch systems.

- 95,96 Plot initialized and scaled.
- 97 Subroutine PLOTBT constructs McCabe-Thiele diagram for batch system.
- 99 Diagram titles for constant reflux operation.
- 102 Diagram titles for varying reflux operation.
- 104-116 Titles read and written out.
- (155-159) Job identification number.

(c) Binary continuous systems.

- 118-122 Limits of enthalpy axis in enthalpy-concentration diagram (required for Ponchon-Savarit diagrams).
- 123-124 Plot initialized and scaled.
- 125 Subroutine PLOTMT constructs either McCabe-Thiele or Ponchon-Savarit diagrams as specified through counter NTYPE (see User Manual).
- 127-137,  
(155-159) Read and write title and job identification number on Ponchon-Savarit enthalpy concentration diagram.
- 140-150,  
(155-159) Read and write title and job identification number on McCabe-Thiele or Ponchon-Savarit x,y diagram.
- 160 Plot terminated through CALCOMP subroutine PLOT.

### 6.3 DIAGRAM-CONSTRUCTING SUBROUTINES

#### 6.31 Subroutine PLOTTS

This subroutine plots the temperature profiles for each of the runs stored on file by MULCOM. They are all plotted on a single set of axes for easy comparison and each curve is identified through use of different plotting characters. The runs are plotted in the order they were stored on file.

<u>Line No.</u>	<u>Program Action</u>
19-28	The length of the x-axis (tray numbers) is determined, based on the total number of trays in the column. The maximum value is 8 inches. For columns with less than eight trays a scale of 1" per tray is used. From 8 to 16 trays, $\frac{1}{2}$ " per tray is used.
29	The length of the y-axis is supplied by the mainline from element AXISTITLE. This is presently set at 8" but can be changed by the user if necessary.
30	CALCOMP subroutine PLOT sets the origin of the axes.
31-34	The temperature data file is added through ERTRAN and temperature vs tray number data for the first run are read.
35-36	CALCOMP subroutine SCALE calculates the scale on each axis based on the defined length and the data to be plotted.
37-42	The appropriate axis titles are selected from the arrays XAX and YAX created by the mainline line.

- 43-44 CALCOMP subroutines MAXIS and AXIS set up and labelled X and Y axes respectively.
- 46 CALCOMP subroutine LINE plots data for first run.
- 49-59 'DO' loop 11 reads and plots data for remaining runs.
- 62-72 'DO' loop 20 and 21 select diagram titles from arrays ALXT, ALYT and TITL, provided by the mainline from element TITLES, and write these at the top of the diagram.
- 74-83 Plotting symbol legend written above diagram identifies each run's temperature profile.
- 84-87 The diagram is completed by a boundary line and control is returned to the mainline. ALX1 and ALY1 position pen correctly for next diagram.

#### 6.32 Subroutine PLOTFS

The vapour and liquid flowrate profiles for a single run are plotted by this subroutine. The user specifies the data set to be plotted at the start of the run (see User Manual). Only one set is plotted because the diagram becomes difficult to read if too many lines are drawn. The user is able to extract data from the MULCOM printouts for comparison, and could insert these manually using different colours for clarity. The program selects the specified data set from the flowrate data file using counter NRF.

<u>Line No.</u>	<u>Program Action</u>
18-30	Axes are sized, origin set up and data file added through ERTRAN as described in section 6.31.
31-38	'DO' loop 10 selects the specified data set from the file.

- 39-52            Axes are scaled, set up and labelled as described in section 6.31.
- 53,54           Liquid and vapour profiles are plotted on the diagram.
- 56-66           Diagram headings selected and written.
- 67-77           Symbol legend identifying liquid and vapour profiles inserted.
- 78-82           Boundary line drawn and control returned to the mainline. ALX1 and ALY1 act as before.

### 6.33 Subroutine PLOTDS

This subroutine constructs a bar diagram indicating the component distribution in the distillate product. The specified run data are selected in the same way as the flowrate profile data, using counter NRF.

<u>Line No.</u>	<u>Program Action</u>
18-23	Axis lengths determined. X-axis (component number) is limited to 18 inches which is the maximum width of the full multicomponent system plot (see Chapter 8). The Y-axis is set at 5" with a scale of 0,2 mole fraction per inch.
24-30	'DO' loop 10 selects axis titles from XAX and YAX and PLOT sets origin of axes.
31-36	The data file is added through ERTRAN. 'DO' loop 11 selects specified data set.
37,38	Origin and interval of plotting on the Y-axis are set.
42,43	PLOT is used to draw X-axis.
45-52	'DO' loop 12 numbers the bar position. The width of a bar is 1" per component for systems of

less than 18 components. For larger systems the 18" length is divided into the appropriate number of equal increments. The component numbers read in order of decreasing volatility with the lightest component numbered 1.

53-55 The X-axis title is written in.  
56 AXIS sets up the Y-axis (mole fraction) in the normal way.  
58-66 'DO' loop 13 draws in the bars of width DEL, using subroutine PLOT.

The remainder of the routine writes the titles above the diagram and constructs a boundary line as described previously (section 6.31 and 6.32).

#### 6.34 Subroutine PLOTCS

This subroutine plots the composition profiles through the column for each component for the specified run.

<u>Line No.</u>	<u>Program Action</u>
19-29	PLOT sets the origin of the axes, ERTRAN adds the data file and 'DO' loop 10 and 11 search for the correct data set by reading through the file until the specified set is reached.
31-35	The data for the light component is then read and the y-axis start value and increment set.
36-49	Axis titles selected, X-axis scaled and axes set up using MAXIS and AXIS. The Y-axis uses the same length and scale as used in subroutine PLOTDS.
50	Data for light component plotted.
52-62	'DO' loop 13 reads and plots data for remaining components.

63-74           The size and format of the legend table identifying the profiles depends on the number of components. A maximum of 30 components is allowed for in three columns of 10 components each. The minimum size of a column is 5 components (see Chapter 8).

76-86           'DO' loops 20 and 21 select and write titles above diagram.

87-105          'DO' loops 22 and 23 write the legend table.

The diagram is then completed with a boundary line and control is returned to the mainline.

#### 6.35 Subroutine PLOTBT

A McCabe-Thiele x,y diagram for binary batch distillation systems operating under either constant or varying reflux conditions is constructed by this subroutine. The plotting procedure is the same in each case and the type of operation is indicated by counter NTYPE (see User Manual). Element AXISMT contains the axis titles. This element and the data file BD PLOT are added through ERTRAN. The overall size of the diagram is set at 10" x 10". This can be scaled down using the CALCOMP subroutine FACTOR rather than by reducing the axis lengths.

<u>Line No.</u>	<u>Program Action</u>
9-10	Length, starting value and plotting increment are set for each axis.
11-16	AXISMT added and axis titles read.
17	PLOT sets origin of axes.
18-19	X and Y-axes set up and labelled using AXIS.
20-22	PLOT completes boundary line around diagram.
23-24	PLOT constructs 45 <sup>o</sup> diagonal on diagram.



- 25-32        Data file added, equilibrium data read and equilibrium curve constructed.
- 33-54        'DO' loops 20 and 21 perform the following for initial and final batch-run conditions.
- (34-36)      Read in number of trays, coordinates of steps on x,y diagram and operating line coordinates.
- (37-40)      Use PLOT to construct operating line.
- (45)         Insert stepped stages on diagram.
- (48-53)      'DO' loop 21 numbers the stages from the bottom of the column.

At this point the diagram is complete for variable reflux operations.

- (56-72)      For constant reflux operations the average distillate product composition is inserted.

Control is then returned to the mainline.

#### 6.36 Subroutine PLOTMT

This subroutine constructs either McCabe-Thiele or Ponchon-Savarit diagrams for binary continuous-distillation systems. The type of diagram required is indicated by counter NTYPE in the calling list.

The first section of the program constructs the enthalpy-concentration section of the Ponchon-Savarit diagram. Element AXISPS contains the axes titles for this section.

The size of the enthalpy-concentration diagram is set at 15 inches for the y-axis and 10 inches for the x-axis. PLOT is used to set the origin of the diagram such that the x,y diagram can be drawn below it, (see Chapter 8).

The enthalpy axis coordinates of the " $\Delta$ " points must be specified by the user (see User Manual). These are

extracted from the result table supplied by program BINARY. The values are requested by the mainline and are included in the calling list. The routine uses the values to determine the scale on the enthalpy axis. The lower value supplied is taken as the starting point for plotting enthalpy data. It is usually convenient to round the values off as this will result in a neater diagram with the  $\Delta$  points lying inside the boundary line.

<u>Line No.</u>	<u>Program Action</u>
(a) Enthalpy-concentration diagram.	
16-23	Axis titles are read and the axis lengths set.
24	PLOT sets the origin of the axis to allow space for the x,y diagram below it.
30-31	Y-axis scaled using $\Delta$ values of enthalpy. Y-axis constructed and labelled using AXIS.
33	X-axis constructed and labelled.
34-51	Data file added through ERTRAN. Coordinates of points representing the feed, reboiler, condenser and $\Delta$ point are read. The lines joining these points are constructed using PLOT.
52-64	These points are labelled on the diagram.
66-73	'DO' loop 11 reads in the enthalpy curve data and plots this on the diagram. Liquid enthalpy is supplied first by the data file.
74-78	Boundary line construction completes diagram.

(b) x,y diagram

The same basic procedure is used for both design methods. Where necessary variations are indicated by counter NTYPE. The size of the diagram is set at 10" x 10". Element AXISMT provides axis titles as described in section 6.35.

- 79-98            Axes set up and diagonal constructed as described in section 6.35.
- 100            Data file added here for McCabe-Thiele designs.
- 102-108        Equilibrium curve constructed as described in section 6.35.
- 110-121        Operating lines constructed on McCabe-Thiele x,y diagram. Complex columns allowed for by NQDAT.
- 124-133        'DO' loop 12 constructs the two operating lines on a Ponchon-Savarit x,y diagram.
- 134-137        Number of trays and coordinates of steps on x,y diagram read for either design method.
- 139-140        Point of intersection of operating lines must be read for Ponchon-Savarit diagrams.
- 143-146        Starting values and plotting increments set for both axes.
- 147            Equilibrium stages stepped off.
- 148-155        'DO' loop 21 numbers the stages from the bottom of the column (reboiler = 0) (see Chapter 8).
- 156-168        'DO' loop 30 labels feed and side stream positions and inserts 'q' lines.

Control is then returned to the mainline.

#### 6.4 GENERAL COMMENTS ON THE PLOTTING FACILITY

Graphical representation of data is useful for comparing and analysing the effects of variables on a system. Constructing operating diagrams or composition profiles manually from tables of data is usually a tedious and time consuming process. The plotting facility developed here enables the user to obtain

graphical solutions to a number of problems in a short period of time. More time can thus be devoted to a critical analysis of the results. This aspect is discussed further in Chapter 8.

The use of data elements to store axis labels and other diagram titles enables the user to alter these without editing the program or subroutines. The size of each diagram is limited so as to prevent wastage of plotting paper, while ensuring that the diagrams are clear. CALCOMP subroutine FACTOR is used to scale the whole diagram to a convenient size such as A4. This procedure results in a better resolution than is obtained if the size of the original plot is reduced. The set of multicomponent plots is limited to a maximum size of 18" (x direction) by 28½" (y direction). The four graphs are arranged as shown in Chapter 8.

#### 6.5 SUMMARY OF PLOTTING PROGRAM

Program name: PLOTTER

Type: Mainline with subroutines.

Operation mode: Demand or Batch.

1. Title: Plotting routines associated with the distillation teaching-aid program package.

2. Solution Type: Graphical representation of data.

3. Application: PLOTTER may be used to construct the following diagrams using data generated by programs in the distillation package.

(A) Multicomponent systems (program MULCOM)

- (1) Temperature profile
- (2) Vapour and liquid flowrate profiles
- (3) Distillate composition bar diagram
- (4) Composition profiles.

(B) Binary systems.

- (1) McCabe-Thiele diagram for batch operation (program BATCH).

(2) McCabe-Thiele diagram for continuous operation  
(program BINARY).

(3) Ponchon-Savarit diagram for continuous operation  
(program BINARY).

The program uses the CALCOMP software and hardware facilities available at the University of Cape Town. Similar facilities would be required to run this program on other computer systems.

#### 4. Program Development:

- a) Computer used UNIVAC 1106 with CALCOMP plotter
- b) Software requirements: CALCOMP plotting software.
- c) Minimum core requirements: 20 K
- d) Language used: FORTRAN IV
- e) Typical run time: CPU : 7 secs  
Plotting: 4 mins (penplotter)

## CHAPTER 7

### DISTILLATION PACKAGE USER MANUAL

#### INTRODUCTION

This manual is designed to guide users in the operation of the distillation programs and associated plotting routines. Extracted from the main text together with Appendix A, the nomenclature tables and list of references, it provides all the necessary information for using the programs to solve standard tutorial problems. The options included in the programs should make them sufficiently versatile for undergraduate tutorial purposes. Where it might be necessary to change the FORTRAN text of a program for research or other purposes, the user is referred to the original thesis. In order to produce a self standing manual a certain amount of duplication of the main text was unavoidable.

The computer programs described here cover the following aspects of distillation taught in an undergraduate course at U.C.T.

- (1) Binary batch distillation in tray towers - section 7.1.
- (2) Binary continuous distillation in tray towers - section 7.2.
- (3) Binary continuous distillation in packed towers - section 7.3.
- (4) Multicomponent continuous distillation in tray towers - section 7.4.

A plotting program is available for graphical representation of results generated by programs 1, 2 and 4 above. (See section 7.5.)

A summary of the general operating procedures and conventions applying to all programs is given in section 7.6.

The computer executive commands listed in the following text apply to the UNIVAC 1106 - EXEC 8 computer system currently installed at U.C.T. It is assumed that the user has a working knowledge of the UNIVAC EXEC 8 control language and is able to execute programs using either a card deck (batch mode) or a computer terminal (demand mode). All distillation programs are contained in a file named \*DISTFILE and all plotting routines in \*PLOTFILE. The project name under which these files are stored on the system is set by the course supervisor. A master copy of both files should be kept on cards for longterm security, and on magnetic tape for normal operating purposes.

## 7.1 BINARY BATCH DISTILLATION

### 7.11 Application

The program BATCH simulates binary batch-distillation systems, operating under either variable or constant reflux conditions. All systems considered are subject to the following limitations:

- (a) The number of trays in the column must be specified.
- (b) Normal McCabe-Thiele assumptions must apply.
- (c) Each tray is assumed to represent an ideal equilibrium stage.
- (d) Hold-up effects in the column are assumed to be negligible.

The program has been developed for operation in demand mode from a computer terminal. It may also be run in batch mode, by means of a card deck, if necessary.

The program consists of a mainline routine, BATCH, and the following subroutines:

<u>Subroutine</u>	<u>Purpose</u>
TEACH	Provides a brief description of the program.
XDSTEP	Steps off stages on the x,y diagram.
BINBPT	Computes bubble points for equilibrium data calculations, using Antoine's equation and a Newton-Raphson convergence procedure.
XYALFA	Supplies relative volatilities corresponding to vapour or liquid mixture compositions. Values are extracted from an internally compiled array using linear interpolation.
SIMPS	Simpson's Rule numerical integration.
SEARCH	Part of a standard iterative search procedure.

The mainline routine reads in data, performs all calculations and prints out results.

For operation in either batch or demand mode, the program must be mapped as follows:

```
@MAP DISTFILE.BATCHM,.BATCH
```

where element BATCHM contains the following statements

```
IN DISTFILE.BATCH  
LIB DISTFILE.
```

Note: Segmented mapping does not reduce the core requirements of this program appreciably.

### 7.12 Operation of Program

7.121 Graphical representation of results: A plotting routine is available for constructing a McCabe-Thiele x-y diagram representing a computer solution to a problem. The program, BATCH, automatically writes the necessary data to a file, which is added to the plotter runstream through the ERTRAN facility. The file-name used by the plotting routine is BDLOT. This file must be assigned to the run, with the



identifying WRITE-key 11, before execution of BATCH.

The following two instructions will correctly assign the data file BDPL0T, and must be entered at the start of the runstream if the plotting facility is to be used.

@ASG,UP BDPL0T.

@USE 11,BDPL0T.

7.122 Execution of BATCH in demand mode, with conversational interaction: The following is a description of the full conversational operating procedure for BATCH.

Note: U: - User instruction

P: - Program response

Where options are available to the user, reference is made to a subsequent section where the options are explained.

To start the program:

U: @XQT DISTFILE.BATCH

P: ANSWER ALL QUESTIONS YES OR NO.

ALL INPUT IN FREE FORMAT.

DO YOU KNOW HOW TO USE THIS PROGRAM?

U: YES or NO

P: THIS IS A BINARY BATCH DISTILLATION SIMULATION PROGRAM.  
SYSTEMS OPERATING UNDER EITHER VARIABLE OR CONSTANT REFLUX CONDITIONS CAN BE CONSIDERED.  
THE PROGRAM IS LIMITED TO SYSTEMS FOR WHICH NORMAL MCCABE-THIELE ASSUMPTIONS HOLD.  
PLOTING ROUTINES ARE AVAILABLE FOR CONSTRUCTING MCCABE-THIELE X-Y DIAGRAMS FOR EITHER OPERATION MODE.

REFER TO THE DISTILLATION PACKAGE MANUAL, SECTION 1, FOR FURTHER INFORMATION.

The program then continues with:

P: MODES OF OPERATION: (1) VARIABLE REFLUX  
(2) CONSTANT REFLUX  
ENTER 1 OR 2 FOR REQUIRED PROCEDURE.

U: 1 or 2 (See section 7.131) (integer)

P: DO YOU WANT TO ADD A DATA ELEMENT?

U: YES - conversational facility is suppressed and a previously prepared data element is added to the run stream. (See section 7.123.)

or NO

P: ENTER NAMES OF COMPONENTS ... 1 PER LINE, MAX 30 CHARACTERS EACH.

U: COMPONENT NAME1 (light component)  
COMPONENT NAME2 (heavy component) (See section 7.132.)

P: ACCEPTED PRESSURE UNITS: 1=MMHG, 2=PSIG, 3=PSIA, 4=KPA, 5=ATM.  
ENTER CONDENSER PRESSURE AND UNIT CODE. (EG. 760.0, 1)

U: Pressure, Unit code (See section 7.133) (real, integer)

P: WILL ANTOINE CONSTANTS OR EQUILIBRIUM DATA BE SUPPLIED?  
ENTER ANT OR EQU.

U: ANT or EQU (See section 7.134 for equilibrium-data specification options.)

P: ENTER NO OF THEORETICAL PLATES IN COLUMN. (MAX=26)

U: Number of theoretical trays. (See section 7.135) (integer)

P: ENTER COMPOSITION AND QUANTITY OF INITIAL CHARGE.

U: Composition, Quantity. (See section 7.136) (real, real)

P: WILL (1) BATCH CYCLE TIME, OR (2) VAPOUR BOIL-UP RATE BE SPECIFIED. ENTER 1 OR 2.

U: 1 or 2 (See section 7.137)

The remaining data-requests depend on whether varying reflux or constant reflux operation is considered. (See section 7.131.)

Finally, for either mode of operation:

P: IS A (1) TOTAL OR (2) PARTIAL CONDENSER USED?  
ENTER 1 OR 2.

U: 1 or 2 (See section 7.135) (integer)

The program proceeds with the calculations and if no errors are detected the next output is the result table. (See section 7.14.)

7.123 Execution of BATCH in demand mode, with data element addition: The user indicates that a data element is to be used. The runstream proceeds as follows:

P: ADD YOUR DATA ELEMENT.

U: @ADD DISTFILE.DATA - where DATA contains all the data required for the run, set out in free format. A typical data element is shown in Table 7.1 with each entry briefly described. The data indicate the options required. Further output, excluding error messages, is suppressed until the problem is solved. A table of results is then printed out. (See section 7.14.)

### 7.13 Selection of optional data inputs

#### 7.131 Variable or constant reflux operation:

(a) U: 1 (Variable reflux)

This number is used by the program to identify the correct calculation procedure and final data requests. For variable reflux these are:

P: ENTER DISTILLATE COMPOSITION.

U: Required product composition as mole fraction of light component. (real)

P: ENTER FINAL RESIDUE CONCENTRATION.

U: Final residue concentration as mole fraction of light component. (real)

For variable reflux operation both the above compositions must be specified. If either is unknown an estimate should be made that can be adjusted later on the basis of computed batch cycle time or boil-up rate.

TABLE 7.1

DATA ELEMENTS FOR OPERATION OF BATCH

AS DESCRIBED IN SECTION 7.123

VARIABLE REFLUX OPERATION

PROBLEM-DATA ELEMENT

1	CARBON DISULPHIDE	Component Names
2	CARBON TETRACHLORIDE	
3	1.,5	Condenser Pressure, unit code
4	EQU	Equilibrium data to be specified
5	NO	Constant relative volatility not assumed
6	19	No. of data pairs
7	@ADD DISTFILE.BATEQU	Data element (see below)
8	6	No. of theoretical trays
9	0.4,50.	Feed composition, quantity (kg moles)
10	2	Vapour boil-up rate specified
11	10.	Vapour boil-up rate (kg moles/hr)
12	0.95	Distillate product composition
13	0.08	Final residue concentration
14	1	Total condenser

EQUILIBRIUM-DATA ELEMENT

	<u>x</u>	<u>y</u>
1	0.05	0.14
2	0.10	0.25
3	0.15	0.34
4	0.20	0.425
5	0.25	0.49
6	0.30	0.55
7	0.35	0.595
8	0.40	0.645
9	0.45	0.69
10	0.50	0.725
11	0.55	0.755
12	0.60	0.79
13	0.65	0.82
14	0.70	0.85
15	0.75	0.877
16	0.80	0.90
17	0.85	0.925
18	0.90	0.95
19	0.95	0.975

(b) U: 2 (Constant reflux)

The program responds as follows:

P: DO YOU WANT TO SPECIFY AN AVERAGE DISTILLATE COMPOSITION?

Two user-responses are allowed

(i) U: YES

P: ENTER AVERAGE VALUE.

U: Required product composition as mole fraction of light component. (real)

(ii) U: NO

P: ENTER FINAL RESIDUE CONCENTRATION.

U: Final residue concentration as mole fraction of light component. (real)

The average distillate composition and the final residue concentration are interdependent. Either may be specified with the unknown value determined by the program.

Finally the constant reflux ratio must be specified.

P: ENTER REFLUX RATIO.

U: Reflux ratio. (real)

7.132 Component names: The length of the component names is limited by the size of the array TITLES(2,5). The names must be entered on separate lines for reading in 5A6 format.

7.133 Condenser Pressure: The specified condenser pressure is assumed to be constant through the column. Pressure drop across trays is not allowed for because the effect is generally small and is not significant in graphical solution procedures. The value is used only if equilibrium data, bubble points, and dew points are computed using Antoine's equation. (Section 7.134.)

The pressure is entered as a real variable and the unit codes as an integer, both in free format.

7.134 Equilibrium data: Three options are available:

(a) U: ANT

P: ENTER ANTOINE CONSTANTS.

U: A,B,C - for light component. (all real)

A,B,C - for heavy component. (all real)

P: ENTER BUBBLE PT OF LIGHT COMPONENT.

U: Bubble point of light component in °C at specified condenser pressure. (real)

The program uses these data to compute vapour-liquid equilibrium data and corresponding relative volatilities, using Antoine's equation and Raoult's law. The specified bubble point is used as a starting value by the Newton-Raphson convergence procedure in subroutine BINBPT.

(b) U: EQU

P: IS THE RELATIVE VOLATILITY A CONSTANT?

Two replies acceptable:

(i) U: YES.

P: ENTER CONSTANT RELATIVE VOLATILITY VALUE

U: Constant relative volatility. (real)

Program computes vapour-liquid equilibrium data using this value.

(ii) U: NO

P: ENTER NO OF SETS OF DATA TO BE SUPPLIED.  
DO NOT INCLUDE X=0.,Y=0. OR X=1.,Y=1.

U: Number of x,y data pairs to be supplied, preferably twenty or more for good accuracy. (integer)

P: ADD EQUILIBRIUM-DATA ELEMENT.

U: @ADD DISTFILE.EQUDAT

where element EQUDAT contains the x,y data pairs, one per line in free format in order of increasing concentration. (See example element, Table 7.1.)

7.135 Number of theoretical trays in column: The number is entered as an integer and must not include either the reboiler or the condenser. The program adjusts the number to include a partial reboiler, and a partial condenser if one is specified, by adding one theoretical stage per device to the specified number of trays.

7.136 Composition and quantity of initial charge: Values are entered as the mole fraction of the light component and the total feed in kg.moles.

7.137 Batch cycle time or vapour boil-up rate: These values are interdependent. The program calculates the variable not specified and adjusts the result table accordingly. The computer response to a user input of either 1 or 2 is

P: ENTER BATCHTIME OR BOIL-UP RATE.

U: Known variable entered. (real)

#### 7.14 Result Table

On completion of the calculations a table of results is printed out. The format depends on the options specified originally by the user, and all results are printed out on the terminal in demand mode. The construction of the complete x,y diagram on the CALCOMP plotter can be initiated from the terminal as shown in section 7.54.

#### 7.15 Execution of BATCH in batch mode.

The same runstream used in demand mode is required for batch mode execution. The run can be entered from the

terminal or by means of a card deck. The following instructions are required in addition to the normal run identification card:

```
@ASG,AX DISTFILE.
@XQT DISTFILE.BATCH
@ADD DISTFILE.BATDATA
```

Element BATDATA must contain the instructions selecting the required procedure, and the problem data as shown in Table 7.1. The conversational output should be suppressed. (See p.195.)

#### 7.16 Error Messages

<u>Origin</u>	<u>Error Message</u>	<u>Description</u>
BATCH	RESIDUE COMPOSITION = ... DISTILLATE COMPOSITION SET TOO LOW.	If $X_{DAVE}$ is specified for a constant reflux operation, the final residue composition may fall below 0,001, which is the minimum allowed.
	SPECIFICATION ERROR ... RESTART PROGRAM	Error in data input format or limits are exceeded, e.g. mole fraction > 1,0 $P_T > 10$ atm. Illegal reply to a program question.
	ARRAY DIMENSION SIZE EXCEEDED	Equilibrium data parameter NPEQ, or integration parameter NPBS too small.

#### 7.17 Dimension parameters and specification limitations

##### (a) Dimension parameters.

There are three parameters associated with BATCH.

##### (1) NPEQ - equilibrium data arrays.

Internal factors DELXAL and NDAT, that generate equilibrium x,y data pairs, set a minimum value of NPEQ = 51.



In general:  $NPEQ \geq NDATA+2$  where NDATA is the number of data pairs supplied, excluding end points 0.,0. and 1.,1.

(2) NPTC - number of trays.

The coordinates of points defining the stepped stages on a McCabe-Thiele x,y diagram are stored for plotting purposes. The arrays X and Y are dimension NPTC where

$$NPTC \geq 2(NT+2)+1$$

where NT is the specified number of trays excluding the reboiler and partial condenser. If a total condenser is specified

$$NPTC \geq 2(NT+1)+1$$

Present value of 55 restricts BATCH to systems with 25 trays and a partial condenser or 26 trays with a total condenser.

(3) NPBS - numerical integration data arrays.

Internal counter NINT determines the number of increments over which integration is performed. In general:

$$NPBT \geq NINT+1$$

where NINT is defined in a DATA statement in BATCH. Other arrays have fixed dimensions to accommodate two components.

(b) Specification limitations.

(1) System pressure

In all the binary system programs in this package, the specified system pressure is assumed to be constant throughout the column for purposes of data calculations. Antoine's equation and other correlations included in the programs (see sections 7.2 and 7.3) are in general limited to system pressures of less than  $10^3$  kPa. Hence the following limitation is imposed.

$$\text{Specified total pressure: } P_T \leq 10^3 \text{ kPa (10 atm)}$$

(2) Number of trays in column.

See (a) (2) above.

## 7.2 BINARY CONTINUOUS DISTILLATION IN TRAY COLUMNS

### 7.21 Application

The program BINARY designs binary continuous distillation systems using either the McCabe-Thiele, or the Ponchon-Savarit, procedure. The dimensions of the column are not calculated hence superficial liquid and vapour flowrates are not considered. The Ponchon-Savarit procedure will consider simple columns whereas the McCabe-Thiele procedure is dimensioned to accommodate complex columns with up to five feed and five side streams. Murphree tray efficiencies may be specified for either procedure, and either total or partial condensers may be considered. Open steam may replace a partial reboiler in the McCabe-Thiele procedure.

Both design procedures compute the number of trays required to achieve a desired separation at a specific reflux ratio. The feed and side streams are positioned to yield the minimum number of trays possible for given operating conditions.

The mainline program BINARY has the following subroutines associated with it.

#### Subroutine

#### Purpose

TEACH

Provides a brief description of the program.

READBD

Reads in all data conversationally or from a data element. Calls CONDEN to determine distillate product compositions for either total or partial condenser operation. Calls FLASHB to determine feed condition for Ponchon-Savarit designs.

CONDEN	Distillate product composition calculations.
FLASHB	Flash calculation for Ponchon-Savarit designs. Calls EQUIL.
EQUIL	Determines equilibrium compositions at flash temperature of a two phase feed stream.
MULTFS	Determines operating conditions for complex columns in McCabe-Thiele designs.
SIMPLE	Same as MULTFS but for simple columns.
PONSAV	Constructs enthalpy-concentration diagram and x,y diagram operating lines for Ponchon-Savarit designs. Calls HTCAP, HFIND, RMIN, OPLINE and FLOWS.
HTCAP	Vapour and liquid stream enthalpies calculated. Calls HTSOLN.
HTSOLN	Determines heat of solution value, if data are supplied for use in liquid stream enthalpy calculations.
HFIND	Linear interpolation search for enthalpy of either vapour or liquid stream of given composition from data compiled in HTCAP.
RMIN	Computes minimum reflux ratio for Ponchon-Savarit design.
OPLINE	Defines operating lines in Ponchon-Savarit x,y diagram.
FLOWS	Computes flowrates and temperatures through Ponchon-Savarit columns. Calls AVEMW.
AVEMW	Computes average molecular weight of streams for mass flowrate calculations. Mass flowrates required in design of packed columns. (See section 7.3.)
CHEC	Searches for pinch points on x,y diagram before attempt is made to step off equilibrium stages.

STEPD	Steps off equilibrium stages on x,y diagram for either design procedures. Calls PSOPXY and FFIND.
PSOPXY	Performs material balance calculation for STEPD in stepping procedure for Ponchon-Savarit designs.
FFIND	Determines flowrates of streams leaving each tray in a Ponchon-Savarit column from data compiled in FLOWS. Determines temperature on each stage as well. Linear interpolation used.
BINRES	Prints result table for BINARY.

BINBPT,BINDPT,XYALFA and SEARCH are also used by BINARY and its subroutines. These are discussed in section 7.11.

In order to minimise core requirements, BINARY must be mapped as follows:

```
@MAP DISTIFILE.BINARYM,.BINARY
```

where element BINARYM contains the following statements:

1	SEG BINARY1
2	IN BINARY
3	SEG READBD1*,(BINARY1)
4	IN READBD
5	SEG SIMPLE1*,(BINARY1)
6	IN SIMPLE
7	SEG PONSAV1*,(BINARY1)
8	IN PONSAV
9	SEG RMIN1*,(PONSAV1)
10	IN RMIN
11	SEG OPLINE1*,(PONSAV1)
12	IN OPLINE
13	SEG FLOWS1*,(PONSAV1)
14	IN FLOWS
15	SEG CHEC1*,(BINARY1)
16	IN CHEC
17	SEG STEPD1*,(BINARY1)
18	IN STEPD
19	SEG PSOPXY1*,(STEPD1)
20	IN PSOPXY
21	SEG FFIND1*,(STEPD1)
22	IN FFIND
23	SEG BINRES1*,(BINARY1)
24	IN BINRES
25	LIB DISTIFILE.

## 7.22 Operating the program

7.221 Graphical representation of results: A plotting routine is available to produce a standard McCabe-Thiele or Ponchon-Savarit diagram representing the computer solution to a problem. Key points on each diagram are marked to simplify interpretation of the results. To make use of this facility the data file BDPLLOT must be assigned to the run as described in section 1.21, before execution of BINARY. The WRITE key in BINARY is 11, hence the necessary instructions are

```
@ASG,UP BDPLLOT.
```

```
@USE 11,BDPLLOT.
```

The WRITE key is defined in a DATA statement in BINARY and may be changed if necessary.

7.222 Execution of BINARY in demand mode, with conversational program user interaction: Execution started by:

```
U: @XQT DISTFILE.BINARY
```

The program responds as shown in section 7.122. If the user replies NO to the first question, the following description of the program is printed out.

```
P: THIS IS A BINARY CONTINUOUS DISTILLATION DESIGN PROGRAM.  
  IT WILL COMPUTE THE NUMBER OF THEORETICAL STAGES RE-  
  QUIRED TO ACHIEVE A SPECIFIED TOP PRODUCT PURITY.  
  A MURPHREE PLATE EFFICIENCY FACTOR MAY BE SPECIFIED.  
  THE DIMENSIONS OF THE COLUMN REQUIRED ARE NOT CALCULATED.  
  SYSTEMS FOR WHICH EITHER MCCABE-THIELE OR PONCHON-  
  SAVARIT ASSUMPTIONS HOLD,MAY BE CONSIDERED.  
  PLOTTING ROUTINES ARE AVAILABLE FOR CONSTRUCTING STAN-  
  DARD DIAGRAMS FOR EITHER DESIGN TECHNIQUE.
```

```
  REFER TO THE DISTILLATION PACKAGE MANUAL,SECTION 2,  
  FOR FURTHER INFORMATION.
```

The execution is then terminated as it is unlikely that a successful run would be completed without preparation. The normal runstream proceeds as follows if the first question is answered YES.

P: DESIGN TECHNIQUES: (1) MCCABE-THIELE  
(2) PONCHON-SAVARIT  
ENTER 1 OR 2 FOR REQUIRED PROCEDURE.

U: 1 or 2 as required. This is read in as counter NOP  
which selects the appropriate options.

The next program response concerns data element addition as shown in section 7.122. The conversational output can be suppressed and a data element added. A sample data element is shown in Table 7.21.

The conversational runstream continues as shown in section 7.122 up to and including the equilibrium data specification. In BINARY, however, equilibrium temperature data corresponding to the composition data, must be supplied if option EQU is used. This is described in section 7.231.

TABLE 7.21

PROBLEM-DATA ELEMENT FOR MCCABE-THIELE DESIGN

1	METHANOL	Component names
2	WATER	
3	760.0,1	Column pressure and unit code.
4	ANT	Antoine constants specified.
5	7.87863,1473.11,230.	Antoine constants
6	7.96681,1668.21,228.	
7	64.7,100.	Pure component boiling points.
8	0.95,38.1	XD,DTOT
9	1.16	R
10	PAR	Partial condenser specified.
11	0.4,0.6	Vap & Liq Fraction in 2 phase distillate
12	NO	'IS XD OVERALL COMPOSITION?'
13	LIQ	Phase defined by XD
14	1	No. of feed streams
15	0	No. of side streams
16	0.39,98.1	Z,FTOT
17	1.04	Q
18	1	Open steam replaces reboiler
19	0.8	Efficiency factor (Murphree tray efficiency)

After supplying equilibrium data the runstream proceeds as follows:

P: ENTER DISTILLATE COMPOSITION AND WITHDRAWAL RATE.

U: XD, FTOT (mole fraction, hg moles/hr, both real)

P: ENTER REFLUX RATIO.

U: R (real)

P: TOTAL OR PARTIAL CONDENSER?...ENTER TOT OR PAR.

U: TOT or PAR. (See section 7.232.)

Counter NOP selects the options corresponding to the design procedure chosen. For NOP=1 the runstream proceeds at (a) below; for NOP=2, the runstream continues at (b).

(a) NOP=1 (McCabe-Thiele design only.)

P: ENTER NO.OF FEED STREAMS.

U: NF (integer)

P: ENTER NO. OF SIDE STREAMS.

U: NS (integer)

If  $NF+NS=1$  (simple columns) the runstream continues at (b) below.

For  $NF+NS>1$  (complex columns):

P: ENTER FEED STREAM COMPOSITIONS.

U:  $Z_{F_k}$   $k=1, NF$  mole fraction of light component (ZF)  
in each feed.

P: ENTER FEED Q VALUES.

U:  $Q_{F_k}$   $k=1, NF$  feed q values in same order as  $Z_F$  (QF)

P: ENTER FEED RATES

U:  $F_{F_k}$   $k=1, NF$  feed flowrates kg mole/hr. (FTOTS)

P: ENTER SIDE STREAM COMPOSITIONS

U:  $Z_{S_k}$   $k=1, NS$  (ZS)

P: ENTER SIDE STREAM Q.VALUES.

U:  $Q_{S_k}$   $k=1,NS$  (QS)

P: ENTER SIDE STREAM WITHDRAWAL RATES.

U:  $F_{S_k}$   $k=1,NS$  (STOTS)

The positions of the feed and side streams in the column are determined by the program to minimise the total number of trays required. Hence the above data may be inserted in any order.

(b) NOP=2 (Ponchon-Savarit designs) or McCabe-Thiele simple columns:

P: ENTER FEED COMPOSITION AND FEED RATE

U: Z, FTOT

A feed q-value (McCabe-Thiele) or a feed temperature (Ponchon-Savarit) must be specified as follows, (See section 7.233):

P: ENTER FEED Q VALUE.

U: Q

or P: ENTER TEMPERATURE OF FEED.. (DEG C).

U: TF

\*P: IF THE HEAVY COMPONENT IS WATER AND OPEN STEAM OPERATION IS REQUIRED, ENTER 1...IF NOT ENTER 0

U: 0 (partial reboiler)

or 1 (open steam).

P: ENTER MURPHREE STAGE EFFICIENCY FACTOR  
(=1.0 IF PERFECT TRAYS ARE ASSUMED)

U:  $E_{V_m}$  (real)

This is the last entry required for McCabe-Thiele designs.

\* McCabe-Thiele designs only.



Enthalpy data are required for Ponchon-Savarit designs only and must be added to the runstream in a data element.

P: ADD ENTHALPY DATA ELEMENT.

U: @ADD DISTFILE.ENTHDAT (See section 7.234.)

P: Feed q-value and temperature data are printed out for inspection.

This is the final entry required for Ponchon-Savarit designs. If no errors are detected during computation, the next output is the result table (section 7.24).

### 7.23 Input of problem specifications.

7.231 Equilibrium data: The procedure and options are described in section 7.134 with the following differences:

(1) Temperature data corresponding to the equilibrium data, are required for Ponchon-Savarit designs. Zeros or fictitious values must be included if a McCabe-Thiele design is required. The data are read in as  $x_i, y_i, T_i$ , with one set of three values per line. The table starts at low concentrations with the values referring to the light component.

(2) The constant relative volatility option is not available in Ponchon-Savarit designs because temperature data are required.

(3) If Antoine constants are specified, the bubble points of both components must be specified; light component value first. Temperature data are then generated using the bubble point routine.

7.232 Total or Partial condenser operation: Specification of a total condenser requires no further input related to the distillate product. If, however, a partial condenser is used, and a two phase product withdrawn, the previously entered composition value XD must be defined as one of three options:

- (a) Overall distillate product composition
- (b) Vapour phase composition
- (c) Liquid phase composition

After specifying a partial condenser, the runstream proceeds as follows:

P: ENTER FRACTION OF DISTILLATE WITHDRAWN AS VAP AND LIQ RESP.  
FOR ALL VAPOUR DISTILLATE ENTER 1.,0.

U: Fraction of each phase, vapour phase value first.  
e.g. 0.5,0.5 for equimolar withdrawal of each phase.

If the distillate is to be withdrawn as a vapour only the runstream continues with:

\*P: ENTER NO. OF FEED STREAM (McCabe-Thiele designs)

or P: ENTER FEED COMPOSITION AND RATE (Ponchon-Davarit design)

For a two-phase distillate product system the program responds with

P: IS XD AN OVERALL COMPOSITION VALUE

U: YES (runstream proceeds as shown above\*)

or U: NO

P: IS XD THE COMPOSITION OF THE VAPOUR OR LIQUID FRACTION  
ENTER VAP OR LIQ

U: VAP

or LIQ (run proceeds as shown above\*)

A two-phase distillate product is allowed in McCabe-Thiele designs only.

7.233 Feed temperature or q-value specifications: It is customary to specify the feed condition in terms of a q-value in McCabe-Thiele tutorial problems. The program BINARY requires a specified q-value for each feed and side stream in a McCabe-Thiele design.

In Ponchon-Savarit designs the knowledge of enthalpy data permits the determination of the feed condition by flash calculation. To perform a flash calculation, the overall composition and temperature of the feed must be known. The feed  $q$ -value is determined from equilibrium data if the temperature of the feed is found to lie between the dew point and the bubble point. The following expression is used.

$$q = \frac{y - z_F}{y - x} \quad (7.21)$$

where  $y$  and  $x$  are the equilibrium concentrations at the feed temperature.

If the feed temperature indicates that the feed is either superheated or supercooled at the column pressure, the calculation of  $q$  is performed according to its definition as the heat required to convert one mole of feed from its condition  $H_F$  to a saturated vapour, divided by the molal latent heat

$$q = \frac{H_V - H_F}{H_V - H_L} \quad (7.22)$$

where  $H_V$  and  $H_L$  are the enthalpies of the feed at its dew point and bubble point respectively.

The above two equations may be used to check computer results.

7.234 Enthalpy calculations: In Ponchon-Savarit designs the effect of enthalpy on the vapour and liquid flow profiles is considered. The program constructs an enthalpy-concentration diagram for the system, and uses the standard graphical construction procedure to determine the positions of the operating lines on the  $x$ - $y$  diagram.

The vapour and liquid enthalpy curves are determined using the heat capacity correlation of Kobe [55] and Watson's [56] correlation for extrapolating latent heat data with

respect to temperature. Heat of solution effects may be included if data are available.

(a) Kobe heat capacity polynomial.

$$Cp_i = a_i + b_i T_j + c_i T_j^2 + d_i T_j^3 \quad (7.23)$$

Coefficients  $a, b, c$  and  $d$  for a number of compounds are listed in Hougen, Watson and Ragatz, Part II [57]. Coefficients for other compounds can be found in the original articles of Kobe [55], or may be determined by applying a least squares fit of the above polynomial to available data. The number of compounds listed in Hougen et al. should be sufficient for setting tutorial problems dealing with a wide range of systems. Coefficients  $b_i, c_i$  and  $d_i$  are quoted as (coeff.  $\times 10^{n_k}$ ). The user enters the coefficients and the corresponding negative exponents ( $-n_k$ ) separately (section 7.234(f)). The program then adjusts the decimal point accordingly. This procedure simplifies the use of free-format for data inputs.

(b) Liquid enthalpy

Enthalpy data are computed in subroutine HTCAP using the definite integral of the above polynomial.

$$h_i = \int_{T_b}^{T_B} Cp_i dT \quad (7.24)$$

where  $T_b$  = a specified base temperature.

Coefficients taken from Hougen et al. are derived for temperature in K and heat capacity in  $\text{cal gmole}^{-1}$ . A number of conversion factors are included to enable data to be specified in convenient units. These are discussed in section 7.235.

(c) Heats of solution

If heat of solution data are available, these may be included in the liquid-enthalpy calculations. Heat of solution data for a few systems are available in International Critical Tables, Vol.5, p.159. These data are quoted as composition vs. heat of solution, determined at a given temperature. The temperature at which these data are available must be specified as the base temperature,  $T_b$ , for enthalpy calculations.

Subroutine HTSOLN determines the heat of solution of a mixture of composition  $x$  using linear interpolation between points in the given data table. Conversion factors adjust the units to correspond with those used in HTCAP. (See section 7.235 below.)

Note: Enthalpy correlation charts often include heat of solution effects. If coefficients for Equation (7.23) are derived from such charts heat of solution data should not be provided separately.

(d) Vapour Enthalpy

Enthalpy data are determined for the vapour phase by adding the latent heat, calculated at the vapour temperature, to the enthalpy of a liquid of the same composition, calculated at the vapour temperature [57]. If the vapour is superheated, the latent heat is calculated at the dewpoint of the vapour.

One value of the latent heat of vaporization for each component must be specified at the normal boiling point of the pure component. Critical temperatures and normal boiling points of each component must be specified in  $^{\circ}\text{C}$ .

Watson's correlation [56] is used to extrapolate a single latent heat value with respect to temperature. The correlation may be used for temperatures within 10 K of  $T_c$  [58]:

$$LH_{i_k} = LH_{i_{BP}} \left[ \frac{T_{C_i} - T_k}{T_{C_i} - T_{BP}} \right]^{0.38} \quad (7.25)$$

(e) Mixture Enthalpy

The enthalpy of liquid mixtures is computed as

$$h_L = \sum x_i h_i + H_S \quad (7.26)$$

The enthalpy of vapour mixtures is computed as

$$H_V = \sum y_i H_i$$

where  $H_i = h_i + LH_{i,T_D}$  (7.27)

with  $h_i$  computed at the temperature of the vapour  $T_D$ .

(f) Data required:

A data element containing all the required enthalpy data must be set up before the program is run. Data are inserted in free format, subscript 1 indicates the light component. Table 7.22 shows problem and enthalpy of data elements for Ponchon-Savarit designs.

Data element ENTHDAT

$a_1, b_1, c_1, d_1$	Values entered as $a, b \times 10^2, c \times 10^5$ , and
$a_2, b_2, c_2, d_2$	$d \times 10^9$ . See Hough, Watson & Ragatz [57].
NBCP, NCCP, NDCP	$-n_k$ , $k=b, c, d$ ; exponents corresponding to $b, c$ and $d$ respectively. E.g. -2, -5, -9
CPCONV, LHCONV,	See section 7.235 below.
HSCONV, HEVABS	" " " "
TBASE, TCONV	" " " "
$LH_1, TBP_1, TC_1$	Latent heat, normal boiling point and
$LH_2, TBP_2, TC_2$	critical temperature for each comp.
$MW_1, CPMW_1, LHMW_1, HSMW_1$	See section 7.235 below
$MW_2, CPMW_2, LHMW_2, HSMW_2$	" " " "

NHS                      Number of heat of solution data pairs.  
 XHS<sub>k</sub>, HSOLN<sub>k</sub> k=1,NHS    Composition - heat of solution data  
                                  if available.

Note: If heat of solution data is not used NHS must be entered as zero.

TABLE 7.22

PROBLEM DATA ELEMENT FOR PONCHON-SAVARIT DESIGNS

1	METHANOL	Component names
2	WATER	
3	760.0,1	Column pressure and code
4	ANT	Antoine constants specified
5	7.87863,1472.11,230	
6	7.96681,1668.21,228	Antoine constants
7	64.7,100.	Pure component boiling points
8	0.915,38.1	XD,DTOT
9	1.16	R
10	TOT	Total condenser specified
11	0.36,98.1	ZF,FTOT
12	57.8	Feed temperature
13	1.0	Efficiency factor (for ideal trays)
14	@ADD DISTFILE.ENTHDAT	Enthalpy data element.

ENTHALPY DATA ELEMENT ENTHDAT

1	4.55,2.186,-0.291,-1.92	
2	7.7,0.04594,0.2521,-0.8587	Kobe function coefficients
3	-2,-5,-9	Exponents relating to coefficients
4	1.8,1.8,430.,-1.	CPCONV,LHCONV,HSCONV,HEVABS
5	19.69,273.	Base temperature, TCONV
6	262.79,64.7,240.	
7	540.,100.,374.15	LH <sub>i</sub> ,T <sub>BPi</sub> ,T <sub>Ci</sub>
8	32.04,1.0,32.04,1.0	
9	18.02,1.0,18.02,1.0	MW <sub>i</sub> CPMW <sub>i</sub> LHMW <sub>i</sub> HSMW <sub>i</sub>
10	0	NHS

7.235 Conversion factors:    Enthalpy data are usually quoted in a variety of units. For example, coefficients for the heat capacity polynomial from Hougen et al. requires

temperature data in K and computes  $C_p$  data in  $\text{cal gmole}^{-1}$ , whereas many of the coefficients shown in the original work of Kobe require temperatures to be specified in  $^{\circ}\text{C}$ .

The calculations performed in the subroutines HTCAP and HTSOLN are independent of the calling routine. Conversion factor ENCONV in PONSAV converts all data returned by HTCAP to the S.I. system of  $\text{KJ Kg mole}^{-1}$ . The value of ENCONV presently inserted in PONSAV programs assumes that enthalpy data are returned in  $\text{Btu/lb mole}$ . CONFAC converts heat duty units from  $\text{kJ/hr}$  to  $\text{kW}$ .

The conversion factors used in HTCAP and HTSOLN must therefore convert the units in which data is supplied to the F.P.S. system. This double conversion has been adopted because many example problems in the literature use the British system of units. By setting CONFAC and ENCONV = 1 in the mainline, results are printed in the British system and this enables the computer solution to be compared to the text solution with no further calculations.

The conversion factors are supplied in three sets

- |                 |                           |
|-----------------|---------------------------|
| (a) CPCONV,CPMW | Heat capacity correlation |
| (b) LHCONV,LHMW | Latent heat correlation   |
| (c) HSCONV,HSMW | Heat of solution data.    |

Each set is used in the same way, as follows:

Consider latent heat data found in the literature in  $\text{cal gm}^{-1}$  [58]. To convert values to  $\text{Btu/lb mole}$  set LHCONV = 1.8 and LHMW = molecular weight of component. These factors will then convert  $\text{cal gm}^{-1}$  to  $\text{Btu lb}^{-1}$  and weight to moles respectively. If either unit is correct the corresponding conversion factor is set equal to 1.0.

Other factors shown in the data list are defined as follows:



TBASE        Base temperature for enthalpy calculations.  
              If heat of solution data are supplied, the  
              temperature at which this is quoted is used  
              as the base temperature. Units: °C.

TCONV        Used to convert temperatures in °C to K for  
              use in the  $C_p$  correlation. If coefficients  
              are specified for a polynomial in °C, TCONV  
              is set equal to zero.

HEVABS       Specified as either -1 if heat of solution  
              is exothermic or +1 if heat of solution is  
              endothermic.

MW<sub>1</sub>, MW<sub>2</sub>    Molecular weight of each component required  
              in mainline program PACKED.

#### 7.24 Results

On completion of the calculations a table of results is printed on the terminal in demand mode, or on the line printer in batch mode. The format of the table is determined by the type of problem solved (McCabe-Thiele or Ponchon-Savarit). The main section of the table contains the standard results required in most tutorial problems. Additional tables of equilibrium and operating line data for manual construction of an x,y diagram is printed on request as follows:

P: DO YOU WANT OPERATING DIAGRAM DATA TABLES?

U: YES or NO

This reply must be included in batch mode runstreams as well.

A McCabe-Thiele x,y diagram or a complete Ponchon-Savarit operating diagram may be constructed on the CALCOMP plotter using the instructions shown in section 7.55.

### 7.25 Execution of BINARY in batch mode

The following statements are required to execute BINARY in batch mode:

@ASG,AZX DISTFILE

@XQT DISTFILE.BINARY

@ADD DISTFILE.BINDATA

where BINDATA is a data element which must conform to the demand mode runstream. The conversational output should be suppressed in batch mode.

If a plot is required the plotting data file must be assigned before the @XQT statement, as shown in section 7.221. The plotting routine can be started in batch mode as shown in section 7.63, but this should not be done until the results have been examined.

### 7.26 Error messages

Certain subroutines are called both by BINARY and by PACKED, and hence the error messages associated with these subroutines are common to both programs.

<u>Origin</u>	<u>Error Message</u>	<u>Description</u>
BINARY	ILLEGAL SPECIFICATION.ERROR TERMINATION	Reply to question other than YES or NO.
SIMPLE, PONSAV	ERROR...BTOT OR XB IS LESS THAN 0.0 OR XB IS GREATER THAN Z	Flowrate or composition specification causes failure of mass balance
MULTFS	ERROR...TWO STREAMS HAVE SAME COMPOSITION	Two streams, feed or side in complex columns have been specified with equal compositions.
	ERROR... BTOT OR XB IS LESS THAN 0.0	Similar to SIMPLE above.

<u>Origin</u>	<u>Error Message</u>	<u>Description</u>
PONSAV	SPECIFIED REFLUX RATIO IS LESS THAN THE CALCULATED MINIMUM RMIN = RSPEC = A VALUE OF 1.5*RMIN WILL BE USED	(Non terminating message.
CHEC	PINCH POINT AT SIDE (OR FEED) STREAM COMPOSITION	Point of intersection of two operating lines detected above equilibrium curve. Change flow-rate or q-value of indicated stream or increase the reflux ratio.
READBD	ERROR IN DATA INPUT...ERROR TERMINATION  ARRAY DIMENSION SIZE EXCEEDED	Data does not conform to specified format or limitations e.g. $P_T > 10$ atm or a reply to a question is illegal. Number of data points exceeds dimension parameter.
STEPD	NUMBER OF TRAYS EXCEEDS: PINCH ZONE IS SUSPECTED...	Dimension parameter NPTC is exceeded.
RMIN	ERROR IN RMIN.FEED ENTER ON TOP TRAY  ERROR IN RMIN ROUTINE THIE LINE PASSING THROUGH POINT(Z,HF) NOT FOUND	Search for RMIN has determined that $\Delta_{DM}$ lies on or below the vapour enthalpy curve on the enthalpy-conc. diagram. Number of iterations allowed for search is 500. This may be exceeded

	HTEST= HF= Y1= X1=	if the convergence limit
	NO.OF ITERATIONS=	ACC in RMIN is too small.
	INCREMENTS=	In this case the displayed
	(Y1 and X1 are the compo-	increment will probably
	sition coordinates of the	be very small and HTEST
	tie line last tested. ATEST	~HF.
	is the iterative enthalpy	
	value that should converge	
	to HF when the tie line	
	passing through HF is found.)	
	ERROR IN RMIN ROUTINE	Similar to above error.
	ENTHALPY OF VAPOUR ENTERING	Result of iterative
	PARTIAL CONDENSER NOT FOUND.	search failure at 500
		iterations. ACC is
		probably too small.
OPLINE	ERROR...NO. OF INTERVALS	Refers to parameter
	CALCULATED IN OPLINE EXCEED	NPPS.
	DIMENSION SPECIFICATIONS	
	NO.OF INTERVALS=	
	ERROR IN OPLINE	
	LOOP 'n' EXCEEDED 500	Similar to search errors
	ITERATIONS.	in RMIN. Increase size
		of ACC in OPLINE.
FLows	TOTAL NUMBER OF OPERATING	NPPS should be increased.
	LINE DATA POINTS EXCEEDS	
	DIMENSION SPECIFICATIONS	
	NEGATIVE LIQUID (OR VAPOUR)	
	FLOW DETECTED...(self	
	explanatory)	

### 7.27 Dimension parameters and specification limitations

(a) Dimension parameters.

There are four parameters associated with BINARY:

- (1) NPEQ - equilibrium data arrays. (See section 7.17.)
- (2) NPTC - number of trays. (See section 7.17.)
- (3) NPEN - Enthalpy data arrays.

In computing enthalpy-concentration data for constructing curves on the Ponchon-Savarit diagram, the interval between points sets the size of the array required. Intervals DELX and DELY in PONSAV of 0,05 mole fraction produce smooth curves for most ideal systems. The parameter can thus be defined as

$$NPEN \geq 1.+DELX^{-1} \text{ or } 1.+DELY^{-1}$$

whichever is larger if DELX  $\neq$  DELY.

(4) NPPS - Arrays associated with Ponchon-Savarit Calculations.

The COMMON section of block BINCOM is used both in BINARY and in PACKED. Arrays dimensioned NPPS are used in Ponchon-Savarit and packed column calculations, where internal counter NINT determines the number of points defining the top operating line. The size of the interval between points is kept the same for both operating lines to maintain a constant level of accuracy in the integration procedure in PACKED. Hence the number of points defining the bottom operating line could exceed NINT. Also, certain arrays contain data for both operating lines. Thus to ensure that sufficient space is available set NPPS as follows:

$$NPPS \geq 3 \times NINT$$

where NINT is defined in a DATA statement in the mainline routine. Counter NPPSC = NPPS-2 is used to check that the array size is not exceeded during calculations.

Other arrays have fixed dimensions to accommodate two components, and complex columns with five feed and five side streams with associated data.

(b) Specification limitations.

(1) Pressure specification. (See section 7.17.)

(2) Number of trays.

Limitation as described in section 7.17 indicated by error message if the designed number of trays exceeds the dimension parameter NPTC.

(3) Complex columns may contain a maximum of five feed and five side streams. No two streams may have the same composition. The streams can not be positioned in the column in a particular order. The program arranges the streams in order of increasing concentration of the light component, moving up the column.

(4) Either a two phase or a single phase distillate product may be withdrawn from a partial condenser. If a two phase product is withdrawn the user must specify whether the distillate composition refers to a particular phase, or the overall product.

(5) Accuracy limits on convergence procedures in RMIN and OPLINE should not be set too low. The present value ( $ACC=0.0001 \times HF$ ) represents an error of 0.01% of the absolute feed-enthalpy value, in each calculation. This produces satisfactory results for normal ideal systems. Lower values increase the CPU time required and may cause failure of the convergence procedure.

### 7.3 BINARY CONTINUOUS DISTILLATION IN PACKED COLUMNS

#### 7.31 Application

The program PACKED determines the diameter and height of a bed of packing required to achieve a given separation. The design is based on a Ponchon-Savarit operating diagram and uses Shulman's [61] mass transfer coefficient correlations in the calculation procedure. The design procedure is described by Treybal [9]. Flooding and loading characteristics are determined using the generalised flooding correlation chart of the U.S. Stoneware Co. (see Figure 1, Appendix A).

It is assumed that the column is fitted with a partial reboiler and either a partial or total condenser. A single phase distillate product is withdrawn. The program uses the same procedure for constructing the Ponchon-Savarit diagram as the program BINARY, see section 7.2. Other subroutines associated with PACKED include

<u>Subroutine</u>	<u>Purpose</u>
PACTYP	Determines type of packing required by the user, and calls PAKING to extract the relevant data.
PAKING	Contains packing data for seven packings. (See Appendix A, Tables 1,2 and 3.) The user may add to this data routine if necessary.
FDATA	Fits polynomials to the curves shown on the flooding correlation chart (Figure 1, Appendix A). The user selects a curve by specifying a column pressure drop.
PACRES	Prints out result table on completion of the design.

The physical property package is also required. The physical property subroutines have been designed so that they can be used by other programs developed in the Department of Chemical Engineering, U.C.T. Program information is entered through a calling list and data required by the subroutine are read from a specially prepared data element through the ERTRAN facility. (See section 7.34.)

In order to minimise core requirements, PACKED must be mapped as follows:

```
@MAP DISTFILE.PACKEDM,.PACKED
```

where element PACKEDM contains the following statements:

```
1          SEG PACKED1
2          IN PACKED
3          SEG READBD1*,(PACKED1)
4          IN READBD
5          SEG PONSAV1*,(PACKED1)
6          IN PONSAV
7          SEG RMIN1*,(PONSAV1)
8          IN RMIN
9          SEG OPLINE1*,(PONSAV1)
10         IN OPLINE
11         SEG FLOWS1*,(PONSAV1)
12         IN FLOWS
13         SEG PACTYP1*,(PACKED1)
14         IN PACTYP
15         SEG PAKING1*,(PACKED1)
16         IN PAKING
17         SEG PACRES1*,(PACKED1)
18         IN PACRES
19         LIB DISTFILE.
```

### 7.32 Operating the program

There are no plotting routines associated with PACKED. The Ponchon-Savarit diagram can be obtained by running the same data in BINARY. This will also yield the number of trays equivalent to the packed height for comparative purposes. These data could be used to calculate the HETP for the system.



PACKED should be executed in demand mode because the user is required to inspect results at intervals during the run. Decisions must be made, based on these results, to direct further computation. These decisions are discussed in the runstream shown below. Subroutine READBD is used by PACKED as well as BINARY to read basic operating data for a problem. The initial part of the runstream is thus similar to that described in section 7.222 for Ponchon-Savarit designs, and requires no further explanation. The brief description of the program provided by TEACH is as follows:

P: THIS PROGRAM DESIGNS PACKED COLUMNS FOR DISTILLATION OF BINARY SOLUTIONS.  
THE PONCHON-SAVARIT METHOD IS USED TO DETERMINE THE OPERATING DIAGRAM.  
THE HEIGHT OF PACKING REQUIRED IN EACH SECTION OF THE COLUMN IS DETERMINED BY INTEGRATION, USING SIMPSON'S RULE.  
MASS TRANSFER COEFFICIENTS ARE COMPUTED USING SHULMANS CORRELATIONS AND PACKING DATA.  
DATA FOR CERAMIC RASCHIG RINGS (4 SIZES) AND BERL SADDLES (3 SIZES) ARE INCLUDED IN THE PACKAGE.  
  
REFER TO THE DISTILLATION PACKAGE MANUAL, SECTION 3, FOR FURTHER INFORMATION.

This terminates the execution as described in section 7.222.

Two data elements may be added to the runstream, with an optional conversational procedure available in each case. The first element concerns the basic problem specifications and is very similar to the element required by BINARY. The second element contains data relating to the type and size of packing required, and other mass transfer calculation details. An example of each element is shown in Table 7.3.

TABLE 7.3

FIRST DATA ELEMENT CONTAINING PROBLEM-DATA

1	METHANOL	Component names
2	WATER	
3	760.0,1	Column pressure and code
4	ANT	Antoine constants specified
5	7.87863,1473.11,230	Antoine constants
6	7.96681,1668.21,228	
7	64.7,100.	Pure component boiling points
8	0.915,38.1	XD,DTOT
9	1.16	R
10	TOT	Total condenser specified
11	0.36,98.1	Z,FTOT
12	57.8	Feed temperature
13	@ADD DISTFILE.ENTHDAT	(See Table 7.22)

SECOND DATA ELEMENT SPECIFYING PACKING AND COMPONENT PROPERTY

DATA

1	NO	'DO YOU WANT TO USE A DIFFERENT TYPE OF PACKING?'
2	1	Type of packing code (Raschig rings)
3	2	Size of packing code (25,4 mm)
4	NO	'ARE SOLUTION-DENSITY DATA ALSO AVAILABLE?'
5	1,1	Component class (see below)

After the first data element has been read, or these data inserted conversationally, the runstream proceeds as follows:

P:FEED Q VALUE AND TEMPERATURE DATA:

Q: 1.0239 FEED TEMP: 57.80 DEG C

DEW PT: 91.83 DEG C BUBBLE PT: 81.88 DEG C

(Calculated feed q-value, dew point and bubble point values and the specified feed temperature.)

P:DATA ACCEPTED. START OF OPERATING DIAGRAM CALCS.

If the specified reflux ratio is too low (see subroutine PONSAY), if the column does not contain a stripping section (see subroutine FLOWS) or if any errors are detected during the construction of the operating diagram, the user is informed here (see section 7.26).

P: OPERATING DIAGRAM COMPLETE.  
START OF PACKING CALCULATIONS.

DO YOU WANT TO ADD A SECOND DATA ELEMENT?

U: YES - for a sample data element, see Table 7.3.  
or NO - the conversational runstream continues.

P: THE PROGRAM DATA PACKAGE INCLUDES PACKING DATA FOR  
CERAMIC RASCHIG RINGS AND BERL SADDLES.  
DO YOU WANT TO USE A DIFFERENT TYPE OF PACKING?

U: YES: (See section 7.331)  
NO:

P: DO YOU WANT TO USE (1) RASCHIG RINGS OR (2) BERL  
SADDLES?  
ENTER 1 OR 2

U: 1 or 2; for 2:

P: THE AVAILABLE SADDLE SIZES ARE:  
(1) 12.7, (2) 25.4, (3) 38.1 (MM)  
ENTER APPROPRIATE CODE NO.

U: 1, 2 or 3

P: COMPONENT-DENSITY DATA MUST BE AVAILABLE AS DESCRIBED  
IN THE DISTILLATION PACKAGE MANUAL, SECTION 3.4  
ARE SOLUTION-DENSITY DATA ALSO AVAILABLE?

U: YES or NO (See section 7.34)

P: COMPOUND CLASSES ARE:  
(1) ALCOHOLS AND WATER  
(2) HYDROCARBONS (SAT., UNSAT. AND AROM.) AND ETHERS  
(3) OTHER ORGANICS  
(4) INORGANIC COMPOUNDS  
ENTER CLASS NO. INTO WHICH THE LIGHT AND HEAVY  
COMPONENTS FALL  
EG. FOR WATER AND NITRIC ACID ENTER 1, 4

U: Appropriate reply depending on components in feed.  
(See section 7.34.)

The runstream now continues conversationally regardless of execution mode. PACKED is designed primarily for demand mode execution where the user is expected to direct the design calculations as shown below.

P: DO YOU WANT TO SPECIFY THE COLUMN DIAMETER?

U: NO: See (b) below (See section 7.332)

or YES: (See section 7.332)

(a) Referring to the flooding correlation chart (see Figure 1, Appendix A), the user may select a diameter on the basis of column pressure drop, as shown below.

P: ENTER COLUMN DIAMETER IN METRES.

U: 1.0

P: OPERATING POINT COORDINATES:

ABSCISSA= .049

ORDINATE= .031

IS THIS POINT WITHIN LOADING REGION?

U: NO

P: ENTER ANOTHER DIAMETER.

U: 0.8

P: OPERATING POINT COORDINATES:

ABSCISSA= .049

ORDINATE= .075

IS THIS POINT WITHIN LOADING REGION?

U: YES

Having selected a diameter, the runstream proceeds at (c) below.

(b)

P: SPECIFY VAP. PRESS. DROP AT LOADING FROM FOLLOWING  
(1) 0.041, (2) 0.082, (3) 0.204, (4) 0.408, (5) 0.816,  
(6) 1.224 (KPA/M)  
NOTE: 0.408 (CODE 4) REPRESENTS APPROX. 60% OF FLOODING  
VALUE.  
ENTER APPROPRIATE CODE NO.

U: 4 (See section 7.332)

P: POLYNOMIAL OR ORDER 10 FITTED.  
SUM OF SQUARED ERRORS= .0000589270

OPERATING POINT ON SELECTED LOADING CURVE:  
ABSCISSA= .039  
ORDINATE= .056

FOR GIVEN VAPOUR PRESSUR-DROP:  
SUPERFICIAL GAS VELOCITY= 4961.07 KG/HR.SQM  
RECOMMENDED TOWER DIAM.= .33 M  
RESULTANT SUP. GAS VEL.= 4897.33 KG/HR.SQM  
RESULTANT SUP. LIQ VEL.= 3573.73 KG/HR.SQM  
ARE THESE FIGURES SATISFACTORY?

U: NO (See section 7.333)

If the figures are rejected as shown above the runstream  
continues with:

P: ENTER NEW PRESSURE DROP CODE NO. FROM FOLLOWING  
(1) 0.041, (2) 0.082, (3) 0.204, (4) 0.408, (5) 0.816,  
(6) 1.224 (KPA/M)

U: 3

P: POLYNOMIAL OR ORDER 6 FITTED.  
SUM OF SQUARED ERRORS= .0000086074

OPERATING POINT ON SELECTED LOADING CURVE:  
ABSCISSA= .039  
ORDINATE= .033

FOR GIVEN VAPOUR PRESSUR-DROP:  
SUPERFICIAL GAS VELOCITY= 3815.86 KG/HR.SQM  
RECOMMENDED TOWER DIAM.= .38 M  
RESULTANT SUP. GAS VEL.= 3693.35 KG/HR.SQM  
RESULTANT SUP. LIQ VEL.= 2695.15 KG/HR.SQM  
ARE THESE FIGURES SATISFACTORY?

U: YES

If the results are acceptable the runstream continues at (d) below.

(c) The final output before the result table is a check on the maximum flowrates within the column.

P: FOR GIVEN VAPOUR PRESSURE-DROP:  
SUPERFICIAL GAS VELOCITY= 4698.14 KG/HR.SQM  
RECOMMENDED TOWER DIAM.= .80 M  
RESULTANT SUP. GAS VEL.= 4474.42 KG/HP.SQM  
RESULTANT SUP. LIQ VEL.= 6638.07 KG/HR.SQM  
ARE THESE FIGURES SATISFACTORY?

U: YES : Calculations proceed - see (d) below

NO : (b) above is repeated until the flowrates  
are acceptable. (See section 7.333.)

(d) If no errors are encountered the next program output is the result table. (See section 7.35.)

### 7.33 Optional data inputs

All options relating to data required for the operating diagram are discussed in section 7.23 for Ponchon-Savarit designs. The only difference in the data input sequence for a packed column design is the absence of a Murphree tray efficiency factor.

7.331 Type and size of packing material: Data for four sizes of Raschig rings and three sizes of Berl saddles are included in the package. These data are stored in subroutine PAKING which is called by the mainline routine. They are: CF, E, AMF1, AMF2, ANF1A, ANF1B, ANF2A, ANF2B, PF1, PF2, DS, PHTW, PHSW, SWP and BETA, as defined in the nomenclature. Most of these factors are empirically derived and relate to Shulman's correlations [61]. The selection of packings included should be sufficient for tutorial purposes, but if users wish to consider other materials, the above factors for the particular packing must be derived or extracted from

literature. These can then be inserted in subroutine PAKING by extending the GO TO statement (line 3) to include another statement number, and listing the data at the end of the subroutine. This can then be accessed, without further alteration to the program, as follows:

P: DO YOU WANT TO USE A DIFFERENT TYPE OF PACKING?

U: YES

P: DO YOU HAVE THE PACKING DATA PREPARED?

(a) U: NO

P: REFER TO THE DISTILLATION PACKAGE MANUAL, SECTION 3.3 FOR FURTHER DETAILS.

The run is then terminated.

or (b)

U: YES

P: ENTER CODES FOR PACKING TYPE AND SIZE

U: 3,N where N is either a GO TO counter if more than one set of data is inserted in PAKING, or a dummy value (integer).

The above runstream is contained in subroutine PACTYP. Data presently included in PAKING are tabulated in Appendix A.

7.332 Column diameter calculations: Two options are available to the user:

(1) Specification of diameter in metres. (maximum 2 m)

If the user specifies the diameter, the program computes the resulting superficial vapour velocity, increases this value by 5% as a safety factor, and calculates the corresponding abscissa and ordinate values on the generalised flooding chart. (Figure 1, Appendix A.) The user then determines whether the point thus defined lies within the loading zone indicated on the chart. The resulting pressure drop

must also be checked at this stage, if a limit has been set on this value. Different values can be tried until a satisfactory design is obtained.

The program must be executed in demand mode in order to use this option effectively.

(2) Specification of allowable pressure drop.

An allowable pressure drop is determined firstly from flooding considerations and secondly in practice by economic analysis of the system.

(a) Flooding and loading.

In most industrial applications the upper limit of loading is 0,816 kPa/m (1,00 in H<sub>2</sub>O/ft) and the lower limit is 0,204 kPa/m [17]. A pressure drop of 0,408 kPa/m represents operation at about 60% of flooding velocities and is often used in practical designs. Six curves are shown in Figure 1, Appendix A, and data for each are contained in elements FLOOD1 (0,041 kPa/m) to FLOOD6 (1,22 kPa/m). The user specifies a particular pressure drop corresponding to one of the six curves. The program fits a polynomial to the specified curve through subroutine FDATA, computes the abscissa and ordinate values, and determines the diameter of the column such that the resultant superficial vapour velocity is less than or equal to the value at the specified pressure drop. Diameters less than 2 m are checked at intervals of 0,01 m.

(b) Economic factors

In general, the lower the pressure drop, the lower the operating costs. For a given duty, however, the diameter of the column must increase in order to reduce the pressure drop, and thus the capital costs increase. An economic balance must be found and the appropriate pressure drop specified, bearing in mind flooding and loading considerations.



Other curves can be drawn on the chart representing specific pressure drop values if required. Data representing this curve must be compiled in a data element similar to those currently used by FDATA. These data elements are added to the runstream automatically through the ERTRAN facility. By calling the specific element FLOOD1, say, the specific curve may be considered without further alterations to the program. The user enters code number 1 to access FLOOD1. The FLOOD-data elements are set up as follows:

ND        No. of data pairs  
 $A_k, B_k$      $k=1, ND$  (one pair per line).

A - abscissa; B - ordinate. Values extracted from Figure 1, Appendix A.

7.333 Vapour and liquid superficial flowrates: Before proceeding with the height of packing calculations the program prints out the superficial vapour and liquid flowrates resulting from the specified or calculated column diameter. This facility enables the user to investigate the effect of column diameter and pressure drop on the values of the flowrates and on each other.

7.334 Conversion factors: All conversion factors relating to the operating diagram are discussed in section 7.2. The following factors apply to mass transfer calculations performed in PACKED.

In the original work of Shulman [61], the F.P.S. system is used predominantly with respect to flowrates and densities. In deriving the correlations, however, a variety of units were used as shown in Appendix A, and conversion factors are required to maintain consistency. They are defined in DATA statements in PACKED.

ALBKGH	- $\text{lb ft}^{-2} \rightarrow \text{kg m}^{-2}$
ASFSM	- $\text{ft} \rightarrow \text{m}$
AMLCON,AMVCON	- viscosity values, see sections 7.345 and 7.346.
DLCON,DVCON	- diffusivity values, see sections 7.343 and 7.344.
RHOCON	- density values, see section 7.231.
SIGCON	- surface tension values, see section 7.342.
GC	- gravitational force constant (Figure 1, Appendix A) = $4.18 \cdot 10^8 \text{ lb}_m/\text{lb}_f\text{sec}^2$

#### 7.34 Physical Property Data Package

The following physical properties must be available to the packed column design program PACKED:

- (a) Liquid density of each component and of the binary mixture.
- (b) Surface tension of binary solution.
- (c) Liquid-phase diffusivity.
- (d) Vapour-phase diffusivity.
- (e) Liquid viscosity of binary solution.
- (f) Vapour viscosity of binary solution.
- (g) Density of water.

Physical property data tables in the literature are often sparse, or non-existent, for many substances. To facilitate collation of data for setting up tutorial problems, or studying real situations, a number of procedures for computing each of the above properties are available to the user.

Data required by each routine is added to the runstream, through the UNIVAC ERTRAN facility. All data elements must be set up in free format. Subscript 1 refers to the light component in the following discussion.

7.341 Liquid Density. Subroutine RHOLIQ: Both pure-component and solution density data, as functions of temperature, are required by PACKED. At least one temperature-density data pair for each pure component must be available. Perry [58] is a good source of density values, usually quoted at 20° or 25°C. Pure component densities are computed in one of two ways for each substance of component density specified at some temperature  $T_1$ .

(1) Single values are extrapolated to higher temperatures  $T_2$  using Goldhammer's correlation [67]:

$$\frac{[\rho_L - \rho_V]_2}{[\rho_L - \rho_V]_1} = \left[ \frac{T_C - T_2}{T_C - T_1} \right]^{\text{EXP}}$$

where EXP takes the following values according to the type of compound considered.

Compound	EXP
Alcohols and Water	0,25
Saturated, Unsaturated and Aromatic	
Hydrocarbons and Ethers	0,29
Other Organic compounds	0,31
Inorganic compounds	0,33

Vapour density  $\rho_V$  is neglected, which assumes that systems considered will be at pressures less than  $10^3$  kPa.

(2) Temperature-density data tables, for a few compounds, can be found in the literature [58]. If such a table is available for either component, the routine will determine the component density at any temperature, within the range of the table, using linear interpolation between the data. The temperature range of the data table should span the column temperature range because end-values are taken for column temperatures falling outside the table limits.

Either of the above procedures may be used for either component. The subroutine selects the appropriate method on the basis of data supplied.

Data elements must be compiled as follows:

(a) For all systems..

(1) Element ROLDA1

AMW <sub>1</sub> , TC <sub>1</sub>	Molecular weight and critical
AMW <sub>2</sub> , TC <sub>2</sub>	temperature data (in °C) for each
	component.
NDAT1, NDAT2	Number of temperature-density data
	points available for each component.

(2) Element ROLDA2

T <sub>1</sub> , ρ <sub>1</sub>	NDAT1 data sets for component 1,
T <sub>1</sub> , ρ <sub>1</sub>	in order of increasing temperature
	(in °C).
T <sub>2</sub> , ρ <sub>2</sub>	NDAT2 data sets for component 2,
T <sub>2</sub> , ρ <sub>2</sub>	in order of increasing temperature
	(in °C).

Note: (i) Conversion factor RHOCON in mainline program PACKED, converts the units in which the density data are specified here, to lb ft<sup>-3</sup> for use in the mass-transfer correlations. The present value of RHOCON assumes that density data are specified in g ml<sup>-1</sup>.

(ii) RHOLIQ is called by VISCL and SURTEN to provide liquid-density data for specific correlation expressions in these subroutines. In each case it is assumed that density data are in units of g ml<sup>-1</sup>. If data are supplied to RHOLIQ in units other than g ml<sup>-1</sup>, these routines and factor RHOCON must be adjusted accordingly.

(iii) NDAT1 and NDAT2 can take any value  $\geq 1$  and need not be equal. For linear interpolation the accuracy depends on the number of data pairs available.

(b) For systems for which composition-density data are available.

(3) Element ROLDA3

NDATA,TBASEC	Number of data pairs and base temperature $^{\circ}\text{C}$
XS,RHOS	NDATA composition-density data pairs in order of increasing composition.
XS,RHOS	XS refers to the light component.

Note: Density data units here must be the same as those used in ROLDA2, so that the conversion factor, RHOCN, applies correctly.

7.342 Surface Tension: Subroutine SURTEN: The surface tension of the binary solution, in  $\text{dyne cm}^{-1}$ , is required by the mainline program, for computing liquid holdup factors.

The surface tension of the mixture is determined either directly from a composition-surface tension data table by linear interpolation or from pure-component surface tension values, using Stakhorsky's expression[69] (see section 4.552, Equation (4.22)). Temperature extrapolation of single values is performed using Lovgren's correlation. (See section 4.552, Equation (4.25).)

Pure-component surface-tension data may be provided as a single temperature-surface tension pair, a table of such pairs for a range of temperature sufficient to cover the column temperature profile, or as molecular weight and parachor data from which the surface tension is computed.

If no pure-component or solution, surface tension data can be found in the literature the Sugden parachor must be

computed as shown in Perry [58].

Data elements must be compiled as follows:

(a) For all systems.

Element SIGDA1

1. NSIGM      Counter specifying availability of composition-surface tension data for mixture.  
                  NSIGM=1 - pure component data to be used.  
                  NSIGM=2 - solution data available.
2. AMW<sub>1</sub>, TC<sub>1</sub>, PARA<sub>1</sub>      TC in °C, PARA 1 and 2 - Sugden
3. AMW<sub>2</sub>, TC<sub>2</sub>, PARA<sub>2</sub>      parachors for each component. (See
4. NDAT1, NDAT2      Perry [58] Table 3.317, p.3-240.)

If NSIGM=2 all values above, except TC<sub>1</sub> and TC<sub>2</sub> can be zeros. If temperature-surface tension data is to be used for either component NDAT1, NDAT2 specifies the number of data pairs available for each component. If parachor values are used for either component the corresponding NDAT is set to zero. If parachors are to be used the molecular weights must also be provided. If either parachor is not used it must be set to -1.

Elements SIGDA2 and SIGDA4

These elements are only required if NDAT1 or NDAT2 are greater than 0.

(a) SIGDA2

T<sub>1k</sub>, SIG1<sub>k</sub>      k=1, NDAT1      T in °C in order of increasing T. SIG1 in dyne cm<sup>-1</sup>

(b) SIGDA4

T<sub>2k</sub>, SIG2<sub>k</sub>      k=1, NDAT2      As above.

Element SIGDA3

This element is only required if NSIGM=2 in SIGDA1.

NDATA,TBASEC	Analogous to ROLDA3 described previously; SIGS in dyne $\text{cm}^{-1}$
$\text{XS}_k, \text{SIGS}_k$	$k=1, \text{NDATA}$

Note: Conversion factor SIGCON in the mainline program converts values supplied by SURTEN to units of dyne  $\text{cm}^{-1}$ . The present value (1,0) assumes that data used by SURTEN is in the correct units.

7.343 Liquid Diffusivity: Subroutine DIFLIQ: Liquid diffusivity data are tabulated in Perry [58] for a few binary systems. The subroutine requires a single diffusivity value and the temperature at which it is quoted. Diffusivities are usually quoted as the diffusivity of component A in component B at infinite dilution of component A. For the purposes of the subroutine DIFLIQ, the light component is component 1. The subroutine uses these data to determine the diffusivity of a solution for which the temperature and composition is provided by the calling routine. If the diffusivity of a system can not be found in the literature, a value should be calculated using an expression such as that of Wilke and Chang [71].

Diffusivity must be specified in  $\text{cm sec}^{-1}$ . A conversion factor in the mainline program, DLCON, converts the temperature corrected value to units of  $\text{ft}^2\text{hr}^{-1}$  for use in the mass transfer calculations. The temperature at which the diffusivity value is determined, is supplied in  $^{\circ}\text{C}$ .

Data element required:

Element DLDAT

DL12,TDL - a single diffusivity-temperature data pair in  $\text{cm}^2\text{sec}^{-1}$  and  $^{\circ}\text{C}$  respectively. DL12 refers to the diffusivity of the light component at infinite dilution in the heavy component.

Note: (i) DL12 above is equivalent to  $D_L \times 10^5$ . The factor  $10^{-5}$  is built into the subroutine because most liquid diffusivities fall within the range  $0,5 \times 10^{-5} - 5,0 \times 10^{-5}$ . For example, if the diffusivity of a system, at  $25^\circ\text{C}$ , is computed to be  $2,6 \times 10^{-5}$ , the values 2.6,25.0 are inserted in the data element DLDAT.

(ii) Present value DLCON=3,875, in mainline program PACKED assumes that data used by DIFLIQ are in  $\text{cm}^2\text{sec}^{-1}$ .

7.344 Vapour Diffusivities: Subroutine DIFVAP: Vapour phase diffusivities, required in PACKED for computing Schmidt numbers are calculated here using Arnold's [72] correlation. The molecular weight, molar volume and normal boiling point of each component must be specified.

The diffusivity calculated using this correlation has units of  $\text{cm}^2\text{sec}^{-1}$ . Conversion factor, DVCON, in the mainline converts the value to units of  $\text{ft}^2\text{hr}^{-1}$  where necessary. The molecular weight, molal volume and normal boiling point for each component must be known.

Data element required:

Element DVDAT

AMW <sub>1</sub> , VB <sub>1</sub> , TB	Molecular weight, molal volume,
AMW <sub>2</sub> , VB <sub>2</sub> , TB	and boiling point ( $^\circ\text{C}$ ), specified
	for each component

Note: (i) Molal volume VB in  $\text{cm}^3\text{gmole}^{-1}$  may be computed as  $\text{MW} \cdot \rho_{L_{bp}}^{-1}$ , where  $\rho_{L_{bp}}$  is the component liquid density at its normal boiling point or by summing contributive atomic volumes as tabulated in Perry [58] Table 3-306, p.3-229.

(ii) DVCON=DLCON in mainline. (See section 7.343.)

7.345 Liquid Viscosity: Subroutine VISCL: Two procedures for determining these data, at given temperatures, are included in VISCL.



The first technique estimate component liquid-viscosities using Thomas's equation, which is considered to be the best generalised expression available [75]. This correlation requires the critical temperature and viscosity constant of each component.

$$\mu_L = 0,1167\rho_L^{0,5}10^\gamma$$

where  $\gamma = B(1-T_r)/T_r$

The viscosity constant, B, is calculated by summing contributory atomic terms tabulated in Perry [58] Table 3-321. This expression may be applied satisfactorily for reduced temperatures,  $T_r \leq 0,7$ .

Units of the calculated viscosity value are centipoise. Conversion factor AMLCON is the mainline, converts the units to  $\text{lb ft}^{-1}\text{hr}^{-1}$  as required.

The alternative procedure for computing liquid viscosity data requires the user to provide a temperature-viscosity data table, the temperature range of which should cover the column temperature-profile.

Liquid mixture viscosity is calculated using the component data computed above and the Kendall-Monroe expression [74].

Data elements required.

Element FMUDAL

TC<sub>1</sub>,B<sub>1</sub>

TC<sub>2</sub>,B<sub>2</sub>

NDAT1,NDAT2

Critical temperature data, TC<sub>1</sub>, supplied in °C. If B factors are not used, zeros must be inserted in the element.

NDAT1,NDAT2 are the numbers of data pairs, if any, to be supplied in FMUDA2 and FMUDA3 respectively.

Element FMUDA2

$T_{1k}$ , AMU1<sub>k</sub>    k=1,NDAT1. FMUDA2 is required only if  $B_1 \leq 0$ . Viscosity data in centipoise.

Element FMUDA3

$T_{2k}$ , AMU2<sub>k</sub>    k=1,NDAT2. FMUDA3 is required only if  $B_2 \leq 0$ . Viscosity data in centipoise.

7.346 Vapour Viscosity: Subroutine VISCV: Vapour viscosity data are calculated in a manner similar to that used in VISCL. The same options are available to the user, although the correlation for pure-component data is different. Arnold's relationship is used [77].

The molecular weight, molar volume and normal boiling point of each component must be specified as for DIFVAP. The computed vapour viscosity is in units of micropoise. Conversion factor AMVCON converts the values to units of  $\text{lb ft}^{-1} \text{hr}^{-1}$  in the mainline program PACKED, where necessary.

Vapour-mixture viscosity data are calculated using the relationship of Herning and Zipperer [76].

If tables of temperature-vapour-viscosity data are available, linear interpolation is used to determine required values.

Data elements required.

Element VMUDAl

AMW<sub>1</sub>, VB<sub>1</sub>, TB<sub>1</sub>

AMW<sub>2</sub>, VB<sub>2</sub>, TB<sub>2</sub>

NDAT1,NDAT2

NDAT1,NDAT2 are zero if data tables are not to be used.

### Elements VMUDA2, VMUDA3

These data elements are analogous to FMUDA2 and FMUDA3 and need be set up only if data tables are available. If they are used the data are inserted as  $T_k, AMU_k$  with the table set out in order of increasing temperature values. Vapour viscosity data must be supplied in micropoise, or factor AMVCON in PACKED must be altered so that the units are converted to  $\text{lb ft}^{-1}\text{hr}^{-1}$ .

7.347 Density of Water: Subroutine H2ORHO: Data element H2ORHO supplies a temperature-density data table for water, which need not be altered by the user. Data in this table are in units of  $^{\circ}\text{C}$  and  $\text{g ml}^{-1}$  respectively. To expand the table, or change the units the format must be as follows:

Data element H2ORHO

NDATA	Number of data pairs in table.
$T_k, \rho W_k$	$k=1, \text{NDATA}$ (in order of increasing temperature.)

### 7.35 Result Table

The format of the result table is similar to that for Ponchon-Savarit design of tray columns. Column diameter and heights of packing required in each section are included. Two data tables may be suppressed if not required, to save output time on the terminal. These are

(i) P: DO YOU WANT COMPOSITION, TEMPERATURE AND FLOWRATE DATA TABLES?

U: YES or NO

(ii) P: DO YOU WANT MASS-TRANSFER DATA TABLES?

U: YES or NO

The tables produced in (i) and (ii) above are similar to those shown in Treybal [9] pp. 365 and 367 respectively.

No plotting routine is associated with PACKED.

### 7.36 Error Messages

<u>Origin</u>	<u>Error Message</u>	<u>Description</u>
PACKED	PINCH POINT OCCURS AT POINT..... ON THE EQUILIBRIUM CURVE. THE REFLUX RATIO OR THE DISTILLATE AND BOTTOMS COMPOSITIONS SHOULD BE ADJUSTED ACCORDINGLY.  DIAMETER OF THE COLUMN EXCEEDS 2 METRES SYSTEM UNSUITABLE FOR PACKED COLUMN DISTILLATION	Pinch point detected.      Limit of 2 metres suggested by various workers [59,60,61].
FDATA	ERROR IN MATRIX. JC=...	Error in UNIVAC routine GJR. Consult UNIVAC MATHPAC manual.

Other error messages occurring in PACKED are described in section 7.26.

### 7.37 Dimension parameters and specification limitations

#### (a) Dimension parameters

There are four parameters associated with PACKED. The COMMON variables are the same as those used in BINARY and consequently the same parameters are used for these. They are NPEQ, NPEN and NPPS. Certain variables appear only in PACKED and use the parameter NPAK which must be equal to NPPS. A separate variable name is used for easy reference when scanning the COMMON block.

#### (b) Specification limitation

The same limitations described in section 7.2 apply here. Most of the physical property correlations included in the package are valid only for systems at less than  $10^3$  kPa pressure. The column diameter is limited to a maximum of 2 m [59,60,62].

## 7.4 MULTICOMPONENT DISTILLATION

### 7.41 Application

The multicomponent distillation program MULCOM is a simulation package based on the calculation procedure proposed by Thiele and Geddes [2]. The  $\theta$  method of convergence, proposed by Lyster et al. [3] is used to obtain a converged solution.

The program can simulate columns with one feed stream and no side streams, using a partial reboiler with either a total or partial condenser. The user may either specify equimolal overflow conditions, or consider the effects of stream enthalpy. Tray efficiency can be included through vaporization efficiency data for each component. For any system the following data must be known or estimated:

- (1) Number of components in the feed stream.
- (2) Number of trays in the column.
- (3) Position of feed tray.
- (4) Condenser operating pressure.
- (5) Pressure drop per tray and pressure drop across the condenser.
- (6) Reboiler and condenser temperature.
- (7) Feed rate, composition, temperature and pressure.
- (8) Feed enthalpy if the pressure of the feed is greater than the pressure on the feed tray.
- (9) Type of condenser (total or partial).
- (10) Enthalpy and equilibrium data (see section 7.43 for details).
- (11) Vaporization efficiency data if ideal trays are not assumed.
- (12) Reflux ratio and distillate product withdrawal rate.

The program is designed to be run in demand mode, with interaction between the program and the user. It can also

be run in batch mode providing the runstream is set up as for demand operation.

Subroutines used by MULCOM include:

<u>Subroutine</u>	<u>Purpose</u>
TEACH	Provides a brief description of the program's capabilities.
READBD	Reads all basic problem data, except that required by subroutine ENTHAL.
EQUCON	Computes equilibrium constants for each component on each tray.
QFLASH	Isobaric or isenthalpic flash calculation.
ENERGY	Standard enthalpy calculation routine.
ENTHAL	Optional enthalpy routine supplied by the user. (See section 7.45.)
BUBCAL,DEWCAL	Bubble point and dew point calculations respectively.
FLOWEQ	Flowrate profiles though column assuming McCabe-Thiele conditions.
FLOWNE	Flowrate profiles considering enthalpy effects.
ABSTRP	Absorption and stripping factor calculations.

The following six subroutines perform the Thiele-Geddes tray-to-tray calculations.

VBLV,VDLD	Material balance and equilibrium calculations.
BDCALC,BDCORR	Feed tray matching.
VLRECT,VLSTRP	Stream compositions.
THECAL	Convergence procedure: calculation of $\theta$ in Holland's $\theta$ method.
RUNRES	Abbreviated table of results.
OUTPUT	Detailed table of results showing composition and molar flowrates of components in each stream.

The mainline program serves to link the subroutines together and to direct the overall calculation procedure.

In order to minimise core requirements, MULCOM must be mapped as follows:

```
@MAP DISTFILE.MULCOMM,.MULCOM
```

where element MULCOMM contains the following statements:

```
1      SEG MULCOM1
2      IN MULCOM
3      SEG READINS*, (MULCOM1)
4      IN READIN
5      SEG QFLASHS*, (MULCOM1)
6      IN QFLASH
7      SEG FLOWEQS*, (MULCOM1)
8      IN FLOWEQ
9      SEG FLOWNES*, (MULCOM1)
10     IN FLOWNE
11     SEG ABSTRPS*, (MULCOM1)
12     IN ABSTRP
13     SEG VBLBS*, (MULCOM1)
14     IN VBLB
15     SEG VDLDS*, (MULCOM1)
16     IN VDLD
17     SEG BDCALCS*, (MULCOM1)
18     IN BDCALC
19     SEG THECAL*, (MULCOM1)
20     IN THECAL
21     SEG BDCORRS*, (MULCOM1)
22     IN BDCORR
23     SEG VLSTRPS*, (MULCOM1)
24     IN VLSTRP
25     SEG VLRECTS*, (MULCOM1)
26     IN VLRECT
27     SEG RUNRESS*, (MULCOM1)
28     IN RUNRES
29     SEG OUTPUTS*, (MULCOM1)
30     IN OUTPUT
31     LIB DISTFILE.
```

#### 7.42 Operations of the program

7.421 Graphical representation of results: The plotting program PLOTTER included in the package (see section 7.5) is able to construct the following diagrams describing a system simulated by MULCOM.

- (a) Temperature profile.
- (b) Liquid and Vapour flowrate profiles.
- (c) Distillate composition bar diagram.
- (d) Component composition profiles.

Within a single execution of MULCOM it is possible to simulate a system several times for different values of key variables. Results for a maximum of five runs may be stored on file for subsequent use by the plotting program. On executing PLOTTER the user must specify the run for which a full plot is required. PLOTTER then selects the specified set of data from the data files and constructs diagrams (b), (c) and (d) above. Diagram (a) will automatically superimpose the temperature profiles for all the runs stored on file, on a single set of axes. These points are discussed in detail in section 7.52.

Files for data storage are assigned by MULCOM during the course of the run, if the user indicates that these are required. (See section 7.435.)

7.422 Execution of MULCOM in demand mode with conversational interaction: The following instruction is entered to start the run:

```
@ADD DISTFILE.STARTMC
```

STARTMC contains the following instructions:

```
@ASG,UP PRINTFILE.
```

```
@USE 16,PRINTFILE.
```

```
@XQT DISTFILE.MULCOM
```

PRINTFILE is assigned to the run for storing detailed results tables than can be sent to the line printer on termination of the run. (See section 7.44.) This file is not used by the plotting program.



The run then proceeds as follows. All instructions and data are entered in free format and it is suggested that bulky tables, such as equilibrium or enthalpy data, be inserted in data-elements for addition to the runstream as required.

The program begins with the standard format.

P: ANSWER ALL QUESTIONS YES OR NO  
ALL INPUT IN FREE FORMAT.

DO YOU KNOW HOW TO USE THIS PROGRAM?

If the user response is NO the following description is printed out and the run is terminated.

P: THIS IS A MULTICOMPONENT DISTILLATION SIMULATION PROGRAM.  
THE CALCULATION PROCEDURE USED IS THAT FIRST PROPOSED BY THIELE AND GEDDES, COMBINED WITH HOLLANDS THETA METHOD OF CONVERGENCE.  
THE PROGRAM IS LIMITED TO SYSTEMS WITH ONE FEED AND NO SIDE STREAMS.

REFER TO THE DISTILLATION PACKAGE MANUAL, SECTION 4, FOR FURTHER INFORMATION.

For a response of YES, the run continues with

P: DO YOU WANT TO ADD A DATA ELEMENT?

As for other programs, a response of YES requires the addition of a previously prepared data element. A sample element is shown in Table 7.41.

If the conversational facility is required the runstream proceeds as follows. Only the program output is shown. Data is entered in free format or by addition of data-elements where options are available or further explanation is required, reference is made to a subsequent section.

TABLE 7.41

PROBLEM-DATA ELEMENTS FOR EXECUTION OF MULCOM

1	6	No. of components
2	14	*No. of trays
3	9	*Feed tray number
4	150.0, 3	Condenser pressure and code.
5	10.0, 1	Tray pressure drop and code.
6	25.0, 1	Condenser pressure drop and code
7	236.0, 2, 115.0, 2	Estimate of reboiler and condenser temperatures with corresponding codes.
8	METHANE	] Component names
9	ETHANE	
10	N-PROPANE	
11	N-BUTANE	
12	N-PENTANE	
13	N-HEXANE	
14	100.0	FTOT
15	0.03, 0.07, 0.15, 0.33, 0.30, 0.12	] Z <sub>i</sub> i=1, NC
16	82.2	
17	YES	T <sub>F</sub> 'ARE ENTHALPY EFFECTS TO BE CONSIDERED?'
18	NO	'IS FEED PRESSURE COLUMN PRESSURE?'
19	PAR	Partial condenser specified.
20	YES	** 'DO YOU WANT TO USE YOUR OWN ENTHALPY ROUTINE?'
21	1	Antoine constants specified.
22	6.61184, 389.93, 266.0	] Antoine constants
23	6.80266, 656.40, 256.0	
24	6.82973, 813.20, 248.0	
25	6.83029, 945.9, 240.0	
26	6.85221, 1064.63, 232.0	
27	6.87776, 1171.53, 224.366	
28	YES	] Convergence limit specification
29	0.0001	
30	NO	] Efficiency data specification (See section 7.434.)
31	2	
32	0.91, 0.92, 0.95, 0.98, 1.05, 1.09	
33	1.2, 38.46	R, DTOT

\* Trays are numbered from the bottom, excluding the reboiler and condenser

\*\* See Table 7.42.

P: ENTER NO. OF COMPONENTS IN FEED.

U: integer value

P: ENTER NO. OF TRAYS IN COLUMN  
EXCLUDE CONDENSER AND REBOILER

U: integer value

P: ENTER FEED TRAY POSN. (NO.FROM BOTTOM,REBOILER=0)

U: integer value

P: ACCEPTED PRESSURE UNITS: 1=MMHG,2=PSIG,3=PSIA,4=KPA,  
5=ATM.  
ENTER CONDENSER PRESSURE AND UNIT CODE. (EG. 760.0,1)

P: ACCEPTED PRESSURE DROP UNITS: 1=MMHG,2=PSI,3=KPA  
ENTER PRESSURE DROP PER TRAY AND UNIT CODE.

P: ENTER PRESSURE DROP ACROSS CONDENSER AND UNIT CODE.

P: ACCEPTED TEMPERATURE UNITS: 1=DEG C,2=DEG F  
ENTER ESTIMATE OF REBOILER AND CONDENSER TEMPERATURES  
WITH UNIT CODES. (EG. 100.0,1,50.0,2)

The pressure inputs have been discussed in section 7.1,  
the temperature input format is shown. The example represents:

reboiler temp. 100°C  
condenser temp. 50°F

P: ENTER NAMES OF COMPONENTS. (ONE PER LINE,MAX 30  
CHARACTERS EACH.)

P: ENTER FEED RATE. (KG MOLES/HR.)

P: ENTER FEED STREAM COMPOSITIONS. (MOL FRACTIONS)

P: ENTER FEED TEMPERATURE. (DEG C)

Component names entered as shown in section 7.1. The  
feed stream data are entered as real numbers in free format.

P: ARE ENTHALPY EFFECTS TO BE CONSIDERED?

U: YES - for options see section 7.431

or NO

P: TOTAL OR PARTIAL CONDENSER?  
ENTER TOT OR PAR.

U: TOT or PAR

At this point enthalpy data are required, if these effects are considered (see section 7.432). Once these data have been inserted, or if enthalpy is not considered, the run proceeds as follows:

P: ENTER EQUIL. DATA AS DESCRIBED IN THE MANUAL, SECT.4,  
OR ADD A DATA ELEMENT.

For options see section 7.433.

P: DO YOU WANT TO SPECIFY THE THETA CONVERGENCE LIMIT?

U: YES

P: ENTER CONVERGENCE LIMIT.

U: real value (e.g. 0.0001)

This is the limit set on the  $\theta$  convergence procedure. If the limit is not supplied by the user a value of 0.00005 is adopted.

P: ARE PERFECT TRAYS ASSUMED?

For options, see section 7.434.

P: ENTER REFLUX RATIO AND DISTILLATE WITHDRAWAL RATE.

U: real values - reflux ratio, distillate rate.

P: DO YOU WANT GRAPHICAL OUTPUT? (SEE MANUAL SECT.4)

For options, see section 7.435.

This completes the input of basic data required to solve a problem. The run is then started and identified through the following two statements.

P: TO PROCEED, TYPE GO OR TO STOP THE PROGRAM, TYPE END.

U: GO

P: ENTER RUN NO. AS: NO.,DAY,MONTH. (INTEGER VALUES)

U: run identification (e.g. 1,4,10)

The acceptance of all data by the program is acknowledged by the statement:

P: START OF COMPUTATION.

Values of  $\theta$  are printed out with each iteration, enabling the user to follow the course of the convergence. On completion of the calculations an abbreviated table of results is printed on the terminal. The runstream then continues:

P: DO YOU WANT A FULL PRINT-OUT?

On the basis of the printed result table, the user may require a full set of results for the run. These are written to the previously assigned file PRINTFILE. (See section 7.44.) The program then continues with:

P: TO PROCEED,TYPE GO OR TO STOP THE PROGRAM,TYPE END.

The user may now terminate the run or solve the problem again for different values of certain selected variables shown below.

U: GO

P: ENTER RUN NO. AS: NO.,DAY,MONTH. (INTEGER VALUES)

U: new run identification

P: ENTER REFLUX RATIO AND DISTILLATE RATE FOR NEW RUN.

P: ENTER FEED TEMPERATURE. (DEG C)

If enthalpy data was supplied originally the feed stream pressure may be specified higher than the column pressure, and its enthalpy supplied, as discussed in section 7.431. The final variable that may be altered is:

P: ENTER FEED TRAY POSN. (NO. FROM BOTTOM,REBOILER=0)

The computation then proceeds as before. Any number of runs may be performed in this manner within a single execution, but a maximum of five full result tables may be requested. This limit is to discourage the generation of excessive amounts of printout.

The program terminates as follows:

P: TO PROCEED,TYPE GO OR TO STOP THE PROGRAM,TYPE END.

U: END

P: SUCCESSFUL EXECUTION COMPLETED.

If full result tables have been stored a set of instructions for sending these to the line printer is also printed out here.

#### 7.43 Input options

7.431 Feed conditions: The feed may be specified as being at the same pressure as the feed tray, or at a higher pressure. A different flash calculation procedure is used in each case. If the pressure of the feed is above that on the feed tray, the flash calculation is isenthalpic and the feed enthalpy must be specified in kJ/kg mole. This procedure obviates the need to provide enthalpy data at both the column pressure and the feed pressure.

The program responds as follows if the user considers enthalpy effects:

P: IS FEED PRESSURE GREATER THAN COLUMN PRESSURE?

U: NO or YES

P: ENTER FEED ENTHALPY. (KJ/KGMOLE)

U: real value

The main runstream continues, as shown in section 7.422, with

P: TOTAL OR PARTIAL CONDENSER?

7.432 Enthalpy data: There are two procedures available for including enthalpy effects in problems:

(a) Library routine:

The user may use one of the functions built into the library enthalpy routine. These functions were chosen because they have been used by a number of workers [4,18,21, 24,25] and thus appropriate coefficients for several problems are readily available. These are three functions available:

$$(1) (H_{i,j})^{\frac{1}{2}} = a_i + b_i T + c_i T^2 \quad T_j \text{ in } ^\circ R \text{ (NENTH=1)} \quad (7.4.1)$$

$$(2) H_{i,j} = a_i + b_i T + c_i T^2 \quad T_j \text{ in } ^\circ F \text{ (NENTH=2)} \quad (7.4.2)$$

$$(3) H_{i,j} = a_i + b_i T + c_i T^2 \quad T_j \text{ in } ^\circ R \text{ (NENTH=3)} \quad (7.4.3)$$

where  $T = T_j/100$  and  $T_j$  is the temperature on stage  $j$ . Similar equations define the liquid enthalpy  $h_{i,j}$ .

Stream enthalpies are computed as

$$H_j = \sum y_{i,j} H_{i,j} \quad (7.4.4)$$

$$\text{and } h_j = \sum x_{i,j} h_{i,j} \quad (7.4.5)$$

The coefficients for the above functions quoted in the literature are derived from enthalpy data in Btu/lb mole. An informal conversion factor ENCONV converts all calculated enthalpies to SI units of kJ/kg mole. Another factor CONFAC converts heat duties from Btu/hr to kilowatts. If coefficients cannot be found they must be derived by fitting one of the two above functions to available data.

It should be noted that the functions are written in terms of  $T_j/100$  rather than  $T_j$ . This facilitates free format data input in the following way:

Suppose a set of coefficients for one of the functions have been derived or extracted from the literature in terms of  $T_j$ . A typical set could be:  $a_i = 687,0$ ;  $b_i = 0,9964$ ;

$c_i = 1,9395 \times 10^{-4}$ . These would be supplied to the program, in free format, as follows:

U: 687.0,99.64,1.9395

The factor  $T_j/100$  will set the decimal point correctly. This procedure was proposed by Holland [21].

(b) User routine:

The user may provide an enthalpy routine incorporating any convenient procedure for computing enthalpies. The routine must conform to certain specifications outlined in section 7.45 which ensure that the routine provides a facility equivalent to the library routine. An example of such a routine, incorporating Kobe's heat capacity functions, is included in the package. (See section 7.45.)

The routine reads in its own data using the UNIVAC ERTRAN facility, and communicates with the calling program through the calling list. Data input is thus not considered in the run-stream.

The derivation of coefficients for the library functions is a time consuming process. It is often more convenient when setting up tutorial problems to use available data directly or to employ some expression for which coefficients are known.

The program determines the procedure required by the user as follows:

P: DO YOU WANT TO USE YOUR OWN ENTHALPY ROUTINE?

U: NO or YES

P: IT IS ASSUMED THAT THE ROUTINE CONFORMS  
TO SPECIFICATIONS SET OUT IN SECT. 4.5  
IN THE MANUAL.

\* (see below)

P: REFER TO MANUAL, SECT. 4.32, FOR ENTHALPY ROUTINE CODES.  
ENTER CODE NO. OF LIBRARY ENTHALPY ROUTINE REQUIRED.



U: integer value (NENTH) corresponding to selected function as shown above.

P: ENTER ENTHALPY DATA AS DESCRIBED IN THE MANUAL, SECT. 4.32 OR ADD A DATA ELEMENT.

For a sample enthalpy data block, see Table 7.42.

\* (see below)

Coefficients for the chosen function must be supplied as follows:

- (a) One set of  $a_i, b_i, c_i$  per line
- (b) In order from light component to heavy component.
- (c) Liquid enthalpy coefficients for all components first, then all vapour enthalpy coefficients.

It is often more convenient to add these data in a block data element.

\* The program then continues with the main runstream. (See section 7.422.)

TABLE 7.42

ENTHALPY DATA FOR LIBRARY ROUTINE ENERGY

The following data are extracted from Holland [21] and correspond to NENTH=2, for a five component system.

1	0.,2521.92,175.417	] Coefficients for liquid enthalpy function $a_i, b_i, c_i$
2	0.,3345.,150.	
3	0.,2960.,400.	
4	0.,3681.33,283.334	
5	0.,3845.,250.	
6	7488.65,1750.51,79.273	] Coefficients for vapour enthalpy function $a_i, b_i, c_i$
7	9592.01,1843.56,221.433	
8	8002.89,4382.70,-401.587	
9	11645.77,2770.55,156.345	
10	12004.88,3168.66,67.456	

7.433 Equilibrium data: Four functions are available in subroutine EQUCON for computing vapour-liquid equilibrium constants.

- NEQUIL=1 (1) Antoine's Equation combined with Raoult's and Henry's Laws.
- NEQUIL=2 (2) Equilibrium constant expressed as a function of temperature.
- NEQUIL=3 (3) Relative volatility and a key component equilibrium constant expressed as function of temperature, with the remaining equilibrium constants determined as the product of these.
- NEQUIL=4 (4) Equilibrium constant expressed as another function of temperature.

The latter three techniques have been included for the same reasons as the library enthalpy routine. If coefficients for these are not available, their derivation is tedious and time consuming. Antoine constant by comparison are readily available for a wide variety of compounds. The use of this equation is, however, highly idealised.

Data for each function must be supplied as follows. Coefficients must be adjusted for functions in terms of  $T/100$  or  $T/1000$  as specified below.

(1) Antoine's Equation

Calculation procedure:

(a) From Antoine's equation

$$P_{V_{i,j}} = 10^{[A_i + B_i / (C_i + T_j)]} \quad T_j \text{ in } ^\circ\text{C} \quad (7.4.6)$$

From Raoult's law

$$y_{i,j} = P_{V_{i,j}} x_{i,j} / P_T$$

From Henry's law

$$y_{i,j} = K_{i,j} x_{i,j}$$

$$(b) \quad K_{i,j} = P_{V_{i,j}} / P_T \quad (7.4.7)$$

Data in free format, preferably in a data element (see Table 7.41):

1 (NEQUIL)

$A_i, B_i, C_i$   $i=1, NC$  (Antoine constants, one set per line light-heavy component order)

### (2) Function of Temperature

This function has been used by Seppala and Luus [18]

$$K_{i,j} = \exp[a_i - (b_i/T) + c_i T] \quad T=T_j/100, \quad (7.4.8)$$

$T_j$  in  $^{\circ}R$

Data in free format, coefficients adjusted for  $T=T_j/100$ .

2 (NEQUIL)

$a_i, b_i, c_i, O^*$   $i=1, NC$  ( $K_{i,j}$  coefficients, one set per line, light-heavy component order)

### (3) Relative volatility and key component K value

Calculation procedure:

$$K_{b_k} = a + bT + cT^2 + dT^3 \quad (7.4.9a)$$

$$\alpha_{i,j} = e_i + f_i T + g_i T^2 \quad T=T_j/100, T_j \text{ in } ^{\circ}F \quad (7.4.9b)$$

and  $K_{i,j} = \alpha_{i,j} \cdot K_{B_k}$  where k signifies the base component.

Data in free format, coefficients adjusted for  $T=T_j/100$

3 (NEQUIL)

$a, b, c, d$  ( $K_{b_k}$  coefficients)

$e_i, f_i, g_i, O^*$   $i=1, NC$  ( $\alpha_{i,j}$  coefficients, one set per line, light-heavy component order)

Note:  $\alpha_{i,k}$  coefficients are 1., 0., 0., 0.

#### (4) Function of Temperature

This function has been used by Holland [21]

$$K_{i,j} = T_j [a_i + b_i T + c_i T^2 + d_i T^3]^3 \quad \begin{matrix} T=T_j/1000 \\ T_j \text{ in } ^\circ R \end{matrix} \quad (7.4.10)$$

Data in free format, coefficients adjusted for  $T=T_j/1000$

4 (NEQUIL)

$a_i, b_i, c_i, d_i$  \*  $i=1, NC$  ( $K_{i,j}$  coefficients, one set per line, light-heavy component order)

\*The same READ statement is used for all three sets of coefficients. The fourth array must therefore be filled in each case with real values or zeros as required.

The program then continues with the main runstream.

7.434 Tray efficiency: Tray efficiency is considered in terms of component vaporization efficiency factors defined as

$$E_{i,j} = \frac{Y_{i,j}}{K_{i,j} x_{i,j}} \quad (7.4.11)$$

These factors should be extracted from literature or determined experimentally for simulating real systems. Vaporization efficiency factors have not been determined for many systems, however, and the experimental procedure for obtaining realistic data is extremely complex.

Efficiency factors may be estimated in order to demonstrate the effects of tray performance on a system, but this procedure can not be used to describe real columns in practice [22]. Data may also be derived from real and ideal equilibrium data using the above equation.

If the user's reply to the main runstream question is NO the program responds:

P: ADD EFFICIENCY DATA AS DESCRIBED IN THE MANUAL.

Data may be supplied in two ways, preferably through data element:

(1) Efficiency factors known for each component on each tray

1 (counter MEFF defining this option)

$E_{1,NP1}, E_{2,NP1}, \dots E_{NC,NP1}$  (Condenser stage, factors  
for all components from  
light to heavy on one line)

$E_{1,0}, E_{2,0}, \dots E_{NC,0}$  (Reboiler stage)

(2) Efficiency factors known for each component on one stage; assumed equal on all stages.

This is not the case in practice, but is useful for demonstrating the basic concept of efficiency.

(See Table 7.41)

2 (counter MEFF defines this option)

$E_{1,n}, E_{2,n}, \dots E_{NC,n}$  (Efficiency factors for each  
component on stage n)

Suitable tables of data for a number of systems are given in the literature [23,41,44,45].

The program continues with the main runstream after reading this data. If perfect trays are assumed, efficiency data are not required, and the factors are set to 1,0. This section is then bypassed.

7.435 Data storage for graphical representation: If the graphs described in section 7.421 are required, files must be assigned for storage of data. A maximum of five data sets may be stored during a single execution. (See section 7.44.)

The runstream proceeds as follows if the user replies YES to the main runstream question:

P: DO YOU KNOW HOW TO USE THE PLOT ROUTINE?

U: YES or NO → return to main runstream.

P: ENTER DATA-FILE NAMES.

Four files are required and the name of each must be six characters in length. File names are typed in on one line, spaced with commas.

U: DFILE1,DFILE2,DFILE3,DFILE4

These are then automatically assigned by the program through the ETRAN facility. The program returns to the main runstream.

7.436 Conversion factors: Conversion factors CONFAC and ENCONV are defined in the enthalpy subroutine ENERGY and are discussed in section 7.235. No other conversion factors are used by the standard program MULCOM. Factors required by a user enthalpy subroutine must be included in the subroutine itself. (See section 7.45.)

#### 7.44 Program result tables

An abbreviated table of results is printed out for each converged solution as it is completed. The user may require more detailed result tables for certain runs and these can be stored on file until the program is terminated. A maximum of five such detailed tables may be stored during one execution. Corresponding data required for plotting purposes is stored simultaneously, if graphical representation is required. (See section 7.435.)

The file in which detailed tables are stored must be assigned before the execution is started as shown in section 7.422.

In the event of an execution-terminating error, or if detailed result tables are not requested by the user, the program automatically deletes PRINTFILE on termination. This is done by adding element FINISH to the runstream through the ERTRAN facility. FINISH contains the following statements:

```
@FREE PRINTFILE.  
@DELETE PRINTFILE.
```

If tables are stored but not required, the user must add FINISH to the runstream after termination.

The plotting procedure associated with MULCOM is described in section 7.53.

#### 7.45 Optional enthalpy-calculation subroutine

If the user has enthalpy data available in a form other than those accepted by the standard routine, ENERGY, a separate subroutine may be compiled and used. If the subroutine is named ENTHAL and conforms to the specifications set out below, no alterations to the mainline program, or other subroutines are necessary. Program variables are transferred through the calling list and any data required by the correlations are read through the ERTRAN facility.

The calling program indicates the type of calculation required through counter NEN. The routine is called by the mainline program to determine stream enthalpies throughout the column, or by subroutine QFLASH during the flash calculations.

The enthalpy subroutine must be capable of calculating vapour and liquid enthalpies given composition and temperature data for the streams. It must also compute condenser and reboiler heat duties and an overall feed enthalpy. Enthalpy data returned to the calling program must be in SI units.

An example of a compatible routine is shown below. It may be used directly with MULCOM providing coefficients for the Kobe [55] heat capacity function are available. Alternatively the user may insert other correlations or develop an entirely new routine along the same lines. The heat capacity and latent heat correlations used here are discussed in section 7.234. Both may be extended to multi-component systems that may be assumed to be ideal [55,58]. Heat of mixing is not considered in this routine, however. The FORTRAN text of ENTHAL is attached, and a description of the calculations performed is shown below.

<u>Line No.</u> *	<u>Program Action</u>
1,2	The calling list is compulsory. Variables are defined below.
3-7	Internal variables dimensioned directly. Calling list variables dimensioned by parameters MC and MP from calling program. Parameter MCC $\geq$ MC.
10	Conversion factors: CONFAC - Energy units to kW. ENCONV - Enthalpy units to kJ.
11	Read key for UNIVAC system.
12	Read statements bypassed after first calling of ENTHAL.
14-27	Data required by enthalpy correlations read in from data element ENTHDATA. The data shown here are required by the correlations discussed previously (section 7.234). A sample data element is shown in Table 7.43.
(14)	Coefficients - (Equation (7.23)).
(15)	Exponents relating to coefficients BC, CC and DC.

\*See FORTRAN text between p.268 and 269.



- (16) Latent heats boiling points and critical temperatures.
- (17) Molecular weights required if data are not supplied in molar units. (See section 7.234.)
- (18) Conversion factors to ensure consistent units in enthalpy calculations.
- (19) Base temperature, set at 25°C, adjusted for correlations, temperature units by TCONV.
- (22-26) Correction of coefficients.
- 29,30 Calculation selected by QFLASH.
- 31-49 DO loops 12 and 13 compute stream enthalpies throughout the column: loop 12 - trays; loop 13 - components.
- 50-57 Calculation of condenser duty.
- 58-75 Calculation of feed stream enthalpy.
- 76 Overall heat balance defines reboiler duty.
- 77,78 DM and BM required in FLOWNE.  
DM - total heat removed per mole of distillate product.  
BM - total heat removed per mole of bottoms product.
- 79,80 Conversion of heat duty units to kW.

The remainder of the routine computes enthalpies for QFLASH for superheated or supercooled feeds, as described above. (NEN = 1)

- 84-87 TF = specified feed temperature, TD and TB are the dew point and bubble point of the feed respectively.

88-108 DO loops 21 and 22 compute enthalpies of the feed stream as:

- (1) Superheated vapour at temperature TF (S1)
- (2) Supercooled liquid at temperature TF (S2)
- (3) Dew point vapour at temperature TD (S3)
- (4) Bubble point liquid at temperature TB (S4)

S1, S2, S3 and S4 are used in QFLASH to determine the feed q-value.

TABLE 7.43

ENTHALPY DATA ELEMENT ENTHDATA

The data shown here correspond to the components shown in Table 7.41.

1	4.750,1.200,0.303,-2.630	Coefficients for Kobe function [55].
2	1.648,4.124,-1.530,1.74	
3	-0.966,7.279,-3.755,7.580	
4	0.945,8.873,-4.38,8.36	
5	1.618,10.85,-5.365,10.10	
6	1.657,13.19,-6.844,13.78	
7	-2,-5,-9	Exponents for coefficients*
8	121.87,-161.6,-82.5	
9	11.87,-88.9,32.1	$LH_i, T_{BP_i}, T_{C_i}$
10	101.76,-42.1,96.8	
11	92.09,-0.5,153.0	
12	85.38,36.08,-197.2	
13	80.48,68.74,234.8	
14	1.0,16.04	$CPMW_i, LHMW_i$
15	1.0,30.07	
16	1.0,44.09	
17	1.0,58.12	
18	1.0,72.15	
19	1.0,86.17	CPCONV, LHCONV, TCONV
20	1.8,1.8,273.	

\* See section 7.23 and Table 7.22

Calling list variables for subroutine ENTHAL

<u>Variable</u>	<u>Type</u>	<u>Description</u>	<u>Dim</u>	<u>Para</u>
AL,AV	real	Moles of liquid and vapour in feed.		
BM,DM	real	Total heat removed per mole of bottoms and distillate product.		
B,D,F	real	Total bottoms, distillate and feed rates		
HD	real	Distillate enthalpy		
HFEED,QF	real	Feed enthalpy - QF* used in printing results. (kJ/kg mole)		
HL,HV	real	Stream enthalpies of liquid and vapour leaving each tray	1	MP
NC	int	Number of components		
NF1,NP1,NP2	int	Feed tray number, total number of trays and condenser number		
NV	int	NV=1 Total condenser NV=2 Partial condenser		
QC,QR	real	Condenser and reboiler heat duties (kW)		
TP	real	Tray temperature (°C)	1	MP
VR	real	Vapour rates leaving each tray	1	MP
X,Y	real	Composition of liquid and vapour streams leaving each tray.	2	MC,MP
XF,XY	real	Composition of each phase in feed stream	1	MC
TB,TD,TF	real	Bubble point, dew point and specified temperature of feed		
NEN	int	Counter selects type of calculation required (from QFLASH or MULCOM).		
S1,S2,S3,S4	real	Phase enthalpies for feed stream (see above)		
NTIME	int	Counter causing subroutine data to be read.		
MC,MP	int	Parameters defining size of arrays from calling program.		

All dimensioned variables with parameter MCC are internal variables. Variables used internally are defined in section 7.234.

\*QF may be multiplied by CONFAC to give the feed heat duty in kW.

SUBROUTINE ENTHAL(AL,AV,BM,DM,B,D,F,HD,HFEED,HL,HV,NC,NF1,NP1,NP2,  
INV,QC,QF,QR,TP,VR,X,XF,Y,YF,TB,TD,TF,NEN,S1,S2,S3,S4,NTIME,MC,MP)  
CALCULATION OF VAPOUR AND LIQUID ENTHALPY DATA,FEED ENTHALPY,  
CONDENSER DUTY AND REBOILER DUTY.

THIS ROUTINE REQUIRES THE FOLLOWING DATA FOR EACH COMPONENT:

(1) COEFFICIENTS FOR A MOLAL HEAT CAPACITY EQN. OF THE FORM:

CP=A+B\*T+C\*T\*\*2+D\*T\*\*3 (T IN DEG K,CP IN CAL/GMOL,DEG K)

(K OBE K.A.,THERMOCHEMISTRY FOR THE PETROCHEMICAL INDUSTRY,,

PETRM. REFNR.,JAN 1949-NOV 1954)

SEE ALSO HOUGEN WATSON & RAGATZ,"CHEMICAL PROCESS PRINCIPLES,,

PART 2,P. XXVI-XXIX.

(2) LATENT HEAT AT COMPONENT BOILING PT. (IN CAL/GM.)

(3) BOILING PT. OF COMPONENT (IN DEG C)

(4) CRITICAL TEMPERATURE OF COMPONENT (IN DEG C)

(5) MOLECULAR WT OF COMPONENT

(6) CP,LH AND TEMP CONVERSION FACTORS,AND BASE TEMPERATURE.

FOR DETAILS REFER TO THE MANUAL,SECTION 4.

CONVERSION TO S.I. UNITS (JOULES & WATTS) IS BUILT IN.

PARAMETER MCC=30

DIMENSION AC(MCC),BC(MCC),CC(MCC),CPMW(MCC),DC(MCC)

DIMENSION HL(MP),HV(MP),LH(MCC),LHMW(MCC)

DIMENSION TBP(MCC),TC(MCC),THOLD(4),TP(MP),VR(MP)

DIMENSION X(MC,MP),XF(MC),Y(MC,MP),YF(MC)

REAL LH,LHCONV,LHNMW,LHT

I FORMAT( )

DATA CONFAC,ENCONV/0.00027778,2.326/

NCR=8

IF(NTIME,Eq.99) GO TO 100

CALL ERTRAN(6,"ADD'E DISTFILE,ENTHDATA ,")

READ(NCR,1) (AC(I),BC(I),CC(I),DC(I),I=1,NC)

READ(NCR,1) NBCP,NCCP,NDCCP

READ(NCR,1) (LH(I),TBP(I),TC(I),I=1,NC)

READ(NCR,1) (CPMW(I),LHNMW(I),I=1,NC)

READ(NCR,1) CPCONV,LHCONV,TCONV

TBASE=25.+TCONV

READ(NCR,1,END=101) IDUM

101 CONTINUE

0003  
0004  
0005  
0006  
0007  
0008  
0009  
0010  
0011  
0012  
0013  
0014  
0015  
0016  
0017  
0018  
0019  
0020  
0021

SUBROUTINE ENTHAL

0001  
0002

```

0022 DO 11 I=1,NC
0023 BC(I)=BC(I)*10.**NDCP
0024 CC(I)=CC(I)*10.**NCCP
0025 DC(I)=DC(I)*10.**NDCP
0026
0027
0028
0029
0030
0031
0032
0033
0034
0035
0036
0037
0038
0039
0040
0041
0042
0043
0044
0045
0046
0047
0048
0049
0050
0051
0052
0053
0054

11 CONTINUE
100 CONTINUE
NTIME=99
IF(NEN.EQ.1) GO TO 200
IF(NEN.EQ.2) GO TO 103
CALCULATION OF VAPOUR AND LIQUID STREAM ENTHALPY.
DO 12 J=1,NP2
SUMHL=0.0
SUMHV=0.0
T=TP(J)+TCONV
DO 13 I=1,NC
AFACT=AC(I)*(T-TBASE)
BFACT=BC(I)*(T**2.-TBASE**2.)/2.
CFACT=CC(I)*(T**3.-TBASE**3.)/3.
DFACT=DC(I)*(T**4.-TBASE**4.)/4.
CP=AFACT+BFACT+CFACT+DFACT
H=CP*CPMW(I)*CPCONV
SUMHL=SUMHL+H*X(I,J)
TFACT=(TC(I)-TP(J))/(TC(I)-TBP(I))
LHT=LH(I)*(TFACT**D.38)*LHMW(I)*LHCONV
SUMHV=SUMHV+(H+LHT)*Y(I,J)
13 CONTINUE
HL(J)=SUMHL*ENCONV
HV(J)=SUMHV*ENCONV
12 CONTINUE
CALCULATION OF CONDENSER DUTY.
HD=HL(NP2)
IF(NV.GT.1) GO TO 102
TOTAL CONDENSER OPERATION.
QC=VR(NP1)*(HV(NP1)-HL(NP2))
GO TO 103
102 CONTINUE

```

SUBROUTINE ENTHAL

```

C      PARTIAL CONDENSER OPERATION.
QC=VR(NP1)*HV(NP1)-(VR(NP1)-D)*HL(NP2)-D*HV(NP2)
HD=HV(NP2)
103 CONTINUE
C      CALCULATION OF FEED STREAM ENTHALPY.
SUMFL=0.
SUMFV=0.
T=TF+TCONV
DO 14 I=1,NC
  AFAC=AC(I)*(T-TBASE)
  BFAC=BC(I)*(T**2.-TBASE**2.)/2.
  CFAC=CC(I)*(T**3.-TBASE**3.)/3.
  DFAC=DC(I)*(T**4.-TBASE**4.)/4.
  CP=AFAC+BFAC+CFAC+DFAC
  H=CP*CPMW(I)*CPCONV
  SUMFL=SUMFL+H*XF(I)
  TFAC=(TC(I)-TF)/(TC(I)-TBP(I))
  LHT=LH(I)*(TFAC**0.38)*LHMW(I)*LHCONV
  SUMFV=SUMFV+(H+LHT)*YF(I)
14 CONTINUE
HFEE=(AV/F*SUMFV+AL/F*SUMFL)*ENCONV
IF(NEH.EQ.2) GO TO 999
QFTOT=F*HFEE
C      CALCULATION OF REBOILER DUTY.
QR=D*HD+B*HL(1)+QC-QFTOT
DM=HD+QC/D
BM=HL(1)-QR/B
C      CONVERSION TO S.I. UNITS
QC=QC*CONFAC
QR=QR*CONFAC
QF=HFEE
GO TO 999

```

SUBROUTINE ENTHAL

C 200 CONTINUE  
CALCULATION OF ENTHALPY FOR Q VALUE DETERMINATION.

THOLD(1)=TF  
THOLD(2)=TF  
THOLD(3)=TD  
THOLD(4)=TB

DO 21 J=1.4  
T=THOLD(J)+TCONV

SL=0.  
SV=0.

DO 22 I=1.4  
AFACT=AC(I)\*(T-TBASE)

BFACT=BC(I)\*(T\*2.-TBASE\*\*2.)/2.

CFACT=CC(I)\*(T\*3.-TBASE\*\*3.)/3.

DFACT=DC(I)\*(T\*4.-TBASE\*\*4.)/4.

CP=AFACT+BFACT+CFACT+DFACT

H=CP\*CPMW(I)\*CPCONV

SL=SL+H\*XF(I)

TFACT=(TC(I)-THOLD(J))/(TC(I)-TBP(I))

LHT=LH(I)\*(TFACT\*\*0.38)\*LHMW(I)\*LHCONV

SV=SV+(H+LHT)\*YF(I)

22 CONTINUE

IF(J.EQ.1) S1=ENCONV\*SV

IF(J.EQ.2) S2=ENCONV\*SL

IF(J.EQ.3) S3=ENCONV\*SV

IF(J.EQ.4) S4=ENCONV\*SL

21 CONTINUE

999 CONTINUE

RETURN

END

0083  
0084  
0085  
0086  
0087  
0088  
0089  
0090  
0091  
0092  
0093  
0094  
0095  
0096  
0097  
0098  
0099  
0100  
0101  
0102  
0103  
0104  
0105  
0106  
0107  
0108  
0109  
0110  
0111

SUBROUTINE ENTHAL

### 7.46 Error Messages

<u>Origin</u>	<u>Error Message</u>	<u>Description</u>
MULCOM	DATA NOT ACCEPTABLE ILLEGAL SPECIFICATION ENCOUNTERED.	Error in data input of reply to program question. Program is terminated.
FLOWEQ	ERROR: NEGATIVE LIQUID FLOW IN STRIPPING SECTION CAUSED BY FEED CONDITION Q= ..., LIQUID FLOW= ENTER AN INCREASED REFLUX RATIO TO CONTINUE RUN, OR ENTER 1000. FOR TERMINATION	Mass balance failure. Increase reflux ratio or feed q-value.
FLOWNE	ERROR: NEGATIVE LIQ (OR VAP) FLOW LEAVING TRAY ... POSSIBLE CAUSES OF ERROR... (1)... (2)... (3)...	Enthalpy calculation error probably due to one of the factors suggested by the message or to inconsistent data.
QFLASH	ERROR INFLASH CALCULATION SEARCH FOR FEED TEMPERA- TURE HAS FAILED VAPOUR FRACTION OF FEED= VALUE SHOULD LIE BETWEEN 0.0 and 1.0. Q= ... FEED TEMP= .. DEG C DEW PT= DEG C BUBBLE PT= DEG C	This error could be caused by inconsistent equilibrium or enthalpy data, or a fault within either the flash or enthalpy routines.
READIN	ONLY TOT OR PAR IS ACCEP- TABLE RETYPE NOW.	This is the only data input error searched for. The user is expected to ensure that dimension parameters and data formats are correct.



#### 7.47 Dimension parameters and specification limitations

(a) Dimension parameters.

- (1) MC - number of components
- (2) MP - number of trays

These are the only two parameters used. They are included in the COMMON block of each subroutine, and in subroutine ENTHAL's calling list. Any internal arrays required by ENTHAL must be dimensioned separately within the routine.

(b) Specification limitations.

- (1) Equilibrium and enthalpy data.

Antoine equation is valid for systems at less than  $10^3$  kPa. The coefficients of other functions used are usually quoted at a given system pressure which is taken at the condenser pressure. The coefficients are assumed to be unaffected by the pressure drop across the trays in the column.

- (2) Condenser type.

A total or partial condenser from which a single phase feed is withdrawn, may be considered.

#### 7.5 PLOTTING PROGRAMS

##### 7.51 Application

The plotting program PLOTTER that is included in the distillation package enables the user to construct the following diagrams on a CALCOMP plotter.

1. Multicomponent systems

- (a) Temperature profile.
- (b) Liquid and vapour flowrate profiles.
- (c) Bar diagram showing compositions of distillate product.

(d) Composition profiles through the column.

## 2. Binary systems.

(a) McCabe-Thiele diagrams for:

(i) Batch distillation operating under constant reflux conditions.

(ii) Batch distillation operating under variable reflux conditions.

(iii) Continuous distillation systems.

(b) Ponchon-Savarit enthalpy-concentration and x-y diagrams for continuous distillation systems.

The program requires access to the standard CALCOMP plotting subroutines and other software. Hardware facilities at the University of Cape Town include a CALCOMP plotter and a Tektronics graphics terminal.

For direct use of the drum-plotter from a standard terminal, the program must be mapped as follows:

(a) @MAP PLOTFILE.PLOTMAP,.PLOTTER

where element PLOTMAP. contains the following statements:

```
IN PLOTFILE.PLOTTER
IN PLOTFILE.PLOTTS
IN PLOTFILE.PLOTFS
IN PLOTFILE.PLOTDS
IN PLOTFILE.PLOTCS
IN PLOTFILE.PLOTBT
IN PLOTFILE.PLOTMT
IN PLOTFILE.AXIS
IN PLOTFILE.MAXIS
LIB CALCOMP*SUBR
```

For execution of PLOTTER from the graphics terminal the following map procedure is used:

@MAP PLOTFILE.PLOTNM,.PLOTTER

where element PLOTNM contains the following statements:

```
IN PLOTFILE.PLOTTER
IN PLOTFILE.PLOTTS
IN PLOTFILE.PLOTFS
IN PLOTFILE.PLOTDS
IN PLOTFILE.PLOTCS
IN PLOTFILE.PLOTBT
IN PLOTFILE.PLOTMT
IN PLOTFILE.AXIS
IN PLOTFILE.MAXIS
LIB TEKFAST*CALPREV,TEKFAST*TCSAG
```

Plots can be sent to the drum plotter, directly from the graphics terminal, without remapping by entering the instruction.

```
@XQT TEKFAST*CALPREV.TOPLOTTER
```

#### 7.52 Executing PLOTTER in demand mode

The structure of the plotting system at U.C.T. is such that plotting programs are best executed from a terminal. On a standard terminal the CALCOMP map statement ((a) above) automatically queues the job on tape to be plotted as a batch job. The graphics terminal may be used to run the job in demand mode with the plot displayed on the terminal screen (map (b) above). In this way the diagram can be examined before it is sent to the plotter, and discarded if it is unsatisfactory.

PLOTTER is a conversational program, and the following runstream is required to execute it in demand mode. If the user has signed off after executing either BATCH or BINARY, the data file BDPLLOT must be assigned with

```
@ASG,AZX BDPLLOT.
```

Data files generated by MULCOM are added to the runstream as shown in section 7.53 below.

```
U: @XQT PLOTFILE.PLOTTER.
```

```
P: ENTER JOB NO. AS: NO.,DAY,MONTH(INTEGER VALUES).
```

U: 1,2,3 (Run number identifying job)

P: ENTER TYPE OF PLOT REQUIRED.

U: Integer value (code number as tabulated below)

<u>Code</u>	<u>Plot generated</u>	<u>See section</u>
1	Multicomponent plots	7.53
2	McCabe-Thiele diagram batch distillation Constant reflux	7.541
3	McCabe-Thiele diagram batch distillation Variable reflux	7.542
4	McCabe-Thiele diagram continuous distillation	7.551
5	Ponchon-Savarit diagram continuous distillation	7.552

On entering the plot code number the conversational runstream continues as shown below.

#### 7.53 Multicomponent plots

(U: 1)

P: ENTER TOTAL NUMBER OF RUNS TO BE PLOTTED.

U: Integer value (This is the number of runs completed within a single execution of MULCOM as described in section 7.4. The maximum value is 5.)

P: ENTER NUMBER OF RUN FOR WHICH FULL PLOT IS REQUIRED.

U: Integer value (The program will superimpose the temperature profiles for all the runs on one diagram, but the remaining diagrams will show data for a single run only.)

P: ENTER NUMBER OF TRAYS IN COLUMN.

U: Integer value (exclude reboiler and condenser)

P: ENTER NUMBER OF COMPONENTS IN FEED.

U: Integer value

P: ENTER DATA FILE NAMES.

U: File names used in execution of MULCOM, in correct order. e.g. from section 7.435.

DFILE1,DFILE2,DFILE3,DFILE4

The files are then assigned automatically through ERTRAN. The final output from the computer is generated by EXEC 8.

i.e. P: PLOT ENTERED IN QUEUE.

The user may then sign off with

U: @FIN

The plot is then collected at the Computer Centre (turn-around time ~ 24 hours).

#### 7.54 Binary batch distillation

##### 7.541 Constant reflux operation:

(U: 2)

No further instructions are required. On completion of the plot the computer responds

P: PLOT ENTERED IN QUEUE.

U: @FIN.

##### 7.542 Variable reflux operation:

(U: 3)

As for constant reflux operation above.

#### 7.55 Binary continuous distillation

##### 7.551 McCabe-Thiele design

(U: 4)

As for section 7.541.

### 7.552 Ponchon-Savarit design:

(U: 5)

P: ENTER UNITS OF ENTHALPY AXIS IN ENTH-CONC DIAGRAM.  
NORMALLY THESE ARE DEL POINT VALUES ROUNDED CON-  
VENIENTLY. ENTER UPPER LIMIT FIRST. (REAL MODE,  
FREE FORMAT).

U:  $\Delta_D$ ,  $\Delta_B$  enthalpy values

These values are printed on the result table in E12.6 format as

P: ENTHALPY DIAGRAM DEL POINTS:

DEL D = ..... KJ/KGMOLE

DEL B = ..... KJ/KGMOLE

Consider DELD = .341099+05 and DELB = -.213842+05. The  
user should enter the following values in the plotting run-  
stream

U: 35000.,-22000.

On completion of the plot the computer responds

P: PLOT ENTERED IN QUEUE.

U: @FIN.

For batch mode execution of PLOTTER, see section 7.63.

### 7.56 Diagram titles

All titles written on diagrams constructed by PLOTTER  
are contained in separate elements which are accessed through  
the ERTRAN facility as required. This structure has been  
adopted to enable users to alter the titles without having  
to edit the main program. The format of the elements, as  
shown below, must be rigidly adhered to, however. Each  
element contains certain alphanumeric labels and integer values  
corresponding to the number of characters in each label. The  
elements are stored in PLOTFILE.

(a) Multicomponent diagrams:

TITLE ELEMENT: HEADING  
TOTAL NUMBER OF RUNS PLOTTED:  
FULL PLOT PRODUCED FOR RUN NO:  
JOB NUMBER:  
29,30,11

TITLE ELEMENT: TITLES  
TEMPERATURE PROFILE  
19  
VAPOUR AND LIQUID FLOWRATES  
27  
DISTILLATE COMPOSITIONS  
23  
COMPOSITION PROFILE  
19  
THIELE-GEDDES PROGRAM  
21  
MULTICOMPONENT DISTILATION  
27  
0.1,9.5,0.1,9.0,0.1,8,5  
0.1,9.5,0.1,9.0,0.1,8.5  
0.1,6.5,0.1,6.0,0.1,5.5  
0.1,6.5,0.1,6.0,0.1,5.5

TITLE ELEMENT: AXISTITLE  
TRAY NUMBER  
11  
TEMPERATURE DEG C  
17  
TRAY NUMBER  
11  
LIQ. AND VAP. RATES KGMOL/HR  
28  
COMPONENT NO.  
13  
MOLE FRACTION  
13  
TRAY NUMBER  
11  
MOLE FRACTION  
13  
8.,8.,5.,5.

(b) Binary system diagrams:

TITLE ELEMENT: BBVTTL  
BINARY BATCH DISTILLATION  
VARIABLE REFLUX RATIO  
X-Y DIAGRAM  
JOB NUMBER  
25,21,11,11

TITLE ELEMENT: BBCTTL  
BINARY BATCH DISTILLATION  
CONSTANT REFLUX RATIO  
X-Y DIAGRAM  
JOB NUMBER:  
25,21,11,11

TITLE ELEMENT: AXISMT  
LIQUID COMP., MOLE FRACTION  
VAPOUR COMP., MOLE FRACTION  
26,26

TITLE ELEMENT: AXISPS  
X,Y ,Liq and vap mole frac  
ENTHALPY, KJ/KGMOLE SOLUTION  
26,27

TITLE ELEMENT: BCTITL  
BINARY CONTINUOUS DISTILLATION  
MCCABE-THIELE DIAGRAM  
JOB NUMBER:  
30,21,11

TITLE ELEMENT: PSTITL  
BINARY CONTINUOUS DISTILLATION  
PONCHON-SAVARIT DESIGN  
ENTHALPY-CONCENTRATION DIAGRAM  
JOB NUMBER:  
30,22,30,11  
BINARY CONTINUOUS DISTILLATION  
PONCHON-SAVARIT DESIGN  
X-Y DIAGRAM  
30,22,11

## 7.6 SUMMARY OF GENERAL OPERATING PROCEDURE FOR ALL PROGRAMS

### 7.61 Before signing on at a terminal or compiling a batch job card deck:

(1) Determine whether the program is able to solve the problem and compile a table of all data required.

(2) Check that the conversion factors included in program will convert the units in which data is supplied to those required in the calculations. These conversion factors are defined in DATA statements in the mainline routines and are described in the relevant sections of this manual.

(3) Check that the dimensioning parameters are large enough to accommodate the system considered. These are



defined in PARAMETER statements in the COMMON block of each program.

(4) Set up data elements containing equilibrium, enthalpy or other bulky data tables required during the run. Physical property data required by PACKED must be inserted in specific elements as shown in section 7.34.

7.62 After signing on at a terminal in demand mode:

(1) Assign the data files required by the plotting program, if this facility is to be used. The necessary instructions are given in each section of the manual.

(2) Initiate the run through an @XQT instruction.

(3) Follow the program's conversational instruction carefully.

(4) Execute plotting program only after analysing the program results.

7.63 When compiling a card deck for batch mode execution:

(1) Ensure that the following instructions are included and that the data cards correspond exactly with the program's instructions. The conversational data input facility should be suppressed.

@RUN card

@ASG,AX DISTFILE..

@XQT DISTFILE.program name

Data cards, or @ADD DISTFILE. data element

@FIN

(2) If the plotting facility is executed in batch mode the following instructions are required. The program must be mapped as shown in section 7.51(a).

@RUN card

@ASG,AX BDPL0T (required for binary systems only)

@XQT PLOTFILE.PLOTTER

Data cards for required plotting procedure (see section 7.5).

Note: The computer solution to a problem must always be checked for errors before submitting a plotting job, to prevent wastage of plotting time.

#### 7.64 General rules for data input

(1) All numerical data must be supplied in free format.

(2) Names of components must not exceed 30 characters (5A6 format).

(3) All questions must be answered YES or NO unless otherwise indicated.

(4) Column pressure may be specified in any of the following units. (1) mmHg, (2) psig, (3) psia, (d) kPa, (e) atm.

(5) Flowrates must be specified in kg moles/hr and temperature data in  $^{\circ}\text{C}$  unless otherwise indicated.

(6) All single composition values in binary systems refer to the light component. In multicomponent systems component 1 is the lightest and component NC is the heaviest.

(7) Tray numbering convention: Reboiler = 0; Column Trays = 1-N; Condenser = N+1; Feed tray position = Tray NF from bottom of column.

(8) Data tables based on composition must list data from  $x=0$  to  $x=1$ .

For information concerning the program algorithms, FORTRAN text, or other computer software, the user is referred to the original thesis.

## CHAPTER 8

### PROGRAM ANALYSIS AND APPLICATIONS

#### 8.1 COMPUTER PROGRAMS AS TEACHING AIDS

A comprehensive set of computer programs designed specifically for undergraduate courses in chemical engineering has been published by CACHE. These programs have been developed by a number of workers, however, and do not form an integrated package with interchangeable subroutines. Many of them are examples of programs that must be written by students in tutorial assignments and thus do not cover their specific subjects in much detail. Also, certain topics, such as McCabe-Thiele designs, are considered in several programs whereas others, such as Ponchon-Savarit designs, have been ignored. It was thus decided to develop a fully integrated set of programs covering all the important aspects of distillation considered in the undergraduate course in mass transfer operations at U.C.T.

Students are not required to write programs in this course, but rather to use the programs available to study the behaviour of distillation systems in detail. The programs were therefore to be economical in computer time and core storage requirements, as well as being simple to execute and able to handle a wide variety of problems. As a teaching aid, the program should be able to guide a user by means of descriptive comments and error messages. The previous discussion of each program has shown how error messages and conversational data input facilities have been included to assist users who have no more than a basic knowledge of the principles involved. The techniques used to reduce the computer time and core requirements of each program are discussed below. Conventions adopted to facilitate future alterations to the program texts are also described.

## 8.2 PROGRAMMING CONVENTIONS

In order to increase the versatility of the programs certain options are included that would not normally be required for undergraduate tutorials. Options such as providing a multicomponent enthalpy routine or data for different packing materials require the user to alter the FORTRAN text of the standard routines. To assist users in altering the programs or adapting them for other applications, the following conventions have been used in all mainlines and subroutines.

### (1) READ and WRITE keys:

Read and Write keys are defined in DATA statements at the beginning of each mainline program. They are included in the COMMON blocks and, where necessary, in the calling lists of subroutines without common variables. Names and definitions of these keys, that are common to all programs unless otherwise stated, are listed below.

<u>Name</u>	<u>Integer Used</u>	<u>Purpose</u>
NCR	8	General READ key on UCT's UNIVAC 1106
NTP	5	General WRITE key on UCT's UNIVAC 1106
NG1	11**,12*	Write to file for plotting use
NG2	13*	" " " " " "
NG3	14*	" " " " " "
NG4	15*	" " " " " "
NLP	16*	Write to file for detailed printout of results

\* Used in multicomponent program MULCOM only.

\*\* Used in binary programs BATCH and BINARY.

Data files are not required by the program PACKED.

(2) Statement numbers - general use:

<u>Range</u>	<u>Application</u>
1-9	READ FORMAT statements
1000-9999	WRITE FORMAT statements
10-99	DO LOOPS
100-999	GO TO and IF statement addresses.

(3) Statement numbers - specific use:

The statement numbers indicate the section of program in which they fall and, in general, the type of operation being performed according to the following convention. This has been adopted to facilitate tracing errors or altering the programs. The convention applies throughout the package.

<u>Range</u>	<u>Application</u>
50-59,500-599,5000-5999	Input of data or selection of options.
80-89,800-899,8000-8999	Compilation and writing to file of data for plotting purposes.
90-99,900-999,9000-9999	Error messages and program termination procedures.

All other numbers are used in blocks for each stage of the particular calculation procedure, e.g. in BATCH.

10-19,100-199,1000-1999	Variable reflux operation case.
20-29,200-299,2000-2999	Constant reflux operation case.
40-49,400-499,4000-4999	Printing results.

(4) Choice of variable names and other nomenclature:

The nomenclature list defines most of the variables, appearing in the programs. In general, the same variable names are used throughout the package although in some cases it was convenient to use a binary program variable name for a different purpose in the multicomponent program.

(a) DO loop counters are chosen to correspond to text nomenclature.

I - component number

J - tray number in loops covering the whole column

N - tray number in loops covering the rectifying section only

M - tray number in loops covering the stripping section only

K - general counter used for loops not categorized as above.

(b) Variable names are chosen to describe their application.

<u>Alphanumeric character</u>	<u>Association when used as first or second character in a name</u>
B	Bottoms product
D	Distillate product
F	Feed stream
G	Vapour or steam entering bottom of column
H	Enthalpy
L	Liquid stream (real variables)
P	Pressure
Q	Feed q-value or heat duty
R	Rectifying section
S	Stripping section
T	Temperature
V	Vapour
X	Liquid composition
Y	Vapour composition
Z	Feed composition

In addition variable names similar to standard textbook nomenclature are used where possible.

(5) Tray numbering:

The standard tray numbering convention designates the reboiler as stage 0 and the condenser as stage  $N+1$  [11]. It is computationally convenient to operate on a base number of 1, rather than 0, therefore the internal tray numbering relates to the conventional system as follows:

<u>Conventional system</u>		<u>Internal system</u>
Reboiler	0	1
Bottom tray	1	2
Feed tray	$N_F$	$NF1 = NF+1$
Top tray	$N$	$NP1 = N+1$
Condenser	$N+1$	$NP2 = N+2$

Thus the figures supplied as the feed tray position and the top tray number are numerically equal to the tray below the feed and the tray below the top tray respectively. The user is not concerned with this convention when operating the program. Input and output tables conform to the standard convention.

### 8.3 COMPUTER CORE REQUIREMENTS

As computer systems with large core storage capacities are not always available for teaching purposes, it is, in general, necessary to keep core requirements of teaching programs to a minimum. Although the UNIVAC 1106 machine at U.C.T. is a teaching computer with a large core area (265 K), the annually increasing load imposes limits on the size of program that may be run in demand mode. For this reason, the core requirement of each program has been limited as follows:

(1) Number of source cards reduced by use of subroutines for repetitive calculations.

(2) Use of dimensioning parameters that can be altered by the user to minimise the core requirement for a given

problem. Separate parameters are used for different blocks of variables so that all arrays need not necessarily be expanded for a particular application.

(3) Use of a single array for more than one purpose in the calculations. This has not been extensively applied, however, because most of the arrays in the programs are required throughout the calculations. Arrays have not been equivalenced for the same reason.

(4) Use of the segmented mapping facility on the UNIVAC system at U.C.T. This facility causes subroutines associated with a program to be transferred to core only when they are required. Core savings of 10% or more are realized in this way.

The core requirements of each program compare favourably with those of CACHE programs, which are assumed to be suitable for teaching purposes. CACHE programs with similar capabilities and using similar calculation procedures have been chosen for comparison.

(a) Program BATCH

CACHE program VI-5 is a batch distillation program assuming negligible holdup and McCabe-Thiele conditions. It considers both constant and variable reflux operations in columns with a known number of trays. Mathematical techniques used include linear interpolation and integration by quadrature. The program requires 16 K of core on an IBM 360 machine.

BATCH requires 10 K of core in order to handle columns with up to 25 trays. This number has proved to be sufficient for all tutorial problems found in the literature. Segmented mapping did not reduce the core requirements appreciably in this case (i.e. < 1 K).



Additional facilities offered in BATCH include

- (1) Partial or total condenser operation.
- (2) Specification of average distillate composition for constant reflux operation.
- (3) Three methods for providing equilibrium data. (Linear interpolation of equilibrium data is the only technique offered in the CACHE program.)
- (4) Writing data to file for plotting of x,y diagram for the system.

(b) Program BINARY

Several programs dealing with McCabe-Thiele designs have been published by CACHE. Program VI-6 is similar to the McCabe-Thiele facility in BINARY in that it calculates the number of trays required to achieve a given separation, allowing for a Murphree tray efficiency factor. In addition, the effect of liquid entrainment may be studied and a Newton-Raphson flash calculation is included. Equilibrium data are determined using equilibrium constants expressed as a function of temperature. This program requires 30 K of core on an IBM 360.

BINARY includes both a McCabe-Thiele and a Ponchon-Savarit design procedure and requires 18 K of core with a standard map, and 15 K when segmented. This appears to be a reasonable requirement by comparison with the CACHE program. Additional facilities offered by BINARY include:

- (1) Ponchon-Savarit design procedure.
- (2) Open steam replacement of a partial reboiler in McCabe-Thiele designs.
- (3) Three methods for providing equilibrium data.
- (4) Operating diagram data for plotting purposes.
- (5) Total or partial condenser.

Other CACHE programs require less core than VI-6, but none are as comprehensive. Ponchon-Savarit design is not considered by any of the CACHE programs.

(c) Program PACKED

CACHE do not consider distillation in packed columns. A program for designing packed absorption towers, using mass transfer coefficient correlations, has been published, and this is used for comparative purposes. The program (VI-2) is similar to PACKED in its approach to the design of packed columns. Column diameter and packed height are obtained, based on the characteristics of the packing and the physical properties of the fluids. The mass transfer correlations used, however, restrict the program to systems of light hydrocarbons and water, and a packing of Raschig rings. The program requires 40 K and a CDC 6600 machine.

PACKED requires ~ 27 K of core with a standard map and 23 K when segmented. The correlations and data included enable the program to consider a wide variety of systems, and packings of either Raschig rings or Berl saddles. PACKED also allows the user to specify either the column diameter or an allowable pressure drop and then computes the corresponding unknown.

Like PACKED, CACHE program VI-2 was written as a teaching aid to be used by students. It is, however, considerably larger and is unsuitable for distillation column design.

(d) Program MULCOM

A number of programs dealing with multicomponent systems have been published by CACHE. Program VI-13 uses the Thiele-Geddes calculation procedure combined with Holland's  $\theta$  method of convergence to rate distillation columns of known configuration. The program will consider ideal systems only and computes equilibrium data using Antoine's equation and Raoult's law. The program is thus similar to MULCOM in many

respects. It does not, however, consider enthalpy effects, tray efficiency or two phase feeds. The CACHE program requires 43 K on an IBM 360 machine whereas MULCOM requires 21 K with a standard map and 17 K when segmented. These requirements apply when MULCOM is dimensioned to accommodate ten components and 25 trays which are sufficient for most tutorial problems. If the dimension parameters are increased to thirty components and fifty trays, 32 K of core is required. For teaching purposes, however, this would be excessive.

A program capable of handling complex columns has been published by CACHE. The program (VI-16) uses the Wang Henke [25] algorithm, which is a tridiagonal matrix technique combined with Muller's iterative convergence procedure. This program considers enthalpy effects, column heat losses and an arbitrary number of single phase feed and side streams. A disadvantage of the program is that equilibrium and enthalpy data are computed using polynomials in temperature for which coefficients must be provided. Data for the sample problem shown in the CACHE manual was extracted from the original paper of Amundson and Pontinen [4]. The determination of coefficients for arbitrary tutorial problems is usually time consuming and for this reason optional procedures are offered in MULCOM. Tray efficiency considerations and flash calculations are not included in the CACHE program either. The program requires 29 K of core on a Univac 1108, which is 6 K smaller than MULCOM for systems of the same size. It is, however, more difficult to use for general teaching purposes because of the limited data options.

It can therefore be said that the programs developed in this study offer more facilities and require no more core than comparable CACHE programs.

#### 8.4 CPU TIME REQUIREMENTS

The question of run times is discussed far more extensively in the literature than that of core requirements. In industry, where large computers are usually available, a major cost is the CPU time required to run a program. In a teaching environment, where the number of jobs run is often very high, a minimal use of CPU time is imperative. Direct comparison of run times of programs executed on different machines is not valid because some CPU's are faster than others, and it is not always clear whether a quoted run time refers to CPU time or total time. The mathematical techniques used in this study were chosen:

- (1) to produce stable algorithms,
- (2) to correspond to material presented in the course, and
- (3) to reduce the CPU time required.

The first two points have been discussed previously but each identifiable technique will now be discussed briefly with respect to all three considerations.

(a) Newton's method of convergence.

Also known as Newton's method of tangents or, when applied to multivariable systems, as the Newton-Raphson procedure, this technique is known to be unstable if the starting value is not a reasonable estimate of the end result. This potential instability means that the procedure should be applied only after a careful study of the behaviour of the function in the region of the desired root. If the function is not well behaved, a poor initial estimate often leads to spurious results or divergence. In this study the method is applied to bubble point and dew point calculations, and in Holland's  $\theta$  method. In these applications the functions are well behaved and, in the case of temperature calculations

pure component boiling points are satisfactory starting values. For the  $\theta$  method (see Chapter 5) any positive value close to zero will ensure stable convergence. It is generally acknowledged that this method is the fastest available providing the stability is checked. Newton's method is introduced in the early part of the course when bubble point calculations are discussed. It is also introduced as a computer technique in an elementary course on computer programming.

(b) Direct iteration: (Subroutine SEARCH).

Direct iteration guarantees convergence providing that changes are made in the correct direction. Changes are linear, however, and thus the method is comparatively slow. It has been applied in this study where stability is of paramount importance, and where the behaviour of the function is not always known. Iteration run times of less than a few milliseconds for BATCH and BINARY, where it is used extensively, indicate that it is sufficiently fast for these applications. The method has been applied for similar problems in several CACHE programs. Trial and error methods are the basis of most chemical engineering design procedures, and direct iteration is therefore discussed early in the curriculum.

(c) Linear Interpolation.

This is another basic procedure that is used chiefly in the binary programs. For well-behaved functions such as ideal system equilibrium curves, the accuracy is satisfactory providing the interpolation interval does not exceed 0,05 mole fraction. The alternative of fitting a polynomial to the data is far more time consuming, and the improved accuracy does not justify the increased CPU time requirement. The use of subroutine FDATA is discussed below. A run time of 0,6 sec (CPU) for solving a McCabe-Thiele complex column problem (section 8.5.2) indicates that the procedure is satisfactory. It is used in BATCH and BINARY, as in certain CACHE programs,

(VI-5, VI-8), for stepping off stages on the x-y diagram, as well as other applications discussed previously.

(d) Fitting of polynomials: (Subroutine FDATA).

A least squares procedure is used to fit a polynomial to a specified flooding and loading correlation curve. Linear interpolation cannot be applied satisfactorily to data extracted from a log-log plot. In addition the least squares facility, which tests up to 10th order polynomials, enables users to consider more specific pressure drop correlations if data are available. Although slower than linear interpolation, it is able to handle complex functions with greater accuracy.

(e) Holland's  $\theta$  method.

This has been discussed extensively in Chapter 5. It is acknowledged as one of the fastest convergence procedures for multicomponent distillation systems [34], and the Thiele-Geddes- $\theta$  combination has proved to be stable for a wide variety of simple distillation systems. In this application the inclusion of two convergence criteria reduces the CPU time required for specific problems, where one criterion is met before the other. Limiting the maximum value of  $\theta$ , and preventing negative values, also improved the stability.

(f) Forcing procedures.

Three forcing procedures were found to improve the rate of convergence in MULCOM, and thus reduce the CPU time required:

(1) Limitation of absolute change in flowrate from one iteration to the next (subroutine FLOWNE) as proposed by Lyster [33].

(2) Limitation of absolute change in temperature profile from one iteration to the next (subroutines BUBCAL and DEWCAL) as proposed by Lyster [33].

(3) Averaging of bubble point and dew point on each tray to determine new temperature profile. Previous workers [18, 21,33] have suggested the use of either dew point or bubble point calculations through the column to establish new temperature profiles. Holland also found that some systems converged faster if dew point calculations were used in the rectifying section and bubble point calculations were used in the stripping section of the column. In this study it has been found, in general, that the use of both dew point and bubble point calculations throughout the column, with subsequent averaging of the two values on each tray, reduces the total number of iterations required for convergence by up to 30%. This procedure proved to be stable for all problems attempted.

(g) Simpson's Rule numerical integration.

This method is introduced in an elementary course on computer programming and is thus familiar to undergraduates. It is acknowledged as an accurate numerical method although the absolute accuracy depends largely on the integration interval. From a comparison of run times of the CACHE program VI-5, which integrates the batch distillation equations by quadrature, and BATCH, it appears that the Simpson's Rule procedure is faster. This conclusion is based on the similarity of the other calculation procedures (e.g. stepping off stages using linear interpolation). Comparative run times are

	CACHE VI-5 (IBM 360-67)	BATCH (UNIVAC 1106)
Constant Reflux:	5-10 mins	8-10 secs.
Variable Reflux	1-2 mins	6-8 sec.

It should be noted, however, that CACHE run times are quoted as "typical" run times, and it is not clear whether they are CPU or Total times.

Typical CPU run times for each program developed in this study have been listed in the summary sheet (section 6) of each chapter. These times are all within the limits of 20 secs and 30 K core set by the Computer Centre U.C.T. for demand mode jobs at the time of writing.

## 8.5 PROGRAM APPLICATIONS

In this section solutions generated by each program are discussed and compared where possible with data presented in the literature.

### 8.51 Batch distillation - program BATCH

The program solutions to the following problems demonstrate the operation and capabilities of BATCH.

#### Problem 8.51(a) (Variable reflux operation)

(Data from Treybal [52] p.335.)

Fifty kgmoles of a solution of  $\text{CS}_2$  and  $\text{CCl}_4$  is charged to a batch fractionator kettle. The composition of the charge is 0,4 mole fraction  $\text{CS}_2$  and a distillate containing 0,95 mole fraction  $\text{CS}_2$  is required. The column contains 6 ideal trays and is limited to a maximum boilup rate of 10 kgmoles/hr.

Assuming a total condenser and a column pressure of 101,3 kPa, determine the time required to achieve a final residue concentration of 0,08 mole fraction  $\text{CS}_2$ .

TABLE 8.511

#### SOLUTION TO PROBLEM 8.51(a)

BATCH DISTILLATION OF A BINARY SOLUTION.  
VARIABLE REFLUX OPERATION.

FEED COMPONENTS: (1) CARBON DISULPHIDE  
(2) CARBON TETRACHLORIDE

INITIAL REFLUX RATIO: 1.487 (1.75)  
FINAL REFLUX RATIO: 9.800 (10.88)



VAPOUR BOIL-UP RATE: 10.00 KGMOLE/HR  
BATCH CYCLE TIME: 7.441 HRS (8.06)  
TOTAL VAPOUR BOILED UP DURING CYCLE: 74.412 KGMOLE (80.6)  
RESIDUE: 31.609 KGMOLE (31.6)  
RESIDUE COMPOSITION: .0800 (0.08)  
TOTAL DISTILLATE WITHDRAWN: 18.391 KGMOLE (18.4)  
DISTILLATE COMPOSITION: .9500 MOLE FRAC (0.95)

The problem is solved using Antoine constants to generate equilibrium data. Table 8.511 shows a cycle time of 7.44 Hrs and Figure 8.511 is a McCabe-Thiele diagram, constructed by PLOTTER, showing the initial and final operating conditions.

The results of Treybal [52] agree well as shown by the values in parenthesis in Table 8.511. Discrepancies are probably due to differences in equilibrium data. The system is slightly easier to separate if Antoine constants correctly describe the equilibrium.

Solving the same problem assuming a constant relative volatility of 3.5, yields a cycle time of 5.01 Hrs. This is expected because the shape of the equilibrium curve changes to increase the distance between it and the operating line. Thus the separation is easier and lower reflux ratios may be used (initial  $R = 0.89$ ; final  $R = 5.06$ ).

Problem 8.51(b) (Constant reflux operation)

One hundred kgmoles of  $\text{CCl}_4$  and toluene (composition 0.3 mole fraction  $\text{CCl}_4$ ) is to be distilled at 1 atm, at a constant reflux ratio of 5.0. The column contains 5 theoretical trays and is limited to a vapour boilup rate of 8 kgmoles/hr. A total condenser is used.

1) Determine the quantity of distillate product that may be withdrawn if the average concentration required is 0.90 mole fraction  $\text{CCl}_4$ .

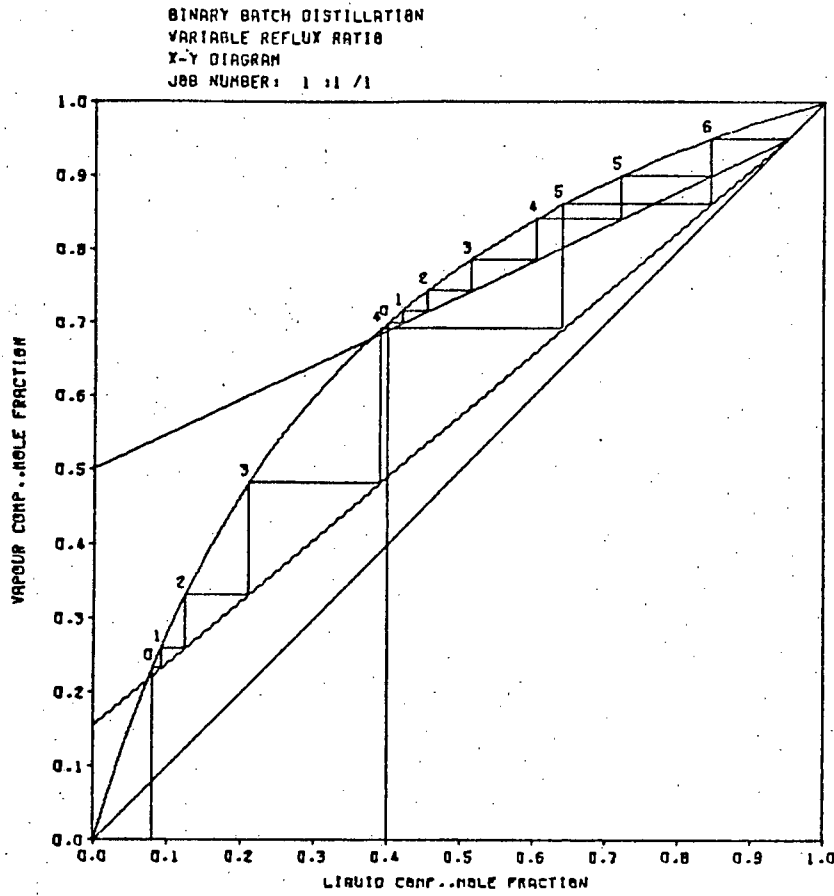


Figure 8.511 Problem 8.51(a) - x,y diagram.

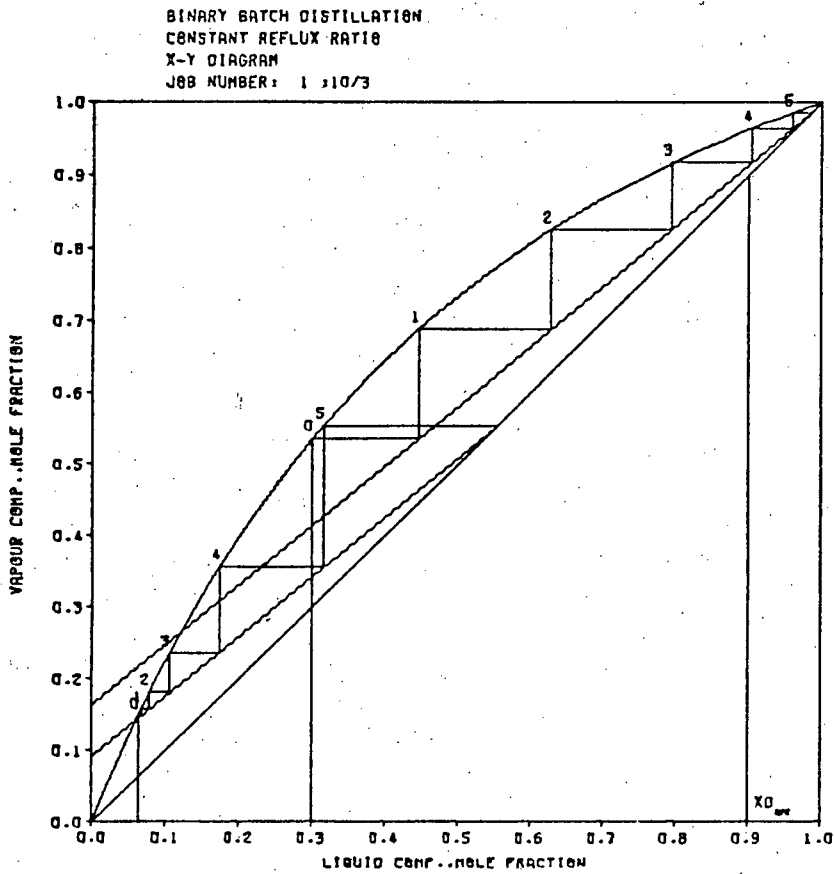


Figure 8.512 Problem 8.51(b1) - x,y diagram.

2) Determine the average distillate composition if the final residue composition required is 0,05 mole fraction  $\text{CCl}_4$ .

TABLE 8.512

SOLUTION TO PROBLEM 8.51(b2)

BATCH DISTILLATION OF A BINARY SOLUTION.  
CONSTANT REFLUX OPERATION.

FEED COMPONENTS: (1) CARBON TETRACHLORIDE  
(2) TOLUENE

REFLUX RATIO: 5.000

VAPOUR BOIL-UP RATE: 10.00 KGMOLE/HR

BATCH CYCLE TIME: 18.253 HRS

TOTAL VAPOUR BOILED UP DURING CYCLE: 182.534 KGMOLE

RESIDUE: 69.578 KGMOLE

RESIDUE COMPOSITION: .0500

TOTAL DISTILLATE WITHDRAWN: 30.422 KGMOLE

INITIAL DISTILLATE COMPOSITION: .9862 MOLE FRAC

FINAL DISTILLATE COMPOSITION: .4517

AVERAGE DISTILLATE COMPOSITION: .8718

Table 8.512 shows an average distillate composition of  $x_{D_{AVE}} = 0,8718$ , with a cycle time of 18,2 Hrs. Repeating the problem for a specified  $x_{D_{AVE}} = 0,9$ , the cycle time is reduced to 16 Hrs whereas the residue composition increased (0,06 mole fraction) and the distillate product yield is reduced to 28 kgmols. Figure 8.512 shows a McCabe-Thiele diagram for the system with  $x_{D_{AVE}}$  specified. Initial and final conditions are shown.

Figures 8.511 and 8.512 are code 3 and 2 plots respectively (see User Manual, section 7.54). In addition to the results shown in Tables 8.511 and 8.512, BATCH prints out a table of either reflux ratio (variable reflux operations) or distillate

composition (constant reflux operations) vs. residue composition. These data may then be used to check the integration manually.

#### 8.52 Binary distillation in tray columns - program BINARY

The graphical design methods of McCabe and Thiele, and Ponchon and Savarit are discussed in detail in the undergraduate course. Program BINARY may be used to demonstrate a number of features associated with these design procedures.

8.521 McCabe-Thiele Problems: Figures 8.521 and 8.522 show complex columns operated under McCabe-Thiele conditions of equimolal overflow. Figure 8.521 shows a system with two feed streams under open steam operation. A Murphree tray efficiency factor of 0,75 is assumed and a total of 7,8 trays are required. If perfect stages are assumed this number is reduced to 5,7 and if a partial reboiler is used only 3,9 ideal trays are required. Figure 8.522 shows a column with a side stream withdrawal and a partial reboiler. Ideal trays are assumed in this case. The result tables produced by BINARY list the flowrates in each section of the column, and the trays on which feeds are entered or from which side streams are withdrawn.

Figures 8.523 and 8.524 demonstrate the use of Antoine constants and equilibrium data respectively for the same system (see Problem 8.521 below). The shape of the equilibrium curve affects the feed tray position, and the total number of trays required.

#### Problem 8.521

(System described by Figure 8.524.)

A solution of methanol and water containing 0,38 mole fraction methanol is distilled at a rate of 100 kgmoles/hr in a column fitted with a partial reboiler and a partial condenser. A two phase distillate product is withdrawn with

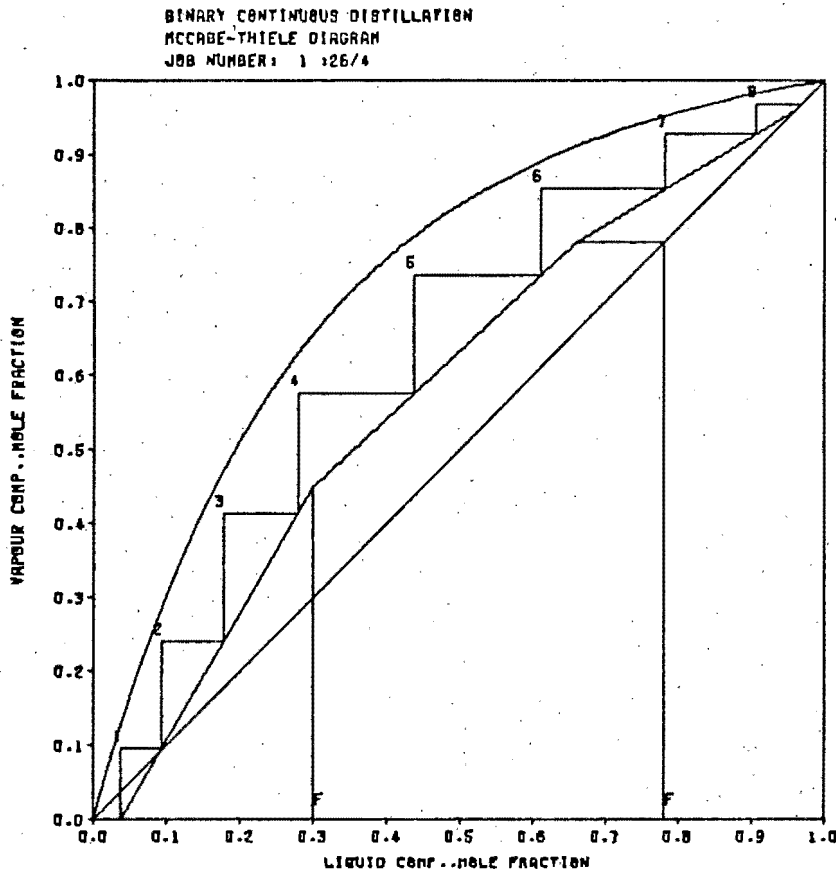


Figure 8.521 Complex column - open steam operation.

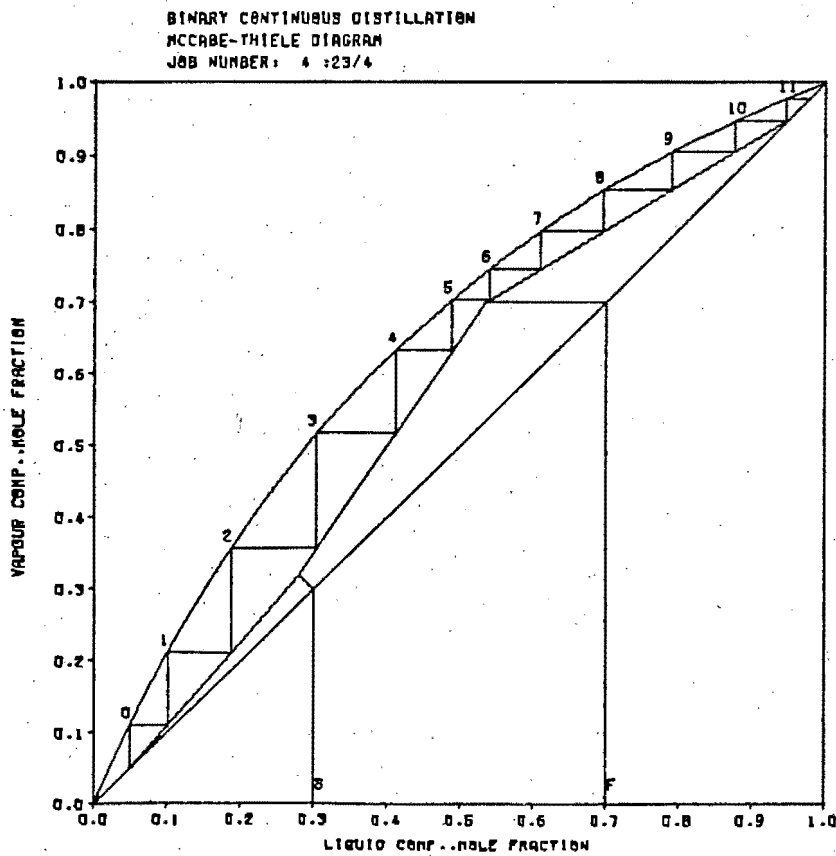


Figure 8.522 Complex column - partial reboiler used.

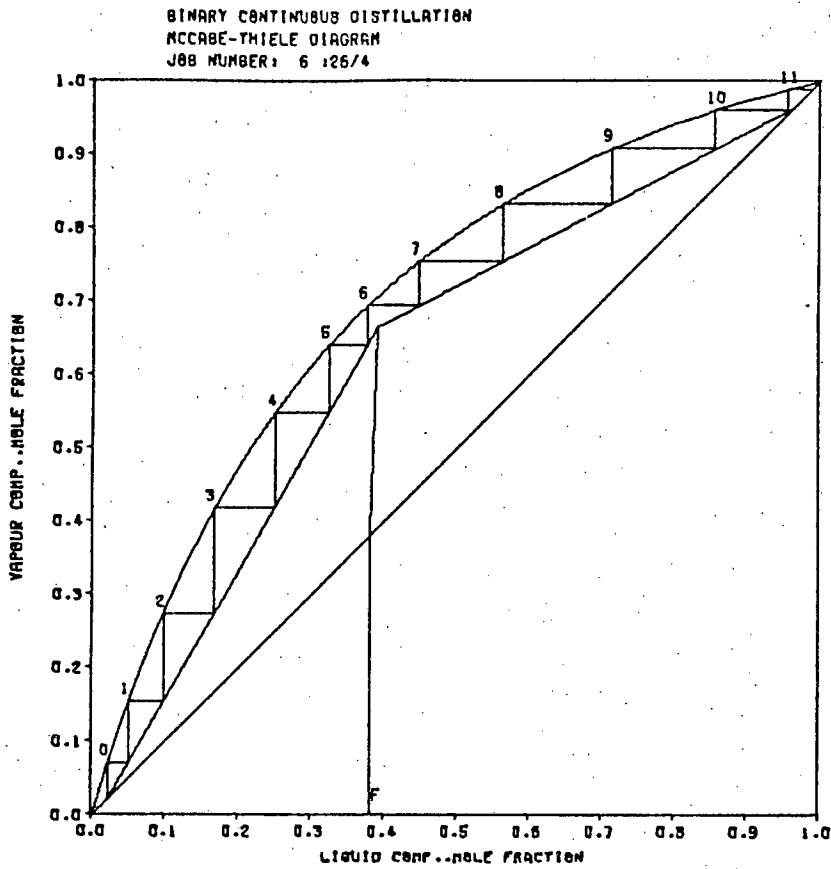


Figure 8.523 Problem 8.521 - Antoine constants.

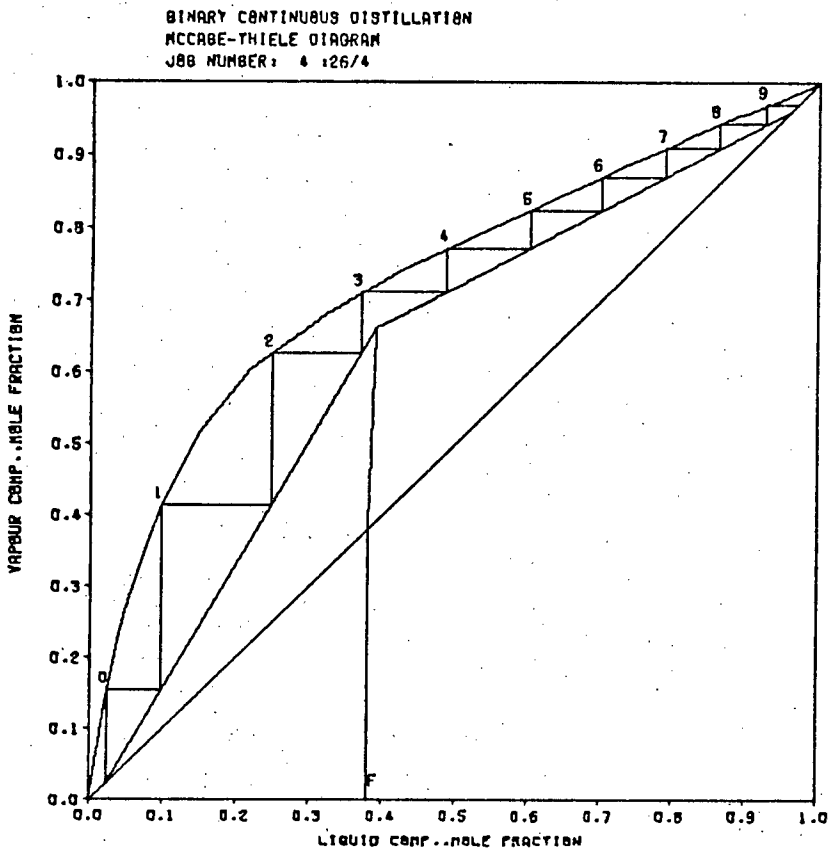


Figure 8.524 Problem 8.521 - x,y equilibrium data.

a liquid phase composition of 0,95 mole fraction. The total distillate withdrawal rate is 38,0 kgmoles/hr, of which 60% is liquid.

Determine the minimum number of ideal trays required in the column, and the feed tray position, if the feed is a liquid just below its bubble point ( $q = 1,04$ ) and the reflux ratio is set at 1,1. Assume that the column operates at atmospheric pressure.

TABLE 8.521

SOLUTION TO PROBLEM 8.521

MCCABE-THIELE DESIGN OF A BINARY DISTILLATION SYSTEM.

FEED COMPONENTS: (1) METHANOL  
(2) WATER

NO. OF FEED STREAMS: 1  
NO. OF SIDE STREAMS: 0

NO. OF THEORETICAL STAGES REQUIRED: 8.4

MURPHREE EFFICIENCY FACTOR USED: 1.00

FEED AND SIDE STREAM SPECIFICATIONS:

STREAM NO	TRAY NO	Q VALUE	FLOWRATE KGMOLE/HR	COMPOSITION MOL FRAC
1 FEED	3	1.04	100.00	.3800

PARTIAL CONDENSER OPERATION  
CONDENSER PRESSURE: 101.32 KPA

REFLUX RATIO 1.1000

DISTILLATE WITHDRAWAL RATE: 38.00 KGMOLE/HR  
DISTILLATE COMPOSITION: .9616

BOTTOMS WITHDRAWAL RATE: 62.00 KGMOLE/HR  
BOTTOMS COMPOSITION: .0235

VAPOUR AND LIQUID RATES THROUGH COLUMN:

SECTION	TRAY	TYPE	VAPOUR (KGMOLE/HR)	LIQUID (KGMOLE/HR)
SECTION 1	REBOILER	- FEED (1)	83.80	145.80
SECTION 2	FEED (1)	- CONDENSER	79.80	41.80

The distillate composition shown is the overall value calculated by the program. In addition to this table, a table of data describing the McCabe-Thiele operating diagram may be requested. These data may be used to construct the diagram manually if a plotting facility is not available.

The effect of feed condition, condenser type, reflux ratio and tray efficiency may be investigated using BINARY. Figure 8.525 shows the system described in problem 8.521 assuming a Murphree tray efficiency factor of 0,8. Note that the reboiler (Stage 0) is assumed to be an ideal stage.

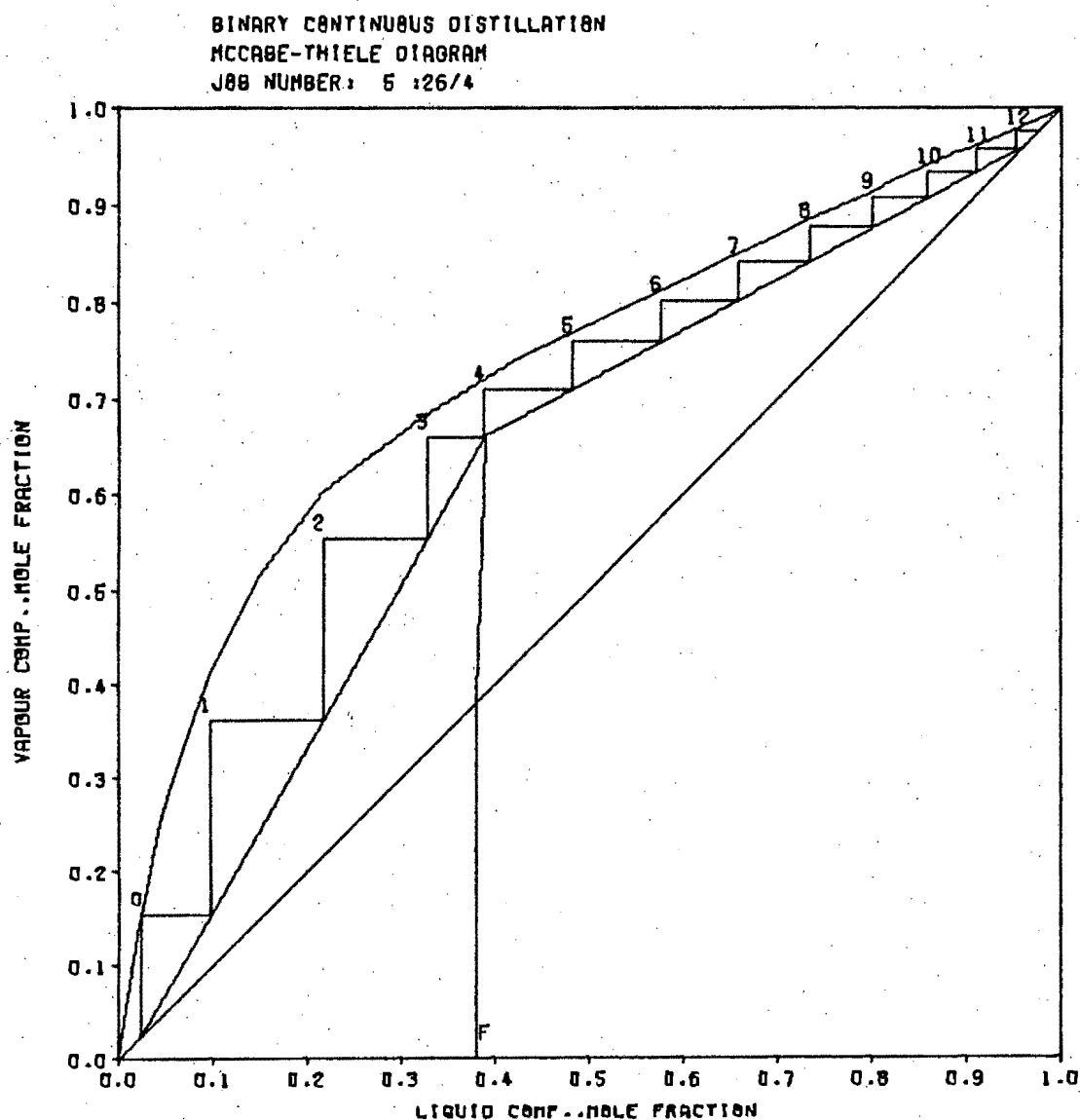


Figure 8.525 Problem 8.521 - Tray efficiency = 0,8.



All diagrams shown in this section are constructed by PLOTTER under code 4 (see section 7.551).

8.522 Ponchon-Savarit problems: Figure 8.526 shows the computer solution to problem 9.8 in Treybal [9] p.326, using Antoine constants. A result table for the same problem using equilibrium data is shown in Table 8.522 below. Agreement with the solution shown in Treybal is good, providing equilibrium data are used. Treybal's results, converted to the SI where necessary, are shown in parenthesis in the table.

TABLE 8.522

SOLUTION TO PROBLEM 9.8 TREYBAL [9] p.326

FEED O VALUE AND TEMPERATURE DATA:

Q: 1.0183 FEED TEMP: 57.80 DEG C  
DEW PT: 89.46 DEG C BUBBLE PT: 76.36 DEG C  
(76.1)

SPECIFIED REFLUX RATIO IS LESS THAN THE CALCULATED MINIMUM.

```

RMIN= .66  R SPEC= .30
A VALUE OF 1.5*RMIN WILL BE USED.

```

# PONCHON-SAVARIT DESIGN OF A BINARY DISTILLATION SYSTEM.

FEED COMPONENTS: (1) METHANOL  
(2) WATER

NO. OF THEORETICAL STAGES REQUIRED: 7.4 (7.9)

MURPHREE EFFICIENCY FACTOR USED: 1.00

FEED AND SIDE STREAM SPECIFICATIONS:

STREAM	TRAY NO	Q VALUE	FLOWRATE	COMPOSITION	
NO	TYPE		KGMOLE/HR	MOL FRAC	
1	FEED	4	1.02	98.10	.3600
		(4)	(1.03)		

FEED TEMPERATURE: 57.80 DEG C (57.77)

TOTAL CONDENSER OPERATION.

CONDENSER PRESSURE: 101.32 KPA  
CONDENSER DUTY: 758.67 KW (753.0)

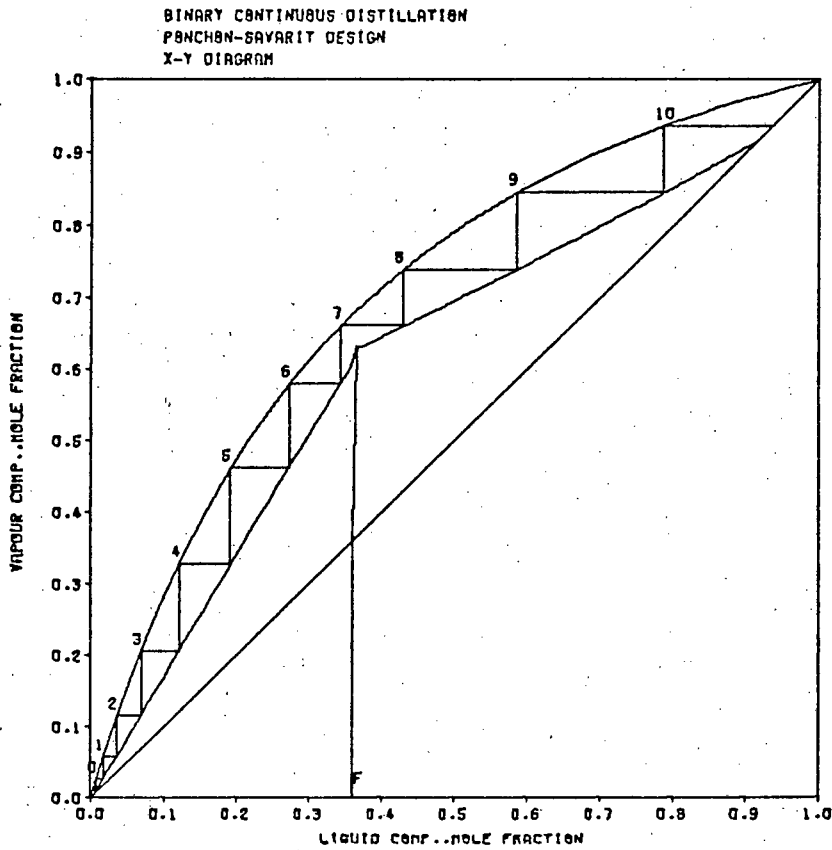
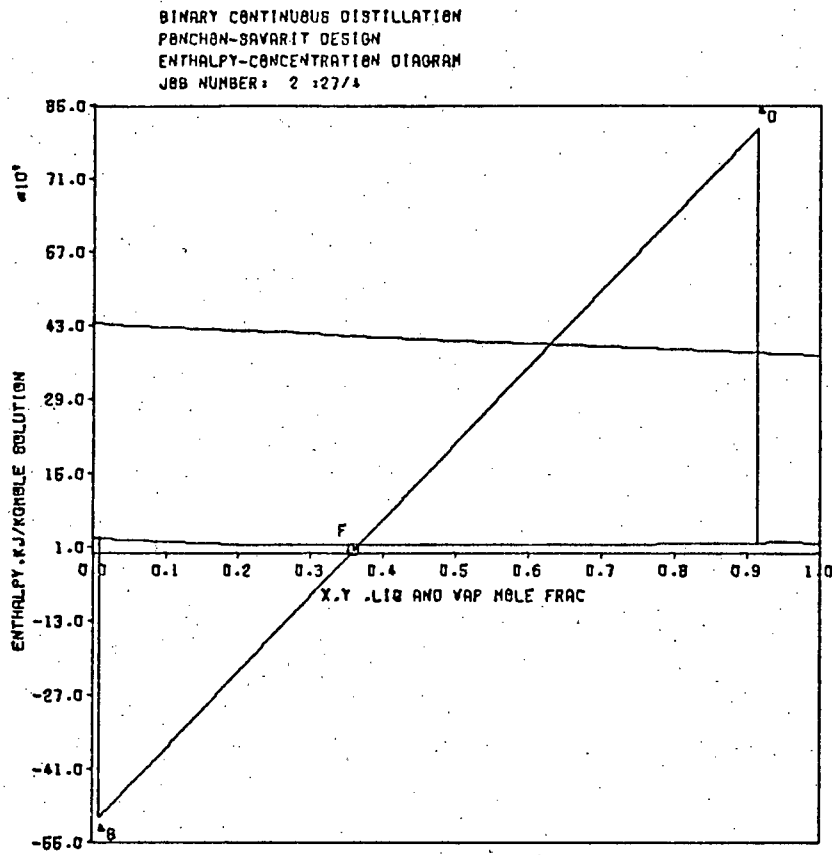


Figure 8.526 Ponchon-Savarit diagram using  
Antoine constants.

REFLUX RATIO: .9932 (1.029)  
MIN REFLUX RATIO: .6621 (0.685)

DISTILLATE WITHDRAWAL RATE: 38.10 KGMOLE/HR  
DISTILLATE COMPOSITION: .9150

BOTTOMS WITHDRAWAL RATE: 60.00 KGMOLE/HR  
BOTTOMS COMPOSITION: .0076 (0.0056)  
REBOILER DUTY: 807.55 KW

COMPOSITION, TEMPERATURE AND FLOWRATE PROFILES:

TRAY	COMPOSITION (MOLE FRAC)		TEMPERATURE (DEG C)	FLOWRATES (KGMOLE/HR)	
	LIQ	VAP	BUBBLE PT	LIQ	VAP
0	.0076	.0076	98.29	60.00	71.57
1	.0339	.0558	94.39	132.14	72.14
2	.1173	.2063	86.56	133.94	73.94
3	.2600	.4566	79.33	137.04	77.04
4	.3598	.6357	76.37	136.75	76.75
FEED TRAY POSITION:					
5	.4773	.7058	73.62	34.86	72.96
6	.6139	.7694	70.90	35.68	73.78
7	.7444	.8315	68.44	36.52	74.62
8	.8661	.8908	66.50	37.44	75.54
COND	.9452	.9452	65.70	37.84	.00

ENTHALPY DIAGRAM DEL PTS:

DEL D= .735637+05 KJ/KGMOLE (74595.0)  
DEL B= -.458130+05 KJ/KGMOLE (-43472.0)

The use of Antoine constants lifts the feed position by 3 trays and increases the total number required to 9,8. The minimum reflux ratio is increased to 0,8 and the feed q-value to 1,02. Table 8.522 shows a minimum reflux ratio of 0,66. The reflux ratio was specified as 0,3 to force the program to use a value of  $1,5 \times R_M$ , which is the procedure adopted in Treybal. The enthalpy and equilibrium data used by Treybal yield a minimum reflux ratio of 0,685. Using Treybal's resultant reflux ratio of 1,029, the reboiler and condenser heat duties increase as expected, although the total number of trays does not alter much (7,2). The discrepancies in the results emphasize the importance of using realistic enthalpy and equilibrium data in distillation designs.

The condenser compositions shown in the composition-profile table (Table 8.522) refer to the top stage stepped off on the x,y diagram (stage 8). The distillate composition refers to the product from stage 7.4. The enthalpy diagram DEL points are the enthalpy coordinates of  $\Delta_D$  and  $\Delta_B$  respectively. These are required by PLOTTER under code 5 (see section 7.552). Tables of operating diagram data may be requested as for McCabe-Thiele problems. The Ponchon-Savarit operating diagram may be plotted manually using these data.

Figure 8.257 demonstrates the use of equilibrium data for a system that deviates appreciably from Raoult's Law (acetone-water). If the feed condition is such that the reboiler stage spans the point of intersection of the operating lines, as shown, the user is informed that the column consists of a rectifying section only. Calculations then continue as normal.

The effects of feed temperature, reflux ratio, condenser type and tray efficiency on Ponchon-Savarit systems can be investigated using BINARY.

#### 8.53 Binary distillation in packed columns - program PACKED

This program uses the procedures described by Treybal [9], who shows only one sample problem. Solving problems manually using this procedure is extremely laborious, which is probably why no further reference to the method could be found. The program PACKED, which is able to solve problems in a matter of seconds, has proved useful in revealing certain discrepancies between results obtained using Shulman's correlations and other design methods. In addition, errors are shown to occur in the calculations of Treybal.

The design of a packed column is apparently more complete than the tray-column design discussed previously, in that the diameter and height of the column are established. Packed column design is based on superficial flowrates and pressure drop, however, therefore diameter and height are necessary features of the calculation procedure.

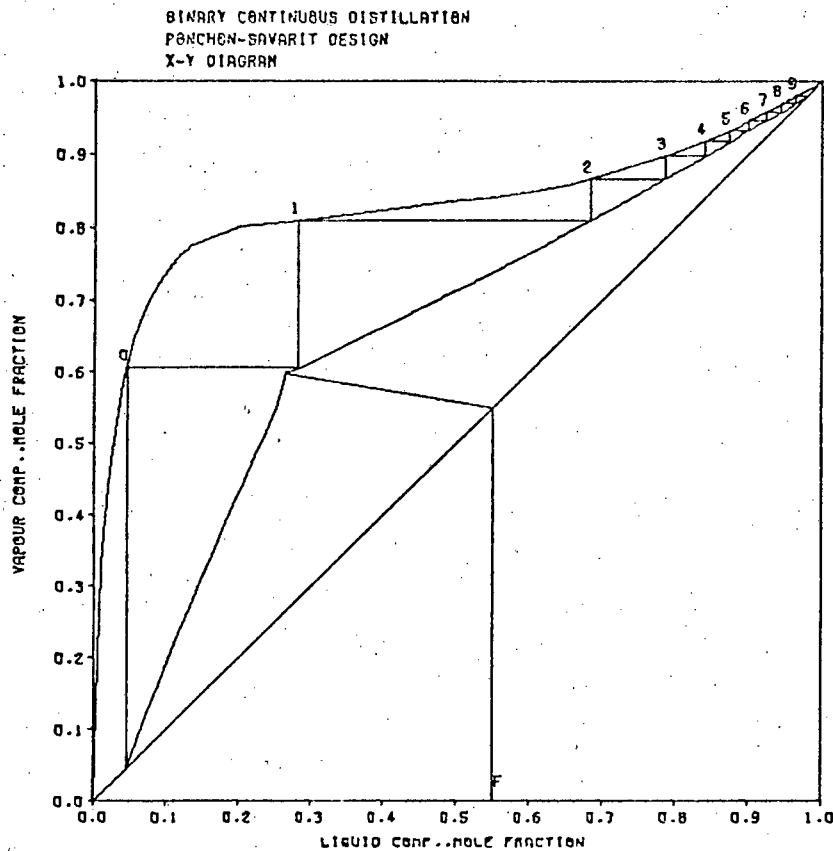
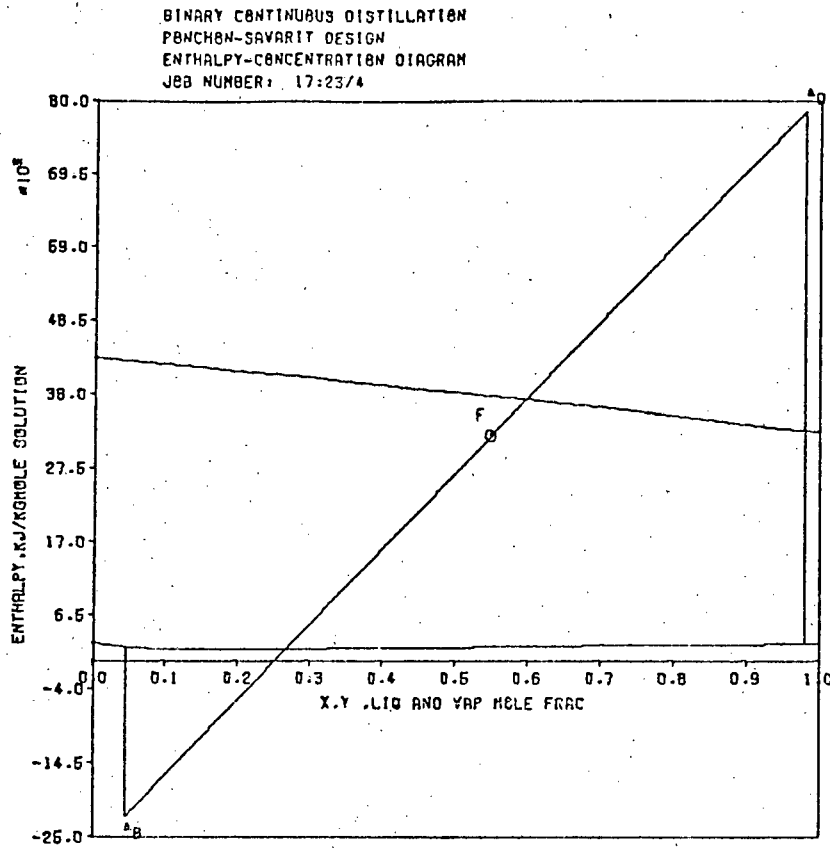


Figure 8.527 Ponchon-Savarit diagram for  
Acetone - Water system.

In Treybal [9] (p.365), problem 9.12 shows the design of a packed column for the system described previously (Table 8.522). The computer solution, using PACKED, (shown in Table 8.531 below) agrees fairly well with data listed in Treybal for 38,1 mm Raschig rings. Treybal's results are shown in parenthesis.

TABLE 8.531

SOLUTION TO PROBLEM 9.12 TREYBAL [9] p.365

DESIGN OF A PACKED DISTILLATION COLUMN.

MASS TRANSFER COEFFICIENT CALCULATIONS BY SHULMANS METHOD

FEED COMPONENTS: (1) METHANOL  
(2) WATER

FEED STREAM SPECIFICATIONS:

FLOWRATE: 98.10 KGMOLE/HR  
COMPOSITION: .3600 MOLE FRAC  
QVALUE: 1.02 (1.03)  
TEMPERATURE: 57.80 DEG C (57.77)

TOTAL CONDENSER OPERATION.

CONDENSER DUTY: 758.67 KW (753)  
CONDENSER PRESSURE: 101.32 KPA

REFLUX RATIO: .9932 (1.029)  
MIN REFLUX RATIO: .6621 (0.685)

DISTILLATE WITHDRAWAL RATE: 38.10 KGMOLE/HR  
DISTILLATE COMPOSITION: .9150

BOTTOMS WITHDRAWAL RATE: 60.00 KGMOLE/HR  
BOTTOMS COMPOSITION: .0076 (0.0056)  
REBOILER DUTY: 807.55 KW (824.7)

VAPOUR PRESSURE-DROP AT LOADING: .408 KPA/M (0.408)

THE COLUMN DIAMETER IS BASED ON CONDITIONS AT THE POINT IN THE COLUMN WITH THE GREATEST LIQUID FLOWRATE.  
(SEE TABLE BELOW)

RECOMMENDED COLUMN DIAMETER: .790 M (0.75)

HEIGHT OF PACKING REQUIRED:

RECTIFYING SECTION: 5.77 M (7.4)  
STRIPPING SECTION: 2.88 M (4.48)

DO YOU WANT COMPOSITION, TEMPERATURE AND FLOWRATE DATA TABLES?

NO

DO YOU WANT MASS-TRANSFER DATA TABLES?

YES

PRINCIPAL MASS TRANSFER RESISTANCE IN VAPOUR PHASE.

SHULAMN MASS TRANSFER CALCULATIONS:

X = LIQUID MOLE FRACTION OF LIGHT COMPONENT

V = VAPOUR RATE IN COLUMN (KGMOLE/HR.SQ M)

A = INTERFACIAL SURFACE AREA OF PACKED SECTION (SQ M/CU M)

KY = VAPOUR MASS TRANSFER COEFFICIENT (KGMOLE/HR.SQ M.MOLE FRAC)

KX = LIQUID MASS TRANSFER COEFFICIENT (KGMOLE/HR.SQ M.MOLE FRAC)

X	V	A	KY	KX	X	A	KY	KX
.9150	154.926	20.949	7.4256	32.0684	(0.915	19.935	5.492	37.8 )
.8601	154.007	20.834	7.3591	31.0438				
.8051	153.137	20.741	7.2957	30.0494				
.7502	152.317	20.673	7.2341	29.0737				
.6953	151.567	20.627	7.1750	28.1515				
.6403	150.854	20.608	7.1179	27.2423	(0.60	21.55	5.56	31.00)
.5854	150.169	20.614	7.0627	26.3676				
.5304	149.483	20.656	7.0068	25.4916				
.4755	148.825	20.732	6.9525	24.6492				
.4206	148.196	20.850	6.8997	23.8242				
.3656	147.606	21.015	6.8483	23.0197	(0.37	21.91	5.565	27.8 )
FEED POSITION								
.3656	151.538	34.884	7.0500	42.5693	(0.37	37.99	5.907	51.26)
.3097	159.363	34.429	7.1251	41.7750				
.2537	156.889	32.878	6.9199	40.3227				
.1978	154.398	31.384	6.7257	38.9371	(0.20	32.8	6.093	53.46)
.1418	151.938	29.942	6.5649	37.6656				
.0859	149.465	28.546	6.5004	36.4664				
.0339	147.215	27.270	6.9029	35.5841	(0.02	22.2	5.712	53.95)

Treybal's results indicate a smaller diameter than that designed using PACKED (0.75 m vs. 0.79 m). This is one cause of the greater heights of packing required. The differences in calculation and integration procedure could also contribute to the discrepancies. The effect of specifying a diameter of 0.75 in PACKED is shown in Table 8.532.

In Table 8.531 it is seen that the interfacial areas are in good agreement with the value computed by Treybal whereas the mass transfer coefficients are greater than Treybal's for the vapour phase, and smaller for the liquid phase. From Shulman's correlations it is probable that physical property data affect these calculations.

Treybal states that if the column is redesigned using 38,1 mm Berl saddles the packing height is reduced by roughly 40%. Using PACKED, it is found that almost twice the bed height is required for Berl saddles, although the pressure drop is less for a column of the same diameter.

By inspection of the equations and correlations of Shulman [61] it can be shown that for a given liquid rate the value of  $H^*$  for Raschig rings is about 60% greater than  $H^*$  for Berl saddles. The specific interfacial surface area for distillation is given by:

$$a_A = a_{AW} \cdot H^* \quad (\text{Equation 4.12, 4.15})$$

It is found that  $H^*$  dominates the equation and thus although  $a_{AW}$  is greater for Berl saddles,  $a_A$  is smaller. The end result is that a greater bed height of Berl saddles is required for a given separation. Referring to the original articles of Shulman, the correlations quoted in Treybal appear to be correct.

Furnas and Taylor [79] have shown that for distillation the bed height required is largely independent of the packing material used. Eckert [17] does not discuss the relative efficiencies of packings, but notes that the lower pressure drop of Berl saddles is an advantage with respect to operating costs. Feller [80] notes that Berl saddles are up to 25% more efficient than Raschig rings when applied to absorption problems. It appears, therefore, that in practice very little



is known about the behaviour of packing materials in distillation applications. Most designs are carried out using simplified HTU methods [17] based on experimental observations or practical experience.

For tutorial purposes the method of Shulman is useful for demonstrating the effects of component physical properties and other variables on packed column operations. It appears that despite the findings of Treybal, the Shulman correlations do not show that Berl saddles are more efficient than Raschig rings. There does not, however, seem to be conclusive evidence in the literature that Berl saddles are more efficient when applied to distillation hence the results obtained using PACKED are possibly not unreasonable. Experimental work would be required to check the validity of Shulman's correlation when applied to distillation processes.

The ultimate decision inevitably rests on the cost factor. With the present emphasis on reducing operating costs, the cost of extra packing would probably be acceptable when compared with the costs of operating at a higher pressure drop in Raschig rings.

PACKED also shows that the design is sensitive to the equilibrium data used, the reflux ratio and the feed condition. Table 8.532 shows the bed heights under different conditions. Data shown in Table 8.531 applies unless otherwise stated.

These results show:

- 1) Efficiency and pressure drop are reduced as the size of the packing increases.
- 2) Higher superficial mass flowrates, caused by reducing the column diameter for a given throughput, improve efficiency.
- 3) Increasing the reflux ratio reduces the height of packing required, and increases the pressure drop if the diameter is not increased.

TABLE 8.532

PACKED COLUMN DESIGN USING PACKED

Solution	Data Used	Packing Type	Size mm	Column Diameter m	Pressure drop kPa	$Z_R$ m	$Z_S$ m
Treybal	basic	R	38,1	0,75	0,408	7,4	4,48
Program	basic	R	38,1	0,79	0,408	5,77	2,88
"	basic	R	38,1	0,75	0,53	6,02	2,
"	basic	B	38,1	0,72	0,408	10,72	3,40
"	basic	B	38,1	0,79	0,25	10,90	4,49
"	Ant.const.	R	38,1	0,82	0,408	4,37	6,82
"	basic	R	25,4	0,90	0,408	3,38	2,04
"	basic	R	25,4	0,79	0,810	3,81	1,98
"	basic	R	50,8	0,72	0,408	5,62	3,0
"	basic	R	50,8	0,79	0,25	5,31	3,04
"	R=2,5	R	38,1	1,01	0,408	4,28	2,12
"	R=2,5	R	38,1	0,79	1,6	3,94	1,80
"	q=0,68	R	38,1	0,64	0,408	6,27	2,58
"	q=0,68	R	38,1	0,79	0,15	6,18	2,95

Table codes:

Packing Type: R - Raschig rings; B = Berl saddles

Data used: basic - data specified for problem 9.11 in Treybal [9] p.365.

Ant.const.- basic with Antoine constants used to generate equilibrium data.

R - basic with reflux ratio changes.

q - basic with feed temperature changed.

Pressure drop: from Figure 1, Appendix A.

$Z_R$ ,  $Z_S$  : height of packing in rectifying and stripping section respectively.

4) Increasing the feed temperature reduces the pressure drop and the efficiency for a fixed diameter and throughput. These points are in agreement with general experience [17,63, 79].

It can also be seen that if Antoine constants are used a greater height of packing is required in the stripping section. This is equivalent to raising the feed position in a tray column as shown in section 8.522.

#### 8.54 Multicomponent distillation - program MULCOM

The time required to solve multicomponent distillation problems manually has lead to an emphasis on short-cut techniques in the undergraduate course. MULCOM, like PACKED, is able to solve complex problems in a matter of seconds, thus enabling students to investigate the effects of several parameters on multicomponent systems.

Solutions to three problems shown below demonstrate the capabilities of MULCOM.

##### Problem 8.541.

(Data from Example 9.13, Treybal [9] p.388)

The system considered consists of a six-component hydrocarbon system in a column containing 14 trays. Table 8.541 shows the abbreviated result table printed on the terminal in demand mode executions. Table 8.542 shows two sections of the detailed result table that may be obtained via PRINTFILE (see section 7.422). The full table shows similar data for each tray in the column, and overall material and enthalpy balance figures. Treybal's results are shown in parenthesis in Table 8.541. The distillate composition and temperature profiles are in close agreement. The discrepancies in the heat duties are due to the enthalpy data chosen.

TABLE 8.541

##### SOLUTION TO EXAMPLE 9.13, TREYBAL [9] p.372

MULTICOMPONENT DISTILLATION BY THIELE-GEDDES METHOD:

RUN NO: 3/15/ 3

##### OVERALL MATERIAL BALANCE:

STREAM	FLOWRATE (KGMOLE/HR)
DISTILLATE	38.4600
BOTTOMS	61.5400
FEED-TOTAL	100.0000
-VAPOUR	37.1501
-LIQUID	62.8499

COMPONENT MOLE FRACTIONS:

COMP.	DISTILLATE	BOTTOMS	FEED STREAM			
			LIQUID	VAPOUR	TOTAL	
1	.0780 (0.0789)	.0000	.0019	.0775	.0300	
2	.1820 (0.1842)	.0000	.0170	.1596	.0700	
3	.3885 (0.3870)	.0010	.0860	.2582	.1500	
4	.3481 (0.3420)	.3187	.3286	.3323	.3300	
5	.0035 (0.0079)	.4853	.3907	.1466	.3000	
6	.0000 (0.0000)	.1750	.1757	.0258	.1200	

TEMPERATURE AND PRESSURE PROFILE:

TRAY NO	TEMPERATURE		PRESSURE
	DEG C	DEG F	KPA
CONDENSER	51.0	123.8 (115)	1034.21
14	62.7	144.9 (138)	1034.21
13	66.4	151.6 (150)	1034.21
12	69.4	157.0 (158)	1034.21
11	73.1	163.7 (165)	1034.21
10	78.1	172.5 (176)	1034.21
FEED	84.6	184.2 (188)	1034.21
8	92.1	197.8 (202)	1034.21
7	96.6	206.0 (206)	1034.21
6	98.6	209.6 (208)	1034.21
5	100.1	212.2 (210)	1034.21
4	101.2	214.2 (212)	1034.21
3	102.2	215.9 (217)	1034.21
2	103.3	218.0 (221)	1034.21
1	105.7	222.3 (227)	1034.21
REBOILER	112.2	234.0 (234)	1034.21

NO. OF ITERATIONS= 14 THETA= 1.000

REFLUX RATIO= .80  
DISTILLATE RATE= 38.46 KGMOLE/HR  
FEED Q VALUE= .6285 (0.67)  
FEED TEMPERATURE= 82.20 DEG C (82.2)  
FEED ENTHALPY= 11798.34 KJ/KGMOLE

CONDENSER DUTY= 153.57 KW (74.5)  
REBOILER DUTY= 206.38 KW (108,5)

PARTIAL CONDENSER OPERATION.

VAPOUR AND LIQUID FLOWRATES LEAVING TRAYS:

TRAY NO	VAPOUR	LIQUID	(KGMOLE/HR)
CONDENSER	38.46	30.77	
14	69.23	31.16	
13	69.62	30.81	
12	69.27	30.26	

11	68.72	29.60
10	68.06	28.98
FEED	67.44	93.93
8	32.39	99.18
7	37.64	95.99
6	34.45	97.59
5	36.05	98.82
4	37.28	99.71
3	38.17	100.30
2	38.76	100.58
1	39.04	100.39
REBOILER	38.85	61.54

Figure 8.541 shows the corresponding diagrams produced by PLOTTER (code 1). The temperature profile diagram shows profiles for three runs at different reflux ratios.

RUN	REFLUX RATIO
1	0,8
2	8,0
3	2,5

The remaining diagrams correspond to run 1 which is equivalent to the example in Treybal.

#### Problem 8.542

A benzene-toluene-xylene column is considered. Figure 8.542 shows the composition profiles for the same system under the following conditions:

- Assuming ideal trays;
- Using a single set of component efficiency data (section 7.434);
- Applying Nord's [81] efficiency data for the B-T-X system.

The efficiency data used in (b) is not representative of the true conditions, but is useful for demonstrating the general effect of efficiency specifications. Efficiency data compiled experimentally, or from a source such as Nord, are probably more representative, but such data are seldom available for

TABLE 8.542

CONDENSER AND FEED TRAY SECTIONS OF THE DETAILED RESULT TABLE.

PARTIAL CONDENSER

COMPONENT NO	VAPOUR DISTILLATE KG/MOLE/HR	LIQUID FLEED MOL FRACTION	VAPOUR LEAVING KG/MOLE/HR	LIQUID REFLUX MOL FRACTION
1	3.00	.730031-01	2.40	.251282-02
2	7.00	.162507+00	5.60	.305821-01
3	14.94	.386522+00	11.95	.234108+00
4	13.38	.348024+00	10.71	.711053+00
5	.13	.344259-02	.11	.217148-01
6	.00	.154456-05	.00	.289260-04
TOTALS:	38.46	1.0000	30.77	1.0000

CONDENSER TEMPERATURE= 50.97 DEG C

CONDENSER PRESSURE= 1034.21 KPA

• FEED TRAY •

TRAY NO COMPONENT NO	VAPOUR FLEED KG/MOLE/HR	LIQUID FLEED MOL FRACTION	VAPOUR LEAVING KG/MOLE/HR	LIQUID LEAVING MOL FRACTION
1	2.06	.779503-01	3.03	.449743-01
2	5.92	.163377+00	7.35	.109046+00
3	9.55	.258680+00	17.70	.262380+00
4	12.24	.331617+00	27.76	.411573+00
5	5.38	.195774+00	10.16	.150677+00
6	.94	.255953-01	1.44	.213490-01
TOTALS:	36.91	1.0000	67.45	1.0000

TRAY TEMPERATURE= 24.51 DEG C

TRAY PRESSURE= 1034.21 KPA

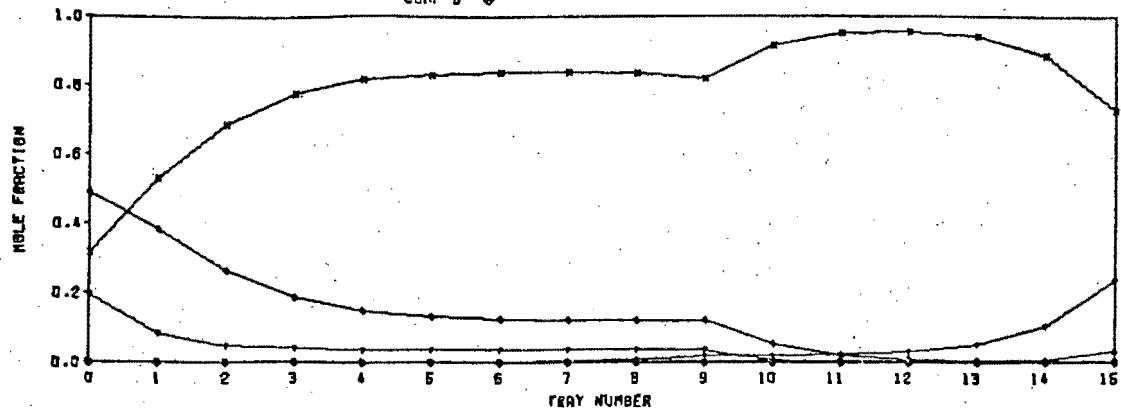
FEED G VALUE= .63

FEED TEMPERATURE= 82.04 DEG C

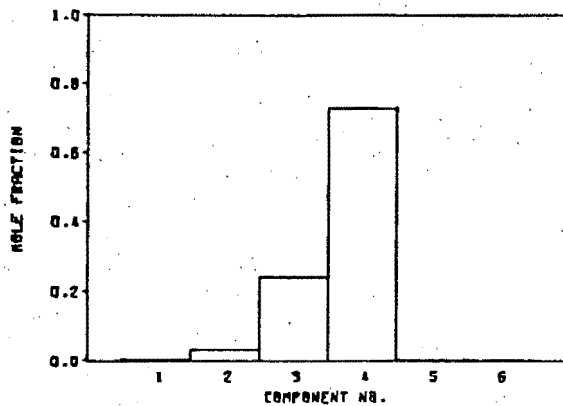
TOTAL NUMBER OF RUNS PLOTTED: 3  
FULL PLOT PRODUCED FOR RUN NO: 1  
JOB NUMBER: 1-122/4

MULTICOMPONENT DISTILLATION  
THIELE-GEDDES PROGRAM  
COMPOSITION PROFILE

COMP 1 ○ COMP 6 ★  
COMP 2 ▲  
COMP 3 +  
COMP 4 x  
COMP 5 ◆

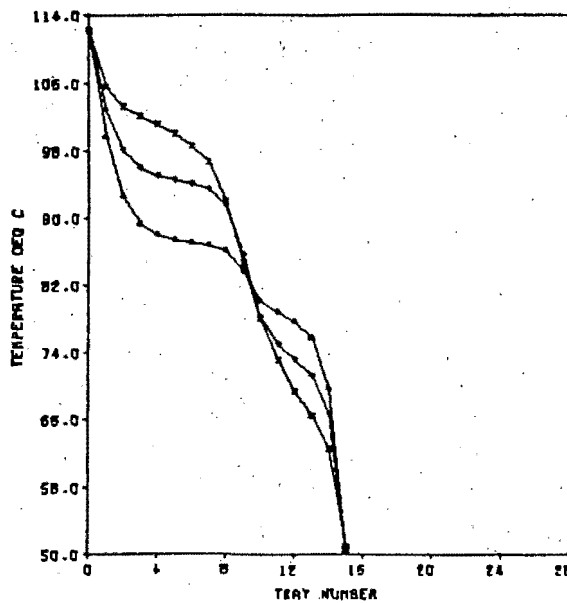


MULTICOMPONENT DISTILLATION  
THIELE-GEDDES PROGRAM  
DISTILLATE COMPOSITIONS



MULTICOMPONENT DISTILLATION  
THIELE-GEDDES PROGRAM  
TEMPERATURE PROFILE

RUN 1 ▲  
RUN 2 x  
RUN 3 ◆



MULTICOMPONENT DISTILLATION  
THIELE-GEDDES PROGRAM  
VAPOR AND LIQUID FLOWRATES

VAP ○  
LIQ ▲

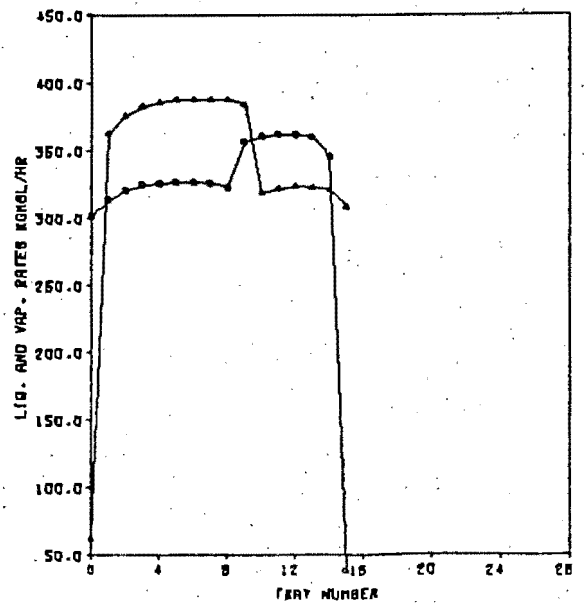


Figure 8.541 Diagrams corresponding to problem 8.541.

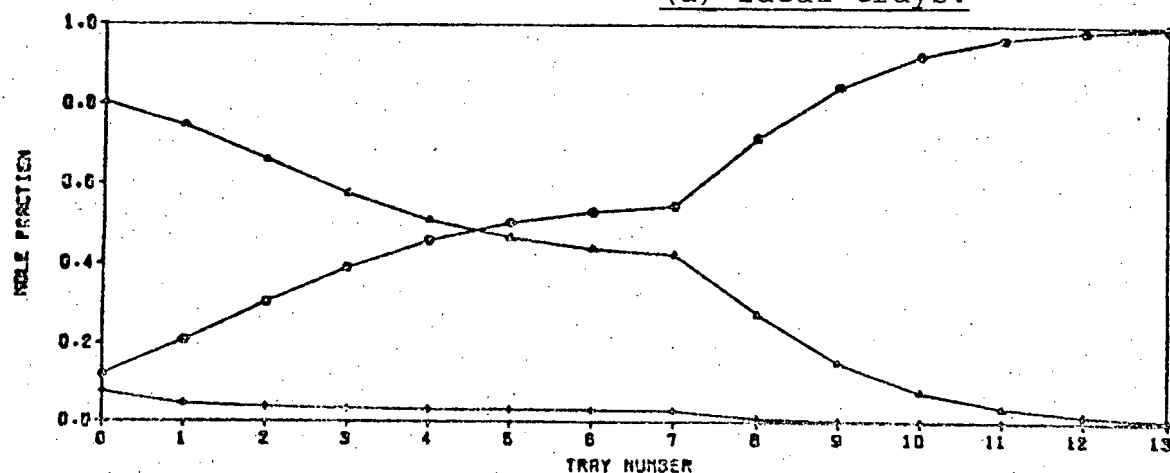
MULTICOMPONENT DISTILLATION

THIELE-GEDDES PROGRAM

COMPOSITION PROFILE

COMP 1 ○  
COMP 2 ▲  
COMP 3 +

(a) Ideal trays.



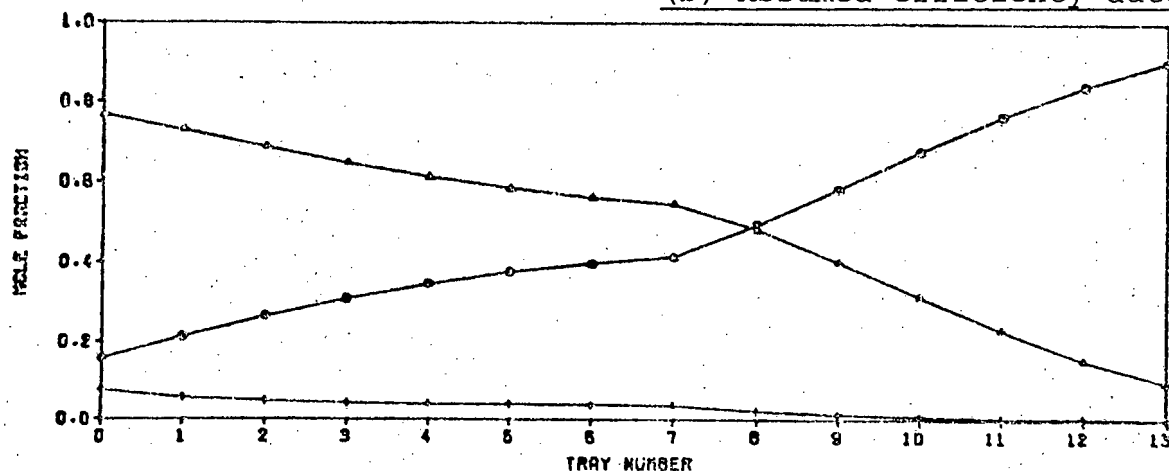
MULTICOMPONENT DISTILLATION

THIELE-GEDDES PROGRAM

COMPOSITION PROFILE

COMP 1 ○  
COMP 2 ▲  
COMP 3 +

(b) Assumed efficiency data.



MULTICOMPONENT DISTILLATION

THIELE-GEDDES PROGRAM

COMPOSITION PROFILE

COMP 1 ○  
COMP 2 ▲  
COMP 3 +

(c) Efficiency data of Nord.

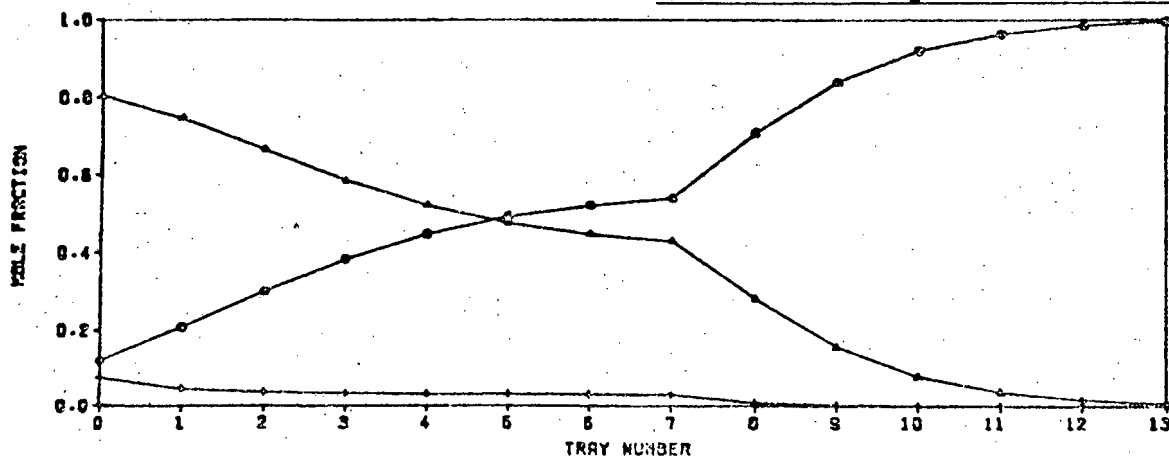


Figure 8.542 Composition profiles for a Benzene-Toluene-Xylene system.



tutorial problems.

The data of Nord cause very little change in the concentration profiles. If efficiency factors are assumed, as in (b), the effect is more noticeable but less realistic.

Problem 8.543

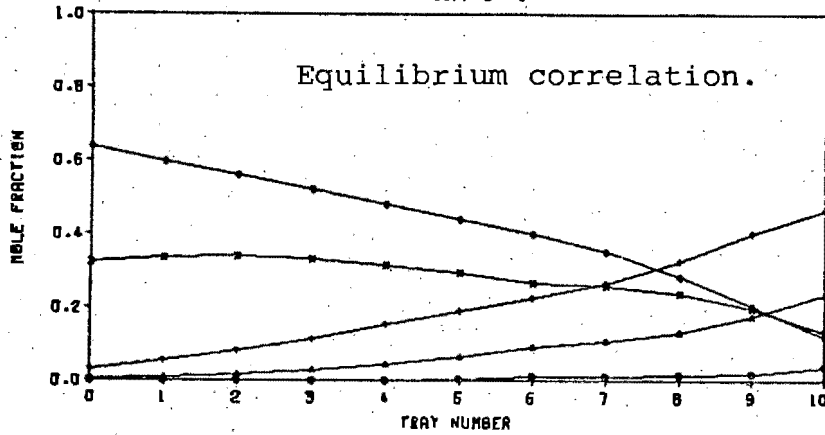
This problem, taken from Holland [21], involves a five component hydrocarbon system. It is used here to compare the use of Antoine constants and equilibrium constant polynomials (section 7.433). Although polynomials in temperature are widely used in industry to correlate equilibrium data, the procedure is cumbersome for setting tutorial problems. Antoine constants are readily available for a wide variety of compounds and are thus easier to use. Figure 8.543 shows the composition and temperature profiles for the two systems. The temperature profiles are superimposed for ease of comparison. It is noticeable that although the composition profiles are very similar, the temperature profile generated by Antoine constants is shallower than that for the temperature polynomials. For standard tutorial problems this is not a serious disadvantage, because the overall behaviour of the system is representative of a real column. The enthalpy data correlation corresponding to NENTH=2 (see section 7.432) was used for both runs. Coefficients were taken from Holland [21].

The effects of feed temperature and pressure, equimolar overflow assumptions, and partial or total condenser operation can also be demonstrated using MULCOM.

TOTAL NUMBER OF RUNS PLOTTED: 1  
FULL PLOT PRODUCED FOR RUN NO: 1  
JOB NUMBER: 12:26/4

MULTICOMPONENT DISTILLATION  
THIELE-GEDES PROGRAM  
COMPOSITION PROFILE

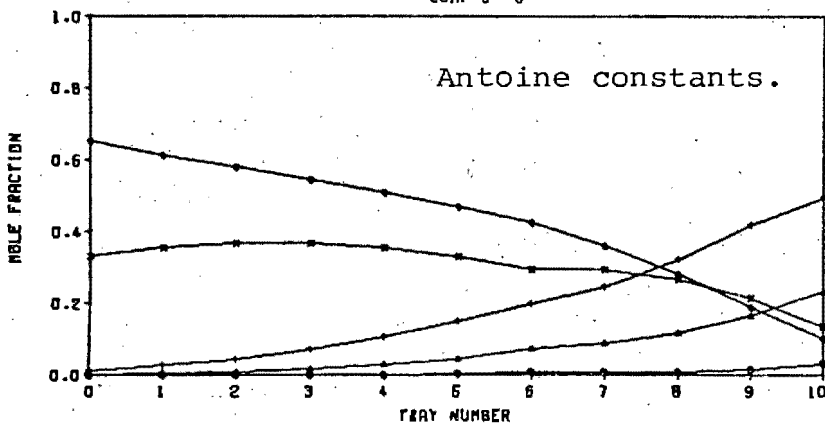
COMP 1	○
COMP 2	▲
COMP 3	+
COMP 4	x
COMP 5	◇



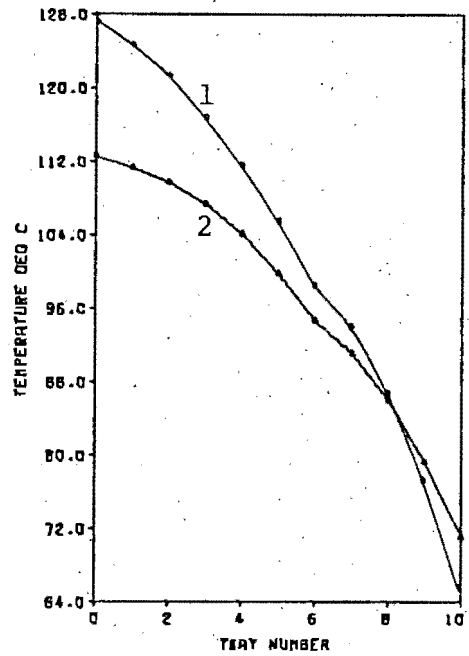
TOTAL NUMBER OF RUNS PLOTTED: 1  
FULL PLOT PRODUCED FOR RUN NO: 1  
JOB NUMBER: 10:26/4

MULTICOMPONENT DISTILLATION  
THIELE-GEDES PROGRAM  
COMPOSITION PROFILE

COMP 1	○
COMP 2	▲
COMP 3	+
COMP 4	x
COMP 5	◇



MULTICOMPONENT DISTILLATION  
THIELE-GEDES PROGRAM  
TEMPERATURE PROFILE



1 - Equilibrium correlation.  
2 - Antoine constants.

Figure 8.543 Diagrams corresponding to problem 8.543.

## 8.6 CONCLUSIONS AND RECOMMENDATIONS

The objectives of this study have been realized in that an integrated computer package has been developed, capable of solving tutorial problems covering all aspects of distillation discussed in the undergraduate course. The programs are generally more economical both in computer time and in core requirements than comparable programs published by CACHE [13]. The programs may be operated both in demand and in batch mode, although the conversational user-program interaction can only be utilised effectively in demand mode.

The design of packed columns is more complete in that the dimensions of the column are determined. This, however, is an integral part of a packed column design, which is based on superficial flowrates and pressure drop considerations. The detailed design of column components was not included in this study and is not necessary for studying the performance of tray columns at an elementary level.

The programs include a number of options that reduce the user's task in compiling data tables and solving problems. The plotting facility is useful for investigating the effects of variables on Ponchon-Savarit diagrams, and multicomponent concentration profiles. These diagrams are normally time consuming to construct manually. Programs for solving Ponchon-Savarit systems, and for designing packed distillation columns do not appear to have been published before. The CACHE programs do not consider enthalpy effects in binary systems. The program PACKED has proved useful in revealing certain errors in published sample solutions. This has suggested that the effect of holdup on the performance of packing materials should be investigated more thoroughly.

The programs have been designed to facilitate alterations and additions in the future. The addition of a tray design program using data directly from MULCOM would complete the

multicomponent design facility. Programs dealing with liquid-liquid extraction and gas absorption should be added to the package. The package would then cover all mass transfer operations considered in the undergraduate course. The following subroutines are designed for general use in future programs:

- (1) Bubble point and dew point calculations (BINBPT, BINDPT, BUBCAL, DEWCAL).
- (2) Flash calculation routines (FLASHB and QFLASH).
- (3) Iterative search procedure (SEARCH).
- (4) Simpson's rule integration (SIMPS).
- (5) Least squares fit of polynomials (FDATA).
- (6) Physical property data package. (Section 4.55)

The present package could be used to conduct experimental investigations on packed and tray columns for distillation purposes, particularly with respect to stage efficiencies. As this is primarily a teaching package, however, emphasis has been placed on the following points:

- (1) Demonstration of the basic principles of distillation.
- (2) Conformity with standard conventions and nomenclature.
- (3) Minimising the effort required to set up and solve tutorial problems.
- (4) Conversational user interaction and descriptive error messages.
- (5) Minimising computer time and core requirements.

The modular structure of the programs should facilitate future additions or alterations to the basic package.

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## APPENDIX A

### PACKED COLUMN CORRELATION DATA

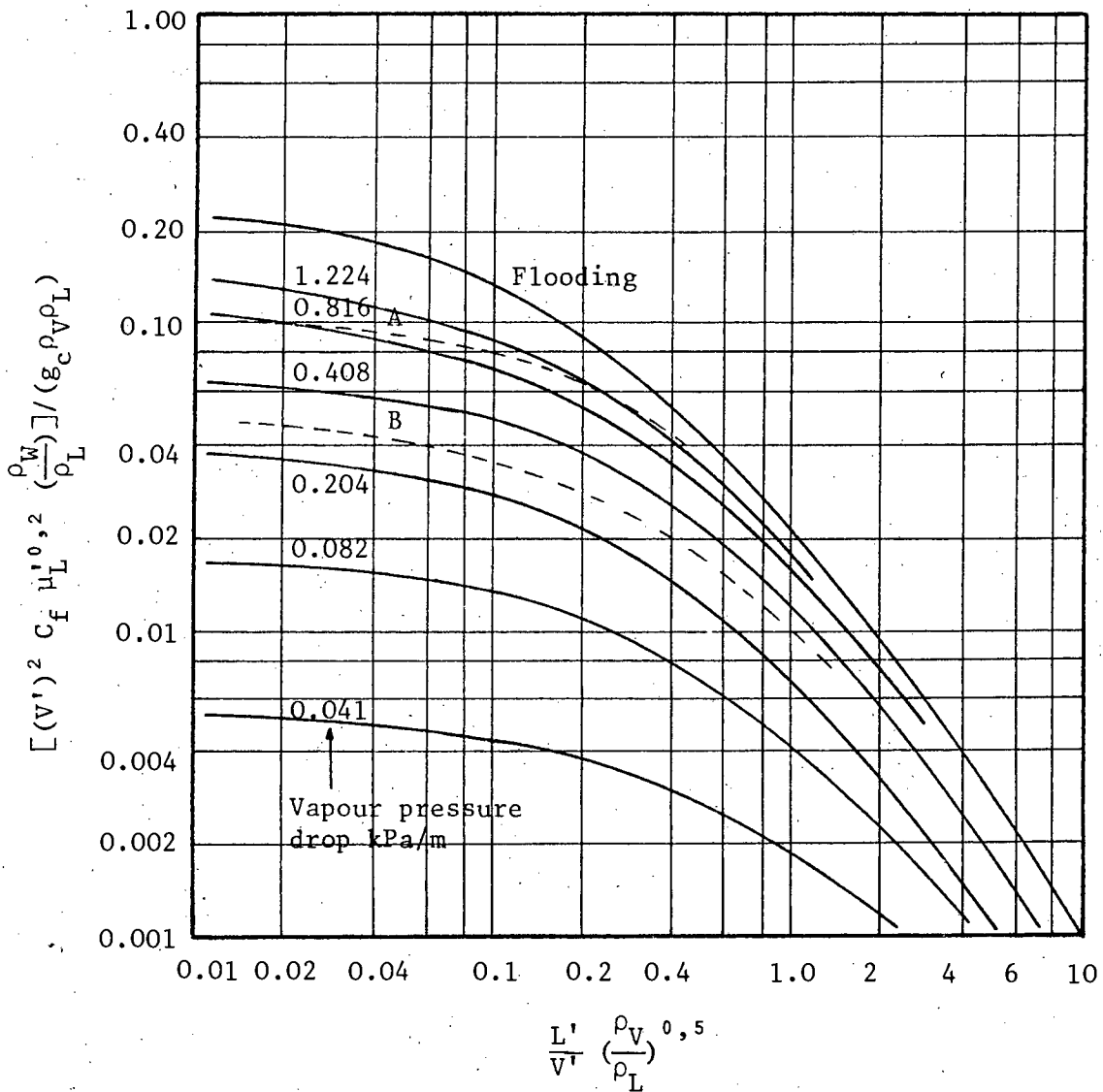


FIGURE 1 FLOODING AND PRESSURE DROP IN PACKED TOWERS

(from Treybal [9], p.160)

Broken lines:

A - upper limit of loading zone [62]

B - lower limit of loading zone [62]

Data used in curve fitting routine FDATA are taken from this diagram and stored in the following data elements. The data elements are accessed by FDATA through the ERTRAN facility.

<u>kPa/m</u>	<u>Element</u>
0.041	FLOOD1
0.084	FLOOD2
0.204	FLOOD3
0.408	FLOOD4
0.816	FLOOD5
1.224	FLOOD6

The user may construct other curves on the diagram and insert corresponding data in a similar element for use in PACKED.

TABLE 1

(Data from Treybal [9] p.168)

Liquid Holdup in Packed Columns:

$$\phi_t = \phi_o + \phi_s ; \quad \phi_{tW} = \phi_{oW} + \phi_{sW} ; \quad \phi_o = \phi_{oW}^{H^*}$$

Correlations:  $\phi_{tW} = \frac{PHTW(10^{-5}) (L')^{BETA}}{d_s^2}$

where  $BETA = 0.965(d_s)^{0.376}$

and  $d_s$  is the diameter of a sphere of same surface area as a single packing particle.

$$\phi_{sW} = \frac{PHSW}{(d_s)^{SWP}}$$

Raschig Rings

Size (mm)

<u>Text symbol</u>	<u>Program Symbol</u>	<u>12,7</u>	<u>25,4</u>	<u>38,1</u>	<u>50,8</u>
$d_s$ (ft)	DS	0,0582	0,1167	0,174	0,238
	PHTW	2,25			
	PHSW	0,00104			
	SWP	1,21			

Berl Saddles

	12,7	25,4	38,1
DS	0,0532	0,105	0,155
PHTW	2,50		
PHSW	0,00032		
SWP	1,56		

These data are contained in subroutine PAKING. Other factors, and the following correlations, are built into PACKED.

(a)  $\phi_s$

$$\text{Raschig rings} \quad \phi_s = \frac{6,85(10^{-5}) \mu_L^{0,02} \sigma^{0,99}}{d_s^{1,21} \rho_L^{0,37}}$$

$$\text{Berl saddles} \quad \phi_s = \frac{1,641(10^{-4}) \mu_L^{0,04} \sigma^{0,55}}{d_s^{1,56} \rho_L^{0,37}}$$

(b)  $H^*$

Ceramic Raschig rings

$$\mu_L' < 12,0 \text{ cp} \quad H^* = \frac{0,897 L^{0,57} \mu_L^{0,13}}{\rho_L^{0,84} (0,1183 L^{0,430} - 1)} \left( \frac{\sigma'}{73} \right)^{0,925 - 0,262 \log L'}$$

$$\mu_L' > 12,0 \text{ cp} \quad H = \frac{0,575 L^{0,57} \mu_L^{0,31}}{\rho_L^{0,84} (0,1183 L^{0,430} - 1)} \left( \frac{\sigma'}{73} \right)^{0,925 - 0,262 \log L'}$$

Ceramic Berl saddles

$$\mu_L' < 20,0 \text{ cp} \quad H = \frac{1,291 L^{0,57} \mu_L^{0,13}}{\rho_L^{0,84} (0,212 L^{0,413} - 1)} \left( \frac{\sigma'}{73} \right)^{1,033 - 0,262 \log L'}$$

$$\mu_L' > 20,0 \text{ cp} \quad H = \frac{0,752 L^{0,57} \mu_L^{0,31}}{\rho_L^{0,84} (0,212 L^{0,413} - 1)} \left( \frac{\sigma'}{73} \right)^{1,033 - 0,262 \log L'}$$

TABLE 2

(Data from Treybal [9] p.166)

Shulman Packing-data Tables.

Interfacial area of packing.  $a_{AW} = mV'^n L'^p$

Range of  $L'$  2441-7323 kg/hr  $m^2$

Packing	Size	Text Symbol	m	n		p
	mm	Prog.Symbol	AMF1	ANF1A	** ANF2A	PF1
Raschig Rings	12,7		8200	3,15(10 <sup>-4</sup> L')	-0,3	-1,04
	25,4		0,274	0,0	0,0	0,552
	38,1		1,82	0,675(10 <sup>-4</sup> L')	-0,1013	0,274
	50,8		0,401	0,0	0,0	0,481
Berl Saddles	12,7		0,0336	0,0	0,0529	0,761
	25,4		15,89	0,686(10 <sup>-4</sup> L')	-0,1029	0,0
	38,1		0,613		-0,0508	0,455

Range of  $L'$  7323-21969 kg/hr  $m^2$

Packing	Size	Text Symbol	m	n		p
	mm	Prog.Symbol	AMF2	ANF1B	ANF2B	PF2
Raschig Rings	12,7		9,32	0,151(10 <sup>-4</sup> L')	+0,148	-0,111
	25,4		463,	0,528(10 <sup>-4</sup> L')	-0,0793	-0,47
	38,1		4,85	0,148(10 <sup>-4</sup> L')	-0,022	0,140
	50,8		0,95	0,0	0,0	0,362
Berl Saddles	12,7		2,54	0,0	+0,0529	0,170
	25,4***		238,	0,42(10 <sup>-4</sup> L')	-0,063	-0,359
	38,1***		465,	0,325(10 <sup>-4</sup> L')	-0,0996	-0,1355

\*\* Factor (10<sup>-4</sup>L') shown with values of ANF1A and ANF1B is included in the program. Only the coefficient is required in the data subroutine (see listing of subroutine PAKING).

\*\*\* Data for  $V' < 3905$  kg/hr  $m^2$ . The values shown above are



included in PAKING. For standard tutorial problems there is no need to change the data if  $V'$  exceeds 3905 [9].

TABLE 3

(Data from Treybal [9] p.162)

Packing Characteristics:

Packing*		Nominal size (mm)			
Raschig Rings		12,7	25,4	38,1	50,8
Symbols		640	160	95	65
Text	Program				
$C_f$	CF				
$\epsilon$	E				
		0,63	0,73	0,71	0,74
Berl Saddles		12,7	25,4	38,1	
Text	Program	380	110	65	
$C_f$	CF				
$\epsilon$	E				
		0,63	0,69	0,75	

\* Both packings are made of ceramic material

## APPENDIX B

### NOMENCLATURE

The nomenclature tables list all major FORTRAN text variables with the corresponding standard symbols used in the text. Where possible, text variables comply with the conventions recommended by the AIChE Report [78]. FORTRAN variables not listed below represent intermediate results in the calculation procedure, and are not carried through the program. Several minor subroutine calling list arguments fall into this category (eg. ALPHA in XYALFA is the value of  $\alpha$  returned to the calling program).

#### B.1 NOMENCLATURE CODES

<u>Code</u>	<u>Program</u>
BAT	BATCH
BIN	BINARY
PAC	PACKED
MUL	MULCOM
PLO	PLOTTER
SPL	Special - variables falling in this category are specific to the routine in which they appear.
T	Variable Type: R - real I - integer A - alphanumeric
DIM	Dimension type 1 - single dimension 2 - two dimensional array
COM	Common variable - variable appears in COMMON block.
PARA	Dimension parameters (see below).

Subscripts/Arguments:

<u>Text</u>	<u>Program</u>	<u>Purpose</u>
i	I	component number
j	J	general tray number
k	K	general counter
m	M	tray number in stripping section
n	N	tray number in rectifying section

B.2 PARAMETER definitions

For binary systems certain variables have fixed dimensions depending on their purpose. All other arrays are dimensioned using a parameter.

Parameter	Purpose	Program
<u>(a) Binary programs</u>		
NPAK	Data used for height-of-packing integration.	PACKED
NPBS	Data used for batch distillation integration.	BATCH
NPEN	Enthalpy data.	BINARY,PACKED
NPEQ	Equilibrium data.	BATCH,BINARY,PACKED
NPPS	Ponchon-Savarit diagram data.	BINARY,PACKED
NPTC	Theoretical tray construction on x,y diagram.	BATCH,BINARY

(b) Multicomponent programs

MC	Components
MP	Trays



PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
ALX1,ALY1		R	Origin of each diagram set relative to point of completion of previous diagram. The overall layout is constant although the overall size may change					*			*	
ALX4,ALY4		R	Final positioning of pen on completion of whole job.						*			
AMAT		R	Array required in least squares fit procedure by UNIVAC library routine "GJR" (see subroutine FDATA)						*	2		20,20
AMF1,AMF2	m	R	Coefficient in equ. 4.16. Defined in Appendix A Table 2			*						
AMLCN,AMVCON		R	Conversion factors: centipoise and micropoise - lb hr <sup>-1</sup> ft <sup>-1</sup> .			*						
AMUML,AMUMV	$\mu_{mL}, \mu_{mv}$	R	Liquid and Vapour phase viscosities (Equ. 4.28 and 4.30) returned by subroutine VISCL and VISCV respectively					*				
AN,BN,CN	$A_1, B_1, C_1$	R	Antoine constants in MULCOM (see equ. 7.4.6)				*			1	*	MC
ANFUNC	n	R	Exponent in equ. 4.16. A function of ANFIA and ANFIB or ANF2A and ANF2B			*						
ANF-1A,1B,2A,2B		R	Factors for computing ANFUNC (see Appendix A Table 2)			*						
AREA		R	Cross-sectional area of packed column			*						
ASFSM		R	Conversion factor: ft <sup>2</sup> ft <sup>-3</sup> m <sup>2</sup> m <sup>-3</sup>			*						
ASIZ		R	Size of character (inches) used on plotted diagrams			*		*			*	
ASTRT,BSTRT		R	Origin of axes for x,y diagram in subroutine PLOTMT						*			
ASURF	$a_A$	R	Computed value of $a_A$ in m <sup>2</sup> m <sup>-3</sup> at points in the column			*				1		NPAK
ATM		R	Conversion factor: atmosphere pressure to lb.in <sup>-2</sup>	*	*	*	*					
AVEL,AVEV		R	Average molecular weight of liquid and vapour phases.					*				
B	$b_1$	R	Molal withdrawals rate of component 1 in Bottoms (equ. 5.16)				*			1	*	MC
BD	$b_1/d_1$	R	Ratio of $b_1$ and $d_1$ as defined in equ. 5.34				*			1	*	MC
BETA	$\beta$	R	Factor defined in Table 1, Appendix A (computed in subroutine PAKING)					*				
BM,DM		R	Factors used in enthalpy balance calculations. See sections 5.53 and 5.562 (from equ. 5.10)				*				*	
BOT,TOP,FEED		R	Limits of integration defining extremities of packed section of column (see Fig. 4.1)			*						
BTOT	B	R	Total molar Bottoms product withdrawal rate (kg moles/hr)	*	*	*	*				*	

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
CB,SLOPB		R	c,m in y=mx+c for operation line in bottom section of column for open steam operation.						*			
CDLDV		R	Molar ratio of liquid and vapour in two phase distillate product.						*			
CF	C <sub>f</sub>	R	Characterization factor of packing (Table 3, Appendix A)			*			*			
CFACCT,SLOPE		R	c,m in y=mx+c used generally in calculations.						*			
CFCT,SLPE		R	c,m in y=mx+c for each operating line on McCabe-Thiele diagram.	*						1	*	12
CHEKL,CHEKV		R	Liquid and vapour flowrates from previous iteration.						*			
CHEKLU,CHEKLL		R	Upper and lower limits constraining new liquid flowrates.						*			
CHEKVU,CHEKVL		R	Upper and lower limits constraining new vapour flowrates.						*			
CMAT		R	Array required by UNIVAC routine GJR (see AMAT)						*	1		20
COEFF		R	Solution vector in least squares curve fitting procedure.						*	1		20
COMPNO		R	Storage of component numbers 1-NC in real form for use by PLOTTER.				*		*	1		MC
CONFAC		R	Conversion factor: kJ/hr - kW						*			
CP	C <sub>p1</sub>	R	Heat capacity defined by Kobe [55] correlation (equ.3.19).						*			
CPCONV,CPMW		R	C <sub>p</sub> conversion factors (see section 7.235).	*	*	*				1	*	2
CRHO	c	R	Molar density of heavy component (liquid phase) (equ.4.9)			*						
D	d <sub>1</sub>	R	Molal withdrawal rate of component 1 in the distillate (equ.5.13)				*			1	*	MC
DEL		R	General difference factor in calculation						*			
DELB,DELD	Δ <sub>B</sub> ,Δ <sub>D</sub>	R	Enthalpy coordinates of Ponchon-Savarit Δ points.	*				*			*	
DELCB,FACT,NFACT		R	Factors used in determining the number of integration intervals in the stripping section based on the size of the interval used in the rectifying section (see subroutine OPLINE)						*			
DELDM	Δ <sub>DM</sub>	R	Δ point defining minimum reflux in subroutine RMIN (equ.3.25)						*			
DELTAC,DELTAP		R	Pressure drop across condenser and trays respectively (in MULCOM)				*				*	
DIAM,DMI		R	Diameter of packed column (metres)			*						
DIFUSL,DIFUSV	D <sub>L</sub> ,D <sub>v</sub>	R	Calculated liquid and vapour diffusivities.			*						

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	DIN	PAC	MUL	PLO	SPL	DIN	COM	PARA
DL, DV		R	Liquid and vapour fractions of a two phase distillate.		*						*	
DLCON, DVCON		R	Conversion factors: $\text{cm}^2\text{sec}^{-1} - \text{ft}^2\text{hr}^{-1}$			*						
DS	$d_s$	R	Diameter of sphere with same surface area as a packing particle. ft. (see Appendix A, Table 1).			*						
DTOT		R	Total molal Distillate product withdrawal rate.	*		*	*				*	
E	$\epsilon$	R	Fractional void volume (dry bed) equ. 4.11 (see Appendix A, Table 3).			*						
EFF	$E_M$	R	Murphree tray efficiency factor for Binary system (equ. 3.7).	*							*	
EFFV	$EV_{1,j}$	R	Holland's vaporization efficiency (equ. 5.61).				*			2	*	MC, MP
ENCONV		R	Conversion factor: Btu/lbmole - kJ/kgmole.		*	*	*					
EO	$\epsilon_o$	R	$\epsilon$ for irrigated packing (equ. 4.11)			*						
ERROR, ERRD		R	Errors in convergence criteria checks (see section 5.58)				*					
EVENS, ODDS		R	Simpson's Rule factors (see section 2.54).						*			
EXAX, EYAX		R	Ponchon-Savarit diagram titles.					*				
EXTEND		R	Variable spacing diagrams for multicomponent systems.					*				
FACTP1, FACTP2		R	Conversion factors: Used to convert specified pressure units to mm Hg						*			
FAL, FA2		R	Factors in calculation of liquid phase mass transfer coefficients ( $FAL \cdot FA2 = k_L$ in equ. 4.9)			*						
FEED	$FZ_1$	R	Moles of component i in feed (subroutine OUTPUT).				*		*	1		MC
FENTH	$H_F$	R	Enthalpy of feed as specified for feeds at a greater pressure than the column.				*					
FL, FV	$F_L, F_V$	R	Mass transfer coefficients defined by equ. 4.9 and 4.8.			*						
FLIQ, FVAP, FTEMP, FX		R	Arrays corresponding to COMMON array LIQR, VAPR, TRL, XROP or LIQS, VAPS, TSL, XSOP when FFIN is called by STEPD.						*	1		NEPS
FLOL, FLOV	$L_j, V_j$	R	Liquid and Vapour flows leaving each tray as calculated in FFIN for Ponchon-Savarit designs.						*			
FLOLIM		R	Factor defining values of CHEKLU, LL etc. in FLOWNE (depends on ICONF.)						*			

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
FLOW		R	Array storing feed and side stream flowrates in positional order as determined in MULTFS.		*					1	*	10
FTOT	F	R	Total feed rate for simple columns (kg moles hr <sup>-1</sup> )	*	*	*	*				*	
FTOTS, STOTS		R	Total feed and side stream rates for complex columns.		*					1	*	5
FUNC		R	Values of function to be integrated by Simpson's Rule.	*						1		NPBS
FUNCR, FUNCS		R	Values of function in rectifying and stripping sections in PACKED, to be integrated.			*				1		NPAK
F <sub>1</sub> , F <sub>2</sub> , F <sub>3</sub> , F <sub>4</sub>		A	Plotting data file names.				*	*				
-	f <sub>i,j</sub>		Molar feed (or withdrawal if negative) of component i to general stage j (equ.1.2).									
G	θ <sub>T</sub> , G	R	Batch time (θ <sub>T</sub> ) or Boil-up rate (G) as specified. Related by equ.2.15.	*								
GC	g <sub>C</sub>	R	Unknown value determined from equ.2.14, 2.17 or 2.20.						*			
GC		R	Gravitational force constant 4,18.10 <sup>8</sup> lb.ft.lb <sub>j</sub> <sup>-1</sup> hr <sup>-2</sup> .			*			*			
GOP		R	Moles of open steam required to replace reboiled vapour.	*					*			
H		R	General enthalpy value used in calling lists in HTSOLN and HFIND.						*			
H		R	Size of integration interval used in BATCH and OPLINE (see also NINT).						*			
HB	h <sub>B</sub>	R	Total enthalpy of Bottoms product.	*	*						*	
HD	h <sub>D</sub>	R	Total enthalpy of Distillate product regardless of state.	*	*		*				*	
HEAD1-HEAD4		A	Diagram headings in PLOTTER.							1		5
HENTH, XYH		R	Arrays corresponding to HLIQ, XH or HVAP, YH in HFIND.					*	*	1		NPEN
HEVABS		R	Factor associated with heat of solution data. Equals +1 or -1 for endothermic or exothermic mixing respectively.	*	*						*	
HF, HFEED	H <sub>F</sub>	R	Calculated feed enthalpy kJ kgmole <sup>-1</sup>	*	*		*				*	
HFUNC	H*	R	Empirical factor in equation 4.15. Defined in Table 1, Appendix A.			*						
HLI, HVD		R	Liquid and vapour phase distillate enthalpies.	*	*	*			*		*	
HLF, HVF		R	Liquid and vapour phase feed enthalpies.	*	*	*			*		*	
HLIQ, HVAP	h <sub>j</sub> , H <sub>j</sub>	R	Enthalpy of liquid and vapour stream in columns. kJ.kgmole <sup>-1</sup> in binary and multicomponent programs.	*	*	*		*	*	1	*	NPEN MP



PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
HL1,HL2,HL3 HV1,HV2,HV3 HS		R	Coefficients for enthalpy functions considered in ENERGY (see section 7.432) Calculated heat of solution included in calculation of $h_j$ (see section 3.59) Conversion factors (see section 7.235)		*		*			1	*	MC
HS CONV, HSMW		R	Heat of solution composition data supplied.	*						1	*	2
HSOLN,XHS		R	Enthalpy factors used in calculations in RMIN HTEST $\neq$ HF within limit ACC after 500 iterations causes termination of search	*						1	*	30
HTEST,HVCM		R	Subscript and 'DO' loop counter referring to components.					*		1		
I	i	I	Array required by CALCOMP plotting software.							1		1000
IBUF		I	Counter determining number of overall iterations.				*			1	*	
ICOUNT		I	Numerical definition of : and / in CALCOMP software									
INTA,B		I	Numerical codes for symbols plotted on diagrams.				*					
INTEQ,I		I	Tray number counter in BUBCAL and DEWCAL									
IT		I	Counter defining run number for PLOTTER.				*					
IUSE		I	Subscript and 'DO' loop counter referring to tray number.				*					
J	j	I	Array required by UNIVAC routine GJR. It is used to return the following data if and/or error occurs in GJR: (a) JC(1)=-I where I = last correctly reduced row. (b) JC(1)=I where I = last row before singularity was detected.							1		20
JC		I	Equilibrium constants defined by equ.5.1				*					MC,MP
K	$K_{1,j}$	R	Equilibrium constants at feed stream				*			2	*	MC
KF	$K_{1,F}$	R	Mass transfer coefficients, liquid and vapour phase.				*			1	*	MC
KX,KY	$K_x, K_y$	R	Liquid and Vapour molar flowrates at points through the packing (kg moles/hr $m^2$ ).			*				1		NPAK
LV	$L_j, V_j$	R	Defined by equ.5.29 and 5.25			*				1		NPAK
LB,LD	$l_{1,m}/b_{1'}$ $l_{1,n}/d_1$	R	Code defining plotting procedure in CALCOMP software.				*			2	*	MC,MP
LFREQ,LFREQ1		I	Latent heat values for each component (binary).					*				
LH	$LH_{1,TBP}$	R		*						1	*	2

			R	Conversion factors: see section 7.235		*	*	*	
LHCONV, LHMW			R						
LHT		$LH_i, T$	R	Latent heat of component i calculated by equ 3.20.					
LIQR, LIOS		$L_n, L_m$	R	Liquid molar flowrates calculated at points (XROP,YROP) and (XSOP,YSOP) in the rectifying and stripping sections.	*				NPPS
LMFR, VMFR		$L_{1,j}, V_{1,i,j}$	R	Molal flowrate of component i leaving tray j in liquid and vapour phase respectively.		*			MC,MP
LPROF		$L_j$	R	Molar liquid profile through multi-component column.		*			MP
LRTOT, LSTOT		$L_n, L_m$	R	Molar liquid flowrates in rectifying and stripping sections.		*			MP
LRWT, LSWT			R	Liquid mass flowrates corresponding to LIQR,S in binary system. Required by PACKED.	*	*			NPPS
M		m	I	Subscript and 'DO' loop counter referring to trays in the stripping section of a column.					
MAIF			I	Counter indicating constant relative volatility (=2) or otherwise (=1). Also used as termination counter (= 1000).	*	*			
MC, MP				See PARAMETER definitions: Section B.2					
MCAN			I	Counter activation (= 1) or suppressing (= 2) the conversational facility.	*	*			
MEFF			I	Counter indicating type of efficiency data supplied.		*			*
MFIN			I	(See NFIN)					
MMD			I	Counter causes efficiency factors to be ignored for total condenser or feed stream bubble point and dew point calculations (see section 5.55).		*			
MN, NCALL			I	Counters directing iterative search procedure (see subroutine SEARCH section 2.55).		*			
MPTS			I	See NPTS					
MPTI			I	MPTS + NFIN (defines bottom of rectifying section relative to NTOT).					
MSTOP			I	Terminates stepping off of McCabe-Thiele stages in XDSTEP.	*				
MT			I	Total number of theoretical stages stepped off on x,y diagram in STEPDP.		*			
MTIMES			I	Counter prevents recalculation of tray pressure profile and activates conversational facility for adjusting variables to be used in second and subsequent runs.					

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	RIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
MW		R	Molecular weight of components in binary systems.		*	*				1	*	2
N	n	I	Subscript and 'DO' loop counter referring to the rectifying section of a column.						*			
NECP, NCCP, NDCP		I	Exponents relating to Kobe function coefficients (see section 7.234)						*			
NBDC		I	Counter selecting bubble point and dew point calculation procedure: = 1 for Antoine constants = 2 for Equilibrium data	*	*							
NC	NC	I	Number of components in multi-component systems.				*				*	
NCERR, NLOOP		I	Counters used in OPLINE to ensure that the number of points chosen does not exceed the dimension limit NPPSC = NPPS - 2						*			
NCODE		I	Counter selects descriptive output in TEACH.						*			
NCOND		I	Counter = 1 - total condenser operation = 2 - partial condenser operation.	*	*	*	*				*	
NCOUNT, MCOUNT		I	Equivalent to NPTS, MPTS in mainline routines. Define number of points on rectifying and stripping section operating lines. (see section 3.512)						*			
NCR		I	READ key (= 8 on UNIVAC system)	*	*	*	*	*			*	
NCURVE		I	Counter selecting correlation curve data for FDATA from Fig 1, Appendix A (Equivalent to NFLOOD from PACKED)			*			*			
NDAT		I	Number of equilibrium data points in XEQUIL. YEQUIL (= NDATA + 2).						*			
NDATA		I	Number of equilibrium data points excluding O., O. and 1., 1.	*	*	*					*	
NEN		I	Counter defining enthalpy calculation (= 1 or 2 from QFLASH, = 3 from MULCOM)						*			
NENTH		I	Counter indicating method of performing enthalpy calculations, see section 7.432 (= 0 if enthalpy is not considered).				*				*	
NENTH		I	Number of enthalpy-composition data points defining lines on Ponchon-Savarit diagram.		*						*	
NEQ		I	Equivalent to NEN for equilibrium calculations (= 1 from QFLASH, = 3 from MULCOM).						*			
NEQUIL		I	Counter indicating equilibrium constant correlation used.				*				*	

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	RAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
NF		I	Feed tray position specified by user or Number of feed streams specified for McCabe-Thiele design.		*		*				*	*
NFACT		I	See DELBC.									
NFIN,MFIN		I	Point on operating line representing the bottom of the column packing. Defined by stepping off the reboiler stage.			*			*			
NFLOOD		I	Counter indicating selected curve on on Fig. 1, Appendix A.			*			*			
NF1,NF2		I	Feed tray position and number of tray above feed tray. (NF1 = NF + 1 in internal coding in MULCOM section 8.2 )				*				*	
NG1,NG2,NG3,NG4		I	Write keys for plotting data files. Defined in DATA statements in each mainline.	*	*		*				*	
NHCAT,NLCAT		I	Counters indicating component category (see section 7.32 ).			*						
NHS		I	Number of HSOLN,XHS data pairs supplied.	*	*						*	
NINT		I	Number of intervals of size H chosen for numerical integration procedure in BATCH and OPLINE.	*	*	*					*	
NJ		I	Counter identifying base components when equilibrium data is supplied in terms of relative volatilities. (section 5.55 ) Also used to terminate the run if errors are detected in subroutines.				*				*	
NLP		I	WRITE key for data file setup for for detailed result table.				*				*	
NN		I	Number of operating lines on McCabe-Thiele x,y diagram (= NF + NS + 1).	*	*				*		*	
NN1		I	Number of points defining the operating lines (= NN + 1).	*	*				*			
NOP		I	Counter: = 1 for McCabe-Thiele diagrams. = 2 for Packed column designs = 3 for Ponchon-Savarit tray tray column designs.		*	*					*	
NOPI		I	Counter defining Batch operation: = 1 Varying reflux = 2 Constant reflux	*								
NORD		I	Order of polynomial with best fit (subroutine FDATA).						*			
NOS		I	Counter: = 0 - partial reboiler = 1 - open steam									
NOUT		I	Counter limiting number of detailed result tables to 5.				*					

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
NOWN		I	Counter indicating enthalpy calculation routine used.				*				*	
NP	N	I	Number of trays in the column specified by the user. Excludes reboiler and condenser.				*				*	
NPACK		I	Counter indicating specified size of packing.			*						
NPH,NPHYS		I	Counters ensure that physical property data are read as required.			*						
NPOINT		I	Indicates point of maximum liquid flowrate in a packed column.			*						
NPOLY		I	Equals (NORD + 1) in FDATA (number of terms in polynomial).					*				
NPTS,NPTS		I	Number of points defined on rectifying and stripping section operating lines.		*	*						
NP1,NP2		I	Top tray and condenser numbers (interval numbering system. See section 8.2).				*				*	
NQVAL		I	Counter indicating feed conditions = 1 - feed at feed tray pressure = 2 - feed at pressure > feed tray pressure.				*				*	
NRDM,NR		I	Counters indicating availability of solution density - composition data, and when this is to be used.									
NRFIN		I	Point on rectifying section operating line representing the bottom of this section of packing defined in NFIN.			*						
NS		I	Number of side streams in complex columns.		*						*	
NSORT		I	Counter indicating the phase controlling the mass transfer.			*						
NSPEC		I	Counter: = 1 if $X_{DAVE}$ is not specified = 2 if $X_{DAVE}$ is specified.	*								
NTC		I	Number of points defining stepped stages on McCabe-Thiele x,y diagram.	*								
NTIME		I	Counter causes subroutine ENTHAL to read data once only.						*			
NTIMES,NTIM		I	Counters in plotting program defining number of curves plotted.					*				
NTOT		I	Total number of points defining Ponchon-Savarit operating lines = NPTS + MPTS.		*	*						
NTP		I	General WRITE key for writing to terminal in demand mode or line printer in batch mode (= 5 on UNIVAC system).	*	*	*	*	*			*	

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	EAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
NTRA		I	Indicates position of feed tray in data tables for Ponchon-Savarit design.		*							
NTRAY		I	Stores feed and side stream tray numbers sequentially during stepping procedure.		*					1	*	12
NTYPE		I	Counter indicating type of packing specified.			*						
ORD		R	Ordinate value of point on flooding correlation chart (see Fig 1, Appendix A).			*						
PAL,P1	$P_1$	R	Vapour pressure of light component (equ.2.9).	*	*			*				
PCOND	$P_T$	R	Condenser pressure in mm Hg.				*				*	
PCONDA	$P_T$	R	Specified condenser pressure in convenient units.	*	*	*	*					
PFACT1,PFACT2		R	Factors in empirical expression defining $\phi_s$ (see Table 1, Appendix A.)			*						
PF1,PF2	$I$	R	Empirical constants shown in Table 2, Appendix A (equ.4.16).			*						
PHIO	$\phi_o$	R	Operating holdup ft <sup>3</sup> liquid/ft <sup>3</sup> packed volume (equ.4.15).			*						
PHIOW	$\phi_{ow}$	R	Operating holdup with respect to pure water.									
PHIS	$\phi_s$	R	Static holdup (see Table 1, Appendix A).			*		*				
PHISW	$\phi_{sw}$	R	Static holdup with respect to water.			*		*				
PHIT	$\phi_t$	R	Total holdup (equ.4.13).			*		*				
PHITW	$\phi_{tw}$	R	Total holdup with respect to water (equ. 4.14).			*		*				
PHSW,PHTW		R	Empirical factors shown in Table 1, Appendix A.			*			*			
POWER		R	Exponent relating to multicomponent enthalpy correlation (see section 7.43)				*		*			
PRESS,PRES		R	Tray pressure profile in mm Hg and kPa respectively.				*			1	*	MP
PT,PT1	$P_T$	R	System pressure in mm Hg and kPa respectively (PT = PCONDA converted to mm Hg).	*	*	*					*	
PVAP	$P_{1,j}$	R	Multicomponent vapour pressure data (equ.2.9) used in subroutine EQUCON.				*		*			
Q	$q_j$	R	Heat loss from general stage j (equ.1.3)									
QB	$q$	R	Feed condition for simple columns.		*	*	*				*	
QC	$Q_B$	R	Reboiler duty (kW) (Binary systems)		*	*	*				*	
QF,QFTOT	$Q_C$	R	Condenser duty (kW)		*	*	*				*	
	$H_F, F, H_F$	R	Molar enthalpy and heat content of multicomponent feed (kJ kgmole <sup>-1</sup> ,kW)		*	*	*				*	

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	DAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
QF, QS	$q_F, q_S$	R	Condition of feed and side streams in complex columns.		*					1	*	5
QFS	$q_F$ and $q_S$	R	Array storing $q_F$ and $q_S$ values as arranged in MULFS.		*					1	*	10
Q MIX, Z MIX		R	q value and composition of "feed" representing all feed and side streams in an overall mass balance for complex columns.		*							
QR		R	Reboiler duty for multicomponent systems (kW).				*				*	
QX, QY		R	Arrays storing coordinates of NN1 points defining operating lines in binary McCabe-Thiele systems.		*					1	*	12
R	R	R	Binary system reflux ratio	*	*						*	
RHOCON		R	Conversion factor for density data: $g\ ml^{-3} \rightarrow lb\ ft^{-3}$ .		*				*			
RHOL, RHOLL, 2		R	Liquid mixture and component densities from RHOLIQ.		*							
RHOV		R	Calculated vapour mixture density									
RHOW		R	Density of water from RHOH2O									
RM	$R_M$	R	Minimum reflux ratio in binary systems	*	*						*	
RR	R	R	Reflux ratio in multicomponent systems				*				*	
S	$S_{i,j}$	R	Stripping factors (equ. 5.22)				*			2	*	MC, MP
SCHL, SCHV	$Sc_L, Sc_V$	R	Liquid and vapour phase Schmidt numbers		*				*			
SIGCON		R	Conversion factor for surface tension data.		*				*			
SIGMA, SIG1, SIG2	$\sigma$	R	Mixture and component surface tension data from SURTEN (see equations 4.22 - 24).		*				*			
SLIQ, SVAP	$El_{i,j}$ $Ev_{i,j}$	R	Summations according to equations 5.43 and 5.44.				*			1	*	MP
SLPE, CFCT		R	Slope and intercept ( $y = mx + c$ ) of McCabe-Thiele operating lines.		*					1	*	12
STR, STR1		A	Names used in printing result table by BINRES.	*	*							
SUM, SUMINT		R	Result of numerical integration from SIMPS.	*	*				*			
SUMB, SUMD	$Ib_{ico}$ $Id_{ico}$	R	Summation of correlated values in BDCORR.				*					
SUMD1	$D_{CALC} = Id_{ico}$	R	Value from BDCALC used in convergence test ( $D_{CALC} - D_{SPEC}$ ).				*					
SUMERR		R	Sum of squared errors in least squares fit (FDATA).		*		*					

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
SWP		R	Exponent in expression for $\phi_s$ (see Table 1, Appendix A).			*			*			
SX,Y,XF,YF	$EX_{1,j}, EY_{1,j}$ $EX_{1,F}, EY_{1,F}$	R	Summation of composition values in OUTPUT shown in detailed result table.				*					
S1,S2,S3,S4		R	Enthalpy data returned to QFLASH for use in flash calculations.				*		*			
T,T1	$T_j$	R	$T = T_j \cdot 10^{-2}$ and $T1$ is $T_j$ converted to units required by equilibrium or enthalpy correlations.				*			1	*	NPEQ
TB,TBPT,TBUB,TBUBPT		R	General bubble point values.						*			
TBASE	$T_b$	R	Base temperature for enthalpy calculations.		*						*	
TBOT,TTOP		R	Estimate of reboiler and condenser temperatures used to obtain initial temperature profile for Thiele Geddes method.				*				*	
TBP	$T_{BP}$	R	Normal boiling points of light and heavy components.		*					1	*	2
TC	$T_C$	R	Critical temperature of each component.		*					1	*	2
TCONV		R	Conversion factor (temperature): $^{\circ}C \rightarrow K$ .						*			
TD,TDPT,TDEW,TDEWPT		R	General dew point values									
TEMP		R	Temperature profiles in $^{\circ}C$ (MULCOM)				*			1	*	MP
TEMPF		R	Temperature profiles in $^{\circ}F$ (subroutine RUNRES).				*			1	*	MP
TF,TFEED	$T_F$	R	Feed temperatures for Binary and Multicomponent systems respectively ( $^{\circ}C$ ).		*		*				*	
TG		R	General temperature value used to start iterative calculation procedures.		*						*	
THETA	$\theta$	R	Computed value of $\theta$ (equ.5.50).				*				*	
THEOLD		R	Starting or previous value of $\theta$ .				*				*	
TIME	$\theta_T$	R	Either batchtime or boil-up rate (see G).	*								
TITLES		A	Names of components in all distillation programs.	*			*			2	*	2,5 or MC,5
TITL,TITLT		A	Titles used in plotting routine.					*				
TNEW,TOLD		R	Forcing procedure limiting change profile in BUBCAL and DEWCAL.				*		*			
TRANK		R	Conversion factor in ENERGY: $^{\circ}F \rightarrow ^{\circ}R$ .				*		*			
TRAYNO		R	Total number of trays, including fractional part of any, stepped off on x,y diagram.		*						*	



PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
TRAYNO		R	Tray numbers stored as real variables for use in plotting routines (CALCOMP software will not accept integers - see COMPNO).				*			1		MP
TRL,TRV TSL,TSV		R R	Temperatures of streams corresponding to flowrates LIQR,VAPR,LIQS,VAPS. Used in PACKED.		*	*				1 1	* *	NPPS NPPS
T2B,T2D		R	Bubble point and dew point of the liquid and vapour mixtures for which HL and HV are required.		*				*			
U1,U2,U3,U4		R	Used to assign WRITE keys NG1 - NG4 to files F1 - F4 in READIN.				*					
V		R	Array required by UNIVAC routine GJR specifying the type of operation required. In this application V(1)=4 (line 54) indicates that the equations in AMAT are to be solved by GJR.			*			*	1		2
VAP	V	R	Total quantity of vapour boiled up during batch cycle (equ. 2.15)	*								
VB,VD	$V_{1,m}/b_1$ $V_{1,n}/d_1$ $V'$	R R R	Ratios defined by equation 5.28 and 5.26.				*			2	*	MC,MP
VLOAD		R	Superficial vapour mass velocity kg mm <sup>2</sup> at NPOINT.			*						
VMOLE	V	R	Superficial vapour molar velocity kgmole hr m <sup>2</sup> at NPOINT.			*						
VPDL		R	Specified vapour pressure-drop through packing. (FDATA fits polynomial to corresponding curve Fig. 1, Appendix A).			*						
VPROF		R	Molar vapour profile through multi-component column.				*			1	*	MP
VRTOT,VSTOT		R	Molar vapour flowrates in rectifying and stripping sections respectively.				*			1	*	MP
VYR,VYS	V.Y or L.x	R	Data stored depends on form of equation 4.3 used for integration. V.Y - for vapour phase limiting mass transfer L.x - for liquid phase limiting mass transfer.			*				1		NPAK
W	W	R	Residue liquid in batch operation	*								
X,Y		R	Coordinates of points defining stepped stages on x,y diagrams.	*	*					1	*	NPTC
X,Y		R	Compositions of liquid and vapour streams leaving each tray (related by equ.5.1).				*			2	*	MC,MP
XALF,YALF		R	Equilibrium composition data defining equilibrium curve on x,y diagram.	*						1	*	NPEQ
XB,YB	$x_B, y_B$	R	Coordinates of point defining lower end of the bottom operating line		*						*	

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	BAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
XC, YC	$x_1, y_1$	R	Point of intersection of top operating line and 45° diagonal on x,y diagram (see Fig. 3.2).		*						*	
XCR, YCR	x	R	Point of intersection of Ponchon-Savarit operating lines (see Fig 3.3)		*	*					*	
XD, YD	$x_D, y_D$	R	Coordinates of point representing distillate product on x,y diagram see (Fig. 3.2).		*						*	
XDATA, YDATA		R	Values used in numerical integration in BATCH.	*						1		NPBT
XDAVE	XDAVE	R	Calculated average distillate composition.	*								
XEQUIL, YEQUIL		R	Copy of XALF, YALF, but includes O., O. and 1., 1. for result table and plotting purposes.	*	*					1	*	NPEQ
XF, YF	$x_F, y_F$	R	Composition of liquid and vapour phase of feed.				*			1	*	MC
XFLO, YFLO	$L_F, x_F$ $V_F, y_F$	R	Moles of each component in liquid and vapour phase of feed.				*		*			
XH, YH		R	Compositions corresponding to HLIQ and HVAP.		*	*				1	*	NPEN
XI, YI	$x_1, y_1$	R	Equilibrium compositions at the liquid-vapour phase interface (see Fig 4.1)			*				1		NPAK
XNTC		R	Computed value of residue composition used in iterative determination of XDATA, YDATA.	*								
XOUT		R	Composition value used in calling list to PSOPXY.		*							
XQM, YQM		R	Point of intersection of overall operating lines in complex columns. (see QMIX and ZMIX)		*				*			
XROP, YROP XSOP, YSOP		R	Coordinates of points defining Ponchon-Savarit operating lines on x,y diagram.		*	*				1 1	* *	NPPS NPPS
XSPEC		R	Specified average distillate composition (see NSPEC).	*								
XW		R	Residue composition at end of batch run.	*								
YINCPT		R	Y-axis intercept (c in $y=mx+c$ ) of operating line on batch x,y diagram.	*				*				
Z	$z_F$	R	Overall composition of feed in binary systems.	*	*	*					*	
Z	$z_1, F$	R	Overall composition of feed in multicomponent systems.				*			1	*	MC
ZF, ZS		R	Overall composition of feed and side streams - complex columns. Arrays ZF1, ZS1 required in MULTFS.	*	*					1	*	5

PROGRAM VARIABLE	TEXT SYMBOL	T	DESCRIPTION	EAT	BIN	PAC	MUL	PLO	SPL	DIM	COM	PARA
ZFS		R	Combined ZF and ZS as arranged in order by MULTFS (see QFS).		*					1	*	10
ZRECT, ZSTRIP	Z <sub>R</sub> , Z <sub>S</sub>	R	Height of packing required in rectifying and stripping sections respectively.			*			*			
ZFLO, ZFS		R	Flowrates and compositions of feed and side streams for plotting purposes.					*		1		15

## APPENDIX C

The FORTRAN listings of all program elements contained in the package are shown here. The name of the mainline or subroutine is displayed on the bottom right hand corner of each page. Line numbers referred to in the main text are shown to the right of the FORTRAN statement.

Programs are listed in the order discussed in the main text.

Subroutine TEACH: This subroutine provides a brief description of each program in the package, excluding PLOTTER (see User Manual, Chapter 7). It is self-explanatory and is not discussed in the text.

### Chapter 2: Program BATCH

Mainline BATCH (section 2.341 of main text)

Subroutines associated with BATCH:

Subroutine	section
XDSTEP	2.51
BINBPT	2.521
BINDPT	2.522
XYALFA	2.53
SIMPS	2.54
SEARCH	2.55

### Chapter 3: Program BINARY

Common block BINCOM

Mainline BINARY (section 3.34 of main text)

Subroutines associated with BINARY:

Subroutine	section
READBD	3.51
CONDEN	3.52
FLASHB	3.53
EQUIL	3.54
MULTFS	3.55
SIMPLE	3.56
PONSAV	3.57
HTCAP	3.58

Subroutine	section
HTSOLN	3.58
HFIND	3.510
RMIN	3.511
OPLINE	3.512
FLWS	3.513
AVEMW	3.514
CHEC	3.515
STPD	3.516
PSOPXY	3.517
FFIND	3.518
BINRES	3.519

#### Chapter 4: Program PACKED

Common block PACCOM

Mainline PACKED (section 4.34 of main text)

Subroutines associated with PACKED:

Subroutine	section
PACRES	4.51
PACTYP	4.52
PAKING	4.53
FDATA	4.54

Physical property subroutines:

RHOLIQ	4.551
SURTEN	4.552
DIFLIQ	4.553
DIFVAP	4.554
VISCL	4.555
VISCV	4.556
RHOH2O	4.557

#### Chapter 5: Program MULCOM

Common block MULTICOM

Mainline MULCOM (section 5.36 of main text)

Subroutines associated with MULCOM:

Subroutine	section
READIN	5.51
EQUCON	5.52
ENERGY	5.53
QFLASH	5.54
BUBCAL,DEWCAL	5.55
FLOWEQ,FLOWNE	5.56
ABSTRP,VBLB,	5.57
VDLD,BDCALC,	"
BDCORR,VLRECT,	"
VLSTRP	"
THECAL	5.58
RUNRES,OUTPUT	5.59

Chapter 6: Program PLOTTER

Mainline PLOTTER (section 6.2 of main text)

Subroutines associated with PLOTTER:

Subroutine	section
PLOTTS	6.31
PLOTFS	6.32
PLOTDS	6.33
PLOTCS	6.34
PLOTBT	6.35
PLOTMT	6.36

```

0001 SURROUTINE TEACH(1)
0002 NTP=5
0003 GO TO (100,200,300,400),I
0004 100 CONTINUE
0005 WRITE(NTP,1000)
0006 1000 FORMAT(/T2,'THIS IS A BINARY BATCH DISTILLATION SIMULATION PROGRAM
0007 1.,./T2,'SYSTEMS OPERATING UNDER EITHER VARIABLE OR CONSTANT REFLUX
0008 2.,./T2,'CONDITIONS CAN BE CONSIDERED.,./T2,'THE PROGRAM IS LIMITED
0009 3TO SYSTEMS FOR WHICH NORMAL.,/T2,'MCCABE-THIELE ASSUMPTIONS HOLD.,
0010 4./T2,'PLOTTING ROUTINES ARE AVAILABLE FOR CONSTRUCTING MCCABE-THIE
0011 5LE.,/T2,'X-Y DIAGRAMS FOR EITHER OPERATION MODE.,')
0012 WRITE(NTP,1001)
0013 1001 FORMAT(/T2,'REFER TO THE DISTILLATION PACKAGE MANUAL,SECTION 1.,./
0014 1T2,'FOR FURTHER INFORMATION.,')
0015 GO TO 999
0016 200 CONTINUE
0017 WRITE(NTP,2000)
0018 2000 FORMAT(/T2,'THIS IS A BINARY CONTINUOUS DISTILLATION DESIGN PROGRA
0019 1M.,./T2,'IT WILL COMPUTE THE NUMBER OF THEORETICAL STAGES REQUIRED
0020 2.,/T2,'TO ACHIEVE A SPECIFIED TOP PRODUCT PURITY.,/T2,'A MURPHREE
0021 3 PLATE EFFICIENCY FACTOR MAY BE SPECIFIED.,/T2,'THE DIMENSIONS OF
0022 4 THE COLUMN REQUIRED ARE NOT CALCULATED.,/T2,'SYSTEMS FOR WHICH EI
0023 5THER MCCABE-THIELE OR PONCHON-SAVARIT.,/T2,'ASSUMPTIONS HOLD,MAY B
0024 6E CONSIDERED.,/T2,'PLOTTING ROUTINES ARE AVAILABLE FOR CONSTRUCTI
0025 7NG STANDARD.,/T2,'DIAGRAMS FOR EITHER DESIGN TECHNIQUE.,')
0026 WRITE(NTP,2001)
0027 2001 FORMAT(/T2,'REFER TO THE DISTILLATION PACKAGE MANUAL,SECTION 2.,./
0028 1T2,'FOR FURTHER INFORMATION.,')
0029 GO TO 999

```

SUBROUTINE TEACH

```

300 CONTINUE
  WRITE(NTP,3000)
3000 FORMAT(/T2,'THIS PROGRAM DESIGNS PACKED COLUMNS FOR DISTILLATION 0
  IF,/T2,'BINARY SOLUTIONS./T2,'THE PONCHON-SAVARIT METHOD IS USED
  2 TO DETERMINE THE OPERATING,/T2,'DIAGRAM./T2,'THE HEIGHT OF PAC
  3 KING REQUIRED IN EACH SECTION OF THE COLUMN./T2,'IS DETERMINED BY
  4 INTEGRATION,USING SIMPSON'S RULE./T2,'MASS TRANSFER COEFFICIENTS
  5 ARE COMPUTED USING SHULMANS./T2,'CORRELATIONS AND PACKING DATA,
  6,/T2,'DATA FOR CERAMIC RASCHIG RINGS (4 SIZES) AND BERL SADDLES'
  7,/T2,'(3 SIZES) ARE INCLUDED IN THE PACKAGE.')
```

```

  WRITE(NTP,3001)
3001 FORMAT(/T2,'REFER TO THE DISTILLATION PACKAGE MANUAL,SECTION 3./T2,
  1T2,'FOR FURTHER INFORMATION.')
```

```

  GO TO 999
400 CONTINUE
  WRITE(NTP,4000)
4000 FORMAT(/T2,'THIS IS A MULTICOMPONENT DISTILLATION SIMULATION PROGR
  1AM./T2,'THE CALCULATION PROCEDURE USED IS THAT FIRST PROPOSED BY
  2,/T2,'THIELE AND GEDDES,COMBINED WITH HOLLANDS THETA METHOD OF,
  3/T2,'CONVERGENCE./T2,'THE PROGRAM IS LIMITED TO SYSTEMS WITH ONE
  4 FEED AND,/T2,'NO SIDE STREAMS.')
```

```

  WRITE(NTP,4001)
4001 FORMAT(/T2,'REFER TO THE DISTILLATION PACKAGE MANUAL,SECTION 4./T2,
  1T2,'FOR FURTHER INFORMATION.')
```

```

999 CONTINUE
  RETURN
  END
```

SUBROUTINE TEACH



```

C      MAINLINE PROGRAM FOR RATING OF BINARY BATCH DISTILLATION SYSTEMS.
C      EITHER CONSTANT OR VARIABLE REFLUX OPERATIONS CAN BE CONSIDERED,
PARAMETER NPEQ=55
PARAMETER NPTC=55
PARAMETER NPBS=101
DIMENSION A(2),B(2),C(2),FUNC(NPBS),TITLES(2,5)
DIMENSION XDATA(NPBS),XEQUIL(NPEQ),YDATA(NPBS),YEQUIL(NPEQ)
COMMON ALFA(NPEQ),ALFCON,R
COMMON MALF,NDATA,NPI,NTC
COMMON X(NPTC),XALF(NPEQ),XD,XNTC,Y(NPTC),YALF(NPEQ)
1 FORMAT( )
2 FORMAT(A3)
3 FORMAT(5A6)
8001 FORMAT(2(2X,F6.4))
8002 FORMAT(I5)
DATA MALF,NCAN,NCODE,NINT,NSPEC/1,1,1,100,1/
DATA NCR,NG1,NTP/8,11,5/
DATA ATM,FACTP1,FACTP2/14,696,51,714752,7,50061683/
C      START OF PROGRAM *****
WRITE(NTP,5101)
5101 FORMAT(/T2,'ANSWER ALL QUESTIONS YES OR NO.,/T2,'ALL INPUT IN FRE
      IE FORMAT.,/T2,'DO YOU KNOW HOW TO USE THIS PROGRAM?')
      READ(NCR,2) ANAME
      IF(ANAME.EQ.'YES') GO TO 511
      CALL TEACH(NCODE)
511 CONTINUE
      WRITE(NTP,5102)
5102 FORMAT(/T2,'MODES OF OPERATION: (1) VARIABLE REFLUX',/T2,
      1'(2) CONSTANT REFLUX',/T2,'ENTER 1 OR 2 FOR REQUIRED PROCEDURE,')
      READ(NCR,1) NOP1
      WRITE(NTP,5103)
5103 FORMAT(/T2,'DO YOU WANT TO ADD A DATA ELEMENT?')
      READ(NCR,2) ANAME
      IF(ANAME.EQ.'NO') GO TO 512
      MCAN=2
      WRITE(NTP,5104)
5104 FORMAT(/T2,'ADD YOUR DATA ELEMENT,')

```

MAINLINE BATCH

```

512 CONTINUE
NINTC=NINT+1
IF(NINTC.GT.NPBS) GO TO 903
C  START OF DATA INPUT *****
5000 FORMAT(/T2,'ENTER NAMES OF COMPONENTS,,.1 PER LINE,MAX 30 CHARACTE
      1RS EACH,')
      READ(NCR,3) ((TTITLE(I,J),J=1,5),I=1,2)
      IF(MCAN,EQ,1) WRITE(NTP,5001)
5001 FORMAT(T4,'ACCEPTED PRESSURE UNITS: 1=MMHG,2=PSIG,3=PSIA,4=KPA,5=A
      1TM,./T2,'ENTER CONDENSER PRESSURE AND UNIT CODE, (EG. 760.0,1)')
      READ(NCR,1) PCONDA,NUNITP
      IF(NUNITP,EQ,1) PT=PCONDA
      IF(NUNITP,EQ,2) PT=(PCONDA+ATM)*FACTP1
      IF(NUNITP,EQ,3) PT=PCONDA*FACTP1
      IF(NUNITP,EQ,4) PT=PCONDA*FACTP2
      IF(NUNITP,EQ,5) PT=PCONDA*ATM*FACTP1
      IF(PT.GT.7600.) GO TO 902
      IF(MCAN,EQ,1) WRITE(NTP,5002)
5002 FORMAT(/T2,'WILL ANTOINE CONSTANTS OR EQUILIBRIUM DATA BE SUPPLIED
      1?./T2,'ENTER ANT OR EQU,')
      READ(NCR,2) ENAME
      IF(ENAME,NE,'ANT',.AND.ENAME,NE,'EQU') GO TO 902
      IF(ENAME,EQ,'EQU') GO TO 502
      IF(MCAN,EQ,1) WRITE(NTP,5003)
5003 FORMAT(/T2,'ENTER ANTOINE CONSTANTS,')
C  ANTOINE CONSTANTS *****
      NBDC=1
      READ(NCR,1) (A(I),B(I),C(I),I=1,2)
      IF(MCAN,EQ,1) WRITE(NTP,5004)
5004 FORMAT(/T2,'ENTER BUBBLE PT OF LIGHT COMPONENT,')
      READ(NCR,1) TBL
      DELXAL=0.02
      XAL=0.0

```

MAINLINE BATCH

```

0067 DO 51 J=1,1000
0068 XAL=XAL+DELXAL
0069 IF (ABS(XAL-1.0).LT.0.001) GO TO 501
0070 CALL BINBPT(A,B,C,NBDC,NDATA,PT,PAL,TALF,TBP,TBL,XALF,XAL,
0071 YAL=PAL*XAL/PT
0072 XALF(J)=XAL
0073 YALF(J)=YAL
0074 XEQUIL(J+1)=XAL
0075 YEQUIL(J+1)=YAL
0076 ALFA(J)=YALF(J)/XALF(J)*(1.-XALF(J))/(1.-YALF(J))
0077
0078 51 CONTINUE
0079 501 NDATA=J-1
0080 NDAT=J+1
0081 XEQUIL(1)=0.0
0082 XEQUIL(NDAT)=1.0
0083 YEQUIL(1)=0.0
0084 YEQUIL(NDAT)=1.0
0085 GO TO 504
0086
0087 502 CONTINUE
0088 IF (MCAN,EQ,1) WRITE(NTP,5005)
0089 FORMAT(/T2,'IS THE RELATIVE VOLATILITY A CONSTANT?')
0090 READ(NCR,2) ANAME
0091 IF (ANAME,NE,'YES',AND,ANAME,NE,'NO') GO TO 902
0092 IF (ANAME,EQ,'NO') GO TO 503
0093 IF (MCAN,EQ,1) WRITE(NTP,5006)
0094 FORMAT(/T2,'ENTER CONSTANT RELATIVE VOLATILITY VALUE,')
0095 C CONSTANT RELATIVE VOLATILITY *****
0096 READ(NCR,1) ALFCOII
0097 MALF=2
0098 NDAT=51
0099 DELX=0.02
0100 XA=0.0
0101 DO 52 I=1,NDAT
0102 XEQUIL(I)=XA
0103 YEQUIL(I)=XA*ALFCON/(1.+XA*(ALFCON-1.))
0104 XA=XA+DELX
0105 52 CONTINUE
0106 GO TO 504

```

MAINLINE BATCH

```

503 CONTINUE
IF(MCAN.EQ.1) WRITE(NTP,5007)
5007 FORMAT(/T2,'ENTER NO. OF SETS OF DATA TO BE SUPPLIED',/T2,'DO NOT
INCLUDE X=0.,Y=0. OR X=1.,Y=1.')
READ(NCR,1) NDATA
IF(MCAN.EQ.1) WRITE(NTP,5008)
5008 FORMAT(/T2,'ADD EQUILIBRIUM DATA ELEMENT,')
C EQUILIBRIUM DATA *****
NBOC=2
READ(NCR,1) (XALF(I),YALF(I),I=1,NDATA)
DO 53 J=1,NDATA
ALFA(J)=YALF(J)/XALF(J)*(1.-XALF(J))/(1.-YALF(J))
53 CONTINUE
XEQUIL(1)=0.0
YEQUIL(1)=0.0
NDAT1=NDATA+1
NDAT=NDATA+2
DO 54 I=2,NDAT1
XEQUIL(I)=XALF(I-1)
YEQUIL(I)=YALF(I-1)
54 CONTINUE
XEQUIL(NDAT)=1.0
YEQUIL(NDAT)=1.0
504 CONTINUE
IF(NDAT.GT.NPEQ) GO TO 903
END OF EQUILIBRIUM DATA CALCULATIONS.
C PROBLEM SPECIFICATIONS.
IF(MCAN.EQ.1) WRITE(NTP,5009)
5009 FORMAT(/T2,'ENTER NO OF THEORETICAL TRAYS IN COLUMN. (MAX=26)')
READ(NCR,1) NP
IF(MCAN.EQ.1) WRITE(NTP,5010)
5010 FORMAT(/T2,'ENTER COMPOSITION AND QUANTITY OF INITIAL CHARGE,')
READ(NCR,1) XF,FTOT
IF(XF.GT.1.0.OR,XF.LT.0.0) GO TO 902

```

MAINLINE BATCH

```

0135 IF(MCAN,EQ.1) WRITE(NTP,5111)
0136 5111 FORMAT(/T2,'WILL (1) BATCH CYCLE TIME,OR (2) VAPOUR BOIL-UP RATE',
0137 1/T2,'BE SPECIFIED. ENTER 1 OR 2.')
```

```

0138 READ(NCR,1) NTIM
0139 IF(NTIM,NE.1.AND.NTIM,NE.2) GO TO 902
0140 IF(MCAN,EQ.1) WRITE(NTP,5011)
0141 5011 FORMAT(/T2,'ENTER BATCH TIME OR BOIL-UP RATE,')
```

```

0142 READ(NCR,1) G
0143 IF(MCAN,EQ.1.AND.NOP1,EQ.1) WRITE(NTP,5012)
0144 5012 FORMAT(/T2,'ENTER DISTILLATE COMPOSITION,')
```

```

0145 IF(NOP1,EQ.1) READ(NCR,1) XD
0146 IF(NOP1,EQ.1) GO TO 505
0147 IF(MCAN,EQ.1) WRITE(NTP,5112)
0148 5112 FORMAT(/T2,'DO YOU WANT TO SPECIFY AN AVERAGE DISTILLATE COMPOSITI
0149 ION?')
```

```

0150 READ(NCR,2) DNAME
0151 IF(DNAME,NE.'YES'.AND.DNAME,NE.'NO') GO TO 902
0152 IF(DNAME,EQ.'NO') GO TO 505
0153 IF(MCAN,EQ.1) WRITE(NTP,5113)
0154 5113 FORMAT(T2,'ENTER AVERAGE VALUE,')
```

```

0155 READ(NCR,1) XSPEC
0156 IF(XSPEC,GT.1.0.OR.XSPEC,LT.XF) GO TO 902
0157 NSPEC=2
0158 505 CONTINUE
```

```

0159 IF(NSPEC,EQ.2) GO TO 506
0160 IF(MCAN,EQ.1) WRITE(NTP,5013)
0161 5013 FORMAT(/T2,'ENTER FINAL RESIDUE CONCENTRATION,')
```

```

0162 READ(NCR,1) XW
0163 IF(XW,GT.XF.OR.XW,LT.0.0) GO TO 902
0164 506 CONTINUE
```

```

0165 IF(MCAN,EQ.1.AND.NOP1,EQ.2) WRITE(NTP,5014)
0166 5014 FORMAT(/T2,'ENTER REFLUX RATIO,')
```

```

0167 IF(NOP1,EQ.2) READ(NCR,1) R
0168 IF(MCAN,EQ.1) WRITE(NTP,5015)
0169 5015 FORMAT(/T2,'IS A (1) TOTAL OR (2) PARTIAL CONDENSER USED?')
```

```

0170 1/T2,'ENTER 1 OR 2.')
```

```

0171 READ(NCR,1) NCOND
```

MAINLINE BATCH



```

C      'DO' LOOP 11 COMPUTES NINT RESIDUE COMPOSITION-REFLUX RATIO DATA SETS. 0198
DO 11 I=1,NINT 0199
XSTEP=XSTEP+H 0200
MN=1 0201
DELR=0.1
'GO TO' LOOP 102 SEARCHES FOR CORRECT R FOR EACH XSTEP VALUE.
102 CONTINUE 0202
CALL XDSTEP 0203
IF(ABS(XNTC-XSTEP).LE.0.00001) GO TO 120 0204
IF(DELR.LT.0.000001) GO TO 120 0205
NCALL=2 0206
CALL SEARCH(XNTC,XSTEP,R,DELR,MN,NCALL) 0207
GO TO 102 0208
120 CONTINUE 0209
IF(I.GT.1) GO TO 103 0210
WRITE FINAL CONDITIONS TO DATA ELEMENT FOR PLOTTING X-Y DIAGRAM.
YINCPT=XD/(R+1.) 0211
WRITE(NG1,8002) NTC 0212
WRITE(NG1,8001) (X(J),Y(J),J=1,NTC) 0213
WRITE(NG1,8001) YINCPT,XD 0214
103 CONTINUE 0215
XDATA(I)=XSTEP 0216
YDATA(I)=R 0217
11 CONTINUE 0218
NINT1=NINT+1 0219
'DO' LOOP 12 CALCULATES VALUES OF THE FUNCTION TO BE INTEGRATED.
DO 12 J=1,NINT1 0220
FUNC(J)=(1.+YDATA(J))/((XD-XDATA(J))*2) 0221
12 CONTINUE 0222
C      SIMPSONS RULE INTEGRATION OF THE FUNCTION.
CALL SIMPS(XF,XW,NINT1,FUNC,SUMINT) 0223
VAP=FTOT*(XD-XF)*SUMINT 0224
TIME=VAP/G 0225
W=FTOT*(XD-XF)/(XD-XW) 0226
DTOT=FTOT-W 0227
GO TO 400 0228

```

MAINLINE BATCH

```

200 CONTINUE
C
C CONSTANT REFLUX OPERATION *****
C
C INITIAL ESTIMATE OF DISTILLATE COMPOSITION XD.
XD=0.99
DELX=0.1
MN=1
NPI=NP+NCOND
GO TO LOOP 201 SEARCHES FOR CORRECT STARTING XD.
201 CONTINUE
CALL XDSTEP
IF (ABS(XNTC-XF).LE.0.00001) GO TO 210
NCALL=1
CALL SEARCH(XNTC,XF,XD,DELX,MN,NCALL)
IF (XD.GT.1.0) XD=0.999
IF (XD.GT.1.0) DELX=0.000001
GO TO 201
210 CONTINUE
WRITE INITIAL CONDITIONS TO DATA ELEMENT FOR PLOTTING X-Y DIAGRAM.
YINCPY=XD/(R+1.)
WRITE (NG1,8002) NDAT
WRITE (NG1,8001) (XEQUIL(I),YEQUIL(I),I=1,NDAT)
WRITE (NG1,8002) NTC
WRITE (NG1,8001) (X(I),Y(I),I=1,NTC)
WRITE (NG1,8001) YINCPY,XD
XDATA(NINT+1)=XF
YDATA(NINT+1)=XD
DELXW=0.01
MM=1
IF (NSPEC.EQ.2) XW=0.1

```

MAINLINE BATCH



C	GO TO, LOOP 204 CONSTRAINS SYSTEM TO SPECIFIED AVERAGE XD VALUE.	0254
C	204 CONTINUE	
C	CALCULATION OF DATA REQUIRED FOR INTEGRATING FUNCTION.	
C	FUNCTION: $D(\text{ALOG}(\text{FTOT}/W)) = D(XW)/(XD-XW)$	
C	LIMITS OF INTEGRATION: $XF \cdot XW \cdot NO$ , OF INTERVALS OF SIZE 'H': NINT	
	$H = (XF - XW) / \text{NINT}$	
	$XSTEP = XW - H$	
	$XD = 0.99$	0255
	$XD = 0.99$	0256
	$XD = 0.99$	0257
C	DO, LOOP 21 COMPUTES NINT RESIDUE-DISTILLATE COMPOSITION DATA SETS.	0258
	DO 21 I=1,NINT	0259
	$XSTEP = XSTEP + H$	0260
	$MN = 1$	0261
	$DELX = 0.1$	0262
C	GO TO, LOOP 202 SEARCHES FOR CORRECT XD FOR EACH XSTEP VALUE.	0263
	202 CONTINUE	0264
	CALL XDSTEP	0265
	$IF(\text{ABS}(XNTC - XSTEP) \cdot LE, 0.00001) \text{ GO TO } 220$	0266
	$NCALL = 1$	0267
	CALL SEARCH(XNTC, XSTEP, XD, DELX, MN, NCALL)	0268
	$IF(XD \cdot GT, 1.0) \text{ XD} = 0.999$	0269
	$IF(XD \cdot GT, 1.0) \text{ DELX} = 0.000001$	0270
	GO TO 202	0271
	220 CONTINUE	
	$IF(1 \cdot GT, 1) \text{ GO TO } 203$	
C	SAVE FINAL CONDITIONS FOR EACH PASS THROUGH LOOP 204.	0272
C	XEQUIL, YEQUIL ARE USED HERE AS STORAGE SPACE.	0273
	$YINCPT = XD / (R + 1.)$	0274
	DO 20 J=1,NTC	0275
	$XEQUIL(J) = X(J)$	0276
	$YEQUIL(J) = Y(J)$	0277
	20 CONTINUE	0278
	$XDHOLD = XD$	0279
	$NTCH = NTC$	0280
203	CONTINUE	0281
	$XDATA(I) = XSTEP$	0282
	$YDATA(I) = XD$	
21	CONTINUE	

MAINLINE BATCH

```

C      NINT1=NINT+1
      *DO* LOOP 22 CALCULATES VALUES OF THE FUNCTION TO BE INTEGRATED,
0283
      DO 22 J=1,NINT1
0284
      FUNC(J)=1./(YDATA(J)-XDATA(J))
0285
0286 22 CONTINUE
      C      SIMPSONS RULE INTEGRATION OF THE FUNCTION.
      CALL SIMPS(XF,XW,NINT1,FUNC,SUMINT)
      W=FTOT/(EXP(SUMINT))
0287
      DTOT=FTOT-W
0288
      VAP=DTOT*(R+1.)
0289
      TIME=VAP/G
0290
      XDAVE=(FTOT*XF-W*XW)/DTOT
0291
      IF(NSPEC.EQ.1) GO TO 230
0292
      IF(ABS(XDAVE-XSPEC).LT.0.0001) GO TO 230
0293
      NCALL=1
0294
      CALL SEARCH(XDAVE,XSPEC,XW,DELXW,MM,NCALL)
0295
      IF(XW.LT.0.001) GO TO 901
0296
      GO TO 204
0297
0298 230 CONTINUE
      WRITE(NG1,8002) NTCH
0299
      WRITE(NG1,8001) (XEQUIL(J),YEQUIL(J),J=1,NTCH)
0300
      WRITE(NG1,8001) YINCPT,XDHOLD
0301
      WRITE(NG1,8001) XDAVE,XD
0302
      WRITE(NG1,8001) XD
0303
0304 400 CONTINUE
      C      PRINT RESULTS *****
0305
      WRITE(NTP,4100)
0306
      4100 FORMAT(/T2,'BATCH DISTILLATION OF A BINARY SOLUTION,')
0307
      IF(NOP1.EQ.1) WRITE(NTP,4101)
0308
      IF(NOP1.EQ.2) WRITE(NTP,4102)
0309
      4101 FORMAT(T2,'VARIABLE REFLUX OPERATION,')
0310
      4102 FORMAT(T2,'CONSTANT REFLUX OPERATION,')

```

MAINLINE BATCH

```

C      COMPONENT TITLES.
      WRITE(NTP,4103) ((TITLES(I,J),J=1,5),I=1,2)
4103  FORMAT(/T2,'FEED COMPONENTS: (1) ',5A6,/T19,'(2) ',5A6)
C      VARIABLE REFLUX OPERATION: INITIAL AND FINAL REFLUX RATIOS.
      IF(NOP1.EQ.1) WRITE(NTP,4104) YDATA(NINT1),YDATA(1)
4104  FORMAT(/T2,'INITIAL REFLUX RATIO:',F7.3,/T2,'FINAL REFLUX RATIO:',
      IF9.3)
C      CONSTANT REFLUX OPERATION: REFLUX RATIO.
      IF(NOP1.EQ.2) WRITE(NTP,4105) R
4105  FORMAT(/T2,'REFLUX RATIO:',F7.3)
C      VAPOUR BOIL-UP RATE AND BATCH CYCLE TIME.
      IF(NTIM.EQ.1) WRITE(NTP,4106) TIME,G
      IF(NTIM.EQ.2) WRITE(NTP,4106) G,TIME
4106  FORMAT(/T2,'VAPOUR BOIL-UP RATE:',F6.2,2X,'KGMOLE/HR',
      1/T2,'BATCH CYCLE TIME:',F10.3,1X,'HRS')
      WRITE(NTP,4108) VAP
4108  FORMAT(/T2,'TOTAL VAPOUR BOILED UP DURING CYCLE:',F8.3,' KGMOLE')
C      RESIDUE DATA.
      WRITE(NTP,4109) W,XW
4109  FORMAT(/T2,'RESIDUE:',9X,F8.3,' KGMOLE',/T2,'RESIDUE COMPOSITION:',
      1,F6.4)
C      DISTILLATE PRODUCT DATA.
      WRITE(NTP,4110) DTOT
4110  FORMAT(/T2,'TOTAL DISTILLATE WITHDRAWN:',F8.3,' KGMOLE')
      IF(NOP1.EQ.1) WRITE(NTP,4111) XD
4111  FORMAT(/T2,'DISTILLATE COMPOSITION:',F6.4,' MOLE FRAC')
      IF(NOP1.EQ.2) WRITE(NTP,4112) YDATA(NINT1),YDATA(1),XDAVE
4112  FORMAT(/T2,'INITIAL DISTILLATE COMPOSITION:',F6.4,' MOLE FRAC',/T2
      1,'FINAL DISTILLATE COMPOSITION:',F8.4,/T2
      2,'AVERAGE DISTILLATE COMPOSITION:',F6.4)
      IF(NOP1.EQ.2) GO TO 401

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MAINLINE BATCH

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C      VARIABLE REFLUX OPERATION: RESIDUE COMPOSITION VS REFLUX RATIO,
      WRITE(NTP,4113)
      4113 FORMAT(/T2,'RESIDUE COMPOSITION VS REFLUX RATIO:','/T6,'XW',9X,'R')
      WRITE(NTP,4114) (XDATA(I),YDATA(I),I=NINT1,1,-10)
      4114 FORMAT(T3,F6,4,3X,F8,4)
      GO TO 999
      401 CONTINUE
      C      CONSTANT REFLUX OPERATION: RESIDUE COMP VS DISTILLATE COMP,
      WRITE(NTP,4115)
      4115 FORMAT(/T2,'RESIDUE COMPOSITION VS DISTILLATE COMPOSITION:','/T6,
      1'XW',9X,'XD')
      WRITE(NTP,4114) (XDATA(I),YDATA(I),I=NINT1,1,-10)
      GO TO 999
      C      ERROR MESSAGES *****
      901 CONTINUE
      WRITE(NTP,9001) XW
      9001 FORMAT(/T2,'RESIDUE COMPOSITION:','F8,4,/T2,'DISTILLATE COMPOSITION
      1 SET TOO LOW.')
      GO TO 999
      902 CONTINUE
      WRITE(NTP,9002)
      9002 FORMAT(/T2,'SPECIFICATION ERROR...RESTART PROGRAM.')
      GO TO 999
      903 CONTINUE
      WRITE(NTP,9003)
      9003 FORMAT(/T2,'DIMENSION SPECIFICATIONS EXCEEDED.')
      999 CONTINUE
      WRITE(NTP,9991)
      9991 FORMAT(/T2,'END OF PROGRAM.')
      END

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MAINLINE BATCH

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C
C
SUBROUTINE XDSTEP
ROUTINE STEPS OFF MCCABE-THIELE STAGES IN THE DESIGN OF BINARY
BATCH DISTILLATION SYSTEMS.
PARAMETER NPEQ=55
PARAMETER NPTC=55
COMMON ALFA(NPEQ),ALFCON,R
COMMON MALF,NDATA,NPI,NTC
COMMON X(NPTC),XALF(NPEQ),XD,XNTC,Y(NPTC),YALF(NPEQ)
X(1)=XD
Y(1)=XD
MSTOP=NPI*2
DO 11 I=2,100,2
Y(I)=Y(I-1)
YA=Y(I)
CALL XYALFA(ALFA,ALFCON,ALF,MALF,NDATA,YALF,YA)
X(I)=Y(I)/(ALF+Y(I)*(1.-ALF))
IF(I.EQ.MSTOP) GO TO 110
Y(I+1)=X(I)*R/(R+1.)+XD/(R+1.)
X(I+1)=X(I)
11 CONTINUE
110 CONTINUE
Y(I+1)=0.0
X(I+1)=X(I)
XNTC=X(I)
NTC=I+1
RETURN
END

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SUBROUTINE XDSTEP

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C      SUBROUTINE BINBPT(A,B,C,N,NDATA,PT,P1,TALF,TBP,T1,XALF,X1)
C      BUBBLE PT. ROUTINE FOR BINARY SYSTEMS.
C      DIMENSION A(2),B(2),C(2),TALF(1),TBP(2),XALF(1),XBPT(2)
C      GO TO (100,200),N
C      BUBBLE PT BY NEWTON'S METHOD,USING ANTOINE CONSTANTS.
100 CONTINUE
T=T1
XBPT(1)=X1
XBPT(2)=1.-X1
DO 10 I=1,1000
SUM1=PT
SUM2=0.0
DO 11 K=1,2
A1=A(K)-B(K)/(C(K)+T)
SUM1=SUM1-XBPT(K)*10.**A1
B1=2.303*B(K)/((C(K)+T)**2.)
SUM2=SUM2+XBPT(K)*B1*10.**A1
11 CONTINUE
DIFF=SUM1/SUM2
IF(ABS(DIFF).LT.0.00001) GO TO 101
T=T+DIFF
10 CONTINUE
101 CONTINUE
FACT=A(1)-B(1)/(C(1)+T)
T1=T
P1=10.**FACT
GO TO 999

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SUBROUTINE BINBPT

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C      BUBBLE PT SEARCH FROM SUPPLIED VAPOUR-LIQUID EQUILIBRIUM DATA.
200  CONTINUE
      IF(X1.GT.XALF(1)) GO TO 201
      XFACT=X1/XALF(1)
      T1=TBP(2)+(TALF(1)-TBP(2))*XFACT
      GO TO 999
201  CONTINUE
      IF(X1.GT.XALF(NDATA)) GO TO 220
      DO 20 I=1,NDATA
      DIFF=X1-XALF(I)
      IF(DIFF.LE.0.0) GO TO 210
20  CONTINUE
      GO TO 220
210  XFACT=(XALF(I)-X1)/(XALF(I)-XALF(I-1))
      T1=TALF(I)+(TALF(I-1)-TALF(I))*XFACT
      GO TO 999
220  CONTINUE
      XFACT=(1.-X1)/(1.-XALF(NDATA))
      T1=TBP(1)+(TALF(NDATA)-TBP(1))*XFACT
999  CONTINUE
      RETURN
      END

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SUBROUTINE BINBPT

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C      SUBROUTINE BINDPT(A,B,C,N,NDATA,PT,P1,TALF,TBP,T1,YALF,Y1)
C      DEW POINT ROUTINE FOR BINARY SYSTEMS.
C      DIMENSION A(2),B(2),C(2),TALF(1),TBP(2),YALF(1),YDPT(2)
C      GO TO (100,200),N
C      DEW PT BY NEWTON'S METHOD USING ANTOINE CONSTANTS.
C      100 CONTINUE
      T=T1
      YDPT(1)=Y1
      YDPT(2)=1.-Y1
      DO 10 I=1,1000
        SUM1=1.
        SUM2=0.0
        DO 11 K=1,2
          A1=A(K)-B(K)/(C(K)+T)
          A2=10.**A1
          SUM1=SUM1-YDPT(K)*PT/A2
          B1=2.303*B(K)/((C(K)+T)**2.)
          SUM2=SUM2+YDPT(K)*B1*PT/A2
11      CONTINUE
        DIFF=SUM1/SUM2
        IF (ABS(DIFF).LT.0.00001) GO TO 101
        T=T-DIFF
10      CONTINUE
101     CONTINUE
        FACT=A(1)-B(1)/(C(1)+T)
        T1=T
        P1=10.**FACT
        GO TO 999

```

SUBROUTINE BINDPT



C DEW PT SEARCH FROM SUPPLIED VAPOUR-LIQUID EQUILIBRIUM DATA.

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200 CONTINUE
    IF(Y1.GT.YALF(1)) GO TO 201
    YFACT=Y1/YALF(1)
    T1=TBP(2)+(TALF(1)-TBP(2))*YFACT
    GO TO 999
201 CONTINUE
    IF(Y1.GT.YALF(NDATA)) GO TO 220
    DO 20 I=1,NDATA
        DIFF=Y1-YALF(I)
        IF(DIFF.LE.0.0) GO TO 210
    20 CONTINUE
    GO TO 220
210 YFACT=(YALF(1)-Y1)/(YALF(1)-YALF(I-1))
    T1=TALF(1)+(TALF(I-1)-TALF(1))*YFACT
    GO TO 999
220 CONTINUE
    YFACT=(1.-Y1)/(1.-YALF(NDATA))
    T1=TBP(1)+(TALF(NDATA)-TBP(1))*YFACT
999 CONTINUE
    RETURN
    END
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SUBROUTINE BINDPT

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C	SUBROUTINE XYALFA(ALFA,ALFCON,ALPHA,MALF,NDATA,XYALF,XYI)	0001
C	ROUTINE SEARCHES FOR THE RELATIVE VOLATILITY CORRESPONDING	
	TO THE SUPPLIED LIQUID OR VAPOUR COMPOSITION VALUE,	
	DIMENSION ALFA(1),XYALF(1)	0002
	IF(MALF.GT.1) GO TO 200	0003
	IF(XYI.GT.XYALF(1)) GO TO 100	0004
	XYFACT=(XYALF(1)-XYI)/XYALF(1)	0005
	ALPHA=ALFA(1)-(ALFA(2)-ALFA(1))*XYFACT	0006
	GO TO 999	0007
	100 CONTINUE	0008
	IF(XYI.GT.XYALF(NDATA)) GO TO 120	0009
	DO 10 I=1,NDATA	0010
	DIFF=XYI-XYALF(I)	0011
	IF(DIFF.LE.0.01) GO TO 110	0012
	10 CONTINUE	0013
	GO TO 120	0014
	110 XYFACT=(XYALF(I)-XYI)/(XYALF(I)-XYALF(I-1))	0015
	ALPHA=ALFA(I)+(ALFA(I-1)-ALFA(I))*XYFACT	0016
	GO TO 999	0017
	120 CONTINUE	0018
	XYFACT=(1.0-XYI)/(1.0-XYALF(NDATA))	0019
	ALPHA=1.0+(ALFA(NDATA)-1.0)*XYFACT	0020
	GO TO 999	0021
	200 ALPHA=ALFCON	0022
	GO TO 999	0023
	999 CONTINUE	0024
	RETURN	0025
	END	0026

SUBROUTINE XYALFA

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C
C
C
SUBROUTINE SIMPS(ALIM,BLIM,N,FUNC,SUM)
ROUTINE INTEGRATES A FUNCTION OVER M INTERVALS BETWEEN LIMITS
ALIM AND BLIM BY SIMPSON'S RULE. VALUES OF THE FUNCTION ARE
STORED IN ARRAY 'FUNC'. INTEGRAL RETURNED AS 'SUM'.
DIMENSION FUNC(1)
M=N-1
EVENS=0.0
ODDS=0.0
H=(ALIM-BLIM)/M
DO 10 J=2,M
  IF(MOD(J,2)) 102,101,102
  101 EVENS=EVENS+FUNC(J)
  GO TO 10
  102 ODDS=ODDS+FUNC(J)
  10 CONTINUE
SUM=H*(FUNC(1)+FUNC(N)+4.*EVENS+2.*ODDS)/3.
RETURN
END

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SUBROUTINE SIMPS

C	SUBROUTINE SEARCH(FA1,FA2,FACTOR,DEL,MM,N)	0001
C	ROUTINE RETURNS A NEW ESTIMATE OF 'C' TO BE USED BY THE MAINLINE	
C	TO CONVERGE THE EQUALITY A=B. 'DEL' IS INCREMENT IN C THAT DECREASES	
C	AS CONVERGENCE IS APPROACHED.	
C	'MM' CAUSES DEL TO BE DECREASED AS A SWITCHES FROM >B TO <B.	
C	'N' DETERMINES THE EFFECT OF C ON CONVERGENCE.	
	IF(FA1.LT.FA2) GO TO 120	0002
	IF(MM.EQ.1) GO TO 111	0003
	DEL=DEL/2.	0004
	MM=1	0005
111	IF(N.EQ.1) FACTOR=FACTOR-DEL	0006
	IF(N.EQ.2) FACTOR=FACTOR+DEL	0007
	GO TO 999	0008
120	IF(MM.EQ.2) GO TO 121	0009
	DEL=DEL/2.	0010
	MM=2	0011
121	IF(N.EQ.1) FACTOR=FACTOR+DEL	0012
	IF(N.EQ.2) FACTOR=FACTOR-DEL	0013
999	CONTINUE	0014
	RETURN	0015
	END	0016

SUBROUTINE SEARCH

C

COMMON BLOCK 'BINCOM' FOR BINARY DISTILLATION PROGRAMS.

```

PARAMETER NPEQ=55
PARAMETER NPPS=150
PARAMETER NPTC=55
PARAMETER NPEN=25
DIMENSION ALFA(NPEQ),HLIQ(NPEN),HVAP(NPEN),LIQR(NPPS),LIQS(NPPS)
DIMENSION LRWT(NPPS),LSWT(NPPS),TALF(NPEQ)
DIMENSION TRL(NPPS),TRV(NPPS),TSL(NPPS),TSV(NPPS)
DIMENSION VAPR(NPPS),VAPS(NPPS),VRWT(NPPS),VSWT(NPPS)
DIMENSION X(NPTC),XALF(NPEQ),XEQUIL(NPEQ),XH(NPEN)
DIMENSION XROP(NPPS),XSOP(NPPS),YROP(NPPS),YSOP(NPPS)
DIMENSION Y(NPTC),YALF(NPEQ),YEQUIL(NPEQ),YH(NPEN)
COMMON A(2),ACP(2),AL(11),ALFA,ALFCON,AV(11)
COMMON B(2),BCP(2),BTOT,C(2),CCP(2),CFCT(12),CPCONV,CPMW(2)
COMMON DCP(2),DELB,DELD,DL,DTOT,DV
COMMON EFF,FLOW(10),FTOT,FTOTS(5)
COMMON HB,HD,HEVABS,HF,HLIQ,HSCONV,HSMW(2),HSOLN(30)
COMMON HVAP,LH(2)
COMMON LHCONV,LHMW(2),LIQR,LIQS,LRWT,LSWT
COMMON MALF,MW(2),NBDC,NCOND,NDATA,NENTH,NF,NHS,NINT
COMMON NN,NOP,NOS,NPPSC,NPTCC,NS,NTRAY(12)
COMMON NCR,NGI,NTP
COMMON PT,Q,QB,QC,QF(5),QS(5),QFS(10),QX(12),QY(12),R,RM
COMMON SLPE(12),STOTS(5),TALF,TBASE,TBP(2),TC(2)
COMMON TCONV,TF,TG,TITLES(2,5),TRAYNO,TRL,TRV
COMMON TSL,TSV,VAPR,VAPS,VRWT,VSWT
COMMON X,XALF,XB,XC,XCR,XD,XEQUIL,XH,XHS(30)
COMMON XROP,XSOP,YROP,YSOP
COMMON Y,YALF,YB,YC,YCR,YD,YEQUIL,YH
COMMON Z,ZF(5),ZF1(5),ZS(5),ZS1(5),ZFS(10)

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COMMON 'BINCOM'

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C MAINLINE PROGRAM FOR SOLUTION OF BINARY CONTINUOUS-DISTILLATION
C PROBLEMS BY THE MCCABE-THIELE OR PONCHON-SAVARIT GRAPHICAL METHOD.
C THE PROGRAM SETS UP THE EQUILIBRIUM CURVE AND OPERATING LINES
C FOR THE GIVEN SYSTEM, AND STEPS OFF THE NUMBER OF STAGES REQUIRED
C FOR THE SPECIFIED SEPARATION.
C AN OUT-PUT FILE OF DATA IS GENERATED THAT CAN BE USED TO PRODUCE
C A MCCABE-THIELE OR PONCHON-SAVARIT DIAGRAM FOR THE SYSTEM, ON THE
C •CALCOMP• PLOTTER.
C *****
C INSERT COMMON BLOCK 'BINCOM' HERE.
C *****
C REAL LH,LHCONV,LHMW,LIQR,LIQS,LRWT,LSWT,MW
      1 FORMAT( )
      2 FORMAT(A3)
      3 FORMAT(SA6)
      DATA HALF,MCAN,NCODE,NINT/1,1,2,50/
      DATA NCR,NG1,NTP/8,11,5/
      NPPSC=NPPS-2
      NPTCC=NPTC
C START OF PROGRAM *****
      WRITE(NTP,5000)
5000 FORMAT(/T2,'ANSWER ALL QUESTIONS YES OR NO','/T2,'ALL INPUT IN FRE
      IE FORMAT.','/T2,'DO YOU KNOW HOW TO USE THIS PROGRAM?')
      READ(NCR,2) ANAME
      IF(ANAME.NE.'YES'.AND.ANAME.NE.'NO') GO TO 901
      IF(ANAME.EQ.'YES') GO TO 500
      CALL TEACH(NCODE)
      GO TO 999
500 CONTINUE
      WRITE(NTP,5010)
5010 FORMAT(/T2,'DESIGN TECHNIQUES: (1) MCCABE-THIELE','/T21,
      1','(2) PONCHON-SAVARIT','/T2,'ENTER 1 OR 2 FOR REQUIRED PROCEDURE,')
      READ(NCR,1) NOP1
      IF(NOP1.EQ.1) NOP=1
      IF(NOP1.EQ.2) NOP=3

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MAINLINE BINARY

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5011 WRITE(NTP,5011)
5011 FORMAT(/T2,'DO YOU WANT TO ADD A DATA ELEMENT?')
5011 READ(NCR,2) CNAME
5011 IF(CNAME.NE.'YES'.AND.CNAME.NE.'NO') GO TO 901
5011 IF(CNAME.EQ.'NO') GO TO 501
5011 MCAN=2
5011 WRITE(NTP,5012)
5012 FORMAT(/T2,'ADD YOUR DATA ELEMENT.')
5011 CONTINUE
5011 *READBD* READS DATA ELEMENT.
5011 CALL READBD(MCAN,NDAT)
5011 IF(MALF.GT.999) GO TO 999
5011 START OF COMPUTATION *****
5011 IF(NOP.GT.1) GO TO 301
5011 MCCABE-THIELE DESIGN TECHNIQUE.
5011 IF(NF.LT.2.AND.NS.EQ.0) GO TO 101
5011 COMPLEX COLUMNS: IE. MORE THAN ONE FEED,AND/OR SIDE STREAMS.
5011 *MULTFS* COMPUTES OPERATING CONDITIONS FOR COMPLEX COLUMNS.
5011 CALL MULTFS(NN1)
5011 IF(MALF.GT.999) GO TO 999
5011 GO TO 302
5011 CONTINUE
5011 SIMPLE COLUMNS: IE. ONLY ONE FEED,NO SIDE STREAMS.
5011 *SIMPLE* COMPUTES OPERATING CONDITIONS FOR SIMPLE COLUMNS.
5011 CALL SIMPLE
5011 IF(MALF.GT.999) GO TO 999
5011 NN1=3
5011 GO TO 302

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MAINLINE BINARY

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301 CONTINUE
C
C PONCHON-SAVARIT DESIGN TECHNIQUE.
C
C *PONSAB* CONSTRUCTS THE PONCHON-SAVARIT OPERATING DIAGRAM.
CALL PONSAB(NPTS,MPTS,NFIN,MFIN)
IF(MALF.GT.999) GO TO 999
NNI=3
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0049
302 CONTINUE
C CONSTRUCTION OF STEPPED STAGES ON X-Y DIAGRAM *****
C *CHEC* ENSURES THAT NO PINCH POINTS WILL BE ENCOUNTERED.
CALL CHEC(ALFA,ALFCON,FLOW,MALF,NDATA,NN,NTP,XALF,QX,QY,ZFS)
IF(MALF.GT.999) GO TO 999
0050
0051
C *STEPP* STEPS OFF STAGES ON X-Y DIAGRAM AND COMPUTES:
CALL STEPD(NII,NPTS,MPTS)
0052
C PRINT RESULTS *****
C *BINRES* PRINTS RESULTS OF RUN.
CALL BINRES(NDAT,NII,NNI,NPTS,MPTS)
0053
C
C WRITE DATA TO FILE FOR PLOTTING X-Y AND ENTHALPY-CONCENTRATION
C DIAGRAMS.
C
8001 FORMAT(2(2X,F6.4))
8002 FORMAT(15)
8003 FORMAT(F7.4,F10.2)
8004 FORMAT(2E12.6)
8005 FORMAT(F6.4,E12.6)
IF(NOP.EQ.1) GO TO 800
C ENTHALPY-CONCENTRATION DIAGRAM DATA.
WRITE(NG1,8004) DELD,DELB
WRITE(NG1,8005) XD,HD
WRITE(NG1,8005) XB,HB
WRITE(NG1,8005) Z,HF
WRITE(NG1,8002) NENTH
WRITE(NG1,8005) (XH(K),HLIQ(K),K=1,NENTH)
WRITE(NG1,8005) (YH(K),HVAP(K),K=1,NENTH)
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MAINLINE BINARY



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800 CONTINUE
C EQUILIBRIUM DATA.
WRITE(NG1,8002) NDAT
WRITE(NG1,8001) (XEQUIL(K),YEQUIL(K),K=1,NDAT)
IF(NOP.EQ.3) GO TO 801
C POINTS OF INTERSECTION OF MCCABE-THIELE OPERATING LINES.
WRITE(NG1,8002) NN1
WRITE(NG1,8001) (QX(K),QY(K),K=1,NN1)
GO TO 802
801 CONTINUE
C OPERATING LINES FOR PONCHON-SAVARIT DESIGNS.
WRITE(NG1,8002) NPTS
WRITE(NG1,8001) (XROP(N),YROP(N),N=1,NPTS)
WRITE(NG1,8002) MPTS
WRITE(NG1,8001) (XSOP(M),YSOP(M),M=1,MPTS)
802 CONTINUE
C POINTS DEFINING EQUILIBRIUM STAGE STEPS ON X,Y DIAGRAM.
WRITE(NG1,8002) N11
WRITE(NG1,8001) (X(K),Y(K),K=1,N11)
WRITE(NG1,8002) NN
C FEED AND SIDE STREAM COMPOSITIONS AND FLOWRATES (COMPLEX COLUMN).
IF(NN.GT.2) WRITE(NG1,8003) (ZFS(K),FLOW(K),K=2,NN)
C FEED STREAM COMPOSITION AND FLOWRATE (SIMPLE COLUMN).
IF(NN.LE.2) WRITE(NG1,8003) Z,FTOT
C POINT OF INTERSECTION OF PONCHON-SAVARIT OPERATING LINES.
IF(NOP.EQ.3) WRITE(NG1,8001) XCR,YCR
GO TO 999
C ERROR MESSAGES *****
901 CONTINUE
WRITE(NTP,9001)
9001 FORMAT(/T2,'ILLEGAL SPECIFICATION. ERROR TERMINATION..')
999 CONTINUE
WRITE(NTP,9999)
9999 FORMAT(/T2,'END OF PROGRAM..')
STOP
END

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MAINLINE BINARY

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0001 SUBROUTINE READBD(MCAN,NDAT)
ROUTINE FOR READING DATA FROM EITHER FILE OR TERMINAL, FOR
C BINARY CONTINUOUS-DISTILLATION PROBLEMS.
C *****
C INSERT COMMON BLOCK 'BINCOM' HERE.
C *****
REAL LH,LHCONV,LHFW,LIQR,LIQS,LRWT,LSWT,MW
1 FORMAT( )
2 FORMAT(A3)
3 FORMAT(5A6)
DATA ATM,FACTP1,FACTP2/14.696,51.714752,7.50061683/
IF(MCAN.EQ.1) WRITE(NTP,5000)
5000 FORMAT(/T2,'ENTER NAMES OF COMPONENTS...1 PER LINE,MAX 30 CHARACTE
1RS EACH,')
READ(NCR,3) ((TITLES(I,J),J=1,5),I=1,2)
IF(MCAN.EQ.1) WRITE(NTP,5001)
5001 FORMAT(T4,'ACCEPTED PRESSURE UNITS: 1=MMHG,2=PSIA,4=KPA,5=A
ITM,./T2,'ENTER CONDENSER PRESSURE AND UNIT CODE, (EG. 760.0,1),')
READ(NCR,1) PCONDA,NUNITP
IF(NUNITP.EQ.1) PT=PCONDA
IF(NUNITP.EQ.2) PT=(PCONDA+ATM)*FACTP1
IF(NUNITP.EQ.3) PT=PCONDA*FACTP1
IF(NUNITP.EQ.4) PT=PCONDA*FACTP2
IF(NUNITP.EQ.5) PT=PCONDA*ATM*FACTP1
IF(PT.GT.7600.) GO TO 901
IF(MCAN.EQ.1) WRITE(NTP,5002)
5002 FORMAT(/T2,'WILL ANTOINE CONSTANTS OR EQUILIBRIUM DATA BE SUPPLIED
1?./T2,'ENTER ANT OR EQU,')
READ(NCR,2) ENAME
IF(ENAME.NE.'ANT'.AND.ENAME.NE.'EQU') GO TO 901
IF(ENAME.EQ.'EQU') GO TO 502
IF(MCAN.EQ.1) WRITE(NTP,5003)
5003 FORMAT(/T2,'ENTER ANTOINE CONSTANTS,')
C ANTOINE CONSTANTS *****
NBDC=1
READ(NCR,1) (A(I),B(I),C(I),I=1,2)
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```

SUBROUTINE READBD

```

5004 IF(MCAN.EQ.1) WRITE(NTP,5004)
      FORMAT(/T2,'ENTER BUBBLE POINTS OF PURE COMPONENTS.')
```

0031

```

      READ(NCR,1) (TBP(I),I=1,2)
      DELXAL=0.02
      XAL=0.0
      TBL=(TBP(1)+TBP(2))/2.
      DO 51 J=1,1000
      XAL=XAL+DELXAL
      IF(ABS(XAL-1.0).LT.0.001) GO TO 501
      CALL BINBPT(A,B,C,NBDC,NDATA,PT,PAL,TALF,TBP,TBL,XALF,XAL)
      YAL=PAL*XAL/PT
      XALF(J)=XAL
      YALF(J)=YAL
      TALF(J)=TBL
      XEQUIL(J+1)=XAL
      YEQUIL(J+1)=YAL
      ALFA(J)=YALF(J)/XALF(J)*(1.-XALF(J))/(1.-YALF(J))
      51 CONTINUE
      501 NDATA=J-1
      NDAT=J+1
      XEQUIL(1)=0.0
      XEQUIL(NDAT)=1.0
      YEQUIL(1)=0.0
      YEQUIL(NDAT)=1.0
      GO TO 504
      502 CONTINUE
      IF(NOP.GT.1) GO TO 503
      IF(MCAN.EQ.1) WRITE(NTP,5005)
      5005 FORMAT(/T2,'IS THE RELATIVE VOLATILITY A CONSTANT?')
```

0032

```

      READ(NCR,2) ANANE
      IF(ANANE.NE.'YES'.AND.ANANE.NE.'NO') GO TO 901
      IF(ANANE.EQ.'NO') GO TO 503
      IF(MCAN.EQ.1) WRITE(NTP,5006)
      5006 FORMAT(/T2,'ENTER CONSTANT RELATIVE VOLATILITY VALUE.')
```

0033

```

      C CONSTANT RELATIVE VOLATILITY *****
      READ(NCR,1) ALFCON
      MALF=2
```

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0066

SUBROUTINE READBD

```

C      SIDE STREAMS.
      IF(MCAN.EQ.1) WRITE(NTP,2010)
2010 FORMAT(/T2,'ENTER SIDE STREAM COMPOSITIONS.')
```

0132  
0133  
0134  
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0141  
0142

```

      READ(NCR,1) (ZS(I),I=1,NS)
      IF(MCAN.EQ.1) WRITE(NTP,2011)
2011 FORMAT(/T2,'ENTER SIDE STREAM Q VALUES.')
```

0143  
0144  
0145  
0146  
0147

```

      READ(NCR,1) (QS(I),I=1,NS)
      IF(MCAN.EQ.1) WRITE(NTP,2012)
2012 FORMAT(/T2,'ENTER SIDE STREAM WITHDRAWAL RATES.')
```

0148  
0149  
0150

```

      READ(NCR,1) (STOTS(I),I=1,NS)
      GO TO 212
211 CONTINUE
C      SIMPLE COLUMNS.
      NF=1
      NS=0
      IF(MCAN.EQ.1) WRITE(NTP,2013)
2013 FORMAT(/T2,'ENTER FEED COMPOSITION AND FEED RATE.')
```

0151  
0152  
0153  
0154  
0155

```

      READ(NCR,1) Z.FTOT
C      FEED Q VALUE MUST BE SPECIFIED FOR MCCABE-THIELE DESIGNS.
      IF(MCAN.EQ.1.AND.NOP.EQ.1) WRITE(NTP,2023)
2023 FORMAT(T2,'ENTER FEED Q VALUE.')
```

0156  
0157  
0158  
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0161

```

      IF(NOP.EQ.1) READ(NCR,1) Q
C      TEMPERATURE OF FEED REQUIRED FOR PONCHON-SAVARIT DESIGNS ONLY.
      IF(MCAN.EQ.1.AND.NOP.GT.1) WRITE(NTP,2014)
2014 FORMAT(T2,'ENTER TEMPERATURE OF FEED ... (DEG C)')
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SUBROUTINE READBD

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C      IF(NOP.EQ.2) GO TO 214
C      EFFICIENCY FACTOR *****
C      IF(MCAN.EQ.1) WRITE(NTP,2016)
2016  FORMAT(/T2,'ENTER MURPHREE STAGE EFFICIENCY FACTOR.',/T2,
1' (I=1.0 IF PERFECT TRAYS ARE ASSUMED).')
      READ(NCR,1) EFF
214  CONTINUE
      IF(NOP.EQ.1) GO TO 999
C      ENTHALPY DATA *****
C      REQUIRED FOR PONCHON-SAVARIT DESIGNS.
      WRITE(NTP,3010)
3010  FORMAT(/T2,'ADD ENTHALPY DATA ELEMENT. (SEE MANUAL SECTION 2).')
C      ENTHALPY FUNCTION COEFFICIENTS AND ASSOCIATED POWERS OF TEN.
      READ(NCR,1) (ACP(I),BCP(I),CCP(I),DCP(I),I=1,2)
      READ(NCR,1) NBCP,NCCP,NDCP
      DO 31 I=1,2
        BCP(I)=BCP(I)*10.**NBCP
        CCP(I)=CCP(I)*10.**NCCP
        DCP(I)=DCP(I)*10.**NDCP
31  CONTINUE
C      CONVERSION FACTORS REQUIRED FOR A CONSISTENT SET OF UNITS.
      READ(NCR,1) CPCONV,LHCONV,HSCONV,HEVABS
C      BASE TEMPERATURE OF ENTHALPY CALCULATIONS AND ABSOLUTE SCALE
C      CONVERSION TEMPERATURE.
      READ(NCR,1) TBASE,TCONV
C      LATENT HEAT, BOILING POINT AND CRITICAL-TEMPERATURE DATA.
      READ(NCR,1) (LH(I),TBP(I),TC(I),I=1,2)
      TG=(TBP(1)+TBP(2))/2.
C      MOLECULAR WEIGHTS: CP-, LH-, HSMW REQUIRED IF RESPECTIVE DATA IS
C      SUPPLIED PER WEIGHT RATHER THAN PER MOLE. (SEE MANUAL)
      READ(NCR,1) (MW(I),CPHW(I),LHMW(I),HSMW(I),I=1,2)
C      HEAT OF MIXING DATA.
      READ(NCR,1) NHS
      IF(NHS.GT.0) READ(NCR,1) (XHS(I),HSOLN(I),I=1,NHS)
      CALL FLASHB
      GO TO 999

```

SUBROUTINE READBD

```

C 901 ERROR MESSAGES *****
    CONTINUE
    WRITE(NTP,9001)
    9001 FORMAT(/T2,'ERROR IN DATA INPUT...ERROR TERMINATION.')
```

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SUBROUTINE READBD

```

    MALF=1000
    GO TO 999
    902 CONTINUE
    WRITE(NTP,9002)
    9002 FORMAT(/T2,'DIMENSION SPECIFICATION EXCEEDED.')
```

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```

    MALF=1000
    999 CONTINUE
    RETURN
    END
```

```

0001 SUBROUTINE CONDENSE(NCAN)
      CALCULATION OF THE COMPOSITIONS OF STREAMS LEAVING THE CONDENSER.
      *****
      C INSERT COMMON BLOCK 'BINCOM' HERE.
      C *****
      C REAL LH,LHCONV,LHHW,LIQR,LIQS,LRWT,LSWT,MW
      1 FORMAT( )
      2 FORMAT(A3)
      GO TO (100,200),NCOND
100 CONTINUE
      C TOTAL CONDENSER
      XC=XD
      YC=XD
      GO TO 999
200 CONTINUE
      C PARTIAL CONDENSER
      IF(NOP.GT.1) GO TO 210
      IF(NCAN.EQ.1) WRITE(NTP,1010)
1010 FORMAT(/T2,'ENTER FRACTION OF TOTAL DISTILLATE TO BE WITHDRAWN AS'
      1./T2,'VAPOUR AND LIQUID RESPECTIVELY. IF ALL VAPOUR ENTER 1.,0.')
      READ(NCR,1) DV,PL
      IF(DL.GT.0.00001) GO TO 211
210 CONTINUE
      C ALL VAPOUR DISTILLATE
      YC=XD
      CALL XYALFA(ALFA,ALFCON,ALPHA,MALF,NDATA,YALF,YC)
      XC=YC/(ALPHA-YC*(ALPHA-1.))
      GO TO 999
211 CONTINUE
      C MIXED VAPOUR AND LIQUID DISTILLATE
      IF(NCAN.EQ.1) WRITE(NTP,2011)
2011 FORMAT(/T2,'IS XD AN OVERALL DISTILLATE COMPOSITION?')
      READ(NCR,2) XONANE
      IF(XONANE.EQ.'YES') GO TO 230
      IF(NCAN.EQ.1) WRITE(NTP,2012)
2012 FORMAT(/T2,'IS XD THE COMPOSITION OF THE VAPOUR OR LIQUID FRACTION
      1./T2,'ENTER VAP OR LIQ.')
      READ(NCR,2) XLNAME
      IF(XLNAME.EQ.'LIQ') GO TO 220

```

SUBROUTINE CONDENSE

```

C      XD GIVEN AS COMPOSITION OF VAPOUR FRACTION
      YCC=XD
      CALL XYALFA(ALFA,ALFCON,ALF,MALF,NDATA,XALF,YALF,YCC)
      XDC=YCC/(ALF-YCC*(ALF-1.))
      CDLDV=DV/DL
      XD=(CDLDV*YCC+XDC)/(1.+CDLDV)
      XC=XDC
      YC=YCC
      GO TO 999
220 CONTINUE
C      XD GIVEN AS COMPOSITION OF LIQUID FRACTION
      XDC=XD
      CALL XYALFA(ALFA,ALFCON,ALF,MALF,NDATA,XALF,XDC)
      YCC=XDC*ALF/(1.+XDC*(ALF-1.))
      CDLDV=DV/DL
      XD=(CDLDV*YCC+XDC)/(1.+CDLDV)
      XC=XDC
      YC=YCC
      GO TO 999
230 CONTINUE
C      XD GIVEN AS OVERALL COMPOSITION VALUE
C      MIXED VAPOUR AND LIQUID DISTILLATE
      CDLDV=DV/DL
      DELX=0.991
      MM=2
      XI=XD-DELX
231 CONTINUE
      CALL XYALFA(ALFA,ALFCON,ALF,MALF,NDATA,XALF,XI)
      YC=XI*ALF/(1.+XI*(ALF-1.))
      FACT=(CDLDV*YC+XI)/(1.+CDLDV)
      IF(ABS(FACT-XD).LT.0.00001) GO TO 232
      NCALL=1
      CALL SEARCH(FACT,XD,XI,DELX,MM,NCALL)
      GO TO 231
232 CONTINUE
      XC=XI
999 CONTINUE
      RETURN
      END

```

SUBROUTINE CONDEN





```

C      SUBROUTINE EQUIL(NDATA,TALF,TBP,TF,XALF,YALF,XI,YI)
C      *EQUIL* SEARCHES FOR EQUILIBRIUM DATA POINTS CORRESPONDING TO
C      TEMPERATURE TF, USING SUPPLIED OR CALCULATED X-Y-T DATA. IF TF NE TALF
C      LINEAR INTERPOLATION IS USED TO FIND VALUES FOR XALF AND YALF.
C      DIMENSION TALF(1),TBP(1),XALF(1),YALF(1)
C      IF(TF.LT.TALF(1)) GO TO 101
C      TFACT=(TBP(2)-TF)/(TBP(2)-TALF(1))
C      XI=0.0+TFACT*XALF(1)
C      YI=TFACT*YALF(1)
C      GO TO 999
101 CONTINUE
C      IF(TF.LT.TALF(NDATA)) GO TO 102
C      DO 10 I=1,NDATA
C      DIFF=TALF(I)-TF
C      IF(DIFF.LE.0.0) GO TO 110
110 CONTINUE
C      GO TO 102
110 CONTINUE
C      TFACT=(TALF(I-1)-TF)/(TALF(I-1)-TALF(I))
C      XI=XALF(I-1)+TFACT*(XALF(I)-XALF(I-1))
C      YI=YALF(I-1)+TFACT*(YALF(I)-YALF(I-1))
C      GO TO 999
102 CONTINUE
C      TFACT=(TALF(NDATA)-TF)/(TALF(NDATA)-TBP(1))
C      XI=XALF(NDATA)+TFACT*(1.0-XALF(NDATA))
C      YI=YALF(NDATA)+TFACT*(1.0-YALF(NDATA))
999 CONTINUE
C      RETURN
C      END

```

SUBROUTINE EQUIL

0001

```

SUBROUTINE MULTFS(NNI)
ROUTINE FOR HANDLING BINARY PROBLEMS WITH MORE THAN ONE FEED
AND/OR SIDE STREAM.
THE ROUTINE: (1) SORTS THE FEED AND SIDE STREAMS INTO AN ORDER OF
INCREASING CONCENTRATION.
(2) COMPUTES BOTTOMS FLOWRATE AND COMPOSITION FROM AN
OVERALL MASS BALANCE.
(3) COMPUTES LIQUID AND VAPOUR FLOWRATES IN EACH
SECTION OF COLUMN.

```

```

*****
INSERT COMMON BLOCK 'BINCOM' HERE.
*****
REAL LH,LHCONV,LHNW,LIQR,LIQS,LRWT,LSWT,MW
OVERALL MASS BALANCE.

```

```

SUMF=0.
SUMFZ=0.
SUMFQ=0.
SUMS=0.
SUMSZ=0.
SUMSQ=0.

```

```

'DO' LOOP 11 COMPUTES OVERALL FEED FLOW, COMPOSITION AND Q VALUE.
DO 11 I=1,MF
SUMF=SUMF+FTOTS(I)
SUMFZ=SUMFZ+FTOTS(I)*ZF(I)
SUMFQ=SUMFQ+FTOTS(I)*QF(I)

```

11 CONTINUE

```

IF(NS.EQ.0) GO TO 101
'DO' LOOP 12 COMPUTES OVERALL SIDE FLOW, COMPOSITION AND Q VALUE.

```

```

DO 12 I=1,NS
SUMS=SUMS+STOTS(I)
SUMSZ=SUMSZ+STOTS(I)*ZS(I)
SUMSQ=SUMSQ+STOTS(I)*QS(I)

```

12 CONTINUE

SUBROUTINE MULTFS

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101 CONTINUE
C RESULTANT OVERALL FEED Q VALUE AND COMPOSITION.
  QMIX=(SUMFQ-SUMSQ)/(SUMF-SUMS)
  ZMIX=(SUMFZ-SUMSZ)/(SUMF-SUMS)
  IF(NOS.EQ.1) GO TO 102
C BOTTOMS FLOWRATE AND COMPOSITION FOR PARTIAL REBOILER OPERATION.
  BTOT=SUMF-SUMS-DTOT
  XB=(SUMFZ-SUMSZ-DTOT*XD)/BTOT
  IF(BTOT.LT.0.0.OR.XB.LT.0.0) GO TO 902
  YB=XB
  GO TO 103
102 CONTINUE
C BOTTOMS FLOWRATE AND COMPOSITION FOR OPEN STEAM OPERATION,
  YB=0.0
  GOP=DTOT*(R+1.)+(QMIX-1.)*(SUMF-SUMS)
  BTOT=SUMF-SUMS-DTOT+GOP
  XB=(SUMFZ-SUMSZ-DTOT*XD)/BTOT
  IF(BTOT.LT.0.0.OR.XB.LT.0.0) GO TO 902
103 CONTINUE
  YD=XD
C POINT OF INTERSECTION OF TOP AND BOTTOM OVERALL OPERATING LINES.
  XQM=((R+1.)*ZMIX+(QMIX-1.)*XD)/(R+QMIX)
  YQM=(R*ZMIX+QMIX*XD)/(R+QMIX)
  SLOPB=(YQM-YB)/(XQM-XB)
C FLOWRATE IN BOTTOM SECTION OF COLUMN : PARTIAL REBOILER OPERATION.
  AV(1)=BTOT/(SLOPB-1.)
  AL(1)=BTOT+AV(1)
C FLOWRATE IN BOTTOM SECTION OF COLUMN : OPEN STEAM OPERATION.
  IF(NOS.EQ.1) AV(1)=GOP
  IF(NOS.EQ.1) AL(1)=BTOT
  CR=YB-SLOPB*XB

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SUBROUTINE MULTFS

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C      NN=NF+NS+1
      DO 21 I=NN,2,-1
      ZMIN=XB
      DO 22 K=1,NS
      ZMAX=ZS(K)
      IF(ZMIN-ZMAX) 201,901,202
201  ZN=ZMAX
      ZMIN=ZMAX
      NI=K
      NNN=1
202  CONTINUE
22  CONTINUE
      DO 23 L=1,NF
      ZMAX=ZF(L)
      IF(ZMIN-ZMAX) 203,901,204
203  ZN=ZMAX
      ZMIN=ZMAX
      NI=L
      NNN=3
204  CONTINUE
23  CONTINUE
      IF(NNN-2) 205,206,206
      COMPOSITION.FLOWRATE AND Q VALUE OF STREAMS IN ORDER OF
      DECREASING COMPOSITION FROM THE TOP OF THE COLUMN.
205  ZFS(I)=ZN
      FLOW(I)=-STOTS(NI)
      QFS(I)=QS(NI)
      ZS1(NI)=ZS(NI)
      ZS(NI)=-1.
      GO TO 21
206  ZFS(I)=ZN
      FLOW(I)=FTOTS(NI)
      QFS(I)=QF(NI)
      ZF1(NI)=ZF(NI)
      ZF(NI)=-1.
21  CONTINUE

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SUBROUTINE MULTFS

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C      SLOPE AND Y-AXIS INTERCEPT OF OPERATING LINES IN EACH SECTION
C      OF THE COLUMN.
      SLPE(1)=SLOPB
      CFCT(1)=CB
      QX(1)=XB
      QY(1)=YB
      NN1=NN+1
      SLPE(NN1)=1.0
      CFCT(NN1)=0.0
C      *DO* LOOP 24 COMPUTES POINTS OF INTERSECTION OF OPERATING LINES*
C      AND LIQUID AND VAPOUR FLOWRATES WITHIN EACH SECTION.
      DO 24 I=2*NN
      II=I-1
      QX(I)=(ZFS(I)-CFCT(II)*(1.-QFS(I)))/(SLPE(II)*(1.-QFS(I))+QFS(I))
      QY(I)=SLPE(II)*QX(I)+CFCT(II)
      AL(II)=AL(II)-QFS(I)*FLOW(I)
      AV(I)=AV(II)+(1.-QFS(I))*FLOW(I)
      SLPE(I)=AL(II)/AV(I)
      CFCT(I)=QY(I)-SLPE(I)*QX(I)
24    CONTINUE
      QX(NN1)=XD
      QY(NN1)=YD
      GO TO 999
C      ERROR MESSAGES *****
901    CONTINUE
      MALF=1000
      WRITE(NTP,9001)
9001  FORMAT(/T2,'ERROR...TWO STREAMS HAVE THE SAME COMPOSITION.')
```

GO TO 999

902 CONTINUE

MALF=1000

WRITE(NTP,9002)

9002 FORMAT(/T2,'ERROR...BTOT OR XB IS LESS THAN O.D.')

999 CONTINUE

RETURN

END

SUBROUTINE MULTFS

0001

```

C SUBROUTINE SIMPLE
C ROUTINE FOR DETERMINING OPERATING CONDITIONS FOR BINARY PROBLEMS
C INVOLVING SIMPLE COLUMNS. (ONE FEED, NO SIDE STREAMS)
C THE ROUTINE: (1) COMPUTES BOTTOMS FLOWRATE AND COMPOSITION FROM AN
C OVERALL MASS BALANCE.
C (2) DETERMINES OPERATING LINE EQUATIONS.
C (3) COMPUTES LIQUID AND VAPOUR FLOWRATES IN EACH
C SECTION OF COLUMN.

```

```

C *****
C INSERT COMMON BLOCK 'BINCOM' HERE.
C *****
C REAL LH,LHCONV,LHMW,LIQR,LIQS,LRWT,LSWT,MW

```

```

ZFS(2)=Z
QFS(2)=Q

```

```

FLOW(2)=FTOT
POINT OF INTERSECTION OF OPERATING LINES.

```

```

XQM=((R+1.)*Z+(Q-1.)*XD)/(R+Q)
YQM=(R*Z+Q*XD)/(R+Q)

```

```

IF(NOS.EQ.1) GO TO 110

```

```

C BOTTOMS FLOWRATE AND COMPOSITION FOR PARTIAL REBOILER OPERATION.

```

```

BTOT=FTOT-DTOT

```

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XB=(FTOT*Z-DTOT*XD)/BTOT

```

```

IF(XB.LT.0.0.OR.XB.GT.Z) GO TO 901

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```

YB=XB

```

```

GO TO 120

```

```

110 CONTINUE

```

```

C BOTTOMS FLOWRATE AND COMPOSITION FOR OPEN STEAM OPERATION.

```

```

YB=0.0

```

```

GOP=DTOT*(R+1.)+(Q-1.)*FTOT

```

```

BTOT=FTOT-DTOT+GOP

```

```

XB=(FTOT*Z-DTOT*XD)/BTOT

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```

IF(XB.LT.0.0.OR.XB.GT.Z) GO TO 901

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```

120 CONTINUE

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YD=XD

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SUBROUTINE SIMPLE

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C      SLOPES AND Y-AXIS INTERCEPTS OF OPERATING LINES.
      SLPE(1)=(YQM-YB)/(XQM-XB)
      CFCT(1)=YQM-SLPE(1)*XQM
      SLPE(2)=R/(R+1.)
      CFCT(2)=XD/(R+1.)
C      VAPOUR AND LIQUID FLOWRATES WITHIN THE COLUMN.
      AV(1)=BTOT/(SLPE(1)-1.)
      AL(1)=BTOT+AV(1)
      IF(NOS.EQ.1) AV(1)=GOP
      IF(NOS.EQ.1) AL(1)=BTOT
      AL(2)=DTOT*R
      AV(2)=DTOT+AL(2)
      SLPE(3)=1.0
      CFCT(3)=0.0
C      POINTS OF INTERSECTION OF OPERATING LINES WITH DIAGONAL AND EACH OTHER.
      QX(1)=XB
      QY(1)=YB
      QX(2)=XQM
      QY(2)=YQM
      QX(3)=XD
      QY(3)=YD
      NN=2
      GO TO 999
C      ERROR MESSAGES *****
901  CONTINUE
      WRITE(NTP,9001)
9001 FORMAT(/T2,'ERROR...BTOT OR XB IS LESS THAN 0.0,./T10,'OR XB IS G
      1REATER THAN Z.')
```

MALF=1000

999 CONTINUE

RETURN

END

SUBROUTINE SIMPLE



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0001
SUBROUTINE PONSAY(IPTS,NPTS,NFIN,MFIN)
ROUTINE FOR CONSTRUCTION OF PONCHON-SAVARIT OPERATING DIAGRAM.
THE ROUTINE: (1) COMPUTES BOTTOMS FLOWRATE AND COMPOSITION FROM AN
OVERALL MASS BALANCE.
(2) CONSTRUCTS ENTHALPY- CONCENTRATION DIAGRAM.
(3) COMPUTES MINIMUM REFLUX RATIO.
(4) DEFINES OPERATING LINES ON X-Y DIAGRAM.
(5) COMPUTES VAPOUR AND LIQUID MASS AND MOLAR FLOWS
THROUGH COLUMN.
(6) DETERMINES CONDENSER AND REBOILER HEAT DUTIES.
*****
INSERT COMMON BLOCK 'BINCOM' HERE.
*****
REAL LH,LHCONV,LHMW,LIQR,LIQS,LRWT,LSWT,MW
DATA CONFAC,ENCONV/0,00027778,2.326/
CALCULATION OF BOTTOMS FLOWRATE AND COMPOSITION.
BTOT=FTOT-DTOT
XB=(FTOT*Z-DTOT*XD)/BTOT
IF(XB.LT.0.0.OR.XB.GT.Z) GO TO 901
TDG=TG
DELY=0.05
DELX=0.05
YH(1)=0.0
XH(1)=0.0
NENTH=0
'DO' LOOP 10 COMPUTES DATA REQUIRED TO CONSTRUCT ENTHALPY-
CONCENTRATION DIAGRAM IN PONCHON-SAVARIT DESIGNS.
DO 10 K=1,100
XG=XH(K)
YG=YH(K)
IF(YG.GT.1.01) GO TO 100
NENTH=NENTH+1
TA=TDG
'BINDPT' CALCULATES DEW POINTS.
CALL BINDPT(A,B,C,NBDC,NDATA,PT,PA,TALF,TBP,TA,YALF,YG)
TDU=TA

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SUBROUTINE PONSAY

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C      TA=TDG
      BINBPT, CALCULATES BUBBLE POINTS.
      CALL BINBPT(A,B,C,NBDC,NDATA,PT,PA,TALF,TBP,TA,XALF,XG)
      TBU=TA
C      HTCAP, COMPUTES VAPOUR AND LIQUID ENTHALPY.
      CALL HTCAP(TDU,TDU,TBU,YG,XG,HV,HL)
      HVAP(K)=HV*ENCONV
      HLIQ(K)=HL*ENCONV
      YH(K+1)=YH(K)+DELY
      XH(K+1)=XH(K)+DELX
      10 CONTINUE
      100 CONTINUE
C      COMPUTATION OF FEED ENTHALPY.
      TDF=TG
      TBF=TG
      CALL BINDPT(A,B,C,NBDC,NDATA,PT,PA,TALF,TBP,TDF,YALF,Z)
      CALL BINBPT(A,B,C,NBDC,NDATA,PT,PA,TALF,TBP,TBF,XALF,Z)
      CALL HTCAP(TDF,TDF,TBF,Z,Z,HVF,HLF)
      HF=(HVF-Q*(HVF-HLF))*ENCONV
      RMIN, COMPUTES THE MINIMUM REFLUX RATIO.
      CALL RMIN(HLD,HVD)
      IF(MALF,GT.999) GO TO 999
C      CHECK IF SPECIFIED REFLUX RATIO IS ACCEPTABLE. IF NOT USE 1.5*RM
      IF(RM,GT.R) GO TO 902
      GO TO 190
      199 R=1.5*RM
      190 CONTINUE
C      SEARCH FOR DISTILLATE ENTHALPY DEPENDENT ON CONDENSER TYPE.
      IF(NCOND.EQ.1) CALL HFIND(XD,HD,HLIQ,NENTH,XH)
      IF(NCOND.EQ.2) CALL HFIND(XB,HD,HVAP,NENTH,YH)
C      CALCULATION OF REBOILER AND CONDENSER HEAT DUTY, AND POSITION OF
C      DEL. POINTS ON ENTHALPY-CONCENTRATION DIAGRAM.
      Q1DASH=R*(HVD-HLD)+HVD
      QC=(Q1DASH-HD)*DTOT
      Q2DASH=(FTOT*HF-DTOT*Q1DASH)/BTOT
C      'HFIND' RETURNS BOTTOMS ENTHALPY HB (AT CONC XB).
      CALL HFIND(XB,HB,HLIQ,NENTH,XH)
      QB=(HB-Q2DASH)*RTOT

```

SUBROUTINE PONSAY

```

DELD=Q1DASH
DELB=Q2DASH
SLOPE=(DELD-HF)/(XD-Z)
CFACT=DELD-XD*SLOPE
DELXF=0.01
MN=1
XG=Z+DELXF
  GO TO 201
  GO TO 202 SEARCH FOR THE POINT OF INTERSECTION OF
  THE OPERATING LINES. (XCR,YCR)
201 CONTINUE
  CALL HFIND(XG,HLC,HVAP,NENTH,XH)
  HG=SLOPE*XG+CFACT
  IF(ABS(HLC-HG).LT.10.) GO TO 210
  NCALL=2
  CALL SEARCH(HLC,HG,XG,DELXF,MN,NCALL)
  GO TO 201
210 CONTINUE
  XCR=XG
  DELYF=0.01
  MN=1
  YG=Z+DELYF
220 CONTINUE
  CALL HFIND(YG,HVC,HVAP,NENTH,YH)
  HG=SLOPE*YG+CFACT
  IF(ABS(HVC-HG).LT.10.) GO TO 220
  NCALL=2
  CALL SEARCH(HVC,HG,YG,DELYF,MN,NCALL)
  GO TO 202
220 CONTINUE
  YCR=YG
  *OPLINE* COMPUTES COORDINATES OF POINTS ON THE OPERATING LINES,
  *USING THE PONCHON-SAVARIT TECHNIQUE.
  CALL OPLINE(NPTS,MPTS)
  IF(MALF.GT.999) GO TO 999
  *FLOWS* COMPUTES LIQUID AND VAPOUR FLOWRATES WITHIN THE COLUMN.
  CALL FLOWS(NPTS,MPTS,NFIN,MFIN)
  IF(MALF.GT.999) GO TO 999

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SUBROUTINE PONSAY

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C      CONVERSION OF HEAT DUTIES AND ENTHALPY TO S.I. UNITS.
C      (QF(1) IS USED TO STORE FEED ENTHALPY,
      QC=QC*CONFAC
      QB=QB*CONFAC
      QF(1)=HF
      IF(NOP.EQ.2) GO TO 999
      YB=XB
      YD=XD
      ZFS(2)=Z
      QFS(2)=Q
      FLOW(2)=FTOT
      NN=2
C      POINTS OF INTERSECTION OF OPERATING LINES WITH DIAGONAL AND EACH OTHER.
      QX(1)=XB
      QY(1)=YB
      QX(2)=XCR
      QY(2)=YCR
      QX(3)=XD
      QY(3)=YD
      GO TO 999
C      ERROR MESSAGES *****
901  CONTINUE
      WRITE(NTP,9001)
9001  FORMAT(/T2,'ERROR...BTOT OR XB IS LESS THAN 0.0,./T10,OR XB IS G
      IREATER THAN Z.')
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      MALF=1000
      GO TO 999
902  CONTINUE
      WRITE(NTP,9002) RM,R
9002  FORMAT(/T2,'SPECIFIED REFLUX RATIO IS LESS THAN THE CALCULATED MIN
      IIMUM.,/T4,'RMIN=',F6.2,'R SPEC=',F6.2,'/T2,
      2'A VALUE OF 1.5*RMIN WILL BE USED.')
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      GO TO 199
999  CONTINUE
      RETURN
      END
```

SUBROUTINE PONSAY

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C      SUBROUTINE HTCAP(TDPT,T2D,T2B,YS,XS,HV,HL)
C      ROUTINE FOR COMPUTING VAPOUR AND LIQUID ENTHALPY FOR BINARY SYSTEMS.
C      *****
C      INSERT COMMON BLOCK 'BINCOM' HERE.
C      *****
C      DIMENSION Cp(2),LHT(2)
C      REAL LH,LHCONV,LHMW,LIQR,LIQS,LRWT,LSWT,MW
C      REAL LHT
C      T=TBASE+TCONV
C      TD=T2D+TCONV
C      TB=T2B+TCONV
C      VAPOUR ENTHALPY CALCULATIONS.
C      DO 10 LOOP 10 COMPUTES HEAT CAPACITY OF EACH COMPONENT AT THE
C      BUBBLE POINT,RELATIVE TO THE BASE TEMPERATURE.
C      DO 10 I=1,2
C      AFACT=ACP(I)*(TD-T)
C      BFACT=BCP(I)*(TD**2,-T**2.)/2.
C      CFACT=CCP(I)*(TD**3,-T**3.)/3.
C      DFACT=DCP(I)*(TD**4,-T**4.)/4.
C      CP(I)=AFACT+BFACT+CFACT+DFACT
C      10 CONTINUE
C      Y1=CP(1)*YS*CPMW(1)*CPCONV
C      Y2=CP(2)*(1.-YS)*CPMW(2)*CPCONV
C      LATENT HEAT CALCULATIONS.
C      DO 11 I=1,2
C      TFACT=(TC(I)-TDPT)/(TC(I)-TBP(I))
C      LHT(I)=LH(I)*(TFACT**0.38)*LHMW(I)*LHCONV
C      11 CONTINUE
C      TOTAL VAPOUR ENTHALPY.
C      HV=Y1+Y2+YS*LHT(1)+(1.-YS)*LHT(2)

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SUBROUTINE HTCAP

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C LIQUID ENTHALPY CALCULATIONS.
C *DO* LOOP 20 COMPUTES HEAT CAPACITY OF EACH COMPONENT AT THE
C DEW POINT RELATIVE TO THE BASE TEMPERATURE.
DO 20 I=1,2
  AFAC=ACP(I)*(TB-T)
  BFAC=BCP(I)*(TB**2.-T**2.)/2.
  CFAC=CCP(I)*(TB**3.-T**3.)/3.
  DFAC=DCP(I)*(TB**4.-T**4.)/4.
  CP(I)=AFAC+BFAC+CFAC+DFAC
20 CONTINUE
C HEAT OF MIXING INCLUDED IF SPECIFIED.
IF(NHS.GT.0) CALL HTSOLN(HS,XS,HEVABS,HSCONV,HSMW,HSOLN,NHS,XHS)
X1=CP(1)*CPMW(1)*CPCONV
X2=CP(2)*CPMW(2)*CPCONV
IF(NHS.EQ.0) HS=0.0
TOTAL LIQUID ENTHALPY.
HL=X1*XS+X2*(1.-XS)+HS
RETURN
END

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SUBROUTINE HTCAP

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C      SUBROUTINE HTSOLN(H,XA,HEVABS,HSCONV,HSMW,HSOLN,NHS,XHS)
C      ROUTINE FOR COMPUTING HEATS OF SOLUTION FOR BINARY MIXTURES.
C      LINEAR INTERPOLATION IS USED BETWEEN DATA POINTS.
      DIMENSION HSMW(1),HSOLN(1),XHS(1)
      IF(XA.GT.XHS(1)) GO TO 100
      H=HSOLN(1)*HSMW(1)+HSCONV*HEVABS*XA
      GO TO 999
100 CONTINUE
      DO 10 I=1,NHS
      DIFF=XA-XHS(I)
      IF(DIFF.LT.0.0) GO TO 110
10 CONTINUE
      GO TO 120
110 XFACT=(XHS(I)-XA)/(XHS(I)-XHS(I-1))
      HFACT=HSCONV*HEVABS*XA
      H=(HSOLN(I)+(HSOLN(I-1)-HSOLN(I))*XFACT)*HSMW(I)*HFACT
      GO TO 999
120 H=HSOLN(NHS)*HSMW(1)+HSCONV*HEVABS*XA
999 CONTINUE
      RETURN
      END
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SUBROUTINE HTSOLN

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C      SUBROUTINE HFIND(XA,H,HENTH,NENTH,XYH)
C      ROUTINE SEARCHES FOR VAPOUR OR LIQUID ENTHALPY VALUE CORRESPONDING
C      TO THE COMPOSITION AND PHASE SUPPLIED.
C      LINEAR INTERPOLATION IS USED BETWEEN ACTUAL DATA POINTS.
      DIMENSION HENTH(1),XYH(1)
      IF(XA.GT.XYH(1)) GO TO 100
      H=HENTH(1)
      GO TO 999
100 CONTINUE
      DO 10 I=1,NENTH
        DIFF=XA-XYH(I)
        IF(DIFF.LT.0.0) GO TO 110
110 CONTINUE
        GO TO 120
110 XFACT=(XYH(I)-XA)/(XYH(I)-XYH(I-1))
      H=HENTH(I)+(HENTH(I-1)-HENTH(I))*XFACT
      GO TO 999
120 H=HENTH(NENTH)
999 CONTINUE
      RETURN
      END
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SUBROUTINE HFIND



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0001 SUBROUTINE RMIN(HLD,HVD)
ROUTINE CALCULATES MINIMUM REFLUX RATIO IN PONCHON-SAVARIT
DESIGN OF BINARY SYSTEMS.
*****
C INSERT COMMON BLOCK 'BINCOM' HERE.
*****
C REAL LH,LHCONV,LHNV,LIQR,LIQS,LRWT,LSWT,MW
ACC=0.0001*HF
DELY=0.02
YI=XD
MN=1
NCOUNT=0
'GO TO' LOOP 100 SEARCHES FOR THE TIE LINE THAT PASSES THROUGH
POINT (Z,HF) ON THE ENTHALPY-CONCENTRATION DIAGRAM.
100 CONTINUE
NCOUNT=NCOUNT+1
IF(NCOUNT.GT.500) GO TO 902
CALL XYALFA(ALFA,ALFCON,ALF,MALF,NDATA,YALF,YI)
X1=YI/(ALF-YI*(ALF-1.))
IF(NCOUNT.GT.1) GO TO 101
IF(X1.LE.Z) GO TO 901
101 CONTINUE
CALL HFIND(X1,HLI,HLIQ,NENTH,XH)
CALL HFIND(YI,HVI,HVAP,NENTH,YH)
SLOPE=(HVI-HLI)/(YI-XI)
CFACT=HLI-SLOPE*X1
HTEST=SLOPE*Z+CFACT
IF(ABS(HTEST-HF).LT.ACC) GO TO 200
NCALL=2
CALL SEARCH(HTEST,HF,YI,DELY,MN,NCALL)
GO TO 100
200 CONTINUE

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SUBROUTINE RMIN

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C X,Y COORDINATES OF TIE LINE REPRESENT A POINT OF INTERSECTION OF
C THE OPERATING LINES THAT FALLS ON THE EQUILIBRIUM CURVE,
C THIS IS A STATE OF MINIMUM REFLUX.
C DELDM=XD*SLOPE+CFAC
C *HFIND* RETURNS ENTHALPY OF REFLUX LIQUID.
C CALL HFIND(XC,HLD,HLIQ,NENTH,XH)
C IF(NCOND.EQ.2) GO TO 210
C TOTAL CONDENSER OPERATION.
C *HFIND* RETURNS ENTHALPY OF VAPOUR ENTERING TOTAL CONDENSER.
C CALL HFIND(XD,HVD,HVAP,NENTH,YH)
C GO TO 220
210 CONTINUE
C PARTIAL CONDENSER OPERATION.
C SLPM=(DELDH-HLD)/(XD-XC)
C CFM=DELDH-XD*SLPM
C DELXM=0.0001
C MN=1
C XI=XC+DELXM
C NCOUNT=0
C *GO TO* LOOP 211 SEARCHES FOR ENTHALPY OF VAPOUR ENTERING PARTIAL
C CONDENSER.
C 211 CONTINUE
C NCOUNT=NCOUNT+1
C IF(NCOUNT.GT.500) GO TO 903
C CALL HFIND(XI,HVD,HVAP,NENTH,YH)
C HVCM=XI*SLPM+CFM
C IF(ABS(HVD-HVCM).LT.ACC) GO TO 220
C NCALL=2
C CALL SEARCH(HVD,HVCM,XI,DELXM,MN,NCALL)
C GO TO 211
220 CONTINUE
C MINIMUM REFLUX RATIO.
C RM=(DELDH-HVD)/(HVD-HLD)
C GO TO 999

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SUBROUTINE RMIN

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C 901 ERROR MESSAGES *****
    CONTINUE
    WRITE(NTP,9001)
    FORMAT(/T2,'ERROR IN RMIN. FEED ENTERS ON TOP TRAY.')
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    MALF=1000
    GO TO 999
    902 CONTINUE
    WRITE(NTP,9002)
    FORMAT(/T2,'ERROR IN RMIN ROUTINE.'/T2,'TIE LINE PASSING THROUGH
    IPOINT (Z,HF) NOT FOUND.')
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    WRITE(NTP,9012) HTEST,HF,Y1,X1,NCOUNT,DELY
    FORMAT(T2,'HTEST=',E12.8,2X,'HF=',E12.8,2X,'Y1=',F6.4,2X,'X1=',
    IF6.4,/T12,'NO. OF ITERATIONS=',I9.2X,'INCREMENT=',E12.8)
    MALF=1000
    GO TO 999
    903 CONTINUE
    WRITE(NTP,9003)
    FORMAT(/T2,'ERROR IN RMIN ROUTINE.'/T2,'ENTHALPY OF VAPOUR ENTERI
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    NG PARTIAL CONDENSER NOT FOUND.')
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    WRITE(NTP,9013) HVD,HVCM,X1,XC,NCOUNT,DELXM
    FORMAT(T2,'HVD=',E12.8,2X,'HVCM=',E12.8,2X,'X1=',F6.4,2X,'XC=',
    IF6.4,/T12,'NO. OF ITERATIONS=',I9.2X,'INCREMENT=',E12.8)
    MALF=1000
    999 CONTINUE
    RETURN
    END
```

SUBROUTINE RMIN

```

0001 SUBROUTINE OPLINE(NPTS,MPTS)
ROUTINE CALCULATES X,Y POINTS USED TO CONSTRUCT THE OPERATING LINES
ON A PONCHON-SAVARIT DIAGRAM.
*****
INSERT COMMON BLOCK 'BINCOM' HERE.
*****
REAL LH,LHCONV,LHNW,LIQR,LIQS,LRWT,LSWT,MW
ACC=0.0001*HF
RECTIFYING SECTION.
END POINTS OF RECTIFYING SECTION ARE XD AND XCR.
THIS INTERVAL IS DIVIDED INTO NINT INCREMENTS OF SIZE H.
NPTS=NINT+1
MN=1
X1=XD
H=(XD-XCR)/NINT
Y1=X1
XROP(NPTS)=X1
YROP(NPTS)=Y1
*GO TO* LOOP 100 USES PONCHON-SAVARIT METHOD TO FIND THE Y-COORDINATE
AT EACH VALUE OF X1, FROM XD TO XCR IN STEPS OF -H.
100 CONTINUE
NPTS=NPTS-1
X1=X1-H
IF(X1.LT.XCR) GO TO 500
*HFIND* RETURNS LIQUID ENTHALPY AT CONCENTRATION X1.
CALL HFIND(X1,HLQ,HLIQ,NENTH,XH)
SLOPE AND Y-AXIS INTERCEPT OF INTERNAL REFLUX RATIO LINE.
SLOPE=(DELD-HLQ)/(XD-X1)
CFACT=HLQ-SLOPE*X1
DELY=0.01
YG=X1
YG=YG+DELY
NLOOP=130
NCERR=0

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SUBROUTINE OPLINE

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C      *GO TO* LOOP 130 FINDS INTERSECTION OF VAPOUR ENTHALPY CURVE AND
C      INTERNAL REFLUX RATIO LINE WHICH YIELDS THE Y-COORDINATE YG OF THE
C      OPERATING LINE POINT AT X1.
      130 CONTINUE
      NCERR=NCERR+1
      IF(NCERR.GT.500) GO TO 902
      CALL HFIND(YG,HVP,HVAP,NENTH,YH)
      HG=SLOPE*YG+CFACT
      IF(ABS(HVP-HG).LT.ACC) GO TO 110
      NCALL=2
      CALL SEARCH(HVP,HG,YG,DELY,MN,NCALL)
      GO TO 130
      110 CONTINUE
      XROP(NPTS)=X1
      YROP(NPTS)=YG
      GO TO 100
      500 CONTINUE
      C      STRIPPING SECTION.
      C      END POINTS OF STRIPPING SECTION ARE XB AND XCR.
      C      NEXT FIVE LINES CALCULATE THE SIZE AND NUMBER OF INCREMENTS IN
      C      THE STRIPPING SECTION,KEEPING THE INCREMENT SIZE AS SIMILAR AS
      C      POSSIBLE TO THAT USED IN THE RECTIFYING SECTION.
      DELBC=XCR-XB
      FACT=DELBC/H
      NFACT=IFIX(FACT)
      IF(NFACT.GT.NPPSC) GO TO 901
      H=DELBC/NFACT
      MN=2
      X1=XB
      Y1=Y1
      XSOP(1)=X1
      YSOP(1)=Y1
      MPTS=1

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SUBROUTINE OPLINE

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C      *GO TO* LOOP 200 USES PONCHON-SAVARIT METHOD TO FIND THE Y-COORDINATE
C      AT EACH VALUE OF X1, FROM XB TO XCR IN STEPS OF H.
200 CONTINUE
X1=X1+H
IF(X1.GT.XCR) GO TO 999
MPTS=MPTS+1
C      *HFIND* RETURNS LIQUID ENTHALPY AT CONCENTRATION X1,
C      CALL HFIND(X1,HLQ,HLIQ,NENTH,XH)
C      SLOPE AND Y-AXIS INTERCEPT OF INTERNAL REFLUX RATIO LINE,
C      SLOPE=(HLIQ-DELB)/(X1-XB)
C      CFACT=HLIQ-SLOPE*X1
C      DELY=D.01
C      YG=Y1
C      YG=YG+DELY
C      NLOOP=NLOOP+1
C      NCERR=0
C      *GO TO* LOOP 230 FINDS INTERSECTION OF VAPOUR ENTHALPY CURVE AND
C      INTERNAL REFLUX RATIO LINE, WHICH YIELDS THE Y-COORDINATE YG OF THE
C      OPERATING LINE POINT AT X1.
230 CONTINUE
NCERR=NCERR+1
IF(NCERR.GT.500) GO TO 902
CALL HFIND(YG,HVP,HVAP,NENTH,YH)
HG=SLOPE*YG+CFACT
IF(ABS(HVP-HG).LT.ACC) GO TO 210
NCALL=NCALL+1
CALL SEARCH(HVP,HG,YG,DELY,MN,NCALL)
GO TO 230
210 CONTINUE
XSOP(MPTS)=X1
YSOP(MPTS)=YG
GO TO 200

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SUBROUTINE OPLINE

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C 901 ERROR MESSAGES *****
    CONTINUE
    WRITE(NTP,9001) NFACT
9001 FORMAT(/T2,ERROR,..NO. OF INTERVALS CALCULATED IN SUBROUTINE OPLI
    INE,./T2,EXCEEDS DIMENSION SPECIFICATIONS: CALCULATED NO.,I4)
    MALF=1000
    GO TO 999
902 CONTINUE
    WRITE(NTP,9002) NLOOP,HVP,HG,DELY
9002 FORMAT(/T2,ERROR IN OPLINE,./T2,LOOP,I4,EXCEEDED 500 ITERATI
    IONS,./T2,HVP=,E12.8,2X,HG=,E12.8,2X,Y-INCREMENT=,F10.8)
    MALF=1000
999 CONTINUE
    YSOP(MPTS)=YCR
    NPTS=NINT+1
    RETURN
    END

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SUBROUTINE OPLINE

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C	SUBROUTINE FLOWS(NPTS,MPTS,NFIN,MFIN)	0001
C	ROUTINE CALCULATES VAPOUR AND LIQUID FLOWRATES THROUGH A BINARY	
C	DISTILLATION COLUMN WHEN CONSTANT MOLAL OVERFLOW IS NOT ASSUMED,	
C	(PONCHON-SAVARIT DESIGN PROCEDURE)	
C	*****	
C	INSERT COMMON BLOCK 'BINCOM' HERE.	
C	*****	
C	REAL LH,LHCONV,LHFW,LIQR,LIQS,LRWT,LSWT,MW	0002
	NTOT=NPTS+MPTS	0003
	IF(NTOT.GT.NPPSC) GO TO 901	0004
	NTOT1=NTOT-1	0005
	NRFIN=MPTS+1	0006
	YD=XD	0007
	T1=TG	0008
	LIQR(NTOT)=DTOT*R	0009
	VAPR(NTOT)=DTOT*(R+1.)	0010
	CALL AVEFW(XD,YD,AVEL,AVEV,MW)	0011
	LRWT(NTOT)=LIQR(NTOT)*AVEL	0012
	VRWT(NTOT)=VAPR(NTOT)*AVEV	0013
	CALL BINBPT(A,B,C,NBDC,NDATA,PT,P1,TALF,TBP,T1,XALF,XD)	0014
	TRL(NTOT)=T1	0015
	CALL BINDPT(A,B,C,NBDC,NDATA,PT,P1,TALF,TBP,T1,YALF,YD)	0016
	TRV(NTOT)=T1	0017

SUBROUTINE FLOWS



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C      'DO' LOOP 11 COMPUTES VAPOUR AND LIQUID MOLAR AND MASS FLOWRATES
C      IN RECTIFYING SECTION, TOGETHER WITH DEW AND BUBBLE POINTS.
      POSN='ABOVE'
      DO 11 I=NTOT1,NRFIN,-1
      NRS=I-MPTS
      LIQR(I)=DTOT*(XD-YROP(NRS))/(YROP(NRS)-XROP(NRS))
      IF(LIQR(I).LT.0.0) GO TO 902
      XAVE=XROP(NRS)
      YAVE=YROP(NRS)
      CALL AVENW(XAVE,YAVE,AVEL,AVEV,MW)
      LRWT(I)=LIQR(I)*AVEL
      VAPR(I)=LIQR(I)*(XD-XROP(NRS))/(XD-YROP(NRS))
      IF(VAPR(I).LT.0.0) GO TO 903
      VRWT(I)=VAPR(I)*AVEV
      CALL BINBPT(A,B,C,NBDC,NDATA,PT,PI,TALF,TBP,TI,XALF,XAVE)
      TRL(I)=TI
      CALL BINDPT(A,B,C,NBDC,NDATA,PT,PI,TALF,TBP,TI,YALF,YAVE)
      TRV(I)=TI
11 CONTINUE
C      'DO' LOOP 12 COMPUTES VAPOUR AND LIQUID MOLAR AND MASS FLOWRATES
C      IN STRIPPING SECTION, TOGETHER WITH DEW AND BUBBLE POINTS.
      POSN='BELOW'
      DO 12 I=MPTS,1,-1
      LIQS(I)=BTOT*(YSOP(I)-XB)/(YSOP(I)-XSOP(I))
      IF(LIQS(I).LT.0.0) GO TO 902
      XAVE=XSOP(I)
      YAVE=YSOP(I)
      CALL AVENW(XAVE,YAVE,AVEL,AVEV,MW)
      LSWT(I)=LIQS(I)*AVEL
      VAPS(I)=LIQS(I)*(XSOP(I)-XB)/(YSOP(I)-XB)
      IF(VAPS(I).LT.0.0) GO TO 903
      VSWT(I)=VAPS(I)*AVEV
      CALL BINBPT(A,B,C,NBDC,NDATA,PT,PI,TALF,TBP,TI,XALF,XAVE)
      TSL(I)=TI
      CALL BINDPT(A,B,C,NBDC,NDATA,PT,PI,TALF,TBP,TI,YALF,YAVE)
      TSV(I)=TI
12 CONTINUE

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SUBROUTINE FLOWS

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C      REBOILER STAGE IS STEPPED OFF FROM XB AS IN A MCCABE-THIELE
C      CONSTRUCTION. STARTING POINT FOR THE PACKING IS XN,YN.
      CALL XYALFA(ALFA,ALFCON,ALF,MALF,NDATA,XALF,XB)
      YN=XB*ALF/(1.+XB*(ALF-1.))
      IF(YN.LT.YSOP(MPTS)) GO TO 202
      WRITE(NTP,2001) YN,XSOP(MPTS),YSOP(MPTS)
2001  FORMAT(/T2,'THE COLUMN CONSISTS OF A RECTIFYING SECTION ONLY.',
1/T2,'COMPOSITION OF VAPOUR LEAVING REBOILER=',F7.4,/T2,'OPERATING
2 LINES INTERSECT AT POINT (X,Y)=',F6.4,'F6.4)
      IF(NOP.NE.2) GO TO 999
      MFIN=1000
      DO 21 I=1,NPTS
      DIFF=YN-YSOP(I)
      IF(DIFF.LT.0.0) GO TO 201
21  CONTINUE
      201  NFIN=I-1
      GO TO 301
202  CONTINUE
      IF(NOP.NE.2) GO TO 999
      DO 22 I=1,MPTS
      DIFF=YN-YSOP(I)
      IF(DIFF.LT.0.0) GO TO 203
22  CONTINUE
      203  MFIN=I-1
      NFIN=I
      GO TO 302

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SUBROUTINE FLOWS

C FLOWRATES AT THE BOTTOM OF THE PACKING.

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301 CONTINUE
YFACT=(YROP(MFIN+1)-YN)/(YROP(MFIN+1)-YROP(MFIN))
XN=XROP(MFIN+1)+(XROP(MFIN)-XROP(MFIN+1))*YFACT
LIQR(MFIN)=BTOT*(YN-XB)/(YN-XN)
VAPR(MFIN)=LIQR(MFIN)*(XN-XB)/(YN-XB)
CALL AVE MW(XN,YN,AVEL,AVEV,MW)
LRWT(MFIN)=LIQR(MFIN)*AVEL
VRWT(MFIN)=VAPR(MFIN)*AVEV
CALL BINBPT(A,B,C,NBDC,NDATA,PT,PI,TALF,TBP,TI,XALF,XN)
TRL(MFIN)=TI
CALL BINDPT(A,B,C,NBDC,NDATA,PT,PI,TALF,TBP,TI,YALF,YN)
TRV(MFIN)=TI
XROP(MFIN)=XN
YROP(MFIN)=YN
GO TO 999

302 CONTINUE
YFACT=(YSOP(MFIN+1)-YN)/(YSOP(MFIN+1)-YSOP(MFIN))
XN=XSOP(MFIN+1)+(XSOP(MFIN)-XSOP(MFIN+1))*YFACT
LIQS(MFIN)=BTOT*(YN-XB)/(YN-XN)
VAPS(MFIN)=LIQS(MFIN)*(XN-XB)/(YN-XB)
CALL AVE MW(XN,YN,AVEL,AVEV,MW)
LSWT(MFIN)=LIQS(MFIN)*AVEL
VSWT(MFIN)=VAPS(MFIN)*AVEV
CALL BINBPT(A,B,C,NBDC,NDATA,PT,PI,TALF,TBP,TI,XALF,XN)
TSL(MFIN)=TI
CALL BINDPT(A,B,C,NBDC,NDATA,PT,PI,TALF,TBP,TI,YALF,YN)
TSV(MFIN)=TI
XSOP(MFIN)=XN
YSOP(MFIN)=YN
GO TO 999

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SUBROUTINE FLOWS

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C  ERROR MESSAGES *****
901  CONTINUE
    WRITE(NTP,9001) NTOT
9001 FORMAT(/T2,•TOTAL NUMBER OF OPERATING LINE DATA POINTS,•,I5,
      1/T2,•EXCEEDS DIMENSION SPECIFICATIONS,•)
    MALF=1000
    GO TO 999
902  CONTINUE
    WRITE(NTP,9002) POSN,I
9002 FORMAT(/T2,•NEGATIVE LIQUID FLOW DETECTED ,•A5,• FEED POSITION,•,
      1/T2,•CHANGE IN FEED Q VALUE OR REFLUX RATIO SUGGESTED,•,/T2,
      2•POINT OF OCCURENCE ON OPERATING LINE:•,I4)
    MALF=1000
    GO TO 999
903  CONTINUE
    WRITE(NTP,9003) POSN,I
9003 FORMAT(/T2,•NEGATIVE VAPOUR FLOW DETECTED ,•A5,• FEED POSITION,•,
      1/T2,•CHANGE IN FEED Q VALUE OR REFLUX RATIO SUGGESTED,•,/T2,
      2•POINT OF OCCURENCE ON OPERATING LINE:•,I4)
    MALF=1000
999  CONTINUE
      RETURN
      END

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SUBROUTINE FLOWS

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SUBROUTINE AVEH(X1,Y1,AVEL,AVEV,NU)  
REAL NU  
DIMENSION NH(2)  
X2=1.-X1  
Y2=1.-Y1  
AVEV=X1*NU(1)+X2*NU(2)  
AVEV=Y1*NU(1)+Y2*NU(2)  
RETURN  
END
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SUBROUTINE AVEH

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C
SUBROUTINE CHEC(ALFA,ALFCON,FLOW,MALF,NDATA,NN,NTP,XALF,XQ,YQ,ZFS)
ROUTINE CHECKS THAT ALL OPERATING LINE INTERSECTION POINTS LIE
BETWEEN THE EQUILIBRIUM CURVE AND THE DIAGONAL ON THE X-Y DIAGRAM.
THIS ENSURES THAT NO PINCH POINTS EXIST.
DIMENSION ALFA(1),FLOW(1),XALF(1),XQ(1),YQ(1),ZFS(1)
DO 10 I=2,NN
XA=XQ(I)
CALL XYALFA(ALFA,ALFCON,ALF,MALF,NDATA,XALF,XA)
YA=XA*ALF/(1.+XA*(ALF-1.))
IF(YA.GT.YQ(I)) GO TO 10
IF(FLOW(I)) 101,102,102
101 CONTINUE
WRITE(NTP,1011) ZFS(I)
1011 FORMAT(/T2,'PINCH POINT AT SIDE STREAM:',F7.4)
GO TO 110
102 CONTINUE
WRITE(NTP,1021) ZFS(I)
1021 FORMAT(/T2,'PINCH POINT AT FEED STREAM:',F7.4)
GO TO 110
10 CONTINUE
GO TO 999
110 CONTINUE
MALF=1000
NN1=NN+1
999 CONTINUE
RETURN
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SUBROUTINE CHEC

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C          SUBROUTINE STEPDI(NTC,NPTS,MPTS)
C          ROUTINE FOR STEPPING OFF STAGES IN MCCABE-THIELE OR PONCHON-SAVARIT
C          DESIGN OF BINARY DISTILLATION TRAY-COLUMNS.
C          ROUTINE COMPUTES: (1) TOTAL NUMBER OF THEORETICAL TRAYS REQUIRED.
C          (2) TRAY POSITIONS OF FEED AND SIDE STREAMS.
C          (3) FLOW AND TEMP. PROFILES FOR P-S DESIGNS.
C          *****
C          INSERT COMMON BLOCK 'BINCOM' HERE.
C          *****
C          REAL LH,LHCONV,LHFW,LHQR,LHQS,LRWT,LSWT,MW
C          DATA I,MT,NI,NP/1,0,2,0/
C          NNI=NN+1
C          X(1)=XB
C          Y(1)=YB
C          XAI=XB
C          YAI=YB
C          XA=XB
C          *DO* LOOP 10 CAUSES STEPPING PROCEDURE TO MOVE FROM ONE OPERATING
C          LINE TO THE NEXT SO AS TO MINIMISE THE TOTAL NUMBER OF THEORETICAL
C          STAGES.
C          FEED AND SIDE STREAMS ARE ALSO POSITIONED HERE.
C          DO 10 J=2,NNI
C          IF(Y(1).GT.QY(J)) GO TO 100
C          *DO* LOOP 11 COMPUTES AND STORES THE COORDINATES OF EACH POINT
C          IN THE STEPPING PROCEDURE AND COUNTS THE NUMBER OF STAGES (MT),
C          DO 11 I=NI,100.2
C          MT=MT+1
C          IF(I.GT.NPTCC) GO TO 901
C          X(I)=X(I-1)
C          XA=X(I)
C          CALL XYALFA(ALFA,ALFCON,ALPHA,MALF,NDATA,XALF,XA)
C          Y(I)=(XA*ALPHA)/(1.+XA*(ALPHA-1.))
C          IF(MT.GT.1) Y(I)=Y(I-1)+(Y(I)-Y(I-1))*EFF
C          IF(Y(I).GT.QY(J)) GO TO 100
C          IF(Y(I).GT.YC) GO TO 200

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SUBROUTINE STEPDI

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C      X(I+1) FOR MCCABE-THIELE DESIGNS.
      IF(NOP.EQ.1) X(I+1)=(Y(I)-CFCT(J-1))/SLPE(J-1)
      IF(NOP.EQ.1) GO TO 111
C      X(I+1) FOR PONCHON-SAVARIT DESIGNS.
      CALL PSOPXY(Y(I),XOUT,XB,XCR,XROP,XSOP,YROP,YB,YC,YCR)
      X(I+1)=XOUT
111 CONTINUE
      Y(I+1)=Y(I)
11 CONTINUE
100 CONTINUE
      IF(NOP.GT.1) GO TO 101
C      X(I+1) FOR MCCABE-THIELE DESIGNS.
      X(I+1)=(Y(I)-CFCT(J))/SLPE(J)
      GO TO 102
101 CONTINUE
      X(I+1) FOR PONCHON-SAVARIT DESIGNS.
      CALL PSOPXY(Y(I),XOUT,XB,XCR,XROP,XSOP,YROP,YB,YC,YCR)
      X(I+1)=XOUT
102 CONTINUE
      Y(I+1)=Y(I)
      NP=NP+1
      TRAY NUMBER POSITIONS OF FEED AND SIDE STREAMS.
      NTRAY(NP)=(I/2)-1
      IF(Y(I).GT.YC) GO TO 200
      NI=I+2
10 CONTINUE
200 CONTINUE
      TFRAC=(YC-Y(I-2))/(Y(I)-Y(I-2))
      TRAYNO=(MT-(I+NCOND))+TFRAC
      NTC=I+1
      IF(NOP.NE.3) GO TO 999

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SUBROUTINE STEP D



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C TEMPERATURE AND VAPOUR-LIQUID FLOWRATE PROFILES FOR P-S DESIGNS,
DO 20 I=1,NTC
XA=X(I)
IF(XA.GT.XCR) GO TO 201
STRIPPING SECTION.
NP=MPTS
MP=D
CALL FFIND(XA,FLOV,FLOL,TEMP,VAPS,LIQS,TSL,XSOP,NP,MP)
LRWT(I)=FLOL
VRWT(I)=FLOV
TRV(I)=TEMP
GO TO 20
201 CONTINUE
RECTIFYING SECTION.
NP=NPTS
MP=MPTS
CALL FFIND(XA,FLOV,FLOL,TEMP,VAPR,LQIR,TRL,XROP,NP,MP)
LRWT(I)=FLOL
VRWT(I)=FLOV
TRV(I)=TEMP
20 CONTINUE
LRWT(I)=BTOT
IF(NCOND.EQ.1) VRWT(NTC)=D.0
IF(NCOND.EQ.2) VRWT(NTC)=DTOT
GO TO 999
ERROR MESSAGES *****
901 CONTINUE
WRITE(NTP,9001) MT
9001 FORMAT(/T2,'NUMBER OF TRAYS EXCEEDS:',I5,/T2,'PINCH ZONE IS SUSPEC
ATED.',/T2,'CURRENT RESULTS WILL BE PRINTED AND THESE SHOULD BE',
2/T2,'USED TO CONSTRUCT A MCCABE-THIELE DIAGRAM.',/T2,'IF A PINCH Z
3ONE IS NOT OBVIOUS THE SIZE OF THE',/T2,'ARRAYS X AND Y IN THE COM
4MON BLOCK MUST BE INCREASED TO',/T2,'ALLOW FOR A GREATER NUMBER OF
5 TRAYS.')
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999 CONTINUE
RETURN
END
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SUBROUTINE STEP D

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C      SUBROUTINE PSOPXY(Y1,XOUT,XB,XCR,XROP,XSOP,YROP,YB,YC,YCR)
C      ROUTINE SEARCHES FOR X-Y COORDINATES ON PONCHON-SAVARIT OPERATING
C      LINES FOR STEPPING OFF THEORETICAL STAGES IN *STEPD*.
C      LINEAR INTERPOLATION IS USED BETWEEN ACTUAL DATA POINTS.
      DIMENSION XROP(1),XSOP(1),YROP(1),YSOP(1)
      IF(Y1.GT.YSOP(2)) GO TO 100
      YFACT=(Y1-YB)/(YSOP(2)-YB)
      XOUT=XB+(XSOP(2)-XB)*YFACT
      GO TO 999
100  CONTINUE
      IF(Y1.GT.YCR) GO TO 200
      DO 10 I=2,500
      DIFF=Y1-YSOP(I)
      IF(DIFF.LE.0.0) GO TO 110
110  CONTINUE
      YFACT=(YSOP(I)-Y1)/(YSOP(I)-YSOP(I-1))
      XOUT=XSOP(I)+(XSOP(I-1)-XSOP(I))*YFACT
      GO TO 999
200  CONTINUE
      IF(Y1.GT.YROP(2)) GO TO 201
      YFACT=(Y1-YCR)/(YROP(2)-YCR)
      XOUT=XCR+(XROP(2)-XCR)*YFACT
      GO TO 999
201  CONTINUE
      IF(Y1.GT.YC) GO TO 300
      DO 20 I=2,500
      DIFF=Y1-YROP(I)
      IF(DIFF.LE.0.0) GO TO 220
220  CONTINUE
      YFACT=(YROP(I)-Y1)/(YROP(I)-YROP(I-1))
      XOUT=XROP(I)+(XROP(I-1)-XROP(I))*YFACT
      GO TO 999
300  CONTINUE
      XOUT=Y1
999  CONTINUE
      RETURN
      END

```

SUBROUTINE PSOPXY

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C SUBROUTINE FFIND(XA,FLOV,FLOL,TEMP,FVAP,FLIQ,FTEMP,FX,NPTS,MPTS)
C ROUTINE CALCULATES LIQUID AND VAPOUR FLOWS RESPECTIVELY LEAVING
C AND ENTERING THEORETICAL STAGES IN PONCHON-SAVARIT DESIGNS.
C LINEAR INTERPOLATION IS USED BETWEEN ACTUAL DATA POINTS.
DIMENSION FVAP(1),FLIQ(1),FTEMP(1),FX(1)
NTOT=NPTS+MPTS
MPTS1=MPTS+2
IF(XA.GT.FX(2)) GO TO 100
XFACT=(FX(2)-XA)/(FX(2)-FX(1))
FLOV=FVAP(NPTS1)-(FVAP(MPTS1+1)-FVAP(MPTS1))*XFACT
FLOL=FLIQ(MPTS1)-(FLIQ(MPTS1+1)-FLIQ(MPTS1))*XFACT
TEMP=FTEMP(MPTS1)-(FTEMP(MPTS1+1)-FTEMP(MPTS1))*XFACT
GO TO 999
100 CONTINUE
DO 10 I=MPTS1,NTOT
NNO=I-MPTS
DIFF=XA-FX(NNO)
IF(DIFF.LE.0.0) GO TO 110
10 CONTINUE
GO TO 120
110 CONTINUE
XFACT=(FX(NNO)-XA)/(FX(NNO)-FX(NNO-1))
FLOV=FVAP(I)+(FVAP(I-1)-FVAP(I))*XFACT
FLOL=FLIQ(I)+(FLIQ(I-1)-FLIQ(I))*XFACT
TEMP=FTEMP(I)+(FTEMP(I-1)-FTEMP(I))*XFACT
GO TO 999
120 CONTINUE
FLOV=FVAP(NTOT)
FLOL=FLIQ(NTOT)
TEMP=FTEMP(NTOT)
999 CONTINUE
RETURN
END

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SUBROUTINE FFIND

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SUBROUTINE BINRES(NDAT,NTC,NNI,NPTS,MPTS)
ROUTINE FOR PRINTING RESULTS OF MCCABE-THIELE OR PONCHON-SAVARIT
SOLUTIONS TO BINARY CONTINUOUS-DISTILLATION PROBLEMS.
*****
C INSERT COMMON BLOCK 'BINCOM' HERE.
*****
C REAL LH,LHCONV,LHNMW,LIQR,LIQS,LRWT,LSWT,MW
CON=COND,
1 FORMAT(A3)
1001 FORMAT(/T2,'MCCABE-THIELE DESIGN OF A BINARY DISTILLATION SYSTEM',
1)
1011 FORMAT(/T2,'PONCHON-SAVARIT DESIGN OF A BINARY DISTILLATION SYSTEM
1,')
1002 FORMAT(/T2,'FEED COMPONENTS: (1) ',5A6,/T19,'(2) ',5A6)
1003 FORMAT(/T2,'NO. OF FEED STREAMS:',I4,/T2,'NO. OF SIDE STREAMS:',
1I4)
1004 FORMAT(/T2,'NO. OF THEORETICAL STAGES REQUIRED:',F6.1)
1005 FORMAT(/T2,'MURPREE EFFICIENCY FACTOR USED:',F5.2)
1006 FORMAT(/T2,'FEED AND SIDE STREAM SPECIFICATIONS:',/T2,
1'STREAM',2X,'TRAY NO.',2X,'Q VALUE',2X,'FLOWRATE',2X,'COMPOSITION',
2/T2,'NO.',1X,'TYPE',T28,'KGMOLE/HR',3X,'MOL FRAC')
1007 FORMAT(/T2,'DISTILLATE WITHDRAWAL RATE:',F7.2,' KGMOLE/HR',/T2,
1'DISTILLATE COMPOSITION:',6X,F7.4)
1008 FORMAT(/T2,'BOTTOMS WITHDRAWAL RATE:',1X,F7.2,' KGMOLE/HR',/T2,
1'BOTTOMS COMPOSITION:',7X,F7.4)
1009 FORMAT(T2,'OPEN STEAM OPERATION.,./T2,'SATURATED STEAM REQUIREMENT
1S:',F7.2,' KGMOLE/HR')
1110 FORMAT(T2,I2,I2,1X,'SIDE',3X,I3,5X,F5.2,4X,F7.2,5X,F6.4)
1120 FORMAT(T2,I2,I2,1X,'FEED',3X,I3,5X,F5.2,4X,F7.2,5X,F6.4)
1201 FORMAT(/T2,'TOTAL CONDENSER OPERATION,')
1202 FORMAT(/T2,'PARTIAL CONDENSER OPERATION,')
1203 FORMAT(T2,'CONDENSER PRESSURE:',F7.2,1X,'KPA')
1204 FORMAT(T2,'FEED TEMPERATURE:',F7.2,1X,'DEG C',/T2,'FEED ENTHALPY',
1F11.2,1X,'KJ/KGMOLE')
1205 FORMAT(T2,'CONDENSER DUTY:',F11.2,1X,'KW')
1206 FORMAT(T2,'REBOILER DUTY:',8X,F10.2,1X,'KW')
1207 FORMAT(/T2,'REFLUX RATIO:',F11.4)
1208 FORMAT(T2,'MIN REFLUX RATIO:',F7.4)

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SUBROUTINE BINRES

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1310 FORMAT(/T2,'VAPOUR AND LIQUID RATES THROUGH COLUMN:','/T35,
1'VAPOUR','3X','LIQUID','3X','(KGMOLE/HR)')
1311 FORMAT(T2,'SECTION','12,' REBOILER - '.A4,' ('.11.').'.3X.F6.2.3X.
IF6.2)
1312 FORMAT(T2,'SECTION','12.2X.A4,' ('.11.').'.A4,' ('.11.').'.3X.F6.2
1'3X.F6.2)
1313 FORMAT(T2,'SECTION','12.2X.A4,' ('.11.').'.CONDENSER'.2X.F6.2.3X.
IF6.2)
1320 FORMAT(/T2,'COMPOSITION,TEMPERATURE AND FLOWRATE PROFILES:')
1321 FORMAT(T2,'TRAY','3X','COMPOSITION'.5X,'TEMPERATURE'.6X,'FLOWRATES'
1'/'T9,'(MOLE FRAC)'.7X,'(DEG C)'.7X,'(KGMOLE/HR)'.',/T9,
2'LIQ'.5X,'VAP'.6X,'BUBBLE PT'.6X,'LIQ'.5X,'VAP')
1322 FORMAT(T2,'13.2X.F6.4.2X.F6.4.6X.F7.2.4X.F7.2.1X.F7.2)
1323 FORMAT(T2,'FEED TRAY POSITION:')
1324 FORMAT(T2,A4.1X.F6.4.2X.F6.4.6X.F7.2.4X.F7.2.1X.F7.2)
1400 FORMAT(/T2,'DO YOU WANT OPERATING DIAGRAM DATA TABLES?')
1401 FORMAT(T5.F6.4.2X.F6.4)
1410 FORMAT(/T2,'EQUILIBRIUM DATA:','/T8,'X'.7X,'Y')
1411 FORMAT(/T2,'ENTHALPY DIAGRAM DEL PTS:','/T4,'DEL D=','E12.6.
11X,'KJ/KGMOLE','/T4,'DEL B=','E12.6.1X,'KJ/KGMOLE')
1420 FORMAT(/T2,'INTERSECTION COORDINATES OF OPERATING LINES:','/T8,
1'X'.7X,'Y')
1421 FORMAT(/T2,'POINTS ON OPERATING LINES:','/T8,'X'.7X,'Y')
1430 FORMAT(/T2,'COORDINATES OF MCCABE-THIELE STAGES:','/T8,'X'.7X,'Y')

C RESULT TABLE HEADING,COMPONENT NAMES AND GENERAL DATA.
C MCCABE-THIELE DESIGNS.
C IF(NOP.EQ.1) WRITE(NTP,1001)
C PONCHON-SAVARIT DESIGNS.
C IF(NOP.EQ.3) WRITE(NTP,1011)
WRITE(NTP,1002) ((TTITLES(I,J),J=1.5),I=1.2)
WRITE(NTP,1003) NF,NS
WRITE(NTP,1004) TRAYNO
WRITE(NTP,1005) EFF
WRITE(NTP,1006)

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SUBROUTINE BINRES

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C      FEED AND SIDE STREAM SPECIFICATIONS AND POSITIONS.
DO 10 I=2,NN
  I1=I-1
  IF(FLOW(I)) 110,10,120
  110 FLO=ABS(FLOW(I))
     WRITE(NTP,1110) I1,NTRAY(I-1),QFS(I),FLO,ZFS(I)
     GO TO 10
  120 WRITE(NTP,1120) I1,NTRAY(I-1),QFS(I),FLOW(I),ZFS(I)
  10 CONTINUE
C      FEED TEMPERATURE AND ENTHALPY FOR P-S DESIGNS.
  IF(NOP.EQ.3) WRITE(NTP,1204) TF,QF(1)
C      CONDENSER SPECIFICATIONS.
  IF(NCOND.EQ.1) WRITE(NTP,1201)
  IF(NCOND.EQ.2) WRITE(NTP,1202)
  PT1=PT/7.50061683
  WRITE(NTP,1203) PT1
  IF(NOP.EQ.3) WRITE(NTP,1205) QC
C      REFLUX CONDITIONS.
  WRITE(NTP,1207) R
  IF(NOP.EQ.3) WRITE(NTP,1208) RM
C      DISTILLATE PRODUCT DATA.
  WRITE(NTP,1007) DTOT,XD
C      BOTTOMS PRODUCT DATA, AND REBOILER SPECIFICATIONS.
  WRITE(NTP,1008) BTOT,XB
  IF(NOS.EQ.1) WRITE(NTP,1009) AV(1)
  IF(NOP.EQ.3) WRITE(NTP,1206) QB
  IF(NOP.EQ.3) GO TO 131
C      FLOWRATES IN EACH SECTION OF COLUMN FOR M-T DESIGNS.
  WRITE(NTP,1310)
  NI=1
  IF(FLOW(2).LT.0.0) STR='SIDE'
  IF(FLOW(2).GT.0.0) STR='FEED'
  WRITE(NTP,1311) NI,STR,NI,AV(1),AL(1)
  NNI=NN-1
  IF(NNI.LT.2) GO TO 130

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SUBROUTINE BINRES

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DO 11 I=2,NMI
NI=NI+1
NIM=NI-1
STR1=STR
IF(FLOW(I+1).LT.0.0) STR='SIDE'
IF(FLOW(I+1).GT.0.0) STR='FEED'
WRITE(NTP,1312) I,STR1,NIM,STR,NI,AV(I),AL(I)
11 CONTINUE
130 CONTINUE
I=NI+1
WRITE(NTP,1313) I,STR,NI,AV(NN),AL(NN)
GO TO 132
131 CONTINUE
C COMPOSITION,TEMPERATURE AND FLOWRATE PROFILES FOR P-S DESIGNS.
WRITE(NTP,1320)
WRITE(NTP,1321)
NTRA=(NTRAY(1)+1)*2
MI=-1
DO 30 I=1,NTRA,2
MI=MI+1
WRITE(NTP,1322) MI,X(I),Y(I),TRV(I),LRWT(I),VRWT(I)
30 CONTINUE
NTRA=NTRA+1
WRITE(NTP,1323)
DO 31 I=NTRA,NTC,2
MI=MI+1
IF(I.EQ.NTC) GO TO 133
WRITE(NTP,1322) MI,X(I),Y(I),TRV(I),LRWT(I),VRWT(I)
GO TO 31
133 WRITE(NTP,1324) CON,X(I),Y(I),TRV(I),LRWT(I),VRWT(I)
31 CONTINUE

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SUBROUTINE BINRES

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132 CONTINUE
C ENTHALPY_CONCENTRATION DIAGRAM *DEL* POINTS,
  IF(NOP.EQ.3) WRITE(NTP,1411) OELD*DELB
  WRITE(NTP,1400)
  READ(NCR,1) ANAME
  IF(ANAME.NE.*YES.*) GO TO 999
C EQUILIBRIUM CURVE DATA,
  WRITE(NTP,1410)
  WRITE(NTP,1401) (XEQUIL(I),YEQUIL(I),I=1,NDAT*5)
  IF(NOP.NE.3) GO TO 141
C OPERATING LINE DATA FOR PONCHON-SAVARIT DESIGNS.
  WRITE(NTP,1421)
  WRITE(NTP,1401) (XSOP(I),YSOP(I),I=1,MPTS*5)
  WRITE(NTP,1323)
  WRITE(NTP,1401) (XROP(I),YROP(I),I=1,NPTS*5)
141 CONTINUE
C OPERATING LINE POINTS OF INTERSECTION.
  WRITE(NTP,1420)
  WRITE(NTP,1401) (QX(I),QY(I),I=1,NN1)
C COORDINATES FOR CONSTRUCTING STAGES ON X-Y DIAGRAM.
  WRITE(NTP,1430)
  WRITE(NTP,1401) (X(I),Y(I),I=1,NTC)
999 CONTINUE
  RETURN
  END

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SUBROUTINE BINRES

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C

COMMON BLOCK 'PACCOM' FOR PACKED-COLUMN DESIGN PROGRAM.

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PARAMETER NPEQ=55
PARAMETER NPPS=150
PARAMETER NPCTC=55
PARAMETER NPEN=25
PARAMETER NPAK=225
DIMENSION ASURF(NPAK),CFAK(NPAK),FUNCR(NPAK),FUNCS(NPAK)
DIMENSION KX(NPAK),KY(NPAK),L(NPAK),SLOPK(NPAK)
DIMENSION V(NPAK),VYR(NPAK),VYS(NPAK),XI(NPAK),YI(NPAK)
DIMENSION ALFA(NPEQ),HLIQ(NPEN),HVAP(NPEN),LIQR(NPPS),LIQS(NPPS)
DIMENSION LRWT(NPPS),LSWT(NPPS),TALF(NPEQ)
DIMENSION TRL(NPPS),TRV(NPPS),TSL(NPPS),TSV(NPPS)
DIMENSION VAPR(NPPS),VAPS(NPPS),VRWT(NPPS),VSWT(NPPS)
DIMENSION X(NPTC),XALF(NPEQ),XEQUIL(NPEQ),XH(NPEN)
DIMENSION XROP(NPPS),XSOP(NPPS),YROP(NPPS),YSOP(NPPS)
DIMENSION Y(NPTC),YALF(NPEQ),YEQUIL(NPEQ),YH(NPEN)
COMMON A(2),ACP(2),AL(11),ALFA,ALFCON,AV(11)
COMMON B(2),BCP(2),BTOT,C(2),CCP(2),CFCT(12),CPCONV,CPMW(2)
COMMON DCP(2),DELB,DELD,DL,DTOT,DV
COMMON EFF,FLOW(10),FTOT,FTOTS(5)
COMMON HB,HD,HEVABS,HF,HLIQ,HSCONV,HSMW(2),HSOLN(30)
COMMON HVAP,LH(2)
COMMON LHCONV,LHMW(2),LIQR,LIQS,LRWT,LSWT
COMMON MALF,MW(2),NBDC,NCOND,NDATA,NENTH,NF,NHS,NINT
COMMON NN,NOP,NOS,NPPSC,NPTCC,NS,NTRAY(12)
COMMON NCR,NG1,NTP
COMMON PT,Q,QB,QC,QF(5),QS(5),QFS(10),QX(12),QY(12),R,RM
COMMON SLPE(12),STOTS(5),TALF,TBASE,TBP(2),TC(2)
COMMON TCONV,TF,TG,TITLES(2,5),TRAYNO,TRL,TRV
COMMON TSL,TSV,VAPR,VAPS,VRWT,VSWT
COMMON X,XALF,XB,XC,XCR,XD,XEQUIL,XH,XHS(30)
COMMON XROP,XSOP,YROP,YSOP
COMMON Y,YALF,YB,YC,YCR,YD,YEQUIL,YH
COMMON Z,ZF(5),ZF1(5),ZS(5),ZS1(5),ZFS(10)

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COMMON 'PACCOM'

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C MAINLINE PROGRAM FOR DESIGN OF PACKED COLUMNS FOR USE IN BINARY
C CONTINUOUS-DISTILLATION OPERATIONS.
C THE PACKAGE INCLUDES DATA FOR CERAMIC RASCHIG RINGS AND BERL SADDLES
C (SOURCE: SHULMAN FROM TREYBAL, MASS TRANSFER OPERATIONS, 2ND ED)
C SPECIFICATIONS FOR PACKINGS OTHER THAN THESE MUST BE SUPPLIED BY
C THE USER.
C *****
C INSERT COMMON BLOCK 'PACCOM' HERE.
C *****
C REAL KX,KY,L,LH,LHCONV,LHMW,LIQR,LIQS,LRWT,LSWT,MW
      1 FORMAT( )
      2 FORMAT(A3)
      3 FORMAT(SA6)
      DATA NCR,NGI,NTP/8,11,5/
      DATA NCODE,NINT,NOP/3,50,2/
      DATA MALF,MCAN/1,1/
      DATA ALBKGH,ASFSM,AMLCON,AMVCON/4,8824,0.3048,2,42,0.000242/
      DATA DLCON,DVCON,RHOCON,SIGCON/3,87504,3.87504,62,43,1.0/
      GC=4.18*(10.**8)
      NPPSC=NPPS-2
      NPTCC=NPTC
C START OF PROGRAM *****
      WRITE(NTP,5000)
5000 FORMAT(/T2,'ANSWER ALL QUESTIONS YES OR NO','/T2,'ALL INPUT IN FRE
      IE FORMAT('','/T2,'DO YOU KNOW HOW TO USE THIS PROGRAM?')
      READ(NCR,2) ANAME
      IF(ANAME.EQ.'YES') GO TO 500
      CALL TEACH(NCODE)
      GO TO 999
500 CONTINUE
      WRITE(NTP,5001)
5001 FORMAT(/T2,'DO YOU WANT TO ADD DATA ELEMENTS?')
      READ(NCR,2) CNAME
      IF(CNAME.EQ.'NO') GO TO 501

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MAINLINE PACKED

MCAN=2	0025
WRITE(NTP,5002)	0026
5002 FORMAT(/T2,'ADD YOUR FIRST DATA ELEMENT.')	0027
501 CONTINUE	0028
C 'READBD' READS DATA ELEMENT,	0029
CALL READBD(MCAH,NDAT)	0030
IF(MALF.GT.999) GO TO 999	*****
START OF COMPUTATION *****	0031
WRITE(NTP,1001)	0032
1001 FORMAT(/T2,'DATA ACCEPTED. START OF OPERATING DIAGRAM CALCS.')	
C 'PONSAB' CONSTRUCTS THE PONCHON-SAVARIT OPERATING DIAGRAM,	0033
CALL PONSAB(NPTS,MPTS,NFIN,MFIN)	0034
IF(MALF.GT.999) GO TO 999	0035
NTOT=NPTS+MPTS	0036
MPT1=MPTS+NFIN	
C	
C START OF PACKING SPECS *****	*****
C	
WRITE(NTP,2001)	0037
2001 FORMAT(/T2,'OPERATING DIAGRAM COMPLETE.'/T2,'START OF PACKING CAL	0038
ICULATIONS.')	0039
WRITE(NTP,5030)	0040
5030 FORMAT(/T2,'DO YOU WANT TO ADD A SECOND DATA ELEMENT?')	0041
READ(NCR,2) CNAME	0042
MCAN=2	0043
IF(CNAME.NE.'YES') MCAN=1	0044
IF(MCAN.NE.1) WRITE(NTP,5031)	0045
5031 FORMAT(/T2,'ADD YOUR SECOND DATA ELEMENT.')	0046
CALL PACTYP(MCAN,NTYPE,NPACK,NCR,NTP)	0047
IF(MCAN.GT.99) GO TO 999	0048
C 'PAKING' RETURNS DESIGN DATA CORRESPONDING TO SPECIFIED TYPE	
C OF PACKING. (EITHER RASCHIG RINGS OR BERL SADDLES)	
CALL PAKING(NTYPE,NPACK,CF,E,AMF1,AMF2,ANF1A,ANF1B,ANF2A,ANF2B,PF1	0049
1,PF2,DS,PHTW,PHSW,SWPOW,BETA)	0050
C END OF PACKING SPECS *****	*****

MAINLINE PACKED

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C      START OF PHYSICAL PROP SPECS *****
NRHOM=Q
IF(MCAN.EQ.1) WRITE(NTP,5032)
5032 FORMAT(/T2,'COMPONENT-DENSITY DATA MUST BE AVAILABLE AS DESCRIBED
      1 IN THE',/T2,'DISTILLATION PACKAGE MANUAL,SECTION 3.4',/T2,'IS SOLU
      2 TION-DENSITY DATA ALSO AVAILABLE?')
      READ(NCR,2) RENAME
      IF(RENAME.EQ.'NO') GO TO 531
      NRHOM=1
531 CONTINUE
      IF(MCAN.EQ.1) WRITE(NTP,5033)
5033 FORMAT(/T2,'COMPOUND CLASSES ARE:',/T2,'(1) ALCOHOLS AND WATER',/T
      12,'(2) HYDROCARBONS (SAT.,UNSAT. AND AROM.) AND ETHERS',/T2,'(3) O
      2 THER ORGANICS',/T2,'(4) INORGANIC COMPOUNDS',/T2,'ENTER CLASS NO.
      3 INTO WHICH THE LIGHT AND HEAVY COMPONENTS FALL',/T2,'EG. FOR WATER
      4 AND NITRIC ACID ENTER 1.4')
      READ(NCR,1) NLCAT,NHCAT
C
C      START OF PACKING HEIGHT CALCULATIONS *****
C      TO 302 CONTINUE, SEARCHES FOR POINT OF MAXIMUM LIQUID FLOW
C      IN COLUMN, RELEVANT FLOW AND TEMPERATURE DATA AT THIS POINT
C      ARE NOTED,
C      SUBSEQUENT CALCULATIONS ARE BASED ON THESE DATA.
      FLOR=0.0
      DO 31 N=NTOT,MPT1,-1
      FL01=LRWT(N)
      IF(FLOR.GT.FL01) GO TO 31
      NFLOR=N
      FLOR=FL01
31 CONTINUE

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IF(MFIN.GT.999) GO TO 301
FLOS=0.0
DO 32 N=MPTS,MFIN,-1
  FL02=LSWT(M)
  IF(FLOS.GT.FL02) GO TO 32
  NFLOS=M
  FLOS=FL02
32 CONTINUE
  IF(FLOR.GT.FLOS) GO TO 301
  FL0L=LSWT(NFLOS)
  FLOV=VSWT(NFLOS)
  FVAP=VAPS(NFLOS)
  TFL0L=TSL(NFLOS)
  TFLOV=TSV(NFLOS)
  XLIQ=XSOP(NFLOS)
  YVAP=YSOP(NFLOS)
  NPOINT=NFLOS
  GO TO 302
301 CONTINUE
  FL0L=LRWT(NFLOR)
  FLOV=VRWT(NFLOR)
  FVAP=VAPR(NFLOR)
  TFL0L=TRL(NFLOR)
  TFLOV=TRV(NFLOR)
  XLIQ=XR0P(NFLOR-MPTS)
  YVAP=YROP(NFLOR-MPTS)
  NPOINT=NFLOR
302 CONTINUE
C  DENSITY OF LIQUID PHASE AT TEMPERATURE *TFL0L*,
  NPHYS=1
  CALL RHOLIQ(NPHYS,NRHOM,NLCAT,NHCAT,TFL0L,XLIQ,RHOL1,RHOL2,RHOL)
  RHOL=RHOL*RHOCON
  RHOV=FLOV*492./((FVAP*359.+(459.4+(TFLOV*1.8+32.)))
  RHOFAC=(RH0V/RHOL)**0.5

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C ABCIS IS THE X-AXIS COORDINATE ON THE U.S. STONEWARE FLOODING  
C AND LOADING CURVE DIAGRAM.

ABCIS=FL0L\*RHO\*FAC/FL0V  
WRITE(NTP,4401)

4401 FORMAT(/T2,'DO YOU WANT TO SPECIFY THE COLUMN DIAMETER?')

READ(NCR,2) DNAME

IF(DNAME.NE.'YES') GO TO 441

WRITE(NTP,4411)

4411 FORMAT(T2,'ENTER COLUMN DIAMETER IN METRES.')

READ(NCR,1) DIAM

IF(DIAM.GT.2) GO TO 902

444 CONTINUE

DM1=DIAM

VPDL=-1.

AREA=0.7854\*(DIAM\*\*2.)

VTO=FL0V/AREA

ALTO=FL0L/AREA

VLOAD=1.05\*VTO

GO TO 442

441 CONTINUE

WRITE(NTP,5040)

5040 FORMAT(/T2,'SPECIFY VAPOUR PRESSURE DROP AT LOADING FROM FOLLOWING

1:./T2,(1) 0.041,(2) 0.082,(3) 0.204,(4) 0.408,(5) 0.816,(6) 1.22

24 (KPA/M)../T2,'NOTE: 0.408 (CODE 4) REPRESENTS APPROX. 60% OF FLO

3ODING VALUE.')

WRITE(NTP,5041)

5041 FORMAT(T2,'ENTER APPROPRIATE CODE NO.,')

READ(NCR,1) NFLOOD

NPHYS=1

\*GO TO\* LOOP 410 SEARCHES FOR ACCEPTABLE VAPOUR MASS-FLOWRATE

C AND COLUMN DIAMETER.

C 410 CONTINUE

IF(NFLOOD.EQ.1.OR.NFLOOD.EQ.2) VPDL=NFLOOD\*0.041

IF(NFLOOD.EQ.3.OR.NFLOOD.EQ.4) VPDL=(NFLOOD-2)\*0.204

IF(NFLOOD.EQ.5.OR.NFLOOD.EQ.6) VPDL=(NFLOOD-3)\*0.408

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C	•FDATA, FITS AN N,TH ORDER POLYNOMIAL TO THE LOADING CURVE	0138
C	CORRESPONDING TO THE SELECTED PRESSURE DROP SPECIFICATIONS,	0139
C	Y-AXIS COORDINATE (ORD) CORRESPONDING TO X-AXIS COORDINATE	0140
C	(ABCSIS) IS RETURNED.	0141
	CALL FDATA(NFLOOD,ABCSIS,ORD)	0142
	WRITE(NTP,4001) ABCIS,ORD	
4001	FORMAT(/T2,•OPERATING POINT ON SELECTED LOADING CURVE:••/T2,	0143
	1,•ABSCISSA=•,F9.3./T2,•ORDINATE=•,F9.3)	0144
442	CONTINUE	0145
C	LIQUID VISCOSITY CALCULATION.	0146
	CALL VISCL(NPHYS,NLCAT,NHCAT,TFLOL,XLIQ,AMU1,AMU2,AMUML)	0147
C	DENSITY OF WATER AT TEMPERATURE •TFLOL•,	0148
	CALL RHOH2O(NPHYS,TFLOL,RHOW)	0149
	RHOW=RHOW*RHOCON	0150
	IF(VPDL.GT.0.0) GO TO 443	0151
	VDASH=VLOAD/ALBKGH	0152
	ORD=((VDASH**2.)*CF*(AMUML**0.2)*((RHOW/RHOL))/((GC*RHOL*RHOV)	0153
	NPHYS=2	0154
	WRITE(NTP,4003) ABCIS,ORD	0155
4003	FORMAT(/T2,•OPERATING POINT COORDINATES:••/T2,•ABSCISSA=•,F9.3,	0156
	1/T2,•ORDINATE=•,F9.3./T2,•IS THIS POINT WITHIN LOADING REGION?•)	0157
	READ(NCR,2) PNAME	0158
	IF(PNAME.EQ.‘YES’) GO TO 400	0159
	WRITE(NTP,4402)	0160
4402	FORMAT(T2,•ENTER ANOTHER DIAMETER.‘)	
	READ(NCR,1) DIAM	
	IF(DIAM.GT.2) GO TO 902	
	GO TO 444	
443	CONTINUE	
C	CALCULATION OF VAPOUR MASS-FLOWRATE AT LOADING,	0161
	VDASH=((ORD*GC*RHOV*RHOL)/(CF*(AMUML**0.2)*((RHOW/RHOL)))*•0.5	0162
	VLOAD=VDASH*ALBKGH	

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DDIAM=0.01
DIAM=2.0
C SEARCH FOR DIAMETER THAT WILL RESULT IN A VAPOUR FLOWRATE
C NOT GREATER THAN THE CALCULATED LOADING VALUE.
DO 40 K=1,200
DIAM=DIAM-DDIAM
CSAREA=0.7854*(DIAM**2.)
VTEST=FLOV/CSAREA
VDIFF=VLOAD-VTEST
IF(K.EQ.1.AND.VDIFF.LE.0.0001) GO TO 902
IF(VDIFF.LE.0.0001) GO TO 400
DMI=DIAM
AREA=CSAREA
VTO=VTEST
ALTD=FL0L/AREA
40 CONTINUE
400 CONTINUE
C PRINTOUT OF RECOMMENDED TOWER DIAMETER,AND RESULTING FLOWRATES.
WRITE(NTP,4002) VLOAD,DMI,VTO,ALTD
4002 FORMAT(/T2,'FOR GIVEN VAPOUR PRESSUR-DROP:',/T2,'SUPERFICIAL GAS V
IELOCITY=',F8.2,X,'KG/HR.SQM',/T2,'RECOMMENDED TOWER DIAM.',F9.2,
22X,'M',/T2,'RESULTANT SUP. GAS VEL.',F9.2,X,'KG/HR.SQM',/T2,'RES
3ULTANT SUP. LIQ VEL.',F9.2,X,'KG/HR.SQM',/T2,'ARE THESE FIGURES
4SATISFACTORY?')
READ(NCR,2) CHNAME
IF(CHNAME.EQ.'YES') GO TO 401
IF ABOVE NOT ACCEPTABLE,REQUEST NEW P.O. AND REPEAT CALCS.
WRITE(NTP,5042)
5042 FORMAT(/T2,'SPECIFY VAPOUR PRESSURE DROP AT LOADING FROM FOLLOWING
1:',/T2,'(1) 0.041,(2) 0.082,(3) 0.204,(4) 0.408,(5) 0.816,(6) 1.22
24 (KPA/M)')
READ(NCR,1) NFLOOD
NPHYS=2
GO TO 410
401 CONTINUE

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C	COLUMN DIAMETER AND MASS FLOWS/UNIT AREA ARE NOW ESTABLISHED.	0194
C	MASS TRANSFER COEFFICIENTS ARE NOW CALCULATED.	
C	NPHYS=1	
C	LOOP 41 CALCULATES THE MASS TRANSFER COEFFICIENTS IN THE STRIPPING	
C	SECTION OF THE COLUMN USING SHULMAN'S PROCEDURE.	
C	EQUATIONS AND LIMITATIONS USED HERE WERE FIRST PROPOSED BY	
C	SHULMAN. A DETAILED DISCUSSION CAN BE FOUND IN TREYBAL, R.E.,	
C	*MASS TRANSFER OPERATIONS, 2ND ED., CHAPTER 6, OR IN THE	
C	DISTILLATION PACKAGE MANUAL, SECTION 3.	
	IF(MFIN.GT.999) MFIN=NPTS	0195
	DO 41 M=MPTS, MFIN, -1	0196
	VT1=VSWT(M)/(AREA*ALBKGH)	0197
	ALT1=LSWT(M)/(AREA*ALBKGH)	0198
	TEMPV=TSV(M)	0199
	TEMPL=TSL(M)	0200
	XLIQ=XSOP(M)	0201
	NPH=2	0202
	CALL RHOLIQ(NPH, NRHOM, NLCAT, NHCAT, TEMPL, XLIQ, RHOL1, RHOL2, RHOL)	0203
	RHOL=RHOL*RHOCON	0204
	RHOV=VSWT(M)*492./(VAPS(M)*359.*(459.4+(TEMPV*1.8+32.)))	0205
	CALL VISCL(NPH, NLCAT, NHCAT, TEMPL, XLIQ, AMU1, AMU2, AMUML)	0206
	IF(ALT1.GT.1500.) GO TO 411	0207
	AM=AMF1	0208
	ANFUNC=ANF1A*(1./((10.**4)))*ALT1+ANF1B	0209
	P=PF1	0210
	GO TO 412	0211
411	CONTINUE	0212
	AM=AMF2	0213
	ANFUNC=ANF2A*(1./((10.**4)))*ALT1+ANF2B	0214
	P=PF2	0215
412	CONTINUE	0216
	IF(ABS(RHOV-0.075).LT.0.002) GO TO 413	0217
	VT1=VT1*((0.075/RHOV)**0.5)	0218
413	CONTINUE	0219
	AAW=AM*(VT1**ANFUNC)*(ALT1**P)	0220

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C
XST=XSOP(M)
SURFACE TENSION CALCULATION.
CALL SURTEN(NPHYS,NLCAT,NHCAT,TEMPL,XST,SIGMA)
SIGMA=SIGMA*SIGCON
IF(NTYPE.EQ.2) GO TO 414
PFACT1=6.85*(1./(10.**5))*(AMUML**0.02)*(SIGMA**0.99)
PFACT2=(DS**1.21)*(RHOL**0.37)
GO TO 415
414 CONTINUE
PFACT1=1.641*(1./(10.**4))*(AMUML**0.04)*(SIGMA**0.55)
PFACT2=(DS**1.56)*(RHOL**0.37)
415 CONTINUE
PHIS=PFACT1/PFACT2
PHISW=PHSW/(DS**SWPOW)
PHITW=PHITW*(1./(10.**5))*(ALTI**BETA)/(DS**2.)
IF(NTYPE.EQ.2) GO TO 417
AFACT2=(RHOL**0.84)*(0.1183*(ALTI**0.43)-1.)
AFACT3=(SIGMA/73.)*(0.925-0.262*ALOG10(ALTI))
IF(AMUML.GT.12.) GO TO 416
AFACT1=0.897*(ALTI**0.57)*(AMUML**0.13)
GO TO 419
416 CONTINUE
AFACT1=0.575*(ALTI**0.57)*(AMUML**0.31)
GO TO 419
417 CONTINUE
AFACT2=(RHOL**0.84)*(0.212*(ALTI**0.413)-1.)
AFACT3=(SIGMA/73.)*(1.033-0.262*ALOG10(ALTI))
IF(AMUML.GT.20.) GO TO 418
AFACT1=1.291*(ALTI**0.57)*(AMUML**0.13)
GO TO 419
418 CONTINUE
AFACT1=0.752*(ALTI**0.57)*(AMUML**0.31)
419 CONTINUE
HFUNC=AFACT1*AFACT3/AFACT2
PHIOW=PHITW-PHISW
PHIO=PHIOW*HFUNC
PHIT=PHIO+PHIS
AA=AAW*PHIO/PHIOW
EO=E-PHIT

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C      VAPOUR DIFFUSIVITY CALCULATION.
      CALL DIFVAP(NPHYS,PT,TEMPV,DIFUSV)
      DIFUSV=DIFUSV*DVCON
      YMU=YSOP(N)
      0259
      0260
      0261

C      VAPOUR VISCOSITY CALCULATION.
      CALL VISCV(NPHYS,TEMPV,YMU,AMUMV)
      AMUMV=AMUMV*AMVCON
      SCHV=AMUMV/(RHOV*DIFUSV)
      0262
      0263
      0264
      VMOLE=VAPS(M)/AREA
      FA1=((AMUMV*(1.-EO))/(DS*VT1))*0.36)
      0265
      0266
      FA2=((VMOLE*1.195/(SCHV**2./3.)))
      0267
      0268
      FV=FA1*FA2
      VAPOUR MASS TRANSFER COEFFICIENT.
      KY(M)=FV
      0269
      0270

C      XDIF=XSOP(N)
      LIQUID DIFFUSIVITY CALCULATION.
      CALL DIFLIQ(NPHYS,NLCAT,NHCAT,TEMPL,XDIF,DIFUSL)
      DIFUSL=DIFUSL*DLCON
      0271
      0272
      XRH0=XSOP(M)
      0273
      0274
      CALL RHOLIQ(NPH,NRHOM,NLCAT,NHCAT,TEMPL,XRH0,RHOL1,RHOL2,RHOL)
      RHOL=RHOL*RHOCOM
      0275
      0276
      AMUML=AMUML*AMLCON
      0277
      SCHL=AMUML/(RHOL*DIFUSL)
      0278
      FA1=((DS*ALT1/AMUML))*0.45
      0279
      FA2=((DIFUSL/DS)*25.1*(SCHL*0.5)
      NR=0
      0280
      CALL RHOLIQ(NPH,NR,NLCAT,NHCAT,TEMPL,XRH0,RHOL1,RHOL2,RHOL)
      CRH0=RHOL2*RHOCOM/NW(2)
      0281
      0282
      FL=FA1*FA2*CRH0
      0283
      LIQUID MASS TRANSFER COEFFICIENT.
      KX(M)=FL*ALBKGH
      0284
      L(M)=LIQS(M)/AREA
      0285
      V(M)=VMOLE
      0286
      ASURF(M)=AA/ASFMS
      0287
      NPHYS=2
      0288
      0289
      0290

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41 CONTINUE
      IF(MFIN.EQ.MPTS) HFIN=1000

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IF(NTYPE.EQ.2) GO TO 424
PFACT1=6.85*(1./((10.**5)))*(AMUML**0.02)*(SIGMA**0.99)
PFACT2=(DS**1.21)*(RHOL**0.37)
GO TO 425
424 CONTINUE
PFACT1=1.641*(1./((10.**4)))*(AMUML**0.04)*(SIGMA**0.55)
PFACT2=(DS**1.56)*(RHOL**0.37)
425 CONTINUE
PHIS=PFACT1/PFACT2
PHISW=PHSW/(DS**SWPOW)
PHITW=PHITW*(1./((10.**5)))*(ALTI**BETA)/(DS**2.)
IF(NTYPE.EQ.2) GO TO 427
AFACT2=(RHOL**0.84)*(0.1183*(ALTI**0.43)-1.)
AFACT3=(SIGMA/73.)*(0.925-0.262*ALOG10(ALTI))
IF(AMUML.GT.12.) GO TO 426
AFACT1=0.897*(ALTI**0.57)*(AMUML**0.13)
GO TO 429
426 CONTINUE
AFACT1=0.575*(ALTI**0.57)*(AMUML**0.31)
GO TO 429
427 CONTINUE
AFACT2=(RHOL**0.84)*(0.212*(ALTI**0.413)-1.)
AFACT3=(SIGMA/73.)*(1.033-0.262*ALOG10(ALTI))
IF(AMUML.GT.20.) GO TO 428
AFACT1=1.291*(ALTI**0.57)*(AMUML**0.13)
GO TO 429
428 CONTINUE
AFACT1=0.752*(ALTI**0.57)*(AMUML**0.31)
GO TO 429
429 CONTINUE
HFUNC=AFACT1*AFACT3/AFACT2
PHIOW=PHITW-PHISW
PHIO=PHIOW*HFUNC
PHIT=PHIO+PHIS
AA=AAW*PHIO/PHIOW
EO=E-PHIT

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CALL DIFVAP(NPHYS,PT,TEMPV,DIFUSV)
DIFUSV=DIFUSV*DVCON
YMU=YROP(NRS)
CALL VISC(NPHYS,TEMPV,YMU,AMUMV)
AMUMV=AMUMV*AMVCON
SCHV=AMUMV/(RHOV*DIFUSV)
VNOLE=VAPR(N)/AREA
FA1=((AMUMV*(1.-EQ)/(DS*VT1))*0.36)
FA2=(VMOLE*1.195/(SCHV**(2./3.)))
FV=FA1*FA2
KY(N)=FV
XDIF=XROP(NRS)
CALL DIFLIQ(NPHYS,NLCAT,NHCAT,TEMPL,XDIF,DIFUSL)
DIFUSL=DIFUSL*DLCON
XRHO=XROP(NRS)
CALL RHOLIQ(NPHYS,NRHOM,NLCAT,NHCAT,TEMPL,XRHO,RHOL1,RHOL2,RHOL)
RHOL=RHOL*RHOCON
AMUML=AMUML*ANLCON
SCHL=AMUML/(RHOL*DIFUSL)
FA1=(DS*ALTI/AMUML)*0.45
FA2=(DIFUSL/DS)*25.1*(SCHL**0.5)
NR=0
CALL RHOLIQ(NPHYS,NR,NLCAT,NHCAT,TEMPL,XRHO,RHOL1,RHOL2,RHOL)
CRHO=RHOL2*RHOCON/NW(2)
FL=FA1*FA2*CRHO
KX(N)=FL*ALBKGH
L(N)=LIQR(N)/AREA
V(N)=VMOLE
ASURF(N)=AA/ASFMS
42 CONTINUE

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PREPARATION FOR HEIGHT-OF-PACKING INTEGRATION.

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DO 43 N=NTOT,MPT1,-1
NRS=N-MPTS
SLOPK(N)=-KX(H)/KY(N)
CFAK(N)=YROP(NRS)-SLOPK(N)*XROP(NRS)
43 CONTINUE
IF(MFIN.GT.999) GO TO 403
DO 44 M=MPTS,MFIN,-1
SLOPK(M)=-KX(M)/KY(M)
CFAK(M)=YSOP(M)-SLOPK(M)*XSOP(M)
44 CONTINUE
403 CONTINUE
DO 45 N=NTOT,MPT1,-1
NRS=N-MPTS
MN=1
DELX=0.001
XG=XROP(NRS)-DELX
SL=SLOPK(N)
CFAC=CFAK(N)
451 CONTINUE
CALL XYALFA(ALFA,ALFCON,ALF,MALF,NDATA,XALF,XG)
YG=ALF*XG/(1.+XG*(ALF-1.))
YTEST=SL*XG+CFAC
IF(ABS(YG-YTEST).LT.0.0001) GO TO 452
NCALL=1
CALL SEARCH(YG,YTEST,XG,DELX,MN,NCALL)
GO TO 451
452 CONTINUE
IF(XG.GE.XROP(NRS).OR.YG.LE.YROP(NRS)) GO TO 901
XI(N)=XG
YI(N)=YG
45 CONTINUE
IF(MFIN.GT.999) GO TO 404

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DO 46 N=MPPTS,MFIN,-1	0417
MN=1	0418
DELX=0.001	0419
XG=XSOP(M)-DELX	0420
SL=SLOPK(M)	0421
CFAC=CFAC(M)	0422
461 CONTINUE	0423
CALL XYALFA(ALFA,ALFCON,ALF,MALF,NDATA,XALF,XG)	0424
YG=ALF*XG/(1.+XG*(ALF-1.))	0425
YTEST=SL*XG+CFAC	0426
IF(ABS(YG-YTEST).LT.0.0001) GO TO 462	0427
NCALL=1	0428
CALL SEARCH(YG,YTEST,XG,DELX,MN,NCALL)	0429
GO TO 461	0430
462 CONTINUE	0431
IF(XG.GE.XSOP(M).OR.YG.LE.YSOP(M)) GO TO 901	0432
XI(M)=XG	0433
YI(M)=YG	0434
46 CONTINUE	0435
404 CONTINUE	0436
C	
C CALCULATION OF VALUES OF THE FUNCTION TO BE INTEGRATED.	
C (RECTIFYING SECTION)	
C IF(KX(NPOINT).LT.KY(NPOINT)) GO TO 611	0437
C *DO* LOOP 61: PRINCIPAL MASS TRANSFER RESISTANCE IN VAPOUR PHASE.	
NSORT=1	
DO 61 N=MPPTS,MFIN,-1	0438
MRS=N+MPPTS	0439
VYR(N)=V(NRS)*YROP(N)	0440
FUNCR(N)=1./(KY(NRS)*ASURF(NRS)*(YI(NRS)-YROP(N)))	0441
61 CONTINUE	0442
GO TO 612	0443
611 CONTINUE	0444
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C      'DO' LOOP 62: PRINCIPAL MASS TRANSFER RESISTANCE IN LIQUID PHASE.
      NSORT=2
      DO 62 N=NPTS,NFIN,-1
      NRS=N+MPTS
      VYR(N)=L(NRS)*XROP(N)
      FUNCN(N)=1./(KX(NRS)*ASURF(NRS)*(XROP(N)-XI(NRS)))
      62 CONTINUE
      612 CONTINUE
      TOP=VYR(NPTS)
      FEED=VYR(1)
      NPTI=NPTS-NFIN+1
      INTEGRATION OVER RECTIFYING SECTION.
      CALL SIMPS(TOP,FEED,NPTI,FUNCR,SUM)
      C      HEIGHT OF PACKING FOR RECTIFYING SECTION (M).
      ZRECT=SUM
      IF(MFIN.GT.999) GO TO 699
      C      CALCULATION OF VALUES OF THE FUNCTION TO BE INTEGRATED.
      C      (STRIPPING SECTION)
      IF(NSORT.EQ.2) GO TO 613
      'DO' LOOP 63: PRINCIPAL MASS TRANSFER RESISTANCE IN VAPOUR PHASE.
      DO 63 M=MPTS,NFIN,-1
      VYS(M)=V(M)*YSOP(M)
      FUNCS(M)=1./(KY(M)*ASURF(M)*(YI(M)-YSOP(M)))
      63 CONTINUE
      GO TO 614
      613 CONTINUE
      C      'DO' LOOP 64: PRINCIPAL MASS TRANSFER RESISTANCE IN LIQUID PHASE.
      DO 64 M=MPTS,NFIN,-1
      VYS(M)=L(M)*XSOP(M)
      FUNCS(M)=1./(KX(M)*ASURF(M)*(XSOP(M)-XI(M)))
      64 CONTINUE
      614 CONTINUE
      FEED=VYS(MPTS)
      BOT=VYS(MFIN)
      C      INTEGRATION OVER STRIPPING SECTION.
      MPTI=MPTS-MFIN+1
      CALL SIMPS(FEED,BOT,MPTI,FUNCS,SUM)

```

MAINLINE PACKED

```

C      HEIGHT OF PACKING FOR STRIPPING SECTION (M).
ZSTRIP=SUM
699  CONTINUE
      CALL PACRES(NDAT,NPTS,MPTS,NFIN,MFIN,NTYPE,NPACK,DMI,ZRECT,ZSTRIP,
1VPDL,ASURF,FUNCR,FUNCS,L,V,VYR,VYS,XI,YI,KX,KY,NSORT)
      GO TO 999
C      ERROR MESSAGES *****
901  CONTINUE
      WRITE(NTP,9001) XG,YG
9001  FORMAT(/T2,'PINCH POINT OCCURS AT POINT',F7.4,'F6.4,' ON THE',
1/T2,'EQUILIBRIUM CURVE',/T2,'THE REFLUX RATIO OR THE DISTILLATE A
2ND BOTTOMS COMPOSITIONS',/T2,'SHOULD BE ADJUSTED ACCORDINGLY,')
      WRITE(NTP,9011)
9011  FORMAT(/T2,'EQUILIRRIUM AND OPERATING LINE DATA WILL BE PRINTED OU
1T',/T2,'THIS SHOULD BE PLOTTED ON AN X-Y DIAGRAM,AND THE SYSTEM',
2/T2,'CHECKED FOR AZEOTROPISM,')
9021  FORMAT(T5,F6.4,2X,F6.4)
9031  FOMAT(/T2,'EQUILIBRIUM DATA:',/T8,'X',7X,'Y,')
9041  FORMAT(/T2,'POINTS ON OPERATING LINES:',/T8,'X',7X,'Y,')
9051  FORMAT(T2,'POINT OF INTERSECTION OF OPERATING LINES:',F7.4,
12X,F6.4)
      WRITE(NTP,9031)
      WRITE(NTP,9021) (XEQUIL(K),YEQUIL(K),K=1,NDAT,5)
      WRITE(NTP,9041)
      WRITE(NTP,9021) (XSOP(M),YSOP(M),M=1,MPTS,5)
      WRITE(NTP,9051) XCR,YCR
      WRITE(NTP,9021) (XROP(N),YROP(N),N=1,NPTS,5)
      GO TO 999
902  CONTINUE
      WRITE(NTP,9002)
9002  FORMAT(/T2,'DIAMETER OF THE COLUMN EXCEEDS 2 METRES,')
1/T2,'SYSTEM UNSUITABLE FOR PACKED COLUMN DISTILLATION,')
999  CONTINUE
      WRITE(NTP,9999)
9999  FORMAT(/T2,'END OF PROGRAM,')
      END

```

MAINLINE PACKED

```

SUBROUTINE PACRES(NDAT,NPTS,MPTS,NFIN,MFIN,NTYPE,NPACK,DIAMM,ZRECT      0001
1,ZSTRIP,VPDL,ASURF,FUR,FUS,L,V,VYR,VYS,XI,YI,KX,KY,NSORT)          0002
ROUTINE FOR PRINTING RESULTS FOR DESIGN OF PACKED COLUMNS.
C *****
C INSERT COMMON BLOCK 'BINCOM' HERE.
C *****
C DIMENSION ASURF(1),FUR(1),FUS(1),KX(1),KY(1),L(1)
C DIMENSION V(1),VYR(1),VYS(1),XI(1),YI(1)
C REAL KX,KY,L,LH,LHCONV,LHMW,LIQR,LIQS,LRWT,LSWT,MW
C NTOT=NPTS+MPTS
C MPT1=MPTS+NFIN
1 FORMAT(A3)
1001 FORMAT(/T2,'DESIGN OF A PACKED DISTILLATION COLUMN',,/T2,
1'MASS TRANSFER COEFFICIENT CALCULATIONS BY SHULMANS METHOD',)
1002 FORMAT(/T2,'FEED COMPONENTS: (1) ',5A6,/T19,'(2) ',5A6)
1003 FORMAT(/T2,'FEED STREAM SPECIFICATIONS:',/T4,'FLOWRATE:',F10.2,3X,
1'KGMOLE/HR',/T4,'COMPOSITION:',F9.4,1X,'MOLE FRAC',/T4,'QVALUE:',
2F12.2,/T4,'TEMPERATURE:',F7.2,3X,'DEG C',/T4,'ENTHALPY:',F10.2,
33X,'KJ/KGMOLE')
1007 FORMAT(/T2,'DISTILLATE WITHDRAWAL RATE:',F7.2,' KGMOLE/HR',/T2,
1'DISTILLATE COMPOSITION:',F13.4)
1008 FORMAT(/T2,'BOTTOMS WITHDRAWAL RATE:',F7.2,' KGMOLE/HR',/T2,
1'BOTTOMS COMPOSITION:',F13.4)
1011 FORMAT(/T2,'SIZE AND TYPE OF PACKING USED:',F6.1,1X,'MM',
13X,'RASCHIG RINGS')
1012 FORMAT(/T2,'SIZE AND TYPE OF PACKING USED:',F6.1,1X,'MM',
13X,'BERL SADDLES')
1201 FORMAT(/T2,'TOTAL CONDENSER OPERATION',)
1202 FORMAT(/T2,'PARTIAL CONDENSER OPERATION',)
1203 FORMAT(T2,'CONDENSER PRESSURE:',F9.2,1X,'KPA')
1205 FORMAT(T2,'CONDENSER DUTY:',F13.2,1X,'KW')
1206 FORMAT(T2,'REBOILER DUTY:',F17.2,1X,'KW')
1207 FORMAT(/T2,'REFLUX RATIO:',F11.4)
1208 FORMAT(T2,'MIN REFLUX RATIO:',F7.4)

```

SUBROUTINE PACRES

```

2010 FORMAT(/T2,'VAPOUR PRESSURE-DROP AT LOADING:',F6.3,1X,'KPA/M')
2011 FORMAT(/T2,'THE COLUMN DIAMETER IS BASED ON CONDITIONS AT THE POIN
IT IN',/T2,'THE COLUMN WITH THE GREATEST LIQUID FLOWRATE.',
2/T2,'(SEE TABLE BELOW)')
2012 FORMAT(/T2,'RECOMMENDED COLUMN DIAMETER:',F6.3,1X,'M')
2013 FORMAT(/T2,'HEIGHT OF PACKING REQUIRED:',/T4,
1'RECTIFYING SECTION:',F7.2,1X,'M',/T4,
2'STRIPPING SECTION:',F8.2,1X,'M')
3000 FORMAT(/T2,'DO YOU WANT COMPOSITION,TEMPERATURE AND FLOWRATE DATA
TABLES?')
3001 FORMAT(/T2,'COMPOSITION,TEMPERATURE AND FLOWRATE PROFILES:')
3010 FORMAT(T4,'COMPOSITION',5X,'TEMPERATURE',6X,'FLOWRATES',
1/T4,'(MOLE FRAC)',7X,'(DEG C)',7X,'(KGMOLE/HR)',/T4,
2'LIQ',5X,'VAP',6X,'BUBBLE PT',6X,'LIQ',5X,'VAP')
3011 FORMAT(T2,F6.4,2X,F6.4,6X,F7.2,4X,F7.2,1X,F7.2)
3012 FORMAT(T2,'FEED POSITION:')
3013 FORMAT(/T2,'SHULMAN MASS TRANSFER CALCULATIONS:',/T4,
1'X =LIQUID MOLE FRACTION OF LIGHT COMPONENT',/T4,
2'V =VAPOUR RATE IN COLUMN (KGMOLE/HR,SQ M)',/T4,
3'A =INTERFACIAL SURFACE AREA OF PACKED SECTION (SQ M/CU M)',/T4,
4'KY=VAPOUR MASS TRANSFER COEFFICIENT (KGMOLE/HR,SQ M*MOLE FRAC)',
5/T4,'KX=LIQUID " " / " ')
3113 FORMAT(/T2,'SHULMAN MASS TRANSFER CALCULATIONS:',/T4,
1'X =LIQUID MOLE FRACTION OF LIGHT COMPONENT',/T4,
2'L =LIQUID RATE IN COLUMN (KGMOLE/HR,SQ M)',/T4,
3'A =INTERFACIAL SURFACE AREA OF PACKED SECTION (SQ M/CU M)',/T4,
4'KY=VAPOUR MASS TRANSFER COEFFICIENT (KGMOLE/HR,SQ M*MOLE FRAC)',
5/T4,'KX=LIQUID " " / " ')
3014 FORMAT(/T4,'X',9X,'V',7X,'A',7X,'KY',7X,'KX')
3114 FORMAT(/T4,'X',9X,'L',7X,'A',7X,'KY',7X,'KX')
3015 FORMAT(T2,F6.4,F10.3,F8.3,F9.4,F9.4)

```

SUBROUTINE PACRES

```

3016 FORMAT(/T2,'DATA FOR NUMERICAL INTEGRATION TO FIND HEIGHT OF PACK1
      1NG:.'/T4,'INTEGRATION OF: D(Z)=D(V,Y)/KY.A.(YI-Y)'/T4,
      2'Y =VAPOUR MOLE FRACTION OF LIGHT COMPONENT.'/T4,
      3'YI=POINT ON EQUIL CURVE CORRESPONDING TO PT Y ON OP. LINE')
3116 FORMAT(/T2,'DATA FOR NUMERICAL INTEGRATION TO FIND HEIGHT OF PACK1
      1NG:.'/T4,'INTEGRATION OF: D(Z)=D(L,X)/KX.A.(X-XI)'/T4,
      2'X =LIQUID MOLE FRACTION OF LIGHT COMPONENT.'/T4,
      3'XI=POINT ON EQUIL CURVE CORRESPONDING TO PT X ON OP. LINE')
3017 FORMAT(/T4,'Y'.8X,'YI'.5X,'I./KY.A.(YI-Y)'.5X,'V.Y')
3117 FORMAT(/T4,'X'.8X,'XI'.5X,'I./KX.A.(X-XI)'.5X,'L.X')
3018 FORMAT(T2,F6.4,3X,F6.4,6X,F7.3,7X,F7.3)
3020 FORMAT(/T2,'DO YOU WANT MASS-TRANSFER DATA TABLES?')
3021 FORMAT(/T2,'PRINCIPAL MASS TRANSFER RESISTANCE IN VAPOUR PHASE,')
3022 FORMAT(/T2,'PRINCIPAL MASS TRANSFER RESISTANCE IN LIQUID PHASE,')
      C RESULT TABLE HEADING,COMPONENT NAMES AND FEED STREAM DATA.
      WRITE(NTP,1001)
      WRITE(NTP,1002) ((TTILES(I,J),J=1,5),I=1,2)
      WRITE(NTP,1003) FTOT,Z,Q,TF,QF(1)
      C CONDENSER SPECIFICATIONS.
      IF(NCOND.EQ.1) WRITE(NTP,1201)
      IF(NCOND.EQ.2) WRITE(NTP,1202)
      WRITE(NTP,1205) QC
      PTK=PT/7.50061683
      WRITE(NTP,1203) PTK
      C REFLUX CONDITIONS,AND DISTILLATE PRODUCT DATA.
      WRITE(NTP,1207) R
      WRITE(NTP,1208) RM
      WRITE(NTP,1007) DTOT,XD
      C BOTTOMS PRODUCT DATA,AND REBOILER SPECIFICATIONS.
      WRITE(NTP,1008) BTOT,XB
      WRITE(NTP,1206) QB
      C COLUMN SPECIFICATIONS.
      WRITE(NTP,2010) VPDL
      WRITE(NTP,2011)
      WRITE(NTP,2012) DIAMM
      IF(MFIN.GT.999) ZSTRIP=0.0
      WRITE(NTP,2013) ZRECT,ZSTRIP

```

SUBROUTINE PACRES

C	SIZE=NPack*12.7	0094
	PACKING SPECIFICATIONS.	
	IF(NTYPE.EQ.1) WRITE(NTP,1011) SIZE	0095
	IF(NTYPE.EQ.2) WRITE(NTP,1012) SIZE	0096
C	COMPOSITION,TEMPERATURE AND FLOWRATE PROFILES.	
	WRITE(NTP,3000)	0097
	READ(NCR,1) ANAME	0098
	IF(ANAME.NE.'YES') GO TO 301	0099
	WRITE(NTP,3001)	0100
	WRITE(NTP,3010)	0101
C	RECTIFYING SECTION.	
	WRITE(NTP,3011) (XROP(I-MPTS),YROP(I-MPTS),TRL(I),LIQR(I),VAPR(I),	0102
	II=NTOT,MPT1,-5)	0103
C	FEED POSITION.	
	WRITE(NTP,3012)	0104
C	STRIPPING SECTION.	
	IF(MFIN,LT.999) WRITE(NTP,3011) (XSOP(I),YSOP(I),TSL(I),LIQS(I),VA	0105
	IPS(I),I=MPTS,MFIN,-5)	0106
	301 CONTINUE	0107
	WRITE(NTP,3020)	0108
	READ(NCR,1) ANAME	0109
	IF(ANAME.NE.'YES') GO TO 999	0110
	IF(NSORT.EQ.2) GO TO 310	0111
C	MASS-TRANSFER DATA FOR SYSTEMS WITH PRINCIPAL RESISTANCE IN	
C	VAPOUR PHASE.	
	WRITE(NTP,3021)	0112
	WRITE(NTP,3013)	0113
	WRITE(NTP,3014)	0114
C	RECTIFYING SECTION.	
	WRITE(NTP,3015) (XROP(I-MPTS),V(I),ASURF(I),KY(I),KX(I),I=NTOT,MPT	0115
	11,-5)	0116
	WRITE(NTP,3012)	0117
C	STRIPPING SECTION.	
	IF(MFIN,LT.999) WRITE(NTP,3015) (XSOP(I),V(I),ASURF(I),KY(I),KX(I)	0118
	1,I=MPTS,MFIN,-5)	0119
	WRITE(NTP,3016)	0120
	WRITE(NTP,3017)	0121
	GO TO 311	0122

SUBROUTINE PACRES

```

310 CONTINUE
C MASS-TRANSFER DATA FOR SYSTEMS WITH PRINCIPAL RESISTANCE IN
C LIQUID PHASE.
WRITE(NTP,3022)
WRITE(NTP,3113)
WRITE(NTP,3114)
C RECTIFYING SECTION.
WRITE(NTP,3015) (XROP(I-MPTS),L(I),ASURF(I),KY(I),KX(I),J=NTOT,MPT
11,-5)
WRITE(NTP,3012)
C STRIPPING SECTION.
IF(MFIN.LT.999) WRITE(NTP,3015) (XSOP(I),L(I),ASURF(I),KY(I),KX(I)
1,I=MPTS,MFIN,-5)
WRITE(NTP,3116)
WRITE(NTP,3117)
311 CONTINUE
C DATA INTEGRATED TO FIND THE HEIGHT OF PACKING IN EACH SECTION.
C RECTIFYING SECTION.
DO 30 I=NTOT,MPT1,-5
IIN=I-MPTS
IF(NSORT.EQ.1) XY1=YROP(IIM)
IF(NSORT.EQ.1) XY2=YI(I)
IF(NSORT.EQ.2) XY1=XROP(IIM)
IF(NSORT.EQ.2) XY2=XI(I)
WRITE(NTP,3018) XY1,XY2,FUR(IIM),VYR(IIM)
30 CONTINUE
WRITE(NTP,3012)
C STRIPPING SECTION.
IF(MFIN.GT.999) GO TO 999
DO 31 I=MPTS,MFIN,-5
IF(NSORT.EQ.1) XY1=YSOP(I)
IF(NSORT.EQ.1) XY2=YI(I)
IF(NSORT.EQ.2) XY1=XSOP(I)
IF(NSORT.EQ.2) XY2=XI(I)
WRITE(NTP,3018) XY1,XY2,FUS(I),VYS(I)
31 CONTINUE
999 CONTINUE
RETURN
END

```

```

SUBROUTINE PACTYP(MCAN,NTYPE,NPACK,NCR,NTP)
  1 FORMAT( )
  2 FORMAT(A3)
  IF(MCAN.EQ.1) WRITE(NTP,1000)
1000 FORMAT(/T2,'THE PROGRAM DATA PACKAGE INCLUDES PACKING DATA FOR CER
  JAMIC RASCHIG',/T2,'RINGS AND BERL SADDLES',/T2,'DO YOU WANT TO US
  2E A DIFFERENT TYPE OF PACKING?')
  READ(NCR,2) PACTYP
  IF(PACTYP.EQ.'YES') GO TO 200
  IF(MCAN.EQ.1) WRITE(NTP,1001)
1001 FORMAT(/T2,'DO YOU WANT TO USE (1) RASCHIG RINGS OR (2) BERL SADDL
  1ES?',/T2,'ENTER 1 OR 2')
  READ(NCR,1) NTYPE
  IF(NTYPE.GT.1) GO TO 101
  IF(MCAN.EQ.1) WRITE(NTP,1002)
1002 FORMAT(/T2,'THE AVAILABLE RING SIZES ARE:',/T4,'(1) 12.7,(2) 25.4,
  1(3) 38.1,(4) 50.8 (MM)',/T2,'ENTER APPROPRIATE CODE NO.')
```

```

  READ(NCR,1) NPACK
  GO TO 999
101 CONTINUE
  IF(MCAN.EQ.1) WRITE(NTP,1003)
1003 FORMAT(/T2,'THE AVAILABLE SADDLE SIZES ARE:',/T4,'(1) 12.7,(2) 25.
  14,(3) 38.1 (MM)',/T2,'ENTER APPROPRIATE CODE NO.')
```

```

  READ(NCR,1) NPACK
  GO TO 999

```

SUBROUTINE PACTYP



```

200 CONTINUE
   IF(MCAN.EQ.1) WRITE(NTP,2001)
2001 FORMAT(/T2,'DO YOU HAVE THE PACKING DATA PREPARED?')
      READ(NCR,2) PNAME
      IF(PNAME.EQ.'YES') GO TO 201
      IF(MCAN.EQ.1) WRITE(NTP,2002)
2002 FORMAT(/T2,'REFER TO THE DISTILLATION PACKAGE MANUAL,SECTION 3.5,F
      1OR FURTHER DETAILS.')
```

```

      MCAN=100
      GO TO 999
```

```

201 CONTINUE
   IF(MCAN.EQ.1) WRITE(NTP,2003)
2003 FORMAT(/T2,'ENTER CODES FOR PACKING TYPE AND SIZE')
```

```

      READ(NCR,1) NTYPE,NPACK
```

```

999 CONTINUE
      RETURN
      END
```

SUBROUTINE PACTYP

SUBROUTINE PAKING(NT,NP,CF,E,AMF1,AMF2,ANF1A,ANF1B,ANF2A,ANF2B,PF1	0001
1,PF2,DS,PHTW,PHSW,SWP,BETA)	0002
C ROUTINE SUPPLIES DATA ON EITHER CERAMIC RASCHIG RING OR BERL SADDLE	
C PACKING MATERIAL TO THE MAINLINE *PACKED*.	
C CHOICE IS DEPENDENT ON TYPE (NT) AND SIZE (NP) OF PACKING SPECIFIED	
C IN MAINLINE PROGRAM.	
C REFERENCE: TREYBAL,R.E.,MASS TRANSFER OPERATIONS.,2ND ED.,	
C MCGRAW-HILL,TABLES 6.2,6.3 AND 6.4.	
GO TO (100,200),NT	0003
100 CONTINUE	0004
C CERAMIC RASCHIG RINGS.	
GO TO (110,120,130,140),NP	0005
110 CONTINUE	0006
C 12.7 MM RINGS (0.5 IN)	
CF=640.	0007
E=0.63	0008
AMF1=8200.	0009
AMF2=9.32	0010
ANF1A=3.15	0011
ANF1B=-0.3	0012
ANF2A=0.151	0013
ANF2B=0.148	0014
PF1=-1.04	0015
PF2=-0.111	0016
DS=0.0582	0017
PHTW=2.25	0018
PHSW=0.00104	0019
SWP=1.21	0020
BETA=0.965*(DS**0.376)	0021
GO TO 999	0022

SUBROUTINE PAKING

```

C 120 CONTINUE
  25.4 MM RINGS (1.0 IN)
  CF=160.
  E=0.73
  AMF1=0.274
  AMF2=463.
  ANF1A=0.0
  ANF1B=0.0
  ANF2A=0.528
  ANF2B=-0.0793
  PF1=0.552
  PF2=-0.47
  DS=0.1167
  PHTW=2.25
  PHSW=0.00104
  SWP=1.21
  BETA=0.965*(DS**0.376)
  GO TO 999

```

```

0023
0024
0025
0026
0027
0028
0029
0030
0031
0032
0033
0034
0035
0036
0037
0038
0039

```

SUBROUTINE PAKING

```

C 130 CONTINUE
  38.1 MM RINGS (1.5 IN)
  CF=95.
  E=0.71
  ANF1=1.82
  ANF2=4.85
  ANF1A=0.675
  ANF1B=-0.1013
  ANF2A=0.148
  ANF2B=-0.022
  PF1=0.274
  PF2=0.140
  DS=0.1740
  PTH=2.25
  PHSW=0.00104
  SWP=1.21
  BETA=0.965*(DS**0.376)
  GO TO 999

```

```

0040
0041
0042
0043
0044
0045
0046
0047
0048
0049
0050
0051
0052
0053
0054
0055
0056

```

SUBROUTINE PAGING

C 140 CONTINUE  
50.8 MM RINGS (2.0 IN)  
CF=65.  
E=0.74  
AMF1=0.401  
AMF2=0.95  
ANF1A=0.0  
ANF1B=0.0  
ANF2A=0.0  
ANF2B=0.0  
PF1=0.481  
PF2=0.362  
DS=0.238  
PHTW=2.25  
PHSW=0.00104  
SVP=1.21  
BETA=0.965\*(DS\*\*0.376)  
GO TO 999

0057  
0058  
0059  
0060  
0061  
0062  
0063  
0064  
0065  
0066  
0067  
0068  
0069  
0070  
0071  
0072  
0073

SUBROUTINE PAKING

```

200 CONTINUE
C CERAMIC BERL SADDLES.
GO TO (210,220,230),NP
210 CONTINUE
C 12.7 MM SADDLES (0.5 IN)
CF=380.
E=0.63
AMF1=0.0336
AMF2=2.54
ANF1A=0.0
ANF1B=0.0529
ANF2A=0.0
ANF2B=0.0529
PF1=0.761
PF2=0.170
DS=0.0532
PHTW=2.5
PHSW=0.00032
SWP=1.56
BETA=0.965*(DS**0.376)
GO TO 999

```

```

0074
0075
0076
0077
0078
0079
0080
0081
0082
0083
0084
0085
0086
0087
0088
0089
0090
0091
0092

```

SUBROUTINE PAKING

```

C 220 CONTINUE
  25.4 MM SADDLES (1.0 IN)
  CF=110.
  E=0.69
  AMF1=15.89
  AMF2=238.
  ANF1A=0.686
  ANF1B=-0.1029
  ANF2A=0.420
  ANF2B=-0.063
  PF1=0.0
  PF2=-0.359
  DS=0.105
  PHTW=2.5
  PHSW=0.00032
  SWP=1.56
  BETA=0.965*(DS**0.376)
  GO TO 999

```

```

0093
0094
0095
0096
0097
0098
0099
0100
0101
0102
0103
0104
0105
0106
0107
0108
0109

```

SUBROUTINE PAKING

```

C 230 CONTINUE
  38.1 MM SADDLES (1.5 IN)
  CF=65.
  E=0.75
  AMF1=0.613
  AMF2=465.
  ANF1A=0.0
  ANF1B=-0.0508
  AMF2A=0.325
  ANF2B=-0.0996
  PF1=0.455
  PF2=-0.1355
  DS=0.155
  PHTW=2.5
  PHSW=0.00032
  SWP=1.56
  BETA=0.965*(DS**0.376)
  GO TO 999
999 CONTINUE
  RETURN
  END

```

SUBROUTINE PAYING

```

0110
0111
0112
0113
0114
0115
0116
0117
0118
0119
0120
0121
0122
0123
0124
0125
0126
0127
0128
0129

```



```

SUBROUTINE FDATA(NCURVE,ABS,ORD)
ROUTINE FITS AN N,TH ORDER POLYNOMIAL TO THE U.S. STONEWARE
COMPANY'S FLOODING AND LOADING CURVES FOR PACKED TOWERS.
REFERENCE: TREYBAL,R.E.,MASS TRANSFER OPERATIONS,2ND ED.,
          MCGRAW-HILL,PG. 160,FIG. 6.26.
THE DESIRED VAPOUR PRESSURE DROP AT LOADING IS SPECIFIED IN
THE MAINLINE,AND COUNTER NCURVE SELECTS THE RELEVANT CURVE.
THE FIT IS BY LEAST SQUARES. POLYNOMIALS UP TO ORDER TEN ARE
TESTED,AND THE BEST FIT IS SELECTED.
THE ROUTINE REQUIRES UNIVAC ,MATH-PAC, LIBRARY SUBROUTINE ,GJR,
FOR SOLVING THE N,TH ORDER MATRICES INVOLVED IN THE LEAST SQUARES
PROCEDURE.
THIS OR A SIMILAR ROUTINE MUST BE PROVIDED.
DIMENSION AMAT(20,20),CMAT(20),COEFF(20),X(50),Y(50)
DIMENSION JC(20),V(2)
DATA NCR,NTP/8,5/
1 FORMAT( )
C ADD DATA FOR CORRELATION CURVE SPECIFIED BY COUNTER NCURVE.
GO TO (100,200,300,400,500,600),NCURVE
100 CONTINUE
CALL ERTRAN(6,'@ADD,E DISTFILE,FLOOD1 , ')
GO TO 700
200 CONTINUE
CALL ERTRAN(6,'@ADD,E DISTFILE,FLOOD2 , ')
GO TO 700
300 CONTINUE
CALL ERTRAN(6,'@ADD,E DISTFILE,FLOOD3 , ')
GO TO 700
400 CONTINUE
CALL ERTRAN(6,'@ADD,E DISTFILE,FLOOD4 , ')
GO TO 700
500 CONTINUE
CALL ERTRAN(6,'@ADD,E DISTFILE,FLOOD5 , ')
GO TO 700
600 CONTINUE
CALL ERTRAN(6,'@ADD,E DISTFILE,FLOOD6 , ')
GO TO 700
700 CONTINUE

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SUBROUTINE FDATA

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C      READ IN DATA.
      READ(NCR,1) NDATA
      READ(NCR,1) (X(I),Y(I),I=1,NDATA)
      READ(NCR,1,END=701) IDUM
701  CONTINUE
C      SET INITIAL VALUES.
      NUM=2
      ERROR=1000.
      NTOT=20
101  NUM1=NUM+1
      *DO* LOOPS 10.11 & 12 CALCULATE TERMS IN MATRICES *AMAT* AND *CMAT*.
      DO 10 K=1,NUM
      DO 11 J=1,NUM
      SUMA=0.0
      SUMC=0.0
      DO 12 I=1,NDATA
      IF(J.EQ.1.AND.K.EQ.1) GO TO 102
      SUMA=SUMA+X(I)**(J+K-2)
      GO TO 103
102  SUMA=NDATA
103  IF(K.EQ.1) GO TO 104
      SUMC=SUMC+Y(I)**(X(I)**(K-1))
      GO TO 12
104  SUMC=SUMC+Y(I)
12  CONTINUE
      AMAT(J,K)=SUMA
11  CONTINUE
      CMAT(K)=SUMC
10  CONTINUE

```

SUBROUTINE FDATA

```

C      INSERTION OF CMAT IN COLUMN NUM1 OF AMAT.
DO 13 I=1,NUM
  AMAT(I,NUM1)=CMAT(I)
13 CONTINUE
C      'GJR' SOLVES EQUATIONS IN 'AMAT' BY GAUSS-JORDAN REDUCTION.
  V(1)=4
  CALL GJR(AMAT,NTOT,NTOT,NUM,NUM1,$900,JC,V)
  SUMERR=0.0
C      CALCULATE SUM OF SQUARED ERRORS.
DO 20 J=1,NDATA
  SUMX=AMAT(I,NUM1)
  DO 21 I=2,NUM
    SUMX=SUMX+AMAT(I,NUM1)*X(J)**(I-1)
21 CONTINUE
  ERR=SUMX-Y(J)
  SQERR=ERR**2.
  SUMERR=SUMERR+SQERR
20 CONTINUE
C      COMPARE SUM OF SQUARED ERRORS WITH PREVIOUS VALUE.
  IF(SUMERR.LT.ERROR) GO TO 110
120 NUM=NUM+1
  IF(NUM.GT.11) GO TO 201
  GO TO 101
110 CONTINUE
C      STORE ERROR AND COEFFICIENTS OF POLYNOMIAL WITH BEST FIT.
  ERROR=SUMERR
  NPOLY=NUM
  DO 30 I=1,NUM
    COEFF(I)=AMAT(I,NUM1)
30 CONTINUE
  GO TO 120
201 CONTINUE

```

SUBROUTINE FDATA

C CALCULATION OF ORDINATE CORRESPONDING TO 'ABS' ON CURVE 'NCURVE'.

```
0078  NPOLY=1
0079  XSUM=COEFF(1)
0080  DO 40 I=2,NPOLY
0081  IC=I-1
0082  XFACT=COEFF(I)*((ABS**IC)
0083  XSUM=XSUM+XFACT
0084  40 CONTINUE
0085  ORD=XSUM
0086  WRITE(NTP,1001) NORD,ERROR
0087  1001 FORMAT(/T2,'POLYNOMIAL OF ORDER',I3,' FITTED.',/,T2,
0088  1'SUM OF SQUARED ERRORS=',F15.10)
0089  GO TO 999
C ERROR MESSAGES *****
C ERROR DETECTED IN 'GJR',
900 CONTINUE
WRITE(NTP,9000) JC(1)
9000 FORMAT(/T2,'ERROR IN MATRIX...',JC=',I3)
999 CONTINUE
RETURN
END
```

SUBROUTINE FDATA

```

C      SUBROUTINE RHOLIQ(NPHY,NR,NL,NH,TG,X,RHO1,RHO2,RHOM)
C      ROUTINE CALCULATES LIQUID DENSITIES OF SINGLE COMPONENTS OR BINARY
C      SOLUTIONS OF COMPOSITION X AT TEMPERATURE TG.
C      KNOWN DATA CAN BE SUPPLIED IN SEVERAL WAYS.
C      FOR DETAILS REFER TO THE PACKAGE MANUAL, SECTION 3.
C      DIMENSION AMW(2),EXP1(4),RHOL1(50),RHOL2(50),RHOS(50)
C      DIMENSION TC(2),TRHO1(50),TRHO2(50),XS(50)
C      I FORMAT( )
C      NCR=8
C      EXP1(1)=0.25
C      EXP1(2)=0.29
C      EXP1(3)=0.31
C      EXP1(4)=0.33
C      IF(NPHY.GT.1) GO TO 600
C      DATA REQUIRED FOR EACH COMPONENT:
C      (A) CRITICAL TEMPERATURE (DEG C)
C      (B) NO. OF TEMPERATURE-DENSITY DATA PAIRS TO BE SUPPLIED.
C      CALL ERTRAN(6,'QADD,E DISTFILE,ROLD A1 . ')
C      READ(NCR,1) (AMW(I),TC(I),I=1,2)
C      TC(1)=TC(1)+273.
C      TC(2)=TC(2)+273.
C      READ(NCR,1) NDAT1,NDAT2
C      READ(NCR,1,END=600) IDUM
C      600 CONTINUE
C      T=TG+273.
C      IF(NDAT1.GT.1) GO TO 100
C      IF(NPHY.GT.1) GO TO 611
C      LIGHT COMPONENT DENSITY. (ONLY ONE DATA PAIR SUPPLIED)
C      CALL ERTRAN(6,'QADD,E DISTFILE,ROLD A2 . ')
C      READ(NCR,1) TRHO1(1),RHOL1(1)
C      TRHO1(1)=TRHO1(1)+273.
C      611 CONTINUE
C      RHOL1=RHOL1(1)*((TC(1)-T)/(TC(1)-TRHO1(1)))*EXP1(NL)
C      GO TO 199

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SUBROUTINE RHOLIQ

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100 CONTINUE
   IF(NPHY.GT.1) GO TO 612
   LIGHT COMPONENT DENSITY, (MORE THAN ONE DATA PAIR SUPPLIED)
   CALL ERTRAN(6,'@ADD-E DISTFILE,ROLOA2 . ')
   READ(NCR,1) (TRH01(I),RHOL1(I),I=1,NDAT1)
612 CONTINUE
   IF(TG.GT.,TRH01(1)) GO TO 110
   RHOL1=RHOL1(1)
   GO TO 199
110 CONTINUE
   DO 10 I=1,NDAT1
   DIFF=TG-TRH01(I)
   IF(DIFF.LE.0.0) GO TO 111
10 CONTINUE
   GO TO 120
111 CONTINUE
   TFACT=(TRH01(I)-TG)/(TRH01(I)-TRH01(I-1))
   RHOL1=RHOL1(I)+(RHOL1(I-1)-RHOL1(I))*TFACT
   GO TO 199
120 CONTINUE
   RHOL1=RHOL1(NDAT1)
199 CONTINUE
   IF(NDAT2.GT.1) GO TO 200
   IF(NPHY.GT.1) GO TO 601
   HEAVY COMPONENT DENSITY, (ONLY ONE DATA PAIR SUPPLIED)
   READ(NCR,1) TRH02(1),RHOL2(1)
   TRH02(1)=TRH02(1)+273.
   READ(NCR,1,END=601) IDUM
601 CONTINUE
   RHOL2=RHOL2(1)*((TC(2)-T)/(TC(2)-TRH02(1)))*EXP1(NH)
   GO TO 299

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SUBROUTINE RHOLIQ

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200 CONTINUE
    IF(NPHY.GT.1) GO TO 602
    HEAVY COMPONENT DENSITY. (MORE THAN ONE DATA PAIR SUPPLIED)
    READ(NCR,1) (TRH02(1),RH02(1),I=1,NDAT2)
    READ(NCR,1,END=602) IDUM
602 CONTINUE
    IF(TG.GT.TRH02(1)) GO TO 210
    RH02=RH02(1)
    GO TO 299
210 CONTINUE
    DO 20 I=1,NDAT2
    DIFF=TG-TRH02(I)
    IF(DIFF.LE.0.0) GO TO 211
20 CONTINUE
    GO TO 220
211 CONTINUE
    TFACT=(TRH02(1)-TG)/(TRH02(1)-TRH02(I-1))
    RH02=RH02(1)+(RH02(I-1)-RH02(1))*TFACT
    GO TO 299
220 CONTINUE
    RH02=RH02(NDAT2)
299 CONTINUE
    IF(NR.EQ.1) GO TO 300
    DENSITY OF SOLUTION OF COMPOSITION X AT TEMPERATURE TG.
    FACT=X*AMW(1)*RH02+(1.-X)*AMW(2)*RH01
    RHOM=RH01*RH02*(X*AMW(1)+(1.-X)*AMW(2))/FACT
    GO TO 999
C

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SUBROUTINE RHOLIQ

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300 CONTINUE
C OPTIONAL FACILITY FOR SYSTEMS WITH COMPOSITION-DENSITY DATA KNOWN
C AT SOME TEMPERATURE ,TBASEC*,
C IF(NPHY.GT.1) GO TO 603
C ,NDATA, COMPOSITION-DENSITY DATA PAIRS,AT TEMPERATURE TBASEC
C MUST BE SUPPLIED.
CALL ERTRAN(6,*,*ADD,E DISTFILE,ROLD A3 . .)
READ(NCR,1) NDATA,TBASEC
READ(NCR,1) (XS(I),RHOS(I),I=1,NDATA)
READ(NCR,1,END=603) IDUM
603 CONTINUE
TBASE=TBASEC+273.
IF(X.GT.XS(1)) GO TO 310
RHOC=RHOS(1)
GO TO 399
310 CONTINUE
DO 30 I=1,NDATA
DIFF=X-XS(I)
IF(DIFF.LE.0.0) GO TO 311
30 CONTINUE
GO TO 320
311 CONTINUE
XFACT=(XS(I)-X)/(XS(I)-XS(I-1))
RHOC=RHOS(1)+(RHOS(I-1)-RHOS(I))*XFACT
GO TO 399
320 CONTINUE
RHOC=RHOS(NDATA)
399 CONTINUE
C DENSITY OF SOLUTION OF COMPOSITION X AT TEMPERATURE TG.
TCP=X*TC(1)+(1.-X)*TC(2)
RHOM=RHOC*((TCP-T)/(TCP-TBASE))*((EXP1(NL)+EXP1(NH))/2.)
999 CONTINUE
RETURN
END

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SUBROUTINE RHOLIQ



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C      SUBROUTINE SURTEN(NPHY,NL,NH,TG,X,SIGM)
C      ROUTINE CALCULATES THE SURFACE TENSION OF BINARY SOLUTIONS OF
C      COMPOSITION X AT TEMPERATURE TG,
C      KNOWN DATA CAN BE SUPPLIED IN SEVERAL WAYS.
C      FOR DETAILS REFER TO THE PACKAGE MANUAL, SECTION 3.
C      DIMENSION AMW(2), PARA(2), SIG1(50), SIG2(50), SIGS(50)
C      DIMENSION T1(2), T2(2), TC(50), XS(50)
C      1 FORMAT( )
C      NCR=8
C      IF(NPHY.GT.1) GO TO 600
C      CALL ERTRAN(6,'@ADD,E DISTFILE,SIGDA1 , ')
C      READ(NCR,1) NSIGM
C      READ(NCR,1) (AMW(I),TC(I),PARA(I),I=1,2)
C      READ(NCR,1) NDATA1,NDAT2
C      TC(1)=TC(1)+273.
C      TC(2)=TC(2)+273.
C      READ(NCR,1,END=600) IDUM
C      600 CONTINUE
C      T=TG+273.
C      IF(NSIGM.EQ.2) GO TO 300
C      IF(PARA(1).GT.0.0) GO TO 198
C      IF(NDAT1.GT.1) GO TO 100
C      IF(NPHY.GT.1) GO TO 601
C      CALL ERTRAN(6,'@ADD,E DISTFILE,SIGDA2 , ')
C      READ(NCR,1) T1(1),SIG1(1)
C      T1(1)=T1(1)+273.
C      READ(NCR,1,END=601) IDUM
C      601 CONTINUE
C      SIGA=SIG1(1)*((TC(1)-T)/(TC(1)-T1(1)))**1.2
C      GO TO 199

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SUBROUTINE SURTEN

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100 CONTINUE
   IF(NPHY.GT.1) GO TO 602
   CALL ERTRAN(6,'@ADD.E DISTFILE.SIGDA2 . ')
   READ(NCR,1) (T1(I),SIG1(I),I=1,NDAT1)
   READ(NCR,1,END=602) IDUM
602 CONTINUE
   IF(TG.GT.T1(1)) GO TO 110
   SIGA=SIG1(1)
   GO TO 199
110 CONTINUE
   DO 10 I=1,NDAT1
     DIFF=TG-T1(I)
     IF(DIFF.LE.0.0) GO TO 111
10 CONTINUE
111 CONTINUE
     GO TO 120
120 CONTINUE
     TFACT=(T1(I)-TG)/(T1(I)-T1(I-1))
     SIGA=SIG1(I)+(SIG1(I-1)-SIG1(I))*TFACT
     GO TO 199
120 CONTINUE
     SIGA=SIG1(NDAT1)
     GO TO 199
198 CONTINUE
     NR=0
     CALL RHOLIQ(NPHY,HR,NL,NH,TG,X,RHO1,RHO2,RHOM)
     SIGA=(PARA(1)*RHO1/AMW(1))*4.
199 CONTINUE
     IF(PARA(2).GT.0.0) GO TO 400
     IF(NDAT2.GT.1) GO TO 200
     IF(NPHY.GT.1) GO TO 604
     CALL ERTRAN(6,'@ADD.E DISTFILE.SIGDA4 . ')
     READ(NCR,1) T2(1),SIG2(1)
     T2(1)=T2(1)+273.
     READ(NCR,1,END=604) IDUM
604 CONTINUE
     SIGB=SIG2(1)*((TC(2)-T)/(TC(2)-T2(1)))*1.2
     GO TO 299

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SUBROUTINE SURTEN

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200 CONTINUE
   IF(NPHY.GT.1) GO TO 605
   CALL ERTRAN(6,'DADD•E DISTFILE•SIGDA4 •')
   READ(MCR,1) (T2(I),SIG2(I),I=1,NDAT2)
   READ(MCR,1,END=605) IDUM
605 CONTINUE
   IF(TG.GT.T2(1)) GO TO 210
   SIGB=SIG2(1)
   GO TO 299
210 CONTINUE
   DO 20 I=1,NDAT2
   DIFF=TG-T2(I)
   IF(DIFF.LE.0.0) GO TO 211
20 CONTINUE
   GO TO 220
211 CONTINUE
   TFACT=(T2(I)-TG)/(T2(I)-T2(I-1))
   SIGB=SIG2(I)+(SIG2(I-1)-SIG2(I))*TFACT
   GO TO 299
220 CONTINUE
   SIGB=SIG2(NDAT2)
299 CONTINUE
   SIGH=SIGA*SIGB/(X*SIGB+(1.-X)*SIGA)
   GO TO 999
300 CONTINUE
C   OPTIONAL FACILITY FOR SYSTEMS WITH COMPOSITION-SURFACE TENSION
C   DATA KNOWN AT SOME TEMPERATURE •TBASEC•.
C   IF(NPHY.GT.1) GO TO 603
C   •NDATA• COMPOSITION-SURFACE TENSION DATA PAIRS•AT TEMPERATURE
C   TBASEC•MUST BE SUPPLIED.

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SUBROUTINE SURTEN

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CALL ERTRAN(6,'@ADD.E DISTFILE,SIGDA3 . ')
READ(NCR,1) NDATA,TBASEC
READ(NCR,1) (XS(I),SIGS(I),I=1,NDATA)
READ(NCR,1,END=603) IDUM
603 CONTINUE
TBASE=TBASEC+273.
IF(X.GT.XS(1)) GO TO 310
SIGC=SIGS(1)
GO TO 399
310 CONTINUE
DO 30 I=1,NDATA
DIFF=X-XS(I)
IF(DIFF.LE.0.0) GO TO 311
30 CONTINUE
GO TO 320
311 CONTINUE
XFACT=(XS(I)-X)/(XS(I)-XS(I-1))
SIGC=SIGS(I)+(SIGS(I-1)-SIGS(I))*XFACT
GO TO 399
320 CONTINUE
SIGC=SIGS(NDATA)
399 CONTINUE
TCP=X*TC(1)+(1.-X)*TC(2)
SIGM=SIGC*((TCP-T)/(TCP-TBASE))*1.2
GO TO 999
400 CONTINUE
NR=0
CALL RHOLIQ(NPHY,NR,NL,NH,TG,X,RHO1,RHO2,RHOM)
SIGB=(PARA(2)*RHO2/AMWL(2))*4.
AMWL=X*AMWL(1)+(1.-X)*AMWL(2)
PARSUM=X*PARA(1)+(1.-X)*PARA(2)
SIGM=(PARSUM*RHOM/AMWL)*4.
999 CONTINUE
RETURN
END

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SUBROUTINE SURTEN

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C      SUBROUTINE DIFLIQ(NPHY,NL,NH,TG,X,DL)
C      ROUTINE CALCULATES LIQUID DIFFUSIVITY OF BINARY SOLUTIONS
C      OF COMPOSITION X AT TEMPERATURE TG.
C      FOR DATA REQUIRED REFER TO PACKAGE MANUAL, SECTION 3.
      1  FORMAT( )
      NCR=B
      IF(NPHY.GT.1) GO TO 600
      CALL ERTRAN(6,'@ADD,E DISTFILE,DLDAT . ')
      READ(NCR,1) DL12,TDL
      TDL=TDL+273.
      READ(NCR,1,END=600) IDUM
      600 CONTINUE
      TAG=TDL-273.
      T=TG+273.
      TA=TDL
      CALL VISCL(NPHY,NL,NH,TAG,X,AMU1,AMU2,AMUM)
      AMUA=AMU2
      DL=DL12/(10.**5)
      CONST=DL*AMUA/TA
      NPH=2
      CALL VISCL(NPH,NL,NH,TG,X,AMU1,AMU2,AMUM)
      DL=CONST*T/AMUM
      999 CONTINUE
      RETURN
      END

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SUBROUTINE DIFLIQ

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SUBROUTINE DIFVAP(NPHY,PT,TG,DV)
ROUTINE CALCULATES VAPOUR DIFFUSIVITY OF BINARY SOLUTIONS
OF COMPOSITION X AT TEMPERATURE TG AND TOTAL PRESSURE PT.
FOR DATA REQUIRED REFER TO PACKAGE MANUAL SECTION 3.
DIMENSION AMW(2),TB(2),VB(2)
1 FORMAT(' ')
NCR=8
IF(NPHY.GT.1) GO TO 600
CALL ERTRAN(6,'@ADD.E DISTFILE,DV DAT . ')
READ(NCR,1) (AMW(I),VB(I),TB(I),I=1,2)
DO 10 I=1,2
VB(I)=VB(I)*0.333
TB(I)=TB(I)+273.
10 CONTINUE
READ(NCR,1,END=600) IDUM
600 CONTINUE
T=TG+273.
P=PT/760.
B=((AMW(1)+AMW(2))/(AMW(1)*AMW(2)))**0.5
F=((2.*(VB(1)*VB(2))**0.5)/(VB(1)+VB(2)))**3.
CI2=1.47*F*((TB(1)*TB(2))**0.5)
DV=((0.00837*B)/(P*((VB(1)+VB(2))**2.)))*((T**2.5)/(T+CI2))
RETURN
END

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SUBROUTINE DIFVAP

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SUBROUTINE VISCL(NPHY,NL,NH,TG,X,AMUA,AMUB,AMUM)
ROUTINE CALCULATES LIQUID VISCOSITIES OF SINGLE COMPONENTS OR
BINARY SOLUTIONS OF COMPOSITION X AT TEMPERATURE TG.
FOR DETAILS REFER TO THE PACKAGE MANUAL, SECTION 3.
DIMENSION AMU1(50),AMU2(50),B(2),RHO(2)
DIMENSION T1(50),T2(50),TC(2),TR(2)
1 FORMAT( )
NCR=8
TIG=TG
T2G=TG
IF(NPHY.GT.1) GO TO 611
CALL ERTRAN(6,*,@ADD,E,DISTFILE,FMUDA1,*)
READ(NCR,1) (TC(I),B(I),I=1,2)
READ(NCR,1) NDAT1,NDAT2
READ(NCR,1,END=601) IDUM
601 CONTINUE
DO 11 I=1,2
TC(I)=TC(I)+273.
11 CONTINUE
611 CONTINUE
T=TG+273.
DO 12 I=1,2
TR(I)=T/TC(I)
12 CONTINUE
IF(B(1).GT.0.0) GO TO 198
IF(NPHY.GT.1) GO TO 602
CALL ERTRAN(6,*,@ADD,E,DISTFILE,FMUDA2,*)
READ(NCR,1) (T1(I),AMU1(I),I=1,NDAT1)
READ(NCR,1,END=602) IDUM
602 CONTINUE
IF(TG.GT.T1(1)) GO TO 110
AMUA=AMU1(1)
GO TO 199
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SUBROUTINE VISCL

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110 CONTINUE
DO 10 I=1,NDAT1
  DIFF=TG-T1(I)
  IF(DIFF.LE.8.0) GO TO 111
10 CONTINUE
GO TO 120
111 CONTINUE
TFACT=(T1(I)-TG)/(T1(I)-T1(I-1))
AMUA=AMU1(I)+(AMU1(I-1)-AMU1(I))*TFACT
GO TO 199
120 CONTINUE
AMUA=AMU1(NDAT1)
GO TO 199
198 CONTINUE
NR=0
CALL RHOLIQ(NPHY,NR,NL,NH,TIG,X,RHO1,RHO2,RHOM)
RHO(1)=RHO1
ALFA=B(1)*(1.-TR(1))/TR(1)
AMUA=(0.1167*(RHO(1)*0.5))*((10.**ALFA)
199 CONTINUE
IF(B(2).GT.0.0) GO TO 300
IF(NPHY.GT.1) GO TO 603
CALL ERTRAN(6.,@ADD,E DISTFILE,FMUDA3 , ')
READ(NCR,1) (T2(I),AMU2(I),I=1,NDAT2)
READ(NCR,1,END=603) IDUM
603 CONTINUE
IF(TG.GT.T2(1)) GO TO 210
AMUR=AMU2(1)
GO TO 299

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SUBROUTINE VISCL

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210 CONTINUE
DO 20 I=1,NDAT2
DIFF=TG-T2(I)
IF(DIFF.LE.0.0) GO TO 211
20 CONTINUE
GO TO 220
211 CONTINUE
TFACT=(T2(I)-TG)/(T2(I)-T2(I-1))
AMUB=AMU2(I)+(AMU2(I-1)-AMU2(I))*TFACT
GO TO 299
220 CONTINUE
AMUB=AMU2(NDAT2)
299 CONTINUE
AMUM=(X*(AMUA*(1./3.))+(1.-X)*(AMUB*(1./3.)))*3.
GO TO 999
300 CONTINUE
NR=0
CALL RHOLIQ(NPHY,NR,NL,NH,T2G,X,RHO1,RHO2,RHOM)
RHO(2)=RHO2
ALFB=B(2)*(1.-TR(2))/TR(2)
AMUB=(0.1167*(RHO(2)*0.5))*(10.**ALFB)
AMUM=(X*(AMUA*(1./3.))+(1.-X)*(AMUB*(1./3.)))*3.
999 CONTINUE
RETURN
END

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SUBROUTINE VISCL

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C      SUBROUTINE VISCV(NPHY,TG,Y,AMUM)
C      ROUTINE CALCULATES VAPOUR VISCOSITIES OF BINARY SOLUTIONS
C      OF COMPOSITION X AT TEMPERATURE TG,
C      FOR DATA REQUIRED REFER TO PACKAGE MANUAL SECTION 3.
      DIMENSION AMU1(50),AMU2(50),AMUV(2),AMW(2)
      DIMENSION T1(50),T2(50),TB(2),VB(2),YI(2)
      1 FORMAT( )
      NCR=8
      YI(1)=Y
      YI(2)=1.-Y
      IF(NPHY.GT.1) GO TO 601
      CALL ERTRAN(6.,@ADD,E DISTFILE,VMUDA1 . .)
      READ(NCR,1) (AMW(I),VB(I),TB(I),I=1,2)
      READ(NCR,1) NDATA1,NDATA2
      TB(1)=TB(1)+273.
      TB(2)=TB(2)+273.
      READ(NCR,1,END=601) IDUM
601 CONTINUE
      T=TG+273.
      IF(VB(1).GT.0.0) GO TO 199
      IF(NPHY.GT.1) GO TO 602
      CALL ERTRAN(6.,@ADD,E DISTFILE,VMUDA2 . .)
      READ(NCR,1) (T1(I),AMU1(I),I=1,NDATA1)
      READ(NCR,1,END=602) IDUM
602 CONTINUE
      IF(TG.GT.T1(1)) GO TO 110
      AMUV(1)=AMU1(1)
      GO TO 199

```

SUBROUTINE VISCV

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110 CONTINUE
DO 10 I=1,NDAT1
DIFF=TG-T1(I)
IF(DIFF.LE.0.0) GO TO 111
10 CONTINUE
GO TO 120
111 CONTINUE
TFACT=(T1(I)-TG)/(T1(I)-T1(I-1))
AMUV(1)=AMU1(I)+(AMU1(I-1)-AMU1(I))*TFACT
GO TO 199
120 CONTINUE
AMUV(1)=AMU1(NDAT1)
199 CONTINUE
IF(VB(2).GT.0.0) GO TO 300
IF(INPHY.GT.1) GO TO 603
CALL ERTRAN(6.,QADD,E DISTFILE,VMUDA3 . .)
READ(NCR,1) (T2(I),AMU2(I),I=1,NDAT2)
READ(NCR,1,END=603) IDUM
603 CONTINUE
IF(TG.GT.T2(1)) GO TO 210
AMUV(2)=AMU2(1)
GO TO 999
210 CONTINUE
DO 20 I=1,NDAT2
DIFF=TG-T2(I)
IF(DIFF.LE.0.0) GO TO 211
20 CONTINUE
GO TO 220

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SUBROUTINE VISCV

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211 CONTINUE
   TFACT=(T2(I)-TG)/(T2(I)-T2(I-1))
   AMUV(2)=AMU2(I)+(AMU2(I-1)-AMU2(I))*TFACT
   GO TO 999

220 CONTINUE
   AMUV(2)=AMU2(NDAT2)
   GO TO 999

300 CONTINUE
   DO 30 I=1,2
     FACT1=27.*(AMW(I)**0.5)*(T**1.5)
     FACT2=(VB(I)**0.667)*(T+1.47*TB(I))
     AMUV(I)=FACT1/FACT2
30 CONTINUE
999 CONTINUE
   SUMA=0.0
   SUMB=0.0
   DO 90 I=1,2
     SUMA=SUMA+YI(I)*(AMW(I)**0.5)*AMUV(I)
     SUMB=SUMB+YI(I)*(AMW(I)**0.5)
90 CONTINUE
   AMUM=SUMA/SUMB
   RETURN
   END

```

SUBROUTINE VI5CV

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```

C      SURROUTINE RHOH2O(NPHY,T,RHO)
C      ROUTINE CALCULATES DENSITY OF WATER AT TEMPERATURE T.
C      A SET OF NDATA TEMPERATURE-DENSITY DATA PAIRS MUST BE SUPPLIED.
      DIMENSION RHOH(200),TW(200)
      1  FORMAT( )
      NCR=8
      IF(NPHY.GT.1) GO TO 600
      CALL ERTRAN(6,'@ADD.E DISTFILE.H2ORHO . ')
      READ(NCR,1) NDATA
      READ(NCR,1) (TW(I),RHOH(I),I=1,NDATA)
      READ(NCR,1,END=600) IDUM
      600 CONTINUE
      IF(T.GT.TW(1)) GO TO 100
      RHO=RHOH(1)
      GO TO 999
      100 CONTINUE
      DO 10 I=1,NDATA
      DIFF=T-TW(I)
      IF(DIFF.LE.0.0) GO TO 111
      10 CONTINUE
      GO TO 120
      111 CONTINUE
      TFACT=(TW(1)-T)/(TW(1)-TW(I-1))
      RHO=RHOH(1)+(RHOH(I-1)-RHOH(1))*TFACT
      GO TO 999
      120 CONTINUE
      RHO=RHOH(NDATA)
      999 CONTINUE
      RETURN
      END

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SUBROUTINE RHOH20

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C DIMENSION AND COMMON BLOCK 'MULTICOM' FOR MULTICOMPONENT DISTILLATION

C PROGRAM,

PARAMETER MC=30

PARAMETER MP=53

DIMENSION A(MC,MP),AN(MC),AK1(MC),AK2(MC),AK3(MC),AK4(MC)

DIMENSION B(MC),BD(MC),BN(MC),CN(MC),D(MC)

DIMENSION EFFV(MC,MP),HLIQ(MP),HVAP(MP),K(MC,MP),KF(MC)

DIMENSION HL1(MC),HL2(MC),HL3(MC),HV1(MC),HV2(MC),HV3(MC)

DIMENSION LB(MC,MP),LD(MC,MP),LMFR(MC,MP),LRTOT(MP),LSTOT(MP)

DIMENSION LPROF(MP),PRESS(MP),PRES(MP),S(MC,MP),SLIQ(MP),SVAP(MP)

DIMENSION TEMP(MP),TTILES(MC,5)

DIMENSION VB(MC,MP),VD(MC,MP),VMFR(MC,MP),VRTOT(MP),VSTOT(MP)

DIMENSION VPROF(MP),X(MC,MP),XF(MC),Y(MC,MP),YF(MC),Z(MC)

COMMON A,ACCLIM,AKB1,AKB2,AKB3,AKB4,AK1,AK2,AK3,AK4,ALIQ,AN,AVAP

COMMON B,BD,BN,BTOT

COMMON CN,D,DELTAC,DELTAP,DM,DTOT,EFFV,FENTH,FTOT

COMMON HD,HFEED,HLIQ,HL1,HL2,HL3,HVAP,HV1,HV2,HV3

COMMON ICOUNT,K,KF,LB,LD,LMFR,LPROF,LRTOT,LSTOT,MEFF

COMMON NC,NCOND,NENTH,NEQUIL,NF,NF1,NF2,NJ,NOWN,NP,NP1,NP2

COMMON NQVAL,NUM1,NUM2,NUM3

COMMON NCR,NG1,NG2,NG3,NG4,NLP,NTP

COMMON PCOND,PRESS,PRES

COMMON Q,QC,QF,QR

COMMON RR,S,SLIQ,SVAP

COMMON TBOT,TEMP,TFEED,THETA,TTILES,TTOP

COMMON VB,VD,VMFR,VPROF,VRTOT,VSTOT

COMMON X,XF,Y,YF,Z

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COMMON 'MULTICOM'

```

C MAINLINE PROGRAM FOR SOLUTION OF MULTICOMPONENT DISTILLATION
C PROBLEMS BY THE THIELE-GEDES METHOD.
C THE HOLLAND ,THETA, METHOD IS USED TO ACCELERATE CONVERGENCE,
C FOR FURTHER DETAILS REFER TO THE DISTILLATION-PACKAGE USER'S MANUAL,
C DEPARTMENT OF CHEMICAL ENGINEERING,UNIVERSITY OF CAPE TOWN.

```

```

C *****
C INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
C *****

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```

C DIMENSION COMPNO(MC),TRAYNO(MP)

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```

C REAL K,KF,LNFR,LRTOT,LSTOT,LD,LB,LPROF

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```

C 1 FORMAT( 1

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```

C 2 FORMAT(A3)

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C 3 FORMAT(5A6)

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```

C DATA IUSEA1,IUSEA2,IUSEB1/1.0,1/

```

```

C DATA MCAN,MTIMES,NCODE,NTIME,NOUT,NOWN/1.1,4.1,0.0/

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```

C DATA NCR,NG1,NG2,NG3,NG4,NLP,NTP/8.12,13.14,15.16,5/

```

```

C READ IN DATA *****

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```

C WRITE(NTP,5000)

```

```

C 5000 FORMAT(/T2,.,ANSWER ALL QUESTIONS YES OR NO,.,/T2,.,ALL INPUT IN FRE

```

```

C 1E FORMAT,.,./T2,.,DO YOU KNOW HOW TO USE THIS PROGRAM?,.)

```

```

C READ(NCR,2) ANAME

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```

C IF(ANAME.EQ.,YES,1) GO TO 501

```

```

C CALL TEACH(NCODE)

```

```

C GO TO 902

```

```

C 501 CONTINUE

```

```

C WRITE(NTP,5001)

```

```

C 5001 FORMAT(/T2,.,DO YOU WANT TO ADD A DATA ELEMENT?,.)

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```

C READ(NCR,2) CNAME

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C IF(CNAME.EQ.,NO,.) GO TO 502

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```

C MCAN=2

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C WRITE(NTP,5002)

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C 5002 FORMAT(/T2,.,ADD YOUR DATA ELEMENT.,.)

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C 502 CONTINUE

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C	CALL READIN(MCAN)	0025
C	IF(NJ.GT.999) GO TO 902	0026
C	DO* LOOPS 51 AND 52 SET UP ARRAYS REQUIRED FOR WRITING DATA TO	
C	FILE FOR PLOTTING PURPOSES.	
	SUMTN=0.	0027
	DO 51 J=1,NP2	0028
	TRAYNO(J)=SUMTN	0029
	SUMTN=SUMTN+1.	0030
51	CONTINUE	0031
	SUMCN=0.	0032
	DO 52 I=1,NC	0033
	SUMCN=SUMCN+1.	0034
	COMPNO(I)=SUMCN	0035
52	CONTINUE	0036
1115	CONTINUE	0037
C		
C	START OF RUN *****	
C		
C	READ RUN VARIABLES *****	
C		
	ICOUNT=1	0038
	WRITE(NTP,5011)	0039
5011	FORMAT(/T2,•TO PROCEED,TYPE GO OR TO STOP THE PROGRAM,TYPE END.•)	0040
	READ(NCR,2) RUN	0041
	IF(RUN.EQ.‘END’) GO TO 901	0042
	WRITE(NTP,5012)	0043
5012	FORMAT(/T2,•ENTER RUN NO. AS: NO. DAY,MONTH, (INTEGER VALUES)•)	0044
	READ(NCR,1) NUM1,NUM2,NUM3	0045
	IF(NUM2.GE.32.OR.NUM3.GE.13) GO TO 991	0046
	IF(TIMES.EQ.1) GO TO 515	0047
	WRITE(NTP,5013)	0048
5013	FORMAT(/T2,•ENTER REFLUX RATIO AND DISTILLATE RATE FOR NEW RUN.•)	0049
	READ(NCR,1) RR,DTOT	0050

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```

5014 WRITE(NTP,5014)
    FORMAT(T2,'ENTER FEED TEMPERATURE. (DEG C)')
    READ(NCR,1) TFEED
    NQVAL=1
    IF(NENTH.EQ.0) GO TO 514
    WRITE(NTP,5015)
5015 FORMAT(T2,'IS FEED PRESSURE GREATER THAN COLUMN PRESSURE?')
    READ(NCR,2) FPRES
    IF(FPRES.EQ.'NO') GO TO 514
    WRITE(NTP,5016)
5016 FORMAT(T2,'ENTER FEED ENTHALPY. (KJ/KGMOLE)')
    READ(NCR,1) FENTH
    NQVAL=2
514 CONTINUE
    WRITE(NTP,5017)
5017 FORMAT(T2,'ENTER FEED TRAY POSN. (NO. FROM BOTTOM REBOILER=0)')
    READ(NCR,1) NF
    NF1=NF+1
    NF2=NF+2
515 CONTINUE
    IF(IRR.LE.0.) GO TO 991
    WRITE(NTP,5111)
5111 FORMAT(/T2,'START OF COMPUTATION.')
    C INITIAL TEMPERATURE PROFILE *****
    DELTAT=(TBOT-TTOP)/NP1
    DO 53 J=1,NP2
    TEMP(J)=TBOT-(J-1)*DELTAT
53 CONTINUE
    C TRAY PRESSURE PROFILE *****
    IF(MTIMES.GT.1) GO TO 516
    PRESS(NP1)=PCOND+DELTAC
    PRESS(NP2)=PCOND
    NCOUNT=0
    DO 54 J=NP1,-1
    NCOUNT=NCOUNT+1
    PRESS(J)=PRESS(NP1)+NCOUNT*DELTAP
54 CONTINUE
516 CONTINUE
    MTIMES=MTIMES+1

```

```

C FLASH CALCULATION TO DETERMINE FEED CONDITION *****
C CALL QFLASH(NTIME,TDEWPT) 0088
C IF(NJ.GT.999) GO TO 902 0089

C INITIAL VAPOUR & LIQUID FLOW RATES IN COLUMN *****
C CALL FLOWEQ 0090
C IF(NJ.GT.999) GO TO 902 0091

C 1111 CONTINUE 0092

C START OF ITERATIONS FORCING THETA TO UNITY *****
C
C CALCULATION OF EQUILIBRIUM CONSTANTS *****
C NEQ=3 0093
C CALL EQUCON(NEQ) 0094

C ENTHALPY CONSIDERATIONS IGNORED FOR FIRST TWO ITERATIONS
C IF(ICOUNT.GT.2) GO TO 111 0095
C GO TO 112 0096

C ENTHALPY CALCULATIONS AND VAPOUR & LIQUID FLOWRATE CORRECTION *****
C 111 CONTINUE 0097
C IF(NENTH.EQ.0) GO TO 112 0098
C ENTHALPY CALCULATIONS *****
C NEN=3 0099
C TDEWP=TDEWPT 0100
C IF(TFEED.LT.TDEWPT) TDEWP=TFEED 0101
C IF(NOWN.EQ.1) CALL ENERGY(TBURP,TDEWP,NEN,S1,S2,S3,S4) 0102
C IF(NOWN.EQ.2) CALL ENTHAL(ALIQ,AVAP,BM,DM,BTOT,DTOT,FTOT,HD,HFEED, 0103
C IHLIQ,HVAP,NC,NFI,NP1,NP2,NCOND,QC,QF,QR,TEMP,VRTOT,X,XF,Y,YF,TBUBP 0104
C 2,TDEWP,TFEED,NEN,S1,S2,S3,S4,NTIME,MC,MP) 0105
C RECALCULATION OF FLOWRATES (NON-EQUI-MOLAR OVERFLOW) *****
C CALL FLOWNE 0106
C IF(NJ.GT.999) GO TO 902 0107
C 112 CONTINUE 0108

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C C ABSORPTION & STRIPPING FACTORS ***** 0109
C CALL ABSTRP *****
C
C V(I)/B(I) & L(I)/B(I) RATIOS ***** 0110
C CALL VBLB(LB,NC,NF1,S,VB,MC,MP) *****
C
C V(I)/D(I) & L(I)/D(I) RATIOS ***** 0111
C CALL VDLB(A,LD,MC,NF2,NP1,NP2,VD,MC,MP) *****
C
C CALCULATED B(I)/D(I) RATIOS ***** 0112
C CALL BDCALC(A,BD,D,FTOT,LB,NC,NF1,SUMDI,VD,Z,MC,MP) *****
C ERRD=DTOT-SUMDI ***** 0113
C
C *THETA* CONVERGENCE TECHNIQUE ***** 0114
C CALL THECAL ***** 0115
C WRITE(NTP,1000) THETA ***** 0116
C 1000 FORMAT(T2,'THETA=',F12.4) ***** 0117
C IF(ABS(THETA).GT.1000.) GO TO 115
C
C CHECK CONVERGENCE (THETA=1.) ***** 0118
C ERROR=1.-THETA ***** 0119
C IF(ABS(ERROR).LE.ACCLIM) GO TO 1112 ***** 0120
C IF(ABS(ERRD).LE.ACCLIM.AND.ABS(ERROR).LE.0.0001) GO TO 1112 ***** 0121
C GO TO 116 ***** 0122
C 115 CONTINUE ***** 0123
C THETA=10000. ***** 0124
C 116 CONTINUE *****
C
C CORRECTED B(I)/D(I) RATIOS AND B(I) & D(I) VALUES ***** 0125
C CALL BDCORR(B,BD,D,FTOT,NC,SLIQ,SUMD,SUMB,THETA,Z,MC,MP) ***** 0126
C DO 11 I=1,NC ***** 0127
C X(I,NP2)=D(I)/SUMD ***** 0128
C X(I,1)=B(I)/SUMB ***** 0129
C 11 CONTINUE

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MAINLINE MULCOM

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C C STRIPPING SECTION: CORRECTED VAPOUR & LIQUID MOLAR FLOWRATES. 0130
C CALL VLSTRP
C
C RECTIFYING SECTION: CORRECTED VAPOUR & LIQUID MOLAR FLOWRATES. 0131
C CALL VLRECT
C
C NFDO=NP1 0132
C IF(MEFF.EQ.0.OR.NCOND.GT.1) GO TO 117 0133
C GO TO 118 0134
C 117 NP1=NP2 0135
C GO TO 119 0136
C 118 CONTINUE 0137
C FOR INEFFICIENT SYSTEMS A TOTAL CONDENSER IS ASSUMED TO BE PERFECT.
C FOR SUCH CASES THE CONDENSER TEMPERATURE IS CALCULATED HERE *****
C MMD=2 0138
C CALL DEWCAL(NP2,MMD,TDDEWP) 0139
C TD=TEMP(NP2) 0140
C CALL BUBCAL(NP2,MMD,TBUBP) 0141
C TEMP(NP2)=(TD+TEMP(NP2))/2, 0142
C NEW TEMPERATURE PROFILE THROUGH THE COLUMN *****
C 119 CONTINUE 0143
C DO 12 J=1,NP1 0144
C MMD=1 0145
C CALL DEWCAL(J,MMD,TDDEWP) 0146
C TD=TEMP(J) 0147
C CALL BUBCAL(J,MMD,TBUBP) 0148
C TEMP(J)=(TD+TEMP(J))/2. 0149
C 12 CONTINUE 0150
C NP1=NFDO 0151
C ICOUNT=ICOUNT+1 0152
C MAXIMUM OF 30 ITERATIONS ALLOWED FOR CONVERGENCE.
C IF(ICOUNT.GE.30) GO TO 1112 0153
C RETURN FOR NEXT ITERATION *****
C GO TO 1111 0154
C 1112 CONTINUE 0155
C WRITE(NTP,5112) 0156
C 5112 FORMAT(/T2,'COMPUTATION COMPLETE.',/) 0157

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C WRITE RESULTS TO TERMINAL ***** 0158
CALL RUNRES *****

C RETURN FOR NEXT RUN WHEN NO DETAILED RESULTS ARE REQUIRED *****
C IF(NOUT.GE.5) GO TO 1115 0159
WRITE(NTP,5018) 0160
5018 FORMAT(/T2,'DO YOU WANT A FULL PRINT-OUT?') 0161
READ(NCR,2) PROUT 0162
IF(PROUT.EQ.'NO') GO TO 1115 0163
NOUT=NOUT+1 0164
IF(NOUT.EQ.5) WRITE(NTP,8091) 0165
8091 FORMAT(T2,'THIS IS THE LAST FULL PRINT-OUT FOR THIS EXECUTION.') 0166

C PRINTOUT OF DETAILED RESULTS *****
C CALL OUTPUT(ERROR,ERRD) 0167

C WRITE TO FILE FOR GRAPHICAL OUTPUT *****
8080 FORMAT(A4) 0168
8081 FORMAT(3I4) 0169
8082 FORMAT(2F10.2) 0170
8083 FORMAT(F10.2,F7.4) 0171
IUSEA2=IUSEA2+2 0172
VPROF(NP2)=DTOT 0173
PRD='DATA' 0174
WRITE(NG1,8080) PRD 0175
WRITE(NG1,8081) NP2,IUSEA1,IUSEA2 0176
WRITE(NG1,8082) (TEMP(J),TRAYNO(J),J=NP2,1,-1) 0177
WRITE(NG2,8080) PRD 0178
WRITE(NG2,8081) NP2,IUSEB1,IUSEA2 0179
WRITE(NG2,8082) (LPROF(J),TRAYNO(J),J=NP2,1,-1) 0180
IUSEB2=IUSEA2-1 0181
WRITE(NG2,8080) PRD 0182
WRITE(NG2,8081) NP2,IUSEB1,IUSEB2 0183
WRITE(NG2,8082) (VPROF(J),TRAYNO(J),J=NP2,1,-1) 0184
WRITE(NG3,8080) PRD 0185
WRITE(NG3,8081) NC,IUSEA1,IUSEA2 0186
WRITE(NG3,8083) (COMPNO(I),X(I,NP2),I=1,NC) 0187

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IUSE1=1
IUSE3=1
DO 86 I=1,NC
WRITE(NG4,8080) PRD
WRITE(NG4,8081) NP2,IUSE1,IUSE3
WRITE(NG4,8083) (TRAYNO(J),X(I,J),J=NP2,1,-1)
IUSE3=IUSE3+1
86 CONTINUE
C RETURN FOR NEXT RUN AFTER WRITING FULL RESULTS TO FILES *****
GO TO 1115
C
C ERROR MESSAGES AND TERMINATION ROUTINES *****
991 CONTINUE
WRITE(NTP,9911)
9911 FORMAT(/T2,'DATA NOT ACCEPTABLE.')
```

```

WRITE(NTP,9912)
9912 FORMAT(/T2,'ILLEGAL SPECIFICATION ENCOUNTERED',/T2,
1'REFER TO THE USER MANUAL FOR THE CORRECT PROCEDURE',/T2,
2'AND START THE PROGRAM AGAIN.')
```

```

CALL ERTRAN(6,'@ADD DISTFILE.FINISH .')
```

```

GO TO 999
901 CONTINUE
WRITE(NTP,9901)
9901 FORMAT(/T2,'SUCCESSFUL EXECUTION COMPLETED.')
```

```

IF(NOUT.LT.1) GO TO 902
WRITE(NTP,9902)
9902 FORMAT(/T2,'TO SEND FULL PRINT-OUT TO LINE PRINTER,ENTER THE',
1/T2,'FOLLOWING COMMANDS:',/T2,'@ADD DISTFILE.PRT',/T2,
2'@ED.U PRINTOUT',/T2,'PRINT:',/T2,'OMIT',/T2,'@ADD DISTFILE.PR2')
```

```

WRITE(NTP,9903)
9903 FORMAT(/T2,'COLLECT FULL PRINT-OUT FROM COMPUTER CENTRE.')
```

```

902 CONTINUE
IF(NOUT.LT.1) CALL ERTRAN(6,'@ADD DISTFILE.FINISH .')
```

```

999 CONTINUE
WRITE(NTP,9999)
9999 FORMAT(/T2,'END OF PROGRAM.',/)
```

```

STOP
END
```

0001

SUBROUTINE READIN(MCAN)  
ROUTINE FOR READING DATA FROM EITHER FILE OR TERMINAL.FOR  
MULTICOMPONENT DISTILLATION PROBLEMS.

\*\*\*\*\*  
INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE,  
\*\*\*\*\*

DIMENSION F1(5),F2(5),F3(5),F4(5),U1(4),U2(4),U3(4),U4(4)  
REAL K,LMFR,LRTOT,LSTOT,LD,LB,LPROF

DATA F1/'@ASG'UP , ,F17 , ,/  
DATA F2/'@ASG'UP , ,F17 , ,/  
DATA F3/'@ASG'UP , ,F17 , ,/  
DATA F4/'@ASG'UP , ,F17 , ,/  
DATA U1/'@USE 12, , , ,/  
DATA U2/'@USE 13, , , ,/  
DATA U3/'@USE 14, , , ,/  
DATA U4/'@USE 15, , , ,/  
1 FORMAT( )  
2 FORMAT(A3)  
3 FORMAT(5A6)  
4 FORMAT(4(A6,1X))

DATA ATM,FACTP1,FACTP2/14.696,51.714752,7.50061683/  
IF(MCAN.EQ.1) WRITE(NTP,1001)

1001 FORMAT(T2,'ENTER ALL DATA IN FREE FORMAT.,./T2,'ENTER NO. OF COMPO  
NENTS IN FEED.')

READ(NCR,1) NC  
IF(MCAN.EQ.1) WRITE(NTP,1002)

1002 FORMAT(T2,'ENTER NO. OF TRAYS IN COLUMN.,./T2,'EXCLUDE CONDENSER A  
ND REBOILER.')

READ(NCR,1) NP

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SUBROUTINE READIN

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0025 IF(MCAN.EQ.1) WRITE(NTP,1003)
0026 FORMAT(T2,ENTER FEED TRAY POSN. (NO. FROM BOTTOM REBOILER=0),)
0027
0028 READ(NCR,1) NF
0029 NP1=NP+1
0030 NP2=NP+2
0031 NF1=NF+1
0032 NF2=NF+2
0033 IF(NC.GT.MC.OR.NP2.GE.MP) GO TO 902
0034 IF(MCAN.EQ.1) WRITE(NTP,1004)
0035 FORMAT(T4,ACCEPTED PRESSURE UNITS: 1=MMHG,2=PSIG,3=PSIA,4=KPA,5=A
0036 ITM,1/T2,ENTER CONDENSER PRESSURE AND UNIT CODE. (EG. 760.0,1),)
0037 READ(NCR,1) PCONDA,NUNITP
0038 IF(NUNITP.EQ.1) PCOND=PCONDA
0039 IF(NUNITP.EQ.2) PCOND=(PCONDA*ATM)*FACTP1
0040 IF(NUNITP.EQ.3) PCOND=PCONDA*FACTP1
0041 IF(NUNITP.EQ.4) PCOND=PCONDA*FACTP2
0042 IF(NUNITP.EQ.5) PCOND=PCONDA*ATM*FACTP1
0043 IF(MCAN.EQ.1) WRITE(NTP,1005)
0044 FORMAT(T4,ACCEPTED PRESSURE DROP UNITS: 1=MMHG,2=PSI,3=KPA,
0045 1/T2,ENTER PRESSURE DROP PER TRAY AND UNIT CODE.,)
0046 READ(NCR,1) DELTPD,NUNIPD
0047 IF(NUNIPD.EQ.1) DELTAP=DELTPD
0048 IF(NUNIPD.EQ.2) DELTAP=DELTPD*FACTP1
0049 IF(NUNIPD.EQ.3) DELTAP=DELTPD*FACTP2
0050 IF(MCAN.EQ.1) WRITE(NTP,1105)
0051 FORMAT(T2,ENTER PRESSURE DROP ACROSS CONDENSER AND UNIT CODE.,)
0052 READ(NCR,1) DELTPC,NUNIPC
0053 IF(NUNIPC.EQ.1) DELTAC=DELTPC
0054 IF(NUNIPC.EQ.2) DELTAC=DELTPC*FACTP1
0055 IF(NUNIPC.EQ.3) DELTAC=DELTPC*FACTP2

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SUBROUTINE READIN



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0055 IF(MCAN.EQ.1) WRITE(NTP,1006)
0056 FORMAT(14,'ACCEPTED TEMPERATURE UNITS: 1=DEG C,2=DEG F,./T2,
0057 1=ENTER ESTIMATE OF REBOILER AND CONDENSER TEMPERATURES WITH UNIT C
0058 20DES.,/T2,.(EG. 100.0,1.50,0.2)')
0059 READ(NCR,1) TBOG,NUTB,TTOPG,NUTT
0060 IF(NUTB.EQ.1) TBOT=TBOG
0061 IF(NUTB.EQ.2) TBOT=(TBOG-32.)/1.8
0062 IF(NUTT.EQ.1) TTOP=TTOPG
0063 IF(NUTT.EQ.2) TTOP=(TTOPG-32.)/1.8
0064 IF(MCAN.EQ.1) WRITE(NTP,1007)
0065 1007 FORMAT(T2,'ENTER NAMES OF COMPONENTS. (ONE PER LINE,MAX 30 CHARACT
0066 IERS EACH.)')
0067 READ(NCR,3) ((TTILES(I,J),J=1,5),I=1,NC)
0068 IF(MCAN.EQ.1) WRITE(NTP,1008)
0069 1008 FORMAT(T2,'ENTER FEED RATE. (KG MOLES/HR.)')
0070 READ(NCR,1) FTOT
0071 IF(MCAN.EQ.1) WRITE(NTP,1009)
0072 1009 FORMAT(T2,'ENTER FEED STREAM COMPOSITIONS, (MOL FRACTIONS)')
0073 READ(NCR,1) (Z(I),I=1,NC)
0074 IF(MCAN.EQ.1) WRITE(NTP,2110)
0075 2110 FORMAT(T2,'ENTER FEED TEMPERATURE. (DEG C)')
0076 READ(NCR,1) TFEED
0077 NQVAL=1
0078 NENTH=10
0079 IF(MCAN.EQ.1) WRITE(NTP,1011)
0080 1011 FORMAT(T2,'ARE ENTHALPY EFFECTS TO BE CONSIDERED?')
0081 READ(NCR,2) ENTHYP
0082 IF(ENTHYP.NE.'YES') NENTH=0
0083 IF(NENTH.EQ.0) GO TO 113
0084 IF(MCAN.EQ.1) WRITE(NTP,2111)
0085 2111 FORMAT(T2,'IS FEED PRESSURE GREATER THAN COLUMN PRESSURE?')
0086 READ(NCR,2) FPRES

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SUBROUTINE READIN

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IF(FPRES.EQ.'NO') GO TO 113
IF(MCAN.EQ.1) WRITE(NTP,2112)
2112 FORMAT(T2,'ENTER FEED ENTHALPY, (KJ/KGMOLE).')
READ(NCR,1) FENTH
NQVAL=2
113 CONTINUE
IF(MCAN.EQ.1) WRITE(NTP,1010)
1010 FORMAT(T4,'TOTAL OR PARTIAL CONDENSER?','/T2,'ENTER TOT OR PAR.')
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110 CONTINUE
READ(NCR,2) CONTYP
IF(CONTYP.NE.'TOT'.AND.CONTYP.NE.'PAR') GO TO 901
IF(CONTYP.EQ.'TOT') GO TO 111
NCOND=2
GO TO 112
111 CONTINUE
NCOND=1
112 CONTINUE
IF(ENTHYP.NE.'YES') GO TO 120
IF(MCAN.EQ.1) WRITE(NTP,1012)
1012 FORMAT(T2,'DO YOU WANT TO USE YOUR OWN ENTHALPY ROUTINE?')
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READ(NCR,2) OWNENT
IF(OWNENT.EQ.'YES') GO TO 114
IF(MCAN.EQ.1) WRITE(NTP,1013)
1013 FORMAT(T4,'REFER TO MANUAL,SECTION 4,FOR ENTHALPY ROUTINE CODES.,',
1/T2,'ENTER CODE NO. OF LIBRARY ENTHALPY ROUTINE REQUIRED.')
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```

READ(NCR,1) NENTH
NOWN=1
IF(MCAN.EQ.1) WRITE(NTP,1113)
1113 FORMAT(T2,'ENTER ENTHALPY DATA AS DESCRIBED IN THE MANUAL,SECTION
14.,/T2,'OR ADD A DATA ELEMENT.')
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READ(NCR,1) (HL1(I),HL2(I),HL3(I),I=1,NC)
READ(NCR,1) (HV1(I),HV2(I),HV3(I),I=1,NC)
GO TO 120
114 CONTINUE
```

SUBROUTINE READIN

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1014 IF(MCAN.EQ.1) WRITE(NTP,1014)
1015 FORMAT(T2,'IT IS ASSUMED THAT THE ROUTINE CONFORMS TO SPECIFICATIO
1016 NS',/T2,'SET OUT IN SECTION 4 IN THE MANUAL.')
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1017 NENTH=5
1018 NOWN=2
1019 CONTINUE
1020 IF(MCAN.EQ.1) WRITE(NTP,1018)
1021 FORMAT(T2,'ENTER EQUIL. DATA AS DESCRIBED IN THE MANUAL,SECTION 4.
1022 ',/T2,'OR ADD A DATA ELEMENT.')
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1023 READ(NCR,1) NEQUIL
1024 IF(NEQUIL.EQ.1) GO TO 131
1025 IF(NEQUIL.EQ.3) GO TO 132
1026 GO TO 133
1027 CONTINUE
1028 READ(NCR,1) (AN(I),RN(I),CN(I),I=1,NC)
1029 GO TO 134
1030 CONTINUE
1031 READ(NCR,1) AKB1,AKB2,AKB3,AKB4
1032 CONTINUE
1033 READ(NCR,1) (AK1(I),AK2(I),AK3(I),AK4(I),I=1,NC)
1034 IF(NEQUIL.NE.3) GO TO 134
1035 DO 10 I=1,NC
1036 IF(ABS(AK2(I)).LT.0.00001) NJ=I
1037 CONTINUE
1038 CONTINUE
1039 IF(MCAN.EQ.1) WRITE(NTP,1019)
1040 FORMAT(T2,'DO YOU WANT TO SPECIFY THE THETA COVERAGE LIMIT?')
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1041 READ(NCR,2) CONLIM
1042 IF(CONLIM.EQ.'YES') GO TO 135
1043 ACCLIM=0.00005
1044 GO TO 136
1045 CONTINUE
1046 IF(MCAN.EQ.1) WRITE(NTP,1020)
1047 FORMAT(T2,'ENTER CONVERGENCE LIMIT.')
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1048 READ(NCR,1) ACCLIM
1049 CONTINUE
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SUBROUTINE READIN

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1021 IF(MCAN.EQ.1) WRITE(NTP,1021)
      FORMAT(T2,'ARE PERFECT TRAYS ASSUMED?')
      EFFVAP=1.0
      MEFF=0
      READ(NCR,2) EFFCRI
      IF(EFFCRI.EQ.'YES') GO TO 137
      IF(MCAN.EQ.1) WRITE(NTP,1022)
1022 FORMAT(T2,'ADD EFFICIENCY DATA AS DESCRIBED IN THE MANUAL SECTION
      14.')
      READ(NCR,1) MEFF
      IF(MEFF.EQ.2) GO TO 138
      READ(NCR,1) (EFFV(I,J),I=1,NC),J=NP2,1.-1)
      GO TO 139
138 CONTINUE
      READ(NCR,1) (EFFV(I,1),I=1,NC)
      DO 11 I=1,NC
      DO 12 J=2,NP2
      EFFV(I,J)=EFFV(I,1)
12 CONTINUE
11 CONTINUE
      GO TO 139
137 CONTINUE
      DO 13 J=1,NP2
      DO 14 I=1,NC
      EFFV(I,J)=EFFVAP
14 CONTINUE
13 CONTINUE
139 CONTINUE
      IF(MCAN.EQ.1) WRITE(NTP,1024)
1024 FORMAT(T2,'ENTER REFLUX RATIO AND DISTILLATE WITHDRAWAL RATE.')
      READ(NCR,1) RR,DTOT

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SUBROUTINE READIN

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1025 WRITE(NTP,1025)
      FORMAT(T2,'DO YOU WANT GRAPHICAL OUTPUT? (SEE MANUAL SECTION 4)')
1026 READ(NCR,2) USEF1
      IF(USEF1.NE.'YES') GO TO 140
      WRITE(NTP,1026)
      FORMAT(T2,'DO YOU KNOW HOW TO USE THE PLOT ROUTINE?')
1027 READ(NCR,2) USEF2
      IF(USEF2.NE.'YES') GO TO 140
      WRITE(NTP,1027)
      FORMAT(T2,'ENTER DATA-FILE NAMES. (SEE MANUAL SECTION 4)')
      READ(NCR,4) F1(3),F2(3),F3(3),F4(3)
      U1(3)=F1(3)
      U2(3)=F2(3)
      U3(3)=F3(3)
      U4(3)=F4(3)
      CALL ERTRAN(6,F1)
      CALL ERTRAN(6,F2)
      CALL ERTRAN(6,F3)
      CALL ERTRAN(6,F4)
      CALL ERTRAN(6,U1)
      CALL ERTRAN(6,U2)
      CALL ERTRAN(6,U3)
      CALL ERTRAN(6,U4)
140 CONTINUE
      GO TO 999
C      ERROR MESSAGES *****
901 CONTINUE
      IF(MCAN.EQ.1) WRITE(NTP,9001)
9001 FORMAT(T2,'ONLY TOT OR PAR IS ACCEPTABLE.'/T2,'RETYPE NOW.')
      GO TO 110
902 CONTINUE
      WRITE(NTP,9002)
9002 FORMAT(T2,'DIMENSION SPECIFICATIONS EXCEEDED.')
      NJ=1000
999 CONTINUE
      RETURN
      END

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SUBROUTINE READIN

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SUBROUTINE EQUCON(NEQ)
C CALCULATION OF EQUILIBRIUM CONSTANTS.
C CHOICE OF FOUR PROCEDURES MADE BY COUNTER NEQUIL
C (1) USE OF ANTOINE'S EQUATION, HENRY'S LAW AND RAOULT'S LAW.
C IE. LOG10(PVAP)=AN-BN/(CN+T), Y=K*X, Y=PVAP*X/PT
C PV=VAPOUR PRESS, PT=TOTAL PRESS, T IN DEG C
C A,B,C ARE ANTOINE CONSTS FOR COMPONENT
C (2) EQUILIBRIUM CONSTANT EXPRESSED AS A FUNCTION OF TEMPERATURE
C IE. K=EXP(AK1-AK2/T+AK3*T) T IN DEG R
C (3) RELATIVE VOLATILITY EXPRESSED AS A FUNCTION OF TEMPERATURE
C WITH A KEY COMPONENT EQUIL CONST AS A FUNCTION OF T, AND THE
C EQUATION: REL VOL=K(I)/AKB (AKB=KEY COMP EQUIL CONST)
C FUNCTIONS ARE: REL VOL=AK1+AK2*T+AK3*T**2 T IN DEG F
C AKB=AKB1+AKB2*T+AKB3*T**2+AKB4*T**3 T IN DEG F
C (4) EQUILIBRIUM CONSTANT EXPRESSED AS A FUNCTION OF TEMPERATURE
C IE. K=T*(AK1+AK2*T+AK3*T**2+AK4*T**3)**3 T IN DEG R
C *****
C INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
C *****
C REAL K,KF,LMFR,LRTOT,LSTOT,LD,LB,LPROF
C GO TO (100,200,300,400),NEQUIL
100 CONTINUE
IF(NEQ.EQ.1) GO TO 110
C CALCULATION USING OPTION (1) ABOVE.
DO 11 J=1,NP2
DO 12 I=1,NC
PVAP=10.**((AN(I)-BN(I))/(CN(I))+TEMP(J)))
K(I,J)=PVAP*EFFV(I,J)/PRESS(J)
12 CONTINUE
11 CONTINUE
110 CONTINUE
DO 13 I=1,NC
PVAP=10.**((AN(I)-BN(I))/(CN(I))+TFEED))
KF(I)=PVAP/PRESS(NFI)
13 CONTINUE
GO TO 999

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SUBROUTINE EQUCON

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200 CONTINUE
  IF(NEQ.EQ.1) GO TO 210
  CALCULATION USING OPTION (2) ABOVE.
  DO 21 J=1,NP2
    T1=TEMP(J)*1.8+32.+459.4
    T=T1/100.
    DO 22 I=1,NC
      K(I,J)=EXP(AK1(I)-AK2(I)/T+AK3(I)*T)*EFFV(I,J)
22 CONTINUE
21 CONTINUE
210 CONTINUE
    T1=TFEED*1.8+32.+459.4
    T=T1/100.
    DO 23 I=1,NC
      KF(I)=EXP(AK1(I)-AK2(I)/T+AK3(I)*T)
23 CONTINUE
    GO TO 999
300 CONTINUE
  IF(NEQ.EQ.1) GO TO 310
  CALCULATION USING OPTION (3) ABOVE.
  CALCULATION USING OPTION (3) ABOVE.
  DO 31 J=1,NP2
    T1=TEMP(J)*1.8+32.
    T=T1/100.
    AKB=AKB1+AKB2*T+AKB3*T**2.+AKB4*T**3.
    DO 32 I=1,NC
      K(I,J)=(AK1(I)+AK2(I)*T+AK3(I)*T**2.)*AKB*EFFV(I,J)
32 CONTINUE
31 CONTINUE
310 CONTINUE
    T1=TFEED*1.8+32.
    T=T1/100.
    AKB=AKB1+AKB2*T+AKB3*T**2.+AKB4*T**3.
    DO 33 I=1,NC
      KF(I)=(AK1(I)+AK2(I)*T+AK3(I)*T**2.)*AKB
33 CONTINUE
    GO TO 999

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400 CONTINUE
IF(NEQ.EQ.1) GO TO 410
C CALCULATION USING OPTION (4) ABOVE.
DO 41 J=1,NP2
T1=TEMP(J)*1.8+32.+459.4
T=T1/1000.
DO 42 I=1,NC
AK=T1*(AK1(I)+AK2(I)+AK3(I)*T**2.+AK4(I)*T**3.)***3.
K(I,J)=AK*EFFV(I,J)
42 CONTINUE
41 CONTINUE
410 CONTINUE
T1=TFEED*1.8+32.+459.4
T=T1/1000.
DO 43 I=1,NC
KF(I)=T1*(AK1(I)+AK2(I)*T+AK3(I)*T**2.+AK4(I)*T**3.)***3.
43 CONTINUE
999 CONTINUE
RETURN
END

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SUBROUTINE EQUCON



0001

SUBROUTINE ENERGY(TBUB,TDew,NEN,S1,S2,S3,S4)  
 CALCULATION OF VAPOUR AND LIQUID ENTHALPY DATA,FEED ENTHALPY,  
 CONDENSER DUTY AND REBOILER DUTY.  
 CHOICE OF THREE AVAILABLE FUNCTIONS MADE BY NENTH,  
 FUNCTIONS ARE:

(A) H=(A+B\*T+C\*T\*\*2)\*\*2 T1 IN DEG R (NENTH=1)  
 (B) H=A+B\*T+C\*T\*\*2 T1 IN DEG F (NENTH=2)  
 (C) H=A+B\*T+C\*T\*\*2 T1 IN DEG R (NENTH=3)

NOTE: T=T1/100. AND T1 IS TEMPERATURE ON TRAY N

H = VAPOUR OR LIQUID MOLAR ENTHALPY FOR COMPONENT,  
 DIFFERENT COEFFICIENTS ARE NEEDED FOR EACH PHASE.

\*\*\*\*\*  
 INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.  
 \*\*\*\*\*

REAL K,KF,LWFR,LRTOT,LSTOT,LD,LB,LPROF  
 DATA CONFAC,ENCONV/0.00027778,2.326/  
 TRANK=0.0  
 POWER=1.

IF(NENTH.EQ.1) POWER=2.  
 IF(NENTH.EQ.1.OR.NENTH.EQ.3) TRANK=459.4  
 IF(NENTH.EQ.1) GO TO 200  
 IF(NENTH.EQ.2) GO TO 103

CALCULATION OF ENTHALPY OF VAPOUR AND LIQUID STREAMS.

DO 10 J=1,NP2  
 T1=TEMP(J)\*1.8+32.+TRANK  
 T=T1/100.

SUMHV=0.  
 SUMHL=0.

DO 11 I=1,NC  
 HV=(HV1(I)+HV2(I)\*T+HV3(I)\*T\*\*2.)\*POWER\*(I,J)  
 HL=(HL1(I)+HL2(I)\*T+HL3(I)\*T\*\*2.)\*POWER\*(I,J)  
 SUMHV=SUMHV+HV  
 SUMHL=SUMHL+HL

11 CONTINUE

HVAP(J)=SUMHV\*ENCONV  
 HLIQ(J)=SUMHL\*ENCONV

10 CONTINUE

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SUBROUTINE ENERGY

C	CALCULATION OF CONDENSER DUTY.	0024
	HD=HLIQ(NP2)	0025
	IF(NCOND.GT.1) GO TO 102	
C	TOTAL CONDENSER OPERATION.	0026
	QC=VRTOT(NP1)*(HVAP(NP1)-HLIQ(NP2))	0027
	GO TO 103	0028
	102 CONTINUE	
C	PARTIAL CONDENSER OPERATION.	0029
	QC=VRTOT(NP1)*HVAP(NP1)-LRTOT(NP2)*HLIQ(NP2)-DTOT*HVAP(NP2)	0030
	HD=HVAP(NP2)	0031
	103 CONTINUE	
C	CALCULATION OF FEED STREAM ENTHALPY.	0032
	SUMFL=D.	0033
	SUMFV=D.	0034
	TI=TFEED*1.8+32.+TRANK	0035
	T=TI/100.	0036
	DO 12 I=1,NC	0037
	HV=(HV1(I)+HV2(I)*T+HV3(I)*T**2.)*POWER*YF(I)	0038
	HL=(HL1(I)+HL2(I)*T+HL3(I)*T**2.)*POWER*XF(I)	0039
	SUMFV=SUMFV+HV	0040
	SUMFL=SUMFL+HL	0041
	12 CONTINUE	0042
	HFEED=(AVAP/FTOT*SUMFV+ALIQ/FTOT*SUMFL)*ENCONV	0043
	IF(NEN.EQ.2) GO TO 999	0044
	QFTOT=FTOT*HFEED	
C	CALCULATION OF REBOILER DUTY.	0045
	QR=DTOT*HD+BTOT*HLIQ(I)+QC-QFTOT	0046
	DM=HD+QC/DTOT	0047
	BM=HLIQ(I)-QR/BTOT	
C	CONVERSION TO S.I. UNITS	0048
	QC=QC*CONFAC	0049
	QR=QR*CONFAC	0050
	QF=HFEED	0051
	GO TO 999	

SUBROUTINE ENERGY

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200 CONTINUE
C    CALCULATION OF ENTHALPY FOR Q VALUE DETERMINATION.
T1=TFEED*1.8+32.+TRANK
T=T1/100.
TB=(TBUB*1.8+32.+TRANK)/100.
TD=(TDEW*1.8+32.+TRANK)/100.
S1=0.
S2=0.
S3=0.
S4=0.
DO 21 I=1,NC
HF1=((HV1(I)+HV2(I)*T+HV3(I)*T**2.)*POWER*Z(I))*ENCONV
HF2=((HL1(I)+HL2(I)*T+HL3(I)*T**2.)*POWER*Z(I))*ENCONV
HV=((HV1(I)+HV2(I)*TD+HV3(I)*TD**2.)*POWER*Z(I))*ENCONV
HL=((HL1(I)+HL2(I)*TB+HL3(I)*TB**2.)*POWER*Z(I))*ENCONV
S1=S1+HF1
S2=S2+HF2
S3=S3+HV
S4=S4+HL
21 CONTINUE
999 CONTINUE
RETURN
END

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SUBROUTINE ENERGY

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0001
SUBROUTINE QFLASH(NTIME,TDEWPT)
C CALCULATION OF COMPOSITION AND FLOWRATE OF THE VAPOUR AND LIQUID
C FRACTIONS OF THE FEED STREAM.
C THE SUBROUTINE CALCULATES THE FEED STREAM Q VALUE FOR EITHER FEED
C TEMPERATURE OR FEED ENTHALPY SPECIFICATIONS.
C IF THE SPECIFIED TEMPERATURE DOES NOT LIE WITHIN THE BUBBLE PT-
C DEW PT RANGE, AND ENTHALPY DATA ARE NOT SUPPLIED THEN A Q VALUE
C MUST BE GIVEN.
C THE REQUIRED PROCEDURE IS INDICATED BY THE COUNTER NQVAL.
C *****
C INSERT DIMENSION AND COMMON BLOCK *MULTICOM* HERE.
C *****
C REAL K,KF,LMFR,LRTOT,LSTOT,LD,LB,LPROF
1 FORMAT( )
NTYP=1
NVAP=NCOND
SUM1=1.
IT=NP2+1
DO 10 I=1,NC
XF(I)=Z(I)
YF(I)=Z(I)
X(I,IT)=Z(I)
Y(I,IT)=Z(I)
10 CONTINUE
PRESS(IT)=PRESS(NF1)
TEMP(IT)=TFEED
MD=2
CALL BUBCAL(IT,MD,TBUBPT)
CALL DEWCAL(IT,MD,TDEWPT)
GO TO (100,300),NQVAL

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SUBROUTINE QFLASH

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C      100 CONTINUE
      ISOBARIC FLASH CALCULATION.
      IF (TFEED.LT.TBOURPT.OR.TFEED.GT.TDEWPT) GO TO 120
      Q=0.01
      DELQ=0.001
      MN=1
      NEQ=1
      CALL EQUCON(NEQ)
      110 CONTINUE
      SUM=0.
      DO 11 I=1,NC
      XF(I)=Z(I)/(KF(I)-Q*KF(I)+Q)
      SUM=SUM+XF(I)
      11 CONTINUE
      IF (ABS(SUM-SUM1).LT.0.0001) GO TO 111
      NCALL=2
      CALL SEARCH(SUM,SUM1,Q,DELQ,MN,NCALL)
      GO TO 110
      111 CONTINUE
      SUMY=0.
      DO 12 I=1,NC
      YF(I)=XF(I)*KF(I)
      SUMY=SUMY+YF(I)
      12 CONTINUE
      DO 13 I=1,NC
      YF(I)=YF(I)/SUMY
      13 CONTINUE
      AVAP=(1.-Q)*FTOT
      ALIQ=Q*FTOT
      GO TO 999

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SUBROUTINE QFLASH

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120 CONTINUE
IF(NENTH.EQ.0) GO TO 200
FEED TEMPERATURE ABOVE DEW PT OR BELOW BUBBLE PT.
NEN=1
IF(NOWN.EQ.1) CALL ENERGY(TBUBPT,TDEWPT,NEN,S1,S2,S3,S4)
IF(NOWN.EQ.2) CALL ENTHAL(ALIQ,AVAP,BM,DM,BTOT,DTOT,FTOT,HD,HFEED,
IHLIQ,HVAP,NC,NF1,NP1,NP2,NVAP,QC,QF,QR,TEMP,VRTOT,X,XF,Y,YF,TBUBPT
2,TDEWPT,TFEED,NEN,S1,S2,S3,S4,NTIME,MC,MP)
121 CONTINUE
IF(TFEED,LT,TBUBPT) GO TO 122
IF(NTYP.EQ.1) Q=(S3-S1)/(S3-S4)
DO 15 I=1,NC
XF(I)=0.0
YF(I)=Z(I)
15 CONTINUE
AVAP=FTOT
ALIQ=0.0
GO TO 999
122 CONTINUE
IF(NTYP.EQ.1) Q=(S3-S2)/(S3-S4)
DO 16 I=1,NC
XF(I)=Z(I)
YF(I)=0.0
16 CONTINUE
AVAP=0.0
ALIQ=FTOT
GO TO 999
200 CONTINUE
C USER MUST SPECIFY FEED Q VALUE IF ENTHALPY EFFECTS ARE IGNORED.
IF(TFEED,LT,TBUBPT) WRITE(NTP,2001) TBUBPT
2001 FORMAT(/T2,'FEED IS BELOW ITS BUBBLE POINT OF',F8.2,1X,'DEG C',
1/T2,'ENTER FEED Q VALUE.')
IF(TFEED,GT,TDEWPT) WRITE(NTP,2002) TDEWPT
2002 FORMAT(/T2,'FEED IS VAPOUR ABOVE ITS DEW POINT OF',F8.2,1X,'DEG C',
1/T2,'ENTER FEED Q VALUE.')
READ(NCR,1) Q
NTYP=2
GO TO 121

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SUBROUTINE QFLASH

300 CONTINUE  
C ISENTHALPIC FLASH CALCULATION.

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NEN=1
IF(NOWN,EQ.1) CALL ENERGY(TBUBPT,TDEWPT,NEN,S1,S2,S3,S4)
IF(NOWN,EQ.2) CALL ENTHAL(ALIQ,AVAP,BM,DM,BTOT,DTOT,FTOT,HD,HFEED,
IHLIQ,HVAP,NC,NF1,NP1,NP2,NVAP,QC,QF,QR,TEMP,VRTOT,X,XF,Y,YF,TBUBPT
2,TDEWPT,TFEED,NEN,S1,S2,S3,S4,NTIME,NC,MP)
IF(S3,LT,FENTH,OR,S4,GT,FENTH) GO TO 320
FACT=0.5
FV=FACT*FTOT
DRLF=0.1
MM=1
DO 31 J=1,1000
TFEED=TBUBPT+FACT*(TDEWPT-TBUBPT)
FV=FACT*FTOT
FL=FTOT-FV
IF(FV,LT,0.0,OR,FL,LT,0.0) GO TO 901
MN=1
DELT=5.
310 CONTINUE
NEQ=1
CALL EQUCON(NEQ)
SUMXF=0.
DO 32 I=1,NC
XF(I)=FTOT*Z(I)/(FL+FV*KF(I))
SUMXF=SUMXF+XF(I)
32 CONTINUE
IF(ABS(SUMXF-SUM1),LT,0.0001) GO TO 311
NCALL=2
CALL SEARCH(SUMXF,SUM1,TFEED,DELT,MN,NCALL)
GO TO 310
311 CONTINUE
SUMYF=0.
DO 33 I=1,NC
YF(I)=XF(I)*KF(I)
SUMYF=SUMYF+YF(I)
33 CONTINUE

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SUBROUTINE QFLASH

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DO 34 I=1,NC
YF(I)=YF(I)/SUNYF
34 CONTINUE
AVAP=FV
ALIQ=FL
Q=FL/FTOT
NEN=2
IF(NOWN.EQ.1) CALL ENERGY(TBUBPT,TDEWPT,NEN,S1,S2,S3,S4)
IF(NOWN.EQ.2) CALL ENTHAL(ALIQ,AVAP,BM,DM,BTOT,DTOT,FTOT,HD,HFEED,
IHLIQ,HVAP,NC,NF1,NP1,NP2,NVAP,QC,QF,QR,TEMP,VRTOT,X,XF,Y,YF,TBUBPT
2,TFEED,TFEED,NEN,S1,S2,S3,S4,NTIME,MC,MP)
IF(ABS(HFEED-FENTH).LT.0.1) GO TO 999
NCALL=1
CALL SEARCH(HFEED,FENTH,FACT,DELFT,MN,NCALL)
31 CONTINUE
320 CONTINUE
IF(S3.LT.FENTH) GO TO 330
FEED TEMP CALC IF FEED IS SUPERCOOLED AT COLUMN PRESSURE,
TFEED=TBUBPT-1.
MN=1
DELT=1.
321 CONTINUE
NEN=1
IF(NOWN.EQ.1) CALL ENERGY(TBUBPT,TDEWPT,NEN,S1,S2,S3,S4)
IF(NOWN.EQ.2) CALL ENTHAL(ALIQ,AVAP,BM,DM,BTOT,DTOT,FTOT,HD,HFEED,
IHLIQ,HVAP,NC,NF1,NP1,NP2,NVAP,QC,QF,QR,TEMP,VRTOT,X,XF,Y,YF,TBUBPT
2,TDEWPT,TFEED,NEN,S1,S2,S3,S4,NTIME,MC,MP)
IF(ABS(S2-FENTH).LT.0.1) GO TO 121
NCALL=1
CALL SEARCH(S2,FENTH,TFEED,DELT,MN,NCALL)
GO TO 321
330 CONTINUE

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SUBROUTINE QFLASH



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C      FEED TEMP CALC IF FEED IS SUPERHEATED AT COLUMN PRESSURE.
      TFEED=TDEWPT+1.
      MN=1
      DELT=10.
331  CONTINUE
      NEN=1
      IF(NOWN.EQ.1) CALL ENERGY(TBUBPT,TDEWPT,NEN,S1,S2,S3,S4)
      IF(NOWN.EQ.2) CALL ENTHAL(ALIQ,AVAP,BM,DM,BTOT,DTOT,FTOT,HD,HFEED,
      IHLIQ,HVAP,NC,NF1,NP1,NP2,NVAP,QC,QF,QR,TEMP,VRTOT,X,XF,Y,YF,TBUBPT
      2,TDEWPT,TFEED,NEN,S1,S2,S3,S4,NTIME,MC,MP)
      IF(ABS(S1-FENTH).LT.0.1) GO TO 121
      NCALL=1
      CALL SEARCH(S1,FENTH,TFEED,DELT,MN,NCALL)
      GO TO 331
C      ERROR MESSAGES *****
901  CONTINUE
      WRITE(NTP,9001) FACT
9001  FORMAT(/T2,'ERROR IN FLASH CALCULATION','/T2,'SEARCH FOR FEED TEMP
      1ERATURE HAS FAILED','/T2,'VAPOUR FRACTION OF FEED=',F8.4,/T2,
      2'VALUE SHOULD LIE BETWEEN 0.0 AND 1.0.')
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      NJ=1000
999  CONTINUE
      WRITE(NTP,4000) Q,TFEED,TDEWPT,TBUBPT
4000  FORMAT(/T2,'FEED Q VALUE AND TEMPERATURE DATA','/T2,'Q:',F8.4,
      115X,'FEED TEMP:',F8.2,1X,'DEG C','/T2,'DEW PT:',F8.2,1X,'DEG C',4X,
      2'BUBBLE PT:',F8.2,1X,'DEG C',/)
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      RETURN
      END
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SUBROUTINE QFLASH

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SUBROUTINE BUBCAL(IT,MMD,TBUB)
BUBBLE POINT ROUTINE.
C FOUR CALCULATION METHODS ARE AVAILABLE. CHOICE DEPENDS ON THE
C TYPE OF EQUILIBRIUM DATA SPECIFIED.
C CHOICE MADE BY COUNTER NEQUIL.
C *****
C INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
C *****
C REAL K,LMFR,LRTOT,LSTOT,LD,LB,LPROF
C GO TO (100,200,300,400),NEQUIL

100 CONTINUE
C CALCULATION USING ANTOINE CONSTANTS.
TOLD=TEMP(IT)
DO 11 KTIMES=1,1000
ASUM=PRESS(IT)
BSUM=0.0
DO 12 I=1,NC
EFF=EFFV(I,IT)
IF(MMD.GT.1) EFF=1.0
ANT=AN(I)-BN(I)/(CN(I)+TOLD)
ASUM=ASUM-X(I,IT)*EFF*10.**ANT
BSQ=2.303*BN(I)/(CN(I)+TOLD)**2.)
BSUM=BSUM+X(I,IT)*EFF*BSQ*10.**ANT
12 CONTINUE
DIV=ASUM/BSUM
IF(ABS(DIV).LE.0.001) GO TO 110
TOLD=TOLD+DIV
IF(TOLD.LT.0.5) GO TO 999
11 CONTINUE
110 CONTINUE
C CHANGE IN TEMPERATURE FROM ONE ITERATION TO THE NEXT IS RESTRICTED
C TO HALF THE CALCULATED CHANGE.
TBUB=TOLD
DIFF=TEMP(IT)-TOLD
IF(DIFF) 113,114,113
113 TEMP(IT)=TOLD+DIFF/2.
GO TO 999
114 TEMP(IT)=TOLD
GO TO 999
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SUBROUTINE BUBCAL

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200 CONTINUE
C CALCULATION USING EQUILIBRIUM CONSTANT DATA, AS SPECIFIED BY
C SUBROUTINE ,EQUCON, (OPTION (2) IN EQUCON).
TOLD=TEMP(IT)*1.8+32.+459.4
250 CONTINUE
TOLD1=TOLD/100.
SUM=0.
SUMD=0.
DO 21 I=1,NC
EFF=EFFV(I,IT)
IF(MMD.GT.1) EFF=1.0
ALFA=EXP(AK1(I)-AK2(I)/TOLD1+AK3(I)*TOLD1)*EFF
DA=AK3(I)+AK2(I)/(TOLD1**2.)
DALFA=(EXP(AK1(I)-AK2(I)/TOLD1+AK3(I)*TOLD1))*DA*EFF
SUM=SUM+(X(I,IT)*ALFA)
SUMD=SUMD+(X(I,IT)*DALFA)
21 CONTINUE
FUNCT=SUM-1.
FUNCTD=SUMD
DIV=FUNCT/FUNCTD
IF(ABS(DIV).LT.0.001) GO TO 210
TOLD=TOLD-DIV
GO TO 250
210 TNEW=(TOLD-32.-459.4)/1.8
C CHANGE IN TEMPERATURE FROM ONE ITERATION TO THE NEXT IS RESTRICTED
C TO HALF THE CALCULATED CHANGE.
TBUB=TNEW
DIFF=TEMP(IT)-TNEW
IF(DIFF) 211,212,211
211 TEMP(IT)=TNEW+DIFF/2.
GO TO 999
212 TEMP(IT)=TNEW
GO TO 999

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SUBROUTINE BUBCAL

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300 CONTINUE
C CALCULATION USING RELATIVE VOLATILITY DATA, AS SPECIFIED BY
C SUBROUTINE ,EQUCON, (OPTION (3) IN EQUCON).
TOLD=TEMP(IT)*1.8+32.
350 CONTINUE
EFF=EFFV(NJ,IT)
IF(NMD.GT.1) EFF=1.0
TOLD1=TOLD/100.
AKB=(AKB1+AKB2*TOLD1+AKB3*TOLD1**2.+AKB4*TOLD1**3.)*EFF
DAKB=(AKB2+2.*AKB3*TOLD1+3.*AKB4*TOLD1**2.)*EFF
SUM=0.
SUMD=0.
DO 31 I=1,NC
SUM=SUM+(X(I,IT)*(AK1(I)+AK2(I)*TOLD1+AK3(I)*TOLD1**2.))
SUMD=SUMD+(X(I,IT)*(AK2(I)+2.*AK3(I)*TOLD1))
31 CONTINUE
FUNCT=AKB-(1./SUM)
FUNCTD=DAKB+(SUMD/(SUM**2.))
DIV=FUNCT/FUNCTD
IF(ABS(DIV).LT.0.001) GO TO 310
TOLD=TOLD-DIV
GO TO 350
310 TNEW=(TOLD-32.)/1.8
C CHANGE IN TEMPERATURE FROM ONE ITERATION TO THE NEXT IS RESTRICTED
C TO HALF THE CALCULATED CHANGE.
TBUB=TNEW
DIFF=TEMP(IT)-TNEW
IF(DIFF) 311,312,311
311 TEMP(IT)=TNEW+DIFF/2.
GO TO 999
312 TEMP(IT)=TNEW
GO TO 999

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SUBROUTINE BU8CAL

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400 CONTINUE
C   CALCULATION USING EQUILIBRIUM CONSTANT DATA, AS SPECIFIED BY
C   SUBROUTINE ,EQUCON, (OPTION (4) IN EQUCON).
TOLD=TEMP(IT)*1.8+32.+459.4
450 CONTINUE
T=TOLD/1000.
SUM=0.
SUMD=0.
DO 41 I=1,NC
EFF=EFFV(I,IT)
IF(MMD.GT.1) EFF=1.0
AK=AK1(I)+AK2(I)*T**2.+AK4(I)*T**3.
ALFA=(AK**3.)*TOLD*EFF
DA=(AK**2.)*(AK1(I)+4.*AK2(I)*T+7.*AK3(I)*T**2.+10.*AK4(I)*T**3.)
DALFA=DA*EFF
SUM=SUM+(X(I,IT)*ALFA)
SUMD=SUMD+(X(I,IT)*DALFA)
41 CONTINUE
FUNCT=SUM-1.
FUNCTD=SUMD
DIV=FUNCT/FUNCTD
IF(ABS(DIV).LT.0.001) GO TO 410
TOLD=TOLD-DIV
GO TO 450
410 TNEW=(TOLD-32.-459.4)/1.8
C   CHANGE IN TEMPERATURE FROM ONE ITERATION TO THE NEXT IS RESTRICTED
C   TO HALF THE CALCULATED CHANGE.
TBUB=TNEW
DIFF=TEMP(IT)-TNEW
IF(DIFF) 411,412,411
411 TEMP(IT)=TNEW+DIFF/2.
GO TO 999
412 TEMP(IT)=TNEW
999 CONTINUE
RETURN
END

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SUBROUTINE BUBCAL

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0001
SUBROUTINE DEWCAL(IT,MMD,TDEW)
DEW POINT ROUTINE.
C FOUR CALCULATION METHODS ARE AVAILABLE. CHOICE DEPENDS ON THE
C TYPE OF EQUILIBRIUM DATA SPECIFIED.
C CHOICE MADE BY COUNTER NEQUIL.
C *****
C INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
C *****
REAL K,LMFR,LRTOT,LSTOT,LD,LB,LPROF
GO TO (100,200,300,400),NEQUIL

100 CONTINUE
C CALCULATION USING ANTOINE CONSTANTS.
TOLD=TEMP(IT)
DO 11 KTIMES=1,1000
ASUM=1.
BSUM=0.
DO 12 I=1,NC
EFF=EFFV(I,IT)
IF(MMD.GT.1) EFF=1.0
ANT=AN(I)-BN(I)/(CN(I)+TOLD)
EANT=(10.**ANT)*EFF
ASUM=ASUM-Y(I,IT)*PRESS(IT)/EANT
BSUM=BSUM+Y(I,IT)*PRESS(IT)*((CN(I)+TOLD)**2.))/EANT
12 CONTINUE
DIV=ASUM/BSUM
IF(ABS(DIV).LE.0.001) GO TO 110
TOLD=TOLD-DIV
11 CONTINUE
110 CONTINUE
C CHANGE IN TEMPERATURE FROM ONE ITERATION TO THE NEXT IS RESTRICTED
C TO HALF THE CALCULATED CHANGE.
TDEW=TOLD
DIFF=TEMP(IT)-TOLD
IF(DIFF) 111,112,111
111 TEMP(IT)=TOLD+DIFF/2.
GO TO 999
112 TEMP(IT)=TOLD
GO TO 999

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SUBROUTINE DEWCAL

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200 CONTINUE
C   CALCULATION USING EQUILIBRIUM CONSTANT DATA, AS SPECIFIED BY
C   SUBROUTINE EQUCON, OPTION (2) IN EQUCON.
TOLD=TEMP(IT)*1.8+32.+459.4
250 CONTINUE
TOLD1=TOLD/100.
SUM=0.
SUMD=0.
DO 21 I=1,NC
EFF=EFFV(I,IT)
IF(MHD.GT.1) EFF=1.0
ALFA=EXP(AK1(I)-AK2(I)/TOLD1+AK3(I)*TOLD1)*EFF
DA=AK3(I)+AK2(I)/(TOLD1**2.)
DALFA=(EXP(AK1(I)-AK2(I)/TOLD1+AK3(I)*TOLD1))*DA*EFF
SUM=SUM+(Y(I,IT)/ALFA)
SUMD=SUMD+(Y(I,IT)/(ALFA**2.))*DALFA
21 CONTINUE
FUNCT=SUM-1.
FUNCTD=-SUMD
DIV=FUNCT/FUNCTD
IF(ABS(DIV).LT.0.001) GO TO 210
TOLD=TOLD-DIV
GO TO 250
210 TNEW=(TOLD-32.-459.4)/1.8
C   CHANGE IN TEMPERATURE FROM ONE ITERATION TO THE NEXT IS RESTRICTED
C   TO HALF THE CALCULATED CHANGE.
TDEW=TNEW
DIFF=TEMP(IT)-TNEW
IF(DIFF) 211,212,211
211 TEMP(IT)=TNEW+DIFF/2.
GO TO 999
212 TEMP(IT)=TNEW
GO TO 999

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SUBROUTINE DEMCAL

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300 CONTINUE
C   CALCULATION USING RELATIVE VOLATILITY DATA, AS SPECIFIED BY
C   SUBROUTINE EQUCON, (OPTION (3) IN EQUCON).
      TOLD=TEMP(IT)*1.8+32.
350 CONTINUE
      EFF=EFFV(NJ,IT)
      IF(NMD.GT.1) EFF=1.0
      TOLD1=TOLD/100.
      AKB=(AKB1+AKB2*TOLD1+AKB3*TOLD1**2.+AKB4*TOLD1**3.)*EFF
      DAKB=(AKB2+2.*AKB3*TOLD1+3.*AKB4*TOLD1**2.)*EFF
      SUM=0.
      SUMD=0.
      DO 31 I=1,NC
        ALFA=AK1(I)+AK2(I)*TOLD1+AK3(I)*TOLD1**2.
        DALFA=AK2(I)+2.*AK3(I)*TOLD1
        SUM=SUM+(Y(I,IT)/ALFA)
        SUMD=SUMD+(Y(I,IT)/(ALFA**2.))*DALFA
31  CONTINUE
      FUNCT=AKB-SUM
      FUNCTD=DAKB+SUMD
      DIV=FUNCT/FUNCTD
      IF(ABS(DIV).LT.0.001) GO TO 310
      TOLD=TOLD-DIV
      GO TO 350
310 TNEW=(TOLD-32.)/1.8
C   CHANGE IN TEMPERATURE FROM ONE ITERATION TO THE NEXT IS RESTRICTED
C   TO HALF THE CALCULATED CHANGE.
      TDEW=TNEW
      DIFF=TEMP(IT)-TNEW
      IF(DIFF) 311,312,311
311 TEMP(IT)=TNEW+DIFF/2.
      GO TO 999
312 TEMP(IT)=TNEW
      GO TO 999

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SUBROUTINE DEFCAL



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400 CONTINUE
C   CALCULATION USING EQUILIBRIUM CONSTANT DATA, AS SPECIFIED BY
C   SUBROUTINE ,EQUCON, (OPTION (4) IN EQUCON).
TOLD=TEMP(IT)*1.8+32.+459.4
450 CONTINUE
T=TOLD/1000.
SUM=0.
SUMD=0.
DO 41 I=1,NC
EFF=EFFV(I,IT)
IF(MMD.GT.1) EFF=1.0
AK=AK1(I)+AK2(I)*T**2.+AK4(I)*T**3.
ALFA=(AK**3.)*TOLD*EFF
DA=(AK**2.)*(AK1(I)+4.*AK2(I)*T+7.*AK3(I)*T**2.+10.*AK4(I)*T**3.)
DALFA=DA*EFF
SUM=SUM+(Y(I,IT)/ALFA)
SUMD=SUMD+(Y(I,IT)/(ALFA**2.))*DALFA
41 CONTINUE
FUNCT=SUM-1.
FUNCTD=-SUMD
DIV=FUNCT/FUNCTD
IF(ABS(DIV).LT.0.001) GO TO 410
TOLD=TOLD-DIV
GO TO 450
410 TNEW=(TOLD-32.-459.4)/1.8
C   CHANGE IN TEMPERATURE FROM ONE ITERATION TO THE NEXT IS RESTRICTED
C   TO HALF THE CALCULATED CHANGE.
TDEW=TNEW
DIFF=TEMP(IT)-TNEW
IF(DIFF) 411,412,411
411 TEMP(IT)=TNEW+DIFF/2.
GO TO 999
412 TEMP(IT)=TNEW
999 CONTINUE
RETURN
END

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SUBROUTINE DEWCAL

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SUBROUTINE FLOWEQ
C CALCULATION OF VAPOUR AND LIQUID FLOWRATES THROUGH THE COLUMN
C BASED ON EQUIMOLAR OVERFLOW ASSUMPTIONS.
C THIS SUBROUTINE IS USED FOR:
C (1) INITIAL FLOWRATE CALCULATIONS
C (2) FLOWRATE CALCULATIONS FOR PROBLEMS THAT IGNORE ENTHALPY EFFECTS
C I.E. EQUIMOLAR OVERFLOW ASSUMPTIONS.
C (3) FLOWRATE CALCULATIONS FOR THE FIRST TWO ITERATIONS IN PROBLEMS
C CONSIDERING ENTHALPY EFFECTS. IE. NON-EQUIMOLAR OVERFLOW CASES.
C *****
C INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
C *****
REAL K,KF,LMFR,LRTOT,LSTOT,LD,LB,LPROF
1 FORMAT( )
C CALCULATION OF LIQUID RATES IN RECTIFYING SECTION.
100 CONTINUE
LRTOT(NP2)=RR*DTOT
DO 11 N=NP1,NF2,-1
LRTOT(N)=LRTOT(N+1)
11 CONTINUE
C CALCULATION OF LIQUID RATES IN STRIPPING SECTION.
LSTOT(1)=FTOT-DTOT
BTOT=LSTOT(1)
DO 12 M=NF1,2,-1
LSTOT(M)=LRTOT(NF2)+Q*FTOT
IF(LSTOT(M).LT.0.0) GO TO 901
12 CONTINUE
LRTOT(NF1)=LSTOT(NF1)
C CALCULATION OF VAPOUR RATES IN STRIPPING SECTION.
DO 13 M=1,NF
VSTOT(M)=LSTOT(2)-RTOT
13 CONTINUE

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SUBROUTINE FLOWEQ

C CALCULATION OF VAPOUR RATES IN RECTIFYING SECTION.

DO 14 N=NF1,NP1  
VRTOT(N)=VSTOT(NF)+(1.-Q)\*FTOT

14 CONTINUE

VSTOT(NF1)=VRTOT(NF1)  
VRTOT(NP2)=0.0  
IF(NCOND.EQ.2) VRTOT(NP2)=DTOT  
LSTOT(1)=BTOT

GO TO 999

C ERROR MESSAGES \*\*\*\*\*

901 CONTINUE

WRITE(NTP,9001) Q,LSTOT(M)

9001 FORMAT(/T2,ERROR: NEGATIVE LIQUID FLOW IN STRIPPING SECTION',  
1/T2,'CAUSED BY FEED CONDITION.',/T2,'Q=',F8.4,'2X','LIQUID FLOW=',  
2F10.2,/T2,'ENTER AN INCREASED REFLUX RATIO TO CONTINUE RUN,OR',  
3/T2,'ENTER 1000. FOR TERMINATION.')

READ(NCR,1) RR

IF(RR.LT.999.) GO TO 100

NJ=1000

999 CONTINUE

RETURN

END

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SUBROUTINE FLOWEQ

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SUBROUTINE FLOWNE
C CALCULATION OF VAPOUR AND LIQUID FLOWRATES THROUGH THE COLUMN
C BASED ON NON-EQUIMOLAR OVERFLOW.
C THIS SUBROUTINE IS USED FOR THE THIRD AND SUBSEQUENT ITERATIONS
C IN PROBLEMS INCLUDING ENTHALPY CONSIDERATIONS.
C THE CHANGE IN FLOWRATE FROM ONE ITERATION TO THE NEXT IS
C RESTRICTED AS SUGGESTED BY HOLLAND.
C REFERENCE: HOLLAND,C.D.,MULTICOMPONENT DISTILLATION,P 123,125.
C *****
C INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
C *****
C REAL K,KF,LMFR,LRTOT,LSTOT,LD,LB,LPROF
C IF(ICOUNT,LT,7) FLOLIM=1,2
C IF(ICOUNT,GT,6,AND,ICOUNT,LT,12) FLOLIM=1,1
C IF(ICOUNT,GT,11) FLOLIM=1,025
C VAPOUR AND LIQUID FLOWS IN RECTIFYING SECTION.
C DO 11 N=NP2,NF2,-1
C   CHEKL=LRTOT(N)
C   CHEKLU=LRTOT(N)*FLOLIM
C   CHEKLL=LRTOT(N)/FLOLIM
C   LRTOT(N)=(DM-HVAP(N-1))*DTOT/(HVAP(N-1)-HLIQ(N))
C   IF(LRTOT(N),LT,0.0) GO TO 901
C   IF(LRTOT(N),LT,CHEKLU,AND,LRTOT(N),GT,CHEKLL) GO TO 111
C   THE NEW LIQUID FLOWRATE VALUE IS RESTRICTED BETWEEN LIMITS 'FLOLIM'
C   AND '1/FLOLIM' TIMES THE PREVIOUS VALUE.
C   IF(LRTOT(N),GE,CHEKLU) LRTOT(N)=CHEKLU
C   IF(LRTOT(N),LE,CHEKLL) LRTOT(N)=CHEKLL
C   GO TO 112
C 111 CONTINUE
C   LRTOT(N)=LRTOT(N)-1(LRTOT(N)-CHEKL)/2.)
C 112 CONTINUE
C   VRTOT(N-1)=LRTOT(N)+DTOT
C 11 CONTINUE

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SUBROUTINE FLOWNE

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C      VAPOUR AND LIQUID FLOWS IN STRIPPING SECTION.
DO 21 M=1,NF
  CHEKV=VSTOT(M)
  CHEKVU=VSTOT(M)*FLOLIM
  CHEKVL=VSTOT(M)/FLOLIM
  VSTOT(M)=(HLIQ(M+1)-BM)*BTOT/(HVAP(M)-HLIQ(M+1))
  IF(VSTOT(M).LT.0.0) GO TO 902
  IF(VSTOT(M).LT.CHEKVU.AND.VSTOT(M).GT.CHEKVL) GO TO 211
C    THE NEW VAPOUR FLOWRATE VALUE IS RESTRICTED BETWEEN LIMITS *FLOLIM*
C    AND *1/FLOLIM* TIMES THE PREVIOUS VALUE.
  IF(VSTOT(M).GE.CHEKVU) VSTOT(M)=CHEKVU
  IF(VSTOT(M).LE.CHEKVL) VSTOT(M)=CHEKVL
  GO TO 212
211 CONTINUE
  VSTOT(M)=VSTOT(M)-((VSTOT(M)-CHEKV)/2.)
212 CONTINUE
  LSTOT(M+1)=VSTOT(M)+BTOT
21 CONTINUE
  LRTOT(NF1)=LSTOT(NF1)
  VSTOT(NF1)=VRTOT(NF1)
  VRTOT(NP2)=0.0
  IF(NCOND.EQ.2) VRTOT(NP2)=DTOT
  LSTOT(1)=RTOT
  GO TO 999
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SUBROUTINE FLOWNE

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C 901 ERROR MESSAGES *****
CONTINUE
N=N-1
WRITE(NTP,9001) N
9001 FORMAT(/T2,'ERROR:  NEGATIVE LIQUID FLOW LEAVING TRAY',I4)
GO TO 903
902 CONTINUE
M=M-1
WRITE(NTP,9002) M
9002 FORMAT(/T2,'ERROR:  NEGATIVE VAPOUR FLOW LEAVING TRAY',I4)
903 CONTINUE
WRITE(NTP,9003)
9003 FORMAT(/T2,'POSSIBLE CAUSES OF ERROR:',/T4,'(1) REFLUX RATIO TOO L
OW',/T4,'(2) FEED ENTHALPY TOO HIGH',/T4,'(3) ERROR IN ENTHALPY CA
LCULATION.')
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WRITE(NTP,9013) QF,QC,QR
9013 FORMAT(/T2,'FEED ENTHALPY=',F13.2,'KJ/KGMOLE',/T2,'CONDENSER DU
ITY=',F12.2,'KW',/T2,'REBOILER DUTY=',F13.2,'KW')
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NJ=1000
999 CONTINUE
RETURN
END
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SUBROUTINE FLOWNE

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0001 SUBROUTINE ABSTRP
      CALCULATION OF ABSORPTION AND STRIPPING FACTORS FOR ALL STAGES IN
      THE COLUMN.
      ABSORPTION FACTORS USED IN RECTIFYING SECTION.
      STRIPPING FACTORS USED IN STRIPPING SECTION.
      *****
      INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
      *****
      REAL K,LMFR,LRTOT,LSTOT,LD,LB,LPROF
      IF(NCOND.GT.1) GO TO 110
      ABSORPTION FACTOR FOR TOTAL-CONDENSER OPERATION.
      DO 11 I=1,NC
      A(I,NP2)=RR
      11 CONTINUE
      GO TO 111
      110 CONTINUE
      ABSORPTION FACTOR FOR PARTIAL-CONDENSER OPERATION.
      DO 12 I=1,NC
      A(I,NP2)=LRTOT(NP2)/(K(I,NP2)*DTOT)
      12 CONTINUE
      111 CONTINUE
      ABSORPTION FACTORS FOR STAGES FROM FEED TRAY TO TOP TRAY INCL.
      DO 13 N=NF1,NP1
      DO 14 I=1,NC
      A(I,N)=LRTOT(N)/(VRTOT(N)*K(I,N))
      14 CONTINUE
      13 CONTINUE
      STRIPPING FACTORS FOR STAGES FROM REBOILER TO FEED TRAY INCL.
      DO 21 N=1,NF1
      DO 22 I=1,NC
      S(I,M)=VSTOT(M)*K(I,M)/LSTOT(M)
      22 CONTINUE
      21 CONTINUE
      RETURN
      END
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SUBROUTINE ABSTRP

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SUBROUTINE VBLB(LB,NC,NFI,S,VB,MC,MP)
CALCULATION OF STRIPPING SECTION.
THIS SUBROUTINE SOLVES THE STRIPPING SECTION MASS BALANCE-
EQUILIBRIUM EQUATIONS WORKING FROM THE REBOILER TOWARDS THE
FEED TRAY.
DIMENSION LB(MC,MP),S(MC,MP),VB(MC,MP)
REAL LB
DO 11 I=1,NC
  VB(I,1)=S(I,1)
11 CONTINUE
DO 12 M=2,NFI
  DO 13 I=1,NC
    LB(I,M)=1.+VB(I,M-1)
    VB(I,M)=S(I,M)*LB(I,M)
13 CONTINUE
12 CONTINUE
RETURN
END

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SUBROUTINE VBLB



C	SUBROUTINE VDLD(A,LD,NC,NF2,NP1,NP2,VD,MC,MP)	0001
C	CALCULATION OF RECTIFYING SECTION.	
C	THIS SUBROUTINE SOLVES THE RECTIFYING SECTION MASS BALANCE-	
C	EQUILIBRIUM EQUATIONS WORKING FROM THE CONDENSER TOWARDS THE	
C	FEED TRAY.	
	DIMENSION A(MC,MP),LD(MC,MP),VD(MC,MP)	0002
	REAL LD	0003
	DO 11 I=1,NC	0004
	VD(I,NP1)=1.+A(I,NP2)	0005
	11 CONTINUE	0006
	DO 12 N=NP1,NF2,-1	0007
	DO 13 I=1,NC	0008
	LD(I,N)=A(I,N)*VD(I,N)	0009
	VD(I,N-1)=1.+LD(I,N)	0010
	13 CONTINUE	0011
	12 CONTINUE	0012
	RETURN	0013
	END	0014

SUBROUTINE VDLD

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C
SUBROUTINE RDCALC(A,BD,D,FTOT,LB,NC,NFI,SUMDI,VD,Z,MC,MP)
CALCULATION OF THE B/D RATIOS AT THE FEED TRAY.
THIS SUBROUTINE MATCHES THE RECTIFYING AND STRIPPING SECTIONS
AT THE FEED TRAY.
DIMENSION A(MC,MP),BD(MC),D(MC),LB(MC,MP),VD(MC,MP),Z(MC)
REAL LB
SUMDI=0.0
DO 11 I=1,NC
  BD(I)=A(I,NFI)*VD(I,NFI)/LB(I,NFI)
  D(I)=FTOT*Z(I)/(1.+BD(I))
  SUMDI=SUMDI+D(I)
11 CONTINUE
RETURN
END

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SUBROUTINE RDCALC

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SUBROUTINE BDCORR(B,BD,D,FTOT,NC,SLIQ,SUMD,SUMB,THETA,Z,MC,MP)
CALCULATION OF THE CORRECTED B/D RATIOS.
THETA IS USED TO CORRECT THE CALCULATED B/D RATIOS. THE CORRECTED
VALUES ARE USED TO CALCULATE THE COMPOSITION PROFILES THROUGH
THE COLUMN IN THE NEXT ROUTINE.
DIMENSION B(MC),BD(MC),D(MC),SLIQ(MP),Z(MC)
SUMD=0.
SUMB=0.
DO 10 I=1,NC
  BD(I)=BD(I)*THETA
  D(I)=FTOT*Z(I)/(1.+BD(I))
  B(I)=D(I)*BD(I)
  SUMD=SUMD+D(I)
  SUMB=SUMB+B(I)
10 CONTINUE
  SLIQ(I)=SUMB
  RETURN
END

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SUBROUTINE BDCORR

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C SUBROUTINE VLRECT
C CALCULATION OF THE COMPOSITION PROFILES IN THE REACTIFYING
C SECTION. THE CORRECTED VALUES OF D,THE DISTILLATE MOLAR
C FLOWRATES,ARE USED.
C *****
C INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
C *****
C REAL K,LMFR,LRTOT,LSTOT,LD,LB,LPROF
SUMV=0.
DO 11 I=1,NC
  VMFR(I,NP1)=VD(I,NP1)*D(I)
  SUMV=SUMV+VMFR(I,NP1)
11 CONTINUE
  SVAP(NP1)=SUMV
  IF(INCOND.GT.1) GO TO 111
  TOTAL CONDENSER OPERATION.
  DO 12 I=1,NC
    Y(I,NP1)=VMFR(I,NP1)/SUMV
    Y(I,NP2)=X(I,NP2)
12 CONTINUE
    GO TO 211
111 CONTINUE
  C PARTIAL CONDENSER OPERATION.
  SUMXK=0.
  DO 21 I=1,NC
    Y(I,NP1)=VMFR(I,NP1)/SUMV
    Y(I,NP2)=X(I,NP2)
    X(I,NP2)=VRTOT(NP1)*Y(I,NP1)/(DTOT*K(I,NP2)+LRTOT(NP2))
    SUMXK=SUMXK+X(I,NP2)
21 CONTINUE
    DO 22 I=1,NC
      X(I,NP2)=X(I,NP2)/SUMXK
22 CONTINUE
211 CONTINUE

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SUBROUTINE VLRECT

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C      RECTIFYING SECTION.
DO 30 N=NP1,NF2.-1
SUMV=0.
SUML=0.
DO 31 I=1,NC
VMFR(I,N-1)=VD(I,N-1)*D(I)
LMFR(I,N)=LD(I,N)*D(I)
SUMV=SUMV+VMFR(I,N-1)
SUML=SUML+LMFR(I,N)
31 CONTINUE
SVAP(N-1)=SUMV
SLIQ(N)=SUML
C      MOLE FRACTION CALCULATIONS (VAPOUR & LIQUID)
DO 41 I=1,NC
X(I,N)=LMFR(I,N)/SUML
Y(I,N-1)=VMFR(I,N-1)/SUMV
41 CONTINUE
30 CONTINUE
DO 50 I=1,NC
LMFR(I,NP2)=D(I)*RR
50 CONTINUE
SLIQ(NP2)=DTOT*RR
IF(INCOND.EQ.1) GO TO 999
C      PARTIAL CONDENSER OPERATION.
DO 51 I=1,NC
VMFR(I,NP2)=D(I)
51 CONTINUE
SVAP(NP2)=DTOT
999 CONTINUE
RETURN
END

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SUBROUTINE VLRECT

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SUBROUTINE VLSTRP
CALCULATION OF THE COMPOSITION PROFILES IN THE STRIPPING SECTION.
THE CORRECTED VALUES OF B, THE BOTTOMS MOLAR FLOWRATES, ARE USED.
*****
INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
*****
REAL K, LMFR, LRTOT, LSTOT, LD, LB, LPROF
DO 10 N=1,NF
SUMV=0.
SUML=0.
DO 11 I=1,NC
VMFR(I,M)=VB(I,M)*B(I)
LMFR(I,M+1)=LB(I,M+1)*B(I)
SUMV=SUMV+VMFR(I,M)
SUML=SUML+LMFR(I,M+1)
11 CONTINUE
SVAP(M)=SUMV
SLIQ(M+1)=SUML
C MOLE FRACTION CALCULATIONS (VAPOUR & LIQUID)
DO 21 I=1,NC
X(I,M+1)=LMFR(I,M+1)/SUML
Y(I,M)=VMFR(I,M)/SUMV
21 CONTINUE
10 CONTINUE
DO 12 I=1,NC
LMFR(I,1)=B(I)
12 CONTINUE
RETURN
END

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SUBROUTINE VLSTRP

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C      SUBROUTINE THECAL
C      CALCULATION OF THETA, IN HOLLAND'S CONVERGENCE PROCEDURE.
C      THE SUBROUTINE USES A NEWTON-RAPHSON PROCEDURE TO CALCULATE
C      THETA WHICH REPRESENTS THE DEGREE OF MISMATCH OF THE RECTIFYING
C      AND STRIPPING SECTIONS AT THE FEED TRAY.
C      *****
C      INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE.
C      *****
C      REAL K,LMFR,LRTOT,LSTOT,LD,LB,LPRUF
C      IF(THETA.LT.0.00001.OR.THETA.GT.9999.) GO TO 111
C      IF(ICOUNT-1) 111,111,110
C      111 THETA=D.1
C      110 CONTINUE
C      SUMF=0.
C      SUMDF=0.
C      DO 10 I=1,NC
C      FTHET=FTOT*Z(I)/(1.+THETA*BD(I))
C      DFTHET=FTHET*BD(I)/(1.+THETA*BD(I))
C      SUMF=SUMF+FTHET
C      SUMDF=SUMDF+DFTHET
C      10 CONTINUE
C      FTHET=SUMF-DTOT
C      DFTHET=-SUMDF
C      DIV=FTHET/DFTHET
C      THETA=THETA-DIV
C      IF(ABS(THETA).GT.10000.) GO TO 999
C      IF(THETA.GE.0.0) GO TO 112
C      THETA=THETA*(-1)
C      GO TO 999
C      112 CONTINUE
C      IF(ABS(DIV).GE.0.001) GO TO 110
C      999 CONTINUE
C      RETURN
C      END

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SUBROUTINE THECAL

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0001 SUBROUTINE RUNRES
ROUTINE FOR WRITING ABBREVIATED RESULTS TO THE TERMINAL,
C *****
C INSERT DIMENSION AND COMMON BLOCK 'MULTICOM' HERE,
C *****
C DIMENSION TEMPF(MP)
REAL K,LMFR,LRTOT,LSTOT,LD,LB,LPROF
1010 FORMAT(/T2,'MULTICOMPONENT DISTILLATION BY THIELE-GEDDES METHOD:.',)
1011 FORMAT(/T2,'RUN NO:',2X,I3,'/',I2,'/',I2)
1012 FORMAT(/T2,'OVERALL MATERIAL BALANCE:',/T5,'STREAM',7X,'FLOWRATE',
11X,'(KGMOLE/HR)',/T3,'DISTILLATE',4X,F9.4,/T3,'BOTTOMS',7X,F9.4,
2/T3,'FEED-TOTAL',4X,F9.4,/T7,'-VAPOUR',3X,F9.4,/T7,'-LIQUID',3X,
3F9.4)
1013 FORMAT(/T2,'COMPONENT MOLE FRACTIONS:',/T3,'COMP.',2X,
1'DISTILLATE',3X,'BOTTOMS',10X,'FEED STREAM',/T33,'LIQUID',3X,
2'VAPOUR',4X,'TOTAL')
1014 FORMAT(T4,I2,6X,F6.4,5X,F6.4,4X,F6.4,3X,F6.4,3X,F6.4)
1015 FORMAT(/T2,'TEMPERATURE AND PRESSURE PROFILE:',/T4,'TRAY NO',10X,
1'TEMPERATURE',7X,'PRESSURE',/T18,'DEG C',7X,'DEG F',6X,'KPA',
2/T3,'CONDENSER',5X,F6.1,6X,F6.1,3X,F8.2)
1115 FORMAT(T5,'FEED',8X,F6.1,6X,F6.1,3X,F8.2)
1016 FORMAT(T6,I3,8X,F6.1,6X,F6.1,3X,F8.2)
1116 FORMAT(T4,'REBOILER',5X,F6.1,6X,F6.1,3X,F8.2)
1017 FORMAT(/T2,'NO. OF ITERATIONS=',I4,5X,'THETA=',F7.3)
1018 FORMAT(/T2,'REFLUX RATIO=',F12.2,/T2,'DISTILLATE RATE=',F9.2,2X,
1'KGMOLE/HR',/T2,'FEED Q VALUE=',F14.4,/T2,'FEED TEMPERATURE=',F8.2
2,2X,'DEG C')
1019 FORMAT(/T2,'TOTAL CONDENSER OPERATION.',)
1020 FORMAT(/T2,'PARTIAL CONDENSER OPERATION.',)
1021 FORMAT(T2,'FEED ENTHALPY=',F11.2,2X,'KJ/KGMOLE',/T2,'CONDENSER DU
ITY=',F12.2,2X,'KW',/T2,'REBOILER DUTY=',F13.2,2X,'KW')
1120 FORMAT(/T2,'VAPOUR AND LIQUID FLOWRATES LEAVING TRAYS:',/T4,'TRAY
1NO',7X,'VAPOUR',7X,'LIQUID',4X,'(KGMOLE/HR)',
2/T3,'CONDENSER',5X,F7.2,6X,F7.2)
1121 FORMAT(T6,I3,8X,F7.2,6X,F7.2)
1122 FORMAT(T5,'FEED',8X,F7.2,6X,F7.2)
1123 FORMAT(T4,'REBOILER',5X,F7.2,6X,F7.2)

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SUBROUTINE RUNRES



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C      TEMPERATURE PROFILE IN DEG F.
DO 11 J=1,NP2
TEMPF(J)=TEMP(J)*1.8+32.
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11 CONTINUE
C      COMBINED RECTIFYING AND STRIPPING SECTION FLOW DATA.
DO 12 N=NP2,NF1,-1
LPROF(N)=LRTOT(N)
VPROF(N)=VRTOT(N)
0037
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0039
12 CONTINUE
DO 13 M=NF,1,-1
LPROF(M)=LSTOT(M)
VPROF(M)=VSTOT(M)
0040
0041
0042
13 CONTINUE
C      PRESSURE PROFILE CONVERTED TO S.I. UNITS.
FACTP2=7.50061683
DO 14 J=1,NP2
PRES(J)=PRESS(J)/FACTP2
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0045
14 CONTINUE
C      RESULT TABLE HEADING AND RUN IDENTIFICATION NUMBER.
WRITE(NTP,1010)
WRITE(NTP,1011) NUM1,NUM2,NUM3
0046
0047
0048
C      DISTILLATE,BOTTOMS AND FEED STREAM FLOWRATES.
WRITE(NTP,1012) DTOT,BTOT,FTOT,AVAP,ALIQ
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0050
C      STREAM COMPOSITIONS.
WRITE(NTP,1013)
WRITE(NTP,1014) (I,Y(I,NP2),X(I,1),XF(I),YF(I),Z(I),I=1,NC)
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SUBROUTINE RUNRES

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C      TEMPERATURE AND PRESSURE PROFILES.
      WRITE(NTP,1015) TEMP(NP2),TEMPF(NP2),PRES(NP2)
      DO 15 J=NP1,2,-1
      IF(J.EQ.NF1) GO TO 115
      MMM=J-1
      WRITE(NTP,1016) MMN,TEMP(J),TEMPF(J),PRES(J)
      GO TO 15
115 CONTINUE
      WRITE(NTP,1115) TEMP(NF1),TEMPF(NF1),PRES(NF1)
15 CONTINUE
      WRITE(NTP,1116) TEMP(1),TEMPF(1),PRES(1)
      CONVERGENCE DATA.
      WRITE(NTP,1017) ICOUNT,THETA
      C      REFLEX RATIO,DISTILLATE RATE AND FEED CONDITION.
      WRITE(NTP,1018) RR,DTOT,Q,TFEED
      C      HEAT DUTIES AND CONDENSER TYPE.
      IF(NENTH.GT,0) WRITE(NTP,1021) QF,QC,QR
      IF(NCOND.EQ,1) WRITE(NTP,1019)
      IF(NCOND.EQ,2) WRITE(NTP,1020)
      C      VAPOUR AND LIQUID FLOWRATE PROFILES.
      WRITE(NTP,1120) VPROF(NP2),LPROF(NP2)
      DO 16 J=NP1,2,-1
      IF(J.EQ.NF1) GO TO 116
      MMM=J-1
      WRITE(NTP,1121) MMN,VPROF(J),LPROF(J)
      GO TO 16
116 CONTINUE
      WRITE(NTP,1122) VPROF(NF1),LPROF(NF1)
16 CONTINUE
      WRITE(NTP,1123) VPROF(1),LPROF(1)
      RETURN
      END

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SUBROUTINE RUNRES

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2113 FORMAT(/T31.66(' '),/T20,'TOTALS:',3X,F10.2,5X,F10.4,15X,F10.2,
15X,F10.4,/T31.66(' '),/T21,'CONDENSER TEMPERATURE=',F8.2,2X,
2'DEG C',/T21,'CONDENSER PRESSURE=',F11.2,2X,'KPA')
2121 FORMAT(/T2,'TRAY NO',2X,13,4X,'* FEED TRAY ',/T2,'COMPONENT NO',
1,10X,'VAPOUR FEED',16X,'LIQUID FEED',14X,'VAPOUR LEAVING',13X,'LIQ
UID LEAVING',/T17,'KGMOLE/HR',2X,'MOL FRACTION',4X,'KGMOLE/HR',2X,
3'MOL FRACTION',4X,'KGMOLE/HR',2X,'MOL FRACTION',4X,'KGMOLE/HR',2X,
4'MOL FRACTION')
2122 FORMAT(/T7,12,7X,F10.2,2X,E12.6,3X,F10.2,2X,E12.6,3X,F10.2,2X,
1E12.6,3X,F10.2,2X,E12.6)
2123 FORMAT(/T17,104(' '),/T5,'TOTALS:',4X,F10.2,2X,F10.4,5X,F10.2,2X,
1F10.4,5X,F10.2,2X,F10.4,5X,F10.2,2X,F10.4,/T17,104(' '),/T21,
1'TRAY TEMPERATURE=',F8.2,2X,'DEG C',/T21,'TRAY PRESSURE=',F11.2,
22X,'KPA',/T21,'FEED Q VALUE=',F12.2,/T21,'FEED TEMPERATURE=',
3F8.2,2X,'DEG C')
2211 FORMAT(/T11,'REBOILER',/T11,'COMPONENT NO',14X,
1'VAPOUR LEAVING',26X,'BOTTOM PRODUCT',/T31,'KGMOLE/HR',5X,'MOL FRA
CTION',14X,'KGMOLE/HR',5X,'MOL FRACTION')
2213 FORMAT(/T31.66(' '),/T20,'TOTALS:',3X,F10.2,5X,F10.4,15X,F10.2,
15X,F10.4,/T31.66(' '),/T21,'REBOILER TEMPERATURE=',F8.2,2X,
2'DEG C',/T21,'REBOILER PRESSURE=',F11.2,2X,'KPA')
3011 FORMAT(/T11,'OVERALL MATERIAL BALANCE:',/T11,'MOLAR PRODUCT AND
1 FEED STREAM RATES',4X,'(KGMOLE/HR)')
3012 FORMAT(/T21,'COMPONENT NO',10X,'DISTILLATE',11X,'BOTTOMS',13X,
1'FEED')
3013 FORMAT(/T25,13,16X,F9.4,10X,F9.4,10X,F9.4)
3014 FORMAT(/T41,50(' '),/T21,'TOTAL RATES:',11X,F9.4,10X,F9.4,10X,F9.4,
1/T41,50(' '),)
3015 FORMAT(/T11,'HEAT DUTIES AND FEED ENTHALPY:',
1//T11,'CONDENSER DUTY=',E16.8,2X,'KW',/T11,
2'REBOILER DUTY=',1X,E16.8,2X,'KW',/T11,'FEED ENTHALPY=',
31X,E16.8,2X,'KJ/KGMOLE')
3016 FORMAT(/T11,'END OF PRINTOUT',)

```

SUBROUTINE OUTPUT

```

C
  RUN VARIABLES.
  WRITE(NLP,1110)
  WRITE(NLP,1111) NUM1,NUM2,NUM3
  WRITE(NLP,1010)
  WRITE(NLP,1014) ICOUNT,ERROR,ERRD
  WRITE(NLP,1011) THETA,ACCLIM
  IF(ABS(THETA-1.0).GT.ACCLIM) WRITE(NLP,1015)
  DISTILLATE COMPOSITION.
  WRITE(NLP,1012) RR,DTOT
  WRITE(NLP,1013) (I,(TTITLES(I,J),J=1,5),Y(I,NP2),I=1,NC)
  HEAT BALANCE.
  IF(NOWN.GT.0) WRITE(NLP,3015) QC,QR,QF
  SPECIFICATIONS THROUGH COLUMN.
  DO 10 J=NP2,1.-1
    SX=0.
    SY=0.
    DO 20 I=1,NC
      FEED(I)=FTOT*Z(I)
      SX=SX+X(I,J)
      SY=SY+Y(I,J)
    20 CONTINUE
    MMM=J-1
    IF(J.EQ.NP2) GO TO 501
    IF(J.EQ.NF1) GO TO 510
    IF(J.EQ.1) GO TO 502
    MMM=J-1
    DETAILS OF TRAYS OTHER THAN FEED TRAY.
    WRITE(NLP,2011) MMM
    WRITE(NLP,2012) (I,VMFR(I,J),Y(I,J),LMFR(I,J),X(I,J),I=1,NC)
    WRITE(NLP,2013) SVAP(J),SY,SLIQ(J),SX,TEMP(J),PRES(J)
    GO TO 10
  501 CONTINUE

```

SUBROUTINE OUTPUT

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C      CONDENSER DETAILS.
      IF(NCOND.EQ.1) WRITE(NLP,2111)
      IF(NCOND.EQ.2) WRITE(NLP,2112)
      WRITE(NLP,2012) (I,D(I),Y(I,NP2),LMFR(I,NP2),X(I,NP2),I=1,NC)
      WRITE(NLP,2113) DTOT,SY,SLIQ(NP2),SX,TEMP(NP2),PRES(NP2)
      GO TO 10
502 CONTINUE
C      REBOILER DETAILS.
      WRITE(NLP,2211)
      WRITE(NLP,2012) (I,VMFR(I,1),Y(I,1),LMFR(I,1),X(I,1),I=1,NC)
      WRITE(NLP,2213) SVAP(I),SY,SLIQ(1),SX,TEMP(1),PRES(1)
      GO TO 10
510 CONTINUE
C      FEED TRAY DETAILS.
      WRITE(NLP,2121) NF
      SYF=0,
      SXF=0,
      DO 51 I=1,NC
        YFLO=YF(I)*AVAP
        XFLO=XF(I)*ALIQ
        WRITE(NLP,2122) I,YFLO,YF(I),XFLO,XF(I),VMFR(I,NF1),Y(I,NF1),
           ILMFR(I,NF1),X(I,NF1)
        SYF=SYF+YF(I)
        SXF=SXF+XF(I)
51 CONTINUE
      WRITE(NLP,2123) AVAP,SYF,ALIQ,SXF,SVAP(NF1),SY,SLIQ(NF1),SX,
        ITEMP(NF1),PRES(NF1),Q,TFEED
10 CONTINUE
C      OVERALL MASS BALANCE.
      WRITE(NLP,3011)
      WRITE(NLP,3012)
      WRITE(NLP,3013) (I,D(I),B(I),FEED(I),I=1,NC)
      WRITE(NLP,3014) DTOT,BTOT,FTOT
      WRITE(NLP,3016)
      WRITE(NLP,1110)
      RETURN
      END

```

SUBROUTINE OUTPUT

MAINLINE PROGRAM FOR DISTILLATION-PACKAGE PLOTTING ROUTINES.  
CONTENTS:

(A) MULTICOMPONENT SYSTEMS:

- (1) TEMPERATURE PROFILE THROUGH COLUMN.
- (2) VAPOUR AND LIQUID FLOWRATE PROFILES THROUGH COLUMN.
- (3) DISTILLATE COMPOSITION BAR DIAGRAM.
- (4) TRAY COMPOSITION PROFILES THROUGH COLUMN.

(B) BINARY SYSTEMS:

- (1) BINARY BATCH MCCABE-THIELE DIAGRAM.
- (2) BINARY CONTINUOUS MCCABE-THIELE DIAGRAM.
- (3) BINARY CONTINUOUS PONCHON-SAVARIT DIAGRAM.

```

DIMENSION A1(4),A2(4),A3(4),A4(4)
DIMENSION ALXT(4,5),ALYT(4,5)
DIMENSION F1(4),F2(4),F3(4),F4(4)
DIMENSION HEAD1(5),HEAD2(5),HEAD3(5),HEAD4(5)
DIMENSION IBUF(1000),NCH(10),NCHARX(4),NCHARY(4),TITL(6,5)
DIMENSION X(100),Y(100),XX(100),YY(100),XAX(4,5),YAX(4,5)
DIMENSION YLEN(4)
COMMON ALX1,ALY1,ALXT,ALYT,ASIZ,F1,F2,F3,F4
COMMON NC,NCH,NCHARX,NCHARY,NP,NP2,NR,NRF,TITL
COMMON X,Y,XX,YY,XAX,YAX,YLEN
DATA F1/,@ADD , , , , /
DATA F2/,@ADD , , , , /
DATA F3/,@ADD , , , , /
DATA F4/,@ADD , , , , /
DATA A1/,@ASG,AZX , , , , /
DATA A2/,@ASG,AZX , , , , /
DATA A3/,@ASG,AZX , , , , /
DATA A4/,@ASG,AZX , , , , /
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MAINLINE PLOTTER

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DATA ASIZ,INTA,INTB,NCR,NTP/D,14,67,84,8,5/
1  FORMAT(4(A6,1X))
2  FORMAT(5A6)
3  FORMAT( )
WRITE(NTP,1000)
1000 FORMAT(/T2,ENTER JOB NO, AS: NO.,DAY,MONTH (INTEGER VALUES),')
READ(NCR,3) NUM1,NUM2,NUM3
DNUM1=FLOAT(NUM1)
DNUM2=FLOAT(NUM2)
DNUM3=FLOAT(NUM3)
WRITE(NTP,1010)
1010 FORMAT(/T2,ENTER TYPE OF PLOT REQUIRED, (SEE MANUAL SECTION 5),')
READ(NCR,3) NTYPE
GO TO (100,200,300,300).NTYPE
100 CONTINUE
C MULTICOMPONENT DISTILLATION DATA REPRESENTATION.
WRITE(NTP,1001)
1001 FORMAT(/T2,ENTER TOTAL NUMBER OF RUNS TO BE PLOTTED,')
READ(NCR,3) NR
WRITE(NTP,1002)
1002 FORMAT(/T2,ENTER NUMBER OF RUN FOR WHICH FULL PLOT IS REQUIRED,')
READ(NCR,3) NRF
WRITE(NTP,1003)
1003 FORMAT(/T2,ENTER NUMBER OF TRAYS IN COLUMN,')
READ(NCR,3) NP
NP2=NP+2
WRITE(NTP,1004)
1004 FORMAT(/T2,ENTER NUMBER OF COMPONENTS IN FEED,')
READ(NCR,3) NC

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MAINLINE PLOTTER



```

1005 WRITE(NTP,1005)
      FORMAT(/T2,ENTER DATAFILE NAMES, (SEE MANUAL SECTION 5),)
      READ(NCR,1) F1(3),F2(3),F3(3),F4(3)
      A1(3)=F1(3)
      A2(3)=F2(3)
      A3(3)=F3(3)
      A4(3)=F4(3)
      CALL ERTRAN(6,A1)
      CALL ERTRAN(6,A2)
      CALL ERTRAN(6,A3)
      CALL ERTRAN(6,A4)
      READ AXIS TITLES AND PLOT HEADINGS FOR ALL MULTICOMPONENT PLOTS.
      CALL ERTRAN(6,'@ADD PLOTFILE,HEADING . ')
      READ(NCR,2) (HEAD1(J),J=1,5)
      READ(NCR,2) (HEAD2(J),J=1,5)
      READ(NCR,2) (HEAD3(J),J=1,5)
      READ(NCR,3) NCHH1,NCHH2,NCHH3
      CALL ERTRAN(6,'@ADD PLOTFILE,AXISTITLE . ')
      DO 11 I=1,4
        READ(NCR,2) (XAX(I,J),J=1,5)
        READ(NCR,3) NCHARX(I)
        READ(NCR,2) (YAX(I,J),J=1,5)
        READ(NCR,3) NCHARY(I)
11      CONTINUE
        READ(NCR,3) (YLEN(I),I=1,4)
        CALL ERTRAN(6,'@ADD PLOTFILE,TITLES . ')
        DO 12 I=1,6
          READ(NCR,2) (TITL(I,J),J=1,5)
          READ(NCR,3) NCH(I)
12      CONTINUE
          READ(NCR,3) ((ALXT(I,J),ALYT(I,J),J=1,3),I=1,4)

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MAINLINE PLOTTER

C	CALCOMP ROUTINE ,PLOTS, INITIATES PLOTTING PROCEDURE.	0077
	CALL PLOTS(IBUF,1000,25)	0078
C	CALL FACTOR(0.5)	0079
	SUBROUTINE FOR COLUMN TEMPERATURE PROFILES.	0080
C	CALL PLOTTS	0081
	SUBROUTINE FOR VAPOUR AND LIQUID FLOWRATE PROFILES.	0082
C	CALL PLOTFS	0083
	SUBROUTINE FOR DISTILLATE COMPOSITION BAR DIAGRAM.	0084
C	CALL PLOTDS	0085
	SUBROUTINE FOR COMPOSITION PROFILES THROUGH THE COLUMN.	0086
C	CALL PLOTCS(ALENX)	0087
	READ AND WRITE TITLES ABOVE MULTICOMPONENT PLOTS.	0088
	ANUM1=FLOAT(NR)	0089
	ANUM2=FLOAT(NRF)	0090
	CALL SYMBOL(1.0,7.7,ASIZ,HEAD1,0.0,NCHH1)	0091
	CALL NUMBER(5.2,7.7,ASIZ,ANUM1,0.0,-1)	0092
	CALL SYMBOL(1.0,7.4,ASIZ,HEAD2,0.0,NCHH2)	0093
	CALL NUMBER(5.3,7.4,ASIZ,ANUM2,0.0,-1)	
	CALL SYMBOL(1.0,7.1,ASIZ,HEAD3,0.0,NCHH3)	
	ANMF=7.1	
	ALX4=ALENX+6.0	
	ALY4=0.0	
	GO TO 999	

MAINLINE PLOTTER

```

200 CONTINUE
C   BINARY BATCH-DISTILLATION DIAGRAMS.
CALL PLOTS(IBUF,1000,25)
CALL FACTOR(0,5)
C   SUBROUTINE FOR BINARY BATCH-DISTILLATION DIAGRAMS.
CALL PLOTBT(LENX,NTYPE)
IF(NTYPE.EQ.3) GO TO 210
C   READ AND WRITE TITLES ON BATCH X-Y DIAGRAMS.
C   CONSTANT REFLUX OPERATION.
CALL ERTRAN(6,2,ADD PLOTFILE,BBCTTL . .)
GO TO 211
210 CONTINUE
C   VARIABLE REFLUX OPERATION.
CALL ERTRAN(6,2,ADD PLOTFILE,BBVTTL . .)
211 CONTINUE
READ(NCR,2) (HEAD1(I),I=1,5)
READ(NCR,2) (HEAD2(I),I=1,5)
READ(NCR,2) (HEAD3(I),I=1,5)
READ(NCR,2) (HEAD4(I),I=1,5)
READ(NCR,3) NCHB1,NCHB2,NCHB3,NCHB4
CALL SYMBOL(1,0,11,1,ASIZ,HEAD1,0,0,NCHB1)
CALL SYMBOL(1,0,10,8,ASIZ,HEAD2,0,0,NCHB2)
CALL SYMBOL(1,0,10,5,ASIZ,HEAD3,0,0,NCHB3)
CALL SYMBOL(1,0,10,2,ASIZ,HEAD4,0,0,NCHB4)
ANMF=10.2
ALX4=LENX+6.0
ALY4=0.0
GO TO 999

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MAINLINE PLOTTER

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300 CONTINUE
C   BINARY CONTINUOUS-DISTILLATION DIAGRAMS.
    IF (NTYPE.EQ.5) WRITE(NTP,3001)
3001 FORMAT(1/T2,'ENTER LIMITS OF ENTHALPY AXIS IN ENTH-CONC DIAGRAM.',
1/T2,'NORMALLY THESE ARE DEL PT VALUES ROUNDED CONVENIENTLY.',
2/T2,'ENTER UPPER LIMIT FIRST. (REAL MODE, FREE FORMAT).')
    IF (NTYPE.EQ.5) READ(NCR,3) YST1,YST2
    CALL PLOTS(IBUF,1000,25)
    CALL FACTOR(0.5)
C   SUBROUTINE FOR BINARY CONTINUOUS-DISTILLATION DIAGRAMS.
    CALL PLOTMT(ALENX,NTYPE,YST1,YST2)
    IF (NTYPE.EQ.4) GO TO 301
C   READ AND WRITE TITLES ON PONCHON-SAVARIT ENTHALPY-CONCENTRATION DIAGRAM.
    CALL ERTRAN(6,'@ADD PLOTFILE,PSTITL . ')
    READ(NCR,2) (HEAD1(I),I=1,5)
    READ(NCR,2) (HEAD2(I),I=1,5)
    READ(NCR,2) (HEAD3(I),I=1,5)
    READ(NCR,2) (HEAD4(I),I=1,5)
    READ(NCR,3) NCHB1,NCHB2,NCHB3,NCHB4
    CALL SYMBOL(1,0,28,1,ASIZ,HEAD1,0,0,NCHB1)
    CALL SYMBOL(1,0,27,8,ASIZ,HEAD2,0,0,NCHB2)
    CALL SYMBOL(1,0,27,5,ASIZ,HEAD3,0,0,NCHB3)
    CALL SYMBOL(1,0,27,2,ASIZ,HEAD4,0,0,NCHB4)
    ANMF=27.2
    GO TO 302

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MAINLINE PLOTTER

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301 CONTINUE
C MCCABE-THIELE X-Y DIAGRAM TITLES ADDED TO THE RUNSTREAM,
CALL ERTRAN(6,'@ADD PLOTFILE,RCITL . .)
302 CONTINUE
C READ AND WRITE TITLES ON MCCABE-THIELE OR PONCHON-SAVARIT X-Y DIAGRAM.
READ(NCR,2) (HEAD1(I),I=1,5)
READ(NCR,2) (HEAD2(I),I=1,5)
READ(NCR,2) (HEAD3(I),I=1,5)
READ(NCR,3) NCHB1,NCHB2,NCHB3
CALL SYMBOL(1,0,10,8,ASIZ,HEAD1,0,0,NCHB1)
CALL SYMBOL(1,0,10,5,ASIZ,HEAD2,0,0,NCHB2)
CALL SYMBOL(1,0,10,2,ASIZ,HEAD3,0,0,NCHB3)
IF(NTYPE.EQ.5) GO TO 303
ANMF=10.2
303 CONTINUE
ALX4=ALENX+6,0
ALY4=0,0
999 CONTINUE
CALL NUMBER(2,82,ANMF,ASIZ,DNUM1,0,0,-1)
CALL SYMBOL(3,10,ANMF,ASIZ,INTA,0,0,-1)
CALL NUMBER(3,24,ANMF,ASIZ,DNUM2,0,0,-1)
CALL SYMBOL(3,52,ANMF,ASIZ,INTB,0,0,-1)
CALL NUMBER(3,66,ANMF,ASIZ,DNUM3,0,0,-1)
CALL PLOT(ALX4,ALY4,999)
STOP
END

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MAINLINE PLOTTER

```

C SUBROUTINE PLOTS
C PLOTTING ROUTINE FOR THE TEMPERATURE PROFILE CALCULATED BY THE
C MULTICOMPONENT DISTILLATION PROGRAM ,THIELEGEDES.,
  DIMENSION ALXT(4,5),ALYT(4,5)
  DIMENSION F1(4),F2(4),F3(4),F4(4)
  DIMENSION NCH(10),NCHARX(4),NCHARY(4),TITL(6,5)
  DIMENSION XAXT(5),YAXT(5),TITLT(5)
  DIMENSION X(100),Y(100),XX(100),YY(100),XAX(4,5),YAX(4,5)
  DIMENSION YLEN(4)
  COMMON ALX1,ALY1,ALXT,ALYT,ASIZ,F1,F2,F3,F4
  COMMON NC,NCH,NCHARX,NCHARY,NP,NP2,NR,NRF,TITL
  COMMON X,Y,XX,YY,XAX,YAX,YLEN
  1 FORMAT( )
  2 FORMAT(A4)
  3 FORMAT(3I4)
  4 FORMAT(2F10,2)
  5 FORMAT(5A6)
  DATA NCHR,NCR,NPLOT,NTIMES/3,8,1,0/
  DATA ALXTR,ALXTRI,ALYTR/4,6,5,6,9,7/
  RUN1='RUN'
  C SET AXIS LENGTHS.
  IF(MOD(NP,2)) 102,101,102
101 IF(NP.LT.7) LENX=NP+1
  IF(NP.GT.7.AND.NP.LT.17) LENX=IFIX((NP+1)/2)
  IF(NP.GT.17) LENX=8
  GO TO 103
102 IF(NP.LT.8) LENX=NP+1
  IF(NP.GT.8.AND.NP.LT.16) LENX=(NP+1)/2
  IF(NP.GT.16) LENX=8
103 CONTINUE
  ALENX=FLOAT(LENX)
  ALENY=YLEN(NPLOT)
  C 'PLOT' SETS UP ORIGIN OF AXES.
  CALL PLOT(1,0,1,0,-3)
  C ADD DATA FILE AND READ TEMPERATURE PROFILE DATA FOR FIRST RUN.
  CALL ERTRAN(6,F1)
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0031

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SUBROUTINE PLOTS

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0032 READ(NCR,2) PRD
0033 READ(NCR,3) NMAX,LFREQ,INTEQ
0034 READ(NCR,4) (Y(I),X(I),I=1,NMAX)
      C SCALE X AND Y AXES.
0035 CALL SCALE(X,ALENX,NMAX,1)
0036 CALL SCALE(Y,ALENY,NMAX,1)
      C SELECT AXIS TITLES.
      DO 10 J=1,5
0037   XAXT(J)=XAX(NPLOT,J)
0038   YAXT(J)=YAX(NPLOT,J)
0039
0040   10 CONTINUE
0041   NCY=NCNARY(NPLOT)
0042   NCX=-NCHARX(NPLOT)
      C SET UP X AND Y AXES.
0043   CALL MAXIS(0,0,0,0,XAXT,NCX,ALENX,0,0,X(NMAX+1),X(NMAX+2),-1)
0044   CALL AXIS(0,0,0,0,YAXT,NCY,ALENY,90,0,Y(NMAX+1),Y(NMAX+2))
0045   NTIMES=NTIMES+1
      C CONSTRUCT TEMPERATURE PROFILE FOR FIRST RUN.
0046   CALL LINE(X,Y,NMAX,1,LFREQ,INTEQ)
0047   IF(NR.EQ.1) GO TO 104
      C CONSTRUCT TEMPERATURE PROFILES FOR REMAINING RUNS.
0048   NR1=NR-1
0049   DO 11 J=1,NR1
0050   READ(NCR,2) PRD
0051   READ(NCR,3) NM,LFREQ,INTEQ
0052   READ(NCR,4) (YY(I),XX(I),I=1,NM)
      C NEXT FOUR LINES SET STARTING VALUE AND INCREMENT, FOR EACH DATA
      C SET EQUAL TO THOSE VALUES COMPUTED BY SCALE, FOR THE FIRST SET.
      XX(NM+1)=X(NMAX+1)
      XX(NM+2)=X(NMAX+2)
      YY(NM+1)=Y(NMAX+1)
      YY(NM+2)=Y(NMAX+2)
      NTIMES=NTIMES+1
      CALL LINE(XX,YY,NM,1,LFREQ,INTEQ)
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      11 CONTINUE
      104 CONTINUE

```

SUBROUTINE PLOTTS

C WRITE DIAGRAM HEADING TITLES.

NPL=6

DO 20 I=1.3

ALX=ALXT(NPLOT,I)

ALY=ALYT(NPLOT,I)

NCHT=NCH(NPL)

DO 21 J=1.5

TITLT(J)=TITL(NPL,J)

21 CONTINUE

NPL=NPL-1

IF(NPL.LT.5) NPL=NPLOT

CALL SYMBOL(ALX,ALY,ASIZ,TITLT,0.0,NCHT)

20 CONTINUE

C WRITE OUT SYMBOL LEGEND TABLE ABOVE DIAGRAM,

INTEQ=0

DO 22 I=1.NTIMES

RUNNO=FLOAT(I)

ALYTR=ALYTR+0.3

ALYTRI=ALYTR+0.05

CALL SYMBOL(ALXTR,ALYTR,ASIZ,RUNI,0.0,NCHR)

ALXTRN=ALXTR+0.56

CALL NUMBER(ALXTRN,ALYTR,ASIZ,RUNNO,0.0,-1)

INTEQ=INTEQ+2

CALL SYMBOL(ALXTRI,ALYTRI,ASIZ,INTEQ,0.0,-1)

22 CONTINUE

C COMPLETE DIAGRAM BOUNDARY LINE.

CALL PLOT(0.0,ALENY,3)

CALL PLOT(ALENX,ALENY,2)

CALL PLOT(ALENX,0.0,2)

EXTEND=2.

IF(ALENX.LT.6.) EXTEND=4.

ALXI=ALENX+EXTEND

RETURN

END

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SUBROUTINE PLOTTS



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C
C
C SURROUTINE PLOTES
C PLOTTING ROUTINE FOR THE VAPOUR AND LIQUID FLOWRATE PROFILES
C CALCULATED BY THE MULTICOMPONENT DISTILLATION PROGRAM ,THIELEGEDES.,
  DIMENSION ALXT(4,5),ALYT(4,5)
  DIMENSION F1(4),F2(4),F3(4),F4(4)
  DIMENSION NCH(10),NCHARX(4),NCHARY(4),TITL(6,5)
  DIMENSION XAXT(5),YAXT(5),TITLT(5)
  DIMENSION X(100),Y(100),XX(100),YY(100),XAX(4,5),YAX(4,5)
  DIMENSION YLEN(4)
  COMMON ALX1,ALY1,ALXT,ALYT,ASIZ,F1,F2,F3,F4
  COMMON NC,NCH,NCHARX,NCHARY,NP,NP2,NR,NRF,TITL
  COMMON X,Y,XX,YY,XAX,YAX,YLEN
  1 FORMAT( )
  2 FORMAT(A4)
  3 FORMAT(3I4)
  4 FORMAT(2F10.2)
  5 FORMAT(5A6)
  DATA NCHR,NCR,NPLOT/5,8,2/
  DATA ALXTF,ALXTF1,ALXTR,ALXTR1,ALYTR/4,6,5,6,4,6,4,8,9,0/
  SET AXIS LENGTHS.
  IF(MOD(NP,2)) 102,101,102
101 IF(NP.LT.7) LENX=NP+1
  IF(NP.GT.7.AND.NP.LT.17) LENX=IFIX((NP+1)/2)
  IF(NP.GT.17) LENX=8
  GO TO 103
102 IF(NP.LT.8) LENX=NP+1
  IF(NP.GT.8.AND.NP.LT.16) LENX=(NP+1)/2
  IF(NP.GT.16) LENX=8
103 CONTINUE
  ALENX=FLOAT(LENX)
  ALENY=YLEN(NPLOT)
  *PLOT* SETS THE ORIGIN OF THE AXES.
  CALL PLOT(ALX1,0.0,-3)

```

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0002

0003

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0029

SUBROUTINE PLOTFS

```

C      ADD DATA FILE AND SEARCH FOR THE CORRECT SET OF DATA. (SET 'NRF')
CALL ERTRAN(6,F2)
0030
0031
C      LIQUID PROFILE DATA.
DO 10 J=1,NRF
0032
0033
0034
C      READ(NCR,2) PRD
READ(NCR,3) NMAX,LFREQ,INTEQ
0035
0036
0037
0038
C      READ(NCR,4) (Y(I),X(I),I=1,NMAX)
VAPOUR PROFILE DATA.
0039
0040
C      READ(NCR,2) PRD
READ(NCR,3) NMAX,LFREQ1,INTEQ1
0041
0042
0043
0044
C      READ(NCR,4) (YY(I),XX(I),I=1,NMAX)
10 CONTINUE
0045
0046
0047
0048
0049
0050
C      SCALE AXES.
CALL SCALE(X,ALENX,NMAX,1)
CALL SCALE(Y,ALENY,NMAX,1)
0051
0052
C      SET VAPOUR PROFILE STARTING VALUE, AND INCREMENT EQUAL TO THOSE
C      VALUES COMPUTED BY 'SCALE' FOR THE LIQUID PROFILE DATA.
XX(NMAX+1)=X(NMAX+1)
XX(NMAX+2)=X(NMAX+2)
YY(NMAX+1)=Y(NMAX+1)
YY(NMAX+2)=Y(NMAX+2)
C      SELECT AXIS TITLES.
DO 11 J=1,5
XAXT(J)=XAX(NPLOT,J)
YAXT(J)=YAX(NPLOT,J)
11 CONTINUE
NCY=NCHARY(NPLOT)
NCX=-NCHARX(NPLOT)
C      SET UP X AND Y AXES.
CALL MAXIS(0,0,0,0,XAXT,NCX,ALENX,0,0,X(NMAX+1),X(NMAX+2),-1)
CALL AXIS(0,0,0,0,YAXT,NCY,ALENY,90,0,Y(NMAX+1),Y(NMAX+2))
C      PLOT LIQUID PROFILE.
CALL LINE(X,Y,NMAX,1,LFREQ,INTEQ)
C      PLOT VAPOUR PROFILE.
CALL LINE(XX,YY,NMAX,1,LFREQ1,INTEQ1)
0053
0054

```

SUBROUTINE PLOTFS

```

C      SELECT DIAGRAM HEADING TITLES.
NPL=6
DO 20 I=1,3
  ALX=ALXT(NPLOT,I)
  ALY=ALYT(NPLOT,I)
  NCHT=NCH(NPL)
DO 21 J=1,5
  TITLT(J)=TITL(NPL,J)
21 CONTINUE
  NPL=NPL-1
  IF(NPL.LT.5) NPL=NPLOT
  WRITE TITLES ABOVE DIAGRAM.
C      CALL SYMBOL(ALX,ALY,ASIZ,TITLT,0.0,NCHT)
20 CONTINUE
C      WRITE OUT SYMBOL LEGEND ABOVE DIAGRAM.
  ALIQ=.LIQ.
  AVAP=.VAP.
  NCHF=3
  CALL SYMBOL(ALXTF,ALYTR,ASIZ,AVAP,0.0,NCHF)
  ALXTF1=5.6
  CALL SYMBOL(ALXTF1,ALYTR,ASIZ,ALIQ,0.0,NCHF)
  ALYTR=ALYTR-0.3
  ALYTR1=ALYTR+0.05
  CALL SYMBOL(ALXTR1,ALYTR1,ASIZ,INTEQ1,0.0,-1)
  ALXTR2=ALXTR1+1.0
  CALL SYMBOL(ALXTR2,ALYTR1,ASIZ,INTEQ,0.0,-1)
  COMPLETE DIAGRAM BOUNDARY LINE.
  CALL PLOT(0.0,ALENY,3)
  CALL PLOT(ALENX,ALENY,2)
  CALL PLOT(ALENX,0.0,2)
  ALX1=-ALX1
  ALY1=ALENY+4.0
  RETURN
END

```

SUBROUTINE PLOTFS

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C	SUBROUTINE PLOTDS	0001
C	PLOTTING ROUTINE FOR A DISTILLATE COMPOSITION BAR DIAGRAM, CALCULATED	
	BY THE MULTICOMPONENT DISTILLATION PROGRAM, THIELEGEDES,	
	DIMENSION ALXT(4,5), ALYT(4,5)	0002
	DIMENSION F1(4), F2(4), F3(4), F4(4)	0003
	DIMENSION NCH(10), NCHARX(4), NCHARY(4), TITL(6,5)	0004
	DIMENSION XAXT(5), XPP(35), YAXT(5), TITLT(5)	0005
	DIMENSION X(100), Y(100), XX(100), YY(100), XAX(4,5), YAX(4,5)	0006
	DIMENSION YLEN(4)	0007
	COMMON ALX1, ALY1, ALXT, ALYT, ASIZ, F1, F2, F3, F4	0008
	COMMON NC, NCH, NCHARX, NCHARY, NP, NP2, NR, NRF, TITL	0009
	COMMON X, Y, XX, YY, XAX, YAX, YLEN	0010
	1 FORMAT( )	0011
	2 FORMAT(A4)	0012
	3 FORMAT(3I4)	0013
	4 FORMAT(2F10.2)	0014
	5 FORMAT(5A6)	0015
	DATA NCR, NPLOT/8, 3/	0016
	DATA YMAX1, YMAX2/0.0, 0.2/	0017
C	SET AXIS LENGTHS.	
	LENX=NC+1	0018
	LENAX=NC-1	0019
	IF (LENX.GT.18) LENX=18	0020
	ALENX=FLOAT(LENX)	0021
	ALENAX=FLOAT(LENAX)	0022
	ALENY=YLEN(NPLOT)	0023
	DO 10 J=1,5	0024
	XAXT(J)=XAX(NPLOT,J)	0025
	YAXT(J)=YAX(NPLOT,J)	0026
	10 CONTINUE	0027
	NCX=NCHARX(NPLOT)	0028
	NCY=NCHARY(NPLOT)	0029
C	'PLOT' SETS THE ORIGIN OF THE AXES.	
	CALL PLOT(ALX1, ALY1, -3)	0030

SUBROUTINE PLOTDS

```

C      ADD DATA FILE AND SEARCH FOR THE CORRECT DATA SET. (SET 'NRF')
CALL ERTRAN(6,F3)
DO 11 J=1,NRF
  READ(NCR,2) PRD
  READ(NCR,3) NMAX,LFREQ,INTEQ
  READ(NCR,4) (X(I),Y(I),I=1,NMAX)
11 CONTINUE
  Y(NMAX+1)=YMAX1
  Y(NMAX+2)=YMAX2
  HEIGHT=0.14
  XPAGE=1.0
  YPAGE=-0.3
  CALL PLOT(0.0,0.0,3)
  CALL PLOT(ALENX,0.0,2)
  DEL=ALENX/(NC-1)
  'DO' LOOP 12 SETS OUT AND NUMBERS THE POSITIONS OF THE BARS
  REPRESENTING EACH COMPONENT.
  DO 12 J=1,NC
    XPP(J)=XPAGE-DEL/2.
    COMP=FLOAT(J)
    IF(J.GT.9) XPAGE=XPAGE-0.07
    CALL NUMBER(XPAGE,YPAGE,HEIGHT,COMP,0.0,-1)
    IF(J.GT.9) XPAGE=XPAGE+0.07
    XPAGE=XPAGE+DEL
12 CONTINUE
    XP=(ALENX/2.)-(NCX*HEIGHT/2.)
    YP=-0.6
  WRITE X AXIS TITLE.
  CALL SYMBOL(XP,YP,HEIGHT,XAXT,0.0,NCX)
  SET UP Y AXIS.
  CALL AXIS(0.0,0.0,YAXT,NCY,ALENY,90.0,Y(NMAX+1),Y(NMAX+2))

```

SUBROUTINE PLOTDS

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0056

C	CONSTRUCT BAR DIAGRAM.	0057
	YI=0.0	0058
	DO 13 I=1,NC	0059
	Y(I)=Y(I)*ALENY	0060
	CALL PLOT(XPP(I),YI,3)	0061
	CALL PLOT(XPP(I),Y(I),2)	0062
	XPPP=XPP(I)+DEL	0063
	CALL PLOT(XPPP,Y(I),2)	0064
	CALL PLOT(XPPP,0.0,2)	0065
	YI=Y(I)	0066
13	CONTINUE	
C	SELECT DIAGRAM HEADING TITLES.	0067
	NPL=6	0068
	DO 20 I=1,3	0069
	ALX=ALXT(NPLOT,I)	0070
	ALY=ALYT(NPLOT,I)	0071
	NCHT=NCH(NPL)	0072
	DO 21 J=1,5	0073
	TITLT(J)=TITL(NPL,J)	0074
21	CONTINUE	0075
	NPL=NPL-1	0076
	IF(NPL.LT.5) NPL=NPLOT	
C	WRITE TITLE ABOVE DIAGRAM.	0077
	CALL SYMBOL(ALX,ALY,ASIZ,TITLT,0.0,NCHT)	0078
20	CONTINUE	
C	COMPLETE DIAGRAM BOUNDARY LINE.	0079
	CALL PLOT(0.0,ALENY,3)	0080
	CALL PLOT(ALENX,ALENY,2)	0081
	CALL PLOT(ALENX,0.0,2)	0082
	ALY1=ALENY+4.0	0083
	RETURN	0084
	END	

SUBROUTINE PLOTDS

```

SUBROUTINE PLOTCS(ALENX)
PLOTTING ROUTINE FOR THE LIQUID COMPOSITION PROFILE CALCULATED BY
C THE MULTICOMPONENT DISTILLATION PROGRAM ,THIELEGEDES.,
C
DIMENSION ALXT(4,5),ALYT(4,5)
DIMENSION F1(4),F2(4),F3(4),F4(4)
DIMENSION NCH(10),NCHARX(4),NCHARY(4),TITL(6,5)
DIMENSION XAXT(5),YAXT(5),TITLT(5)
DIMENSION X(100),Y(100),XX(100),YY(100),XAX(4,5),YAX(4,5)
DIMENSION YLEN(4)
COMMON ALX1,ALY1,ALXT,ALYT,ASIZ,F1,F2,F3,F4
COMMON NC,NCH,NCHARX,NCHARY,NP,NP2,NR,NRF,TITL
COMMON X,Y,XX,YY,XAX,YAX,YLEN
1 FORMAT( )
2 FORMAT(A4)
3 FORMAT(3I4)
4 FORMAT(2F10.2)
5 FORMAT(5A6)
DATA NCHR,NCR,NPLOT,NTIMES/4,8,4,0/
DATA ALXTR,ALXTRI,YMAX1,YMAX2/4,6,5,0,0,0,2/
COMPI=.COMP.
C 'PLOT' SETS THE ORIGIN OF THE AXES.
CALL PLOT(0,0,ALY1,-3)
C ADD COMPOSITION DATA FILE.
CALL ERTRAN(6,F4)
NRF=NRFC-1
C SEARCH FOR REQUIRED DATA SET.
IF(NRFC,LT,1) GO TO 100
DO 10 M=1,NRFC
DO 11 J=1,NC
READ(NCR,2) PRD
READ(NCR,3) NMAX,LFREQ,INTEQ
READ(NCR,4) (X(I),Y(I),I=1,NMAX)
11 CONTINUE
10 CONTINUE
100 CONTINUE

```

SUBROUTINE PLOTCS

```

C      READ COMPOSITION DATA FOR LIGHT COMPONENT.
      READ(NCR,2) PRD
      READ(NCR,3) NMAX,LFREQ,INTEQ
      READ(NCR,4) (X(I),Y(I),I=1,NMAX)
      Y(NMAX+1)=YMAX1
      Y(NMAX+2)=YMAX2
C      SELECT AXIS TITLES.
      DO 12 J=1,5
        XAXT(J)=XAX(NPLOT,J)
        YAXT(J)=YAX(NPLOT,J)
12      CONTINUE
        NCX=-NCHARX(NPLOT)
        NCY=NCHARY(NPLOT)
        SET AXIS LENGTHS.
        ALENY=YLEN(NPLOT)
        LENX=NMAX-1
        ALENX=FLOAT(LENX)
        IF(NMAX.GT.19) ALENX=18.
        SCALE X AXIS. (TRAY NO.)
        CALL SCALE(X,ALENX,NMAX,1)
        SET UP X AND Y AXES.
        CALL MAXIS(0.0,0.0,0.0,XAXT,NCX,ALENX,0.0,X(NMAX+1),X(NMAX+2),-1)
        CALL AXIS(0.0,0.0,0.0,YAXT,NCY,ALENY,90.0,Y(NMAX+1),Y(NMAX+2))
        NTIMES=NTIMES+1
C      PLOT LIGHT COMPONENT COMPOSITION DATA.
        CALL LINE(X,Y,NMAX,1,LFREQ,INTEQ)
C      READ AND PLOT DATA FOR REMAINING COMPONENTS.
        NC1=NC-1
        DO 13 J=1,NC1
          READ(NCR,2) PRD
          READ(NCR,3) NM,LFREQ,INTEQ
          READ(NCR,4) (XX(I),YY(I),I=1,NM)
          XX(NM+1)=X(NMAX+1)
          XX(NM+2)=X(NMAX+2)
          YY(NM+1)=Y(NMAX+1)
          YY(NM+2)=Y(NMAX+2)
          NTIMES=NTIMES+1
          CALL LINE(XX,YY,NM,1,LFREQ,INTEQ)
13      CONTINUE

```



C DETERMINE SIZE OF SYMBOL LEGEND TABLE.

```

NCNT=5
ALYTRS=6.7
IF(NTIMES.GT.15) NCNT=6
IF(NTIMES.GT.18) NCNT=7
IF(NTIMES.GT.21) NCNT=8
IF(NTIMES.GT.24) NCNT=9
IF(NTIMES.GT.27) NCNT=10
IF(NCNT.EQ.6) ALYTRS=7.0
IF(NCNT.EQ.7) ALYTRS=7.3
IF(NCNT.EQ.8) ALYTRS=7.6
IF(NCNT.EQ.9) ALYTRS=7.9
IF(NCNT.EQ.10) ALYTRS=8.2
SELECT DIAGRAM HEADING TITLES.
NPL=6
DO 20 I=1,3
  ALX=ALXT(NPLOT,I)
  ALY=ALYT(NPLOT,I)
  NCHT=NCH(NPL)
  DO 21 J=1,5
    TITLT(J)=TITL(NPL,J)
  21 CONTINUE
  NPL=NPL+1
  IF(NPL.LT.5) NPL=NPLOT
  WRITE TITLE ABOVE DIAGRAM.
  CALL SYMBOL(ALX,ALY,ASIZ,TITLT,0,0,NCHT)
  20 CONTINUE

```

C

```

NPL=6
DO 20 I=1,3
  ALX=ALXT(NPLOT,I)
  ALY=ALYT(NPLOT,I)
  NCHT=NCH(NPL)
  DO 21 J=1,5
    TITLT(J)=TITL(NPL,J)
  21 CONTINUE
  NPL=NPL+1
  IF(NPL.LT.5) NPL=NPLOT
  WRITE TITLE ABOVE DIAGRAM.
  CALL SYMBOL(ALX,ALY,ASIZ,TITLT,0,0,NCHT)
  20 CONTINUE

```

C

SUBROUTINE PLOTCS

```

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```

C WRITE OUT SYMBOL LEGEND ABOVE DIAGRAM.

```
INTEQ=0
NTIM=0
DO 22 J=1.3
  ALYTR=ALYTRS
DO 23 I=1.NCNT
  NTIM=NTIM+1
  COMPNO=FLOAT(NTIM)
  ALYTR=ALYTR-0.3
  ALYTRI=ALYTR+0.05
  CALL SYMBOL(ALXTR,ALYTR,ALYTRI,ASIZ,COMP1,0.0,NCHR)
  ALXTRN=ALXTR+0.7
  CALL NUMBER(ALXTRN,ALYTR,ASIZ,COMPNO,0.0,-1)
  INTEQ=INTEQ+1
  CALL SYMBOL(ALXTRI,ALYTRI,ALYTRI,ASIZ,INTEQ,0.0,-1)
  IF(NTIM.EQ.NTIMES) GO TO 210
23 CONTINUE
  ALXTR=ALXTR+1.5
  ALXTRI=ALXTRI+1.5
22 CONTINUE
210 CONTINUE
  COMPLETE DIAGRAM BOUNDARY LINE.
  CALL PLOT(0.0,ALENY,3)
  CALL PLOT(ALENX,ALENY,2)
  CALL PLOT(ALENX,0.0,2)
  RETURN
END
```

SUBROUTINE PLOTCS

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0111

```

C SUBROUTINE PLOTBT(ALENX,NTYPE)
C PLOTTING ROUTINE FOR CONSTRUCTION OF A MCCABE-THIELE DIAGRAM
C FOR A BINARY BATCH-DISTILLATION SYSTEM, OPERATING UNDER EITHER
C CONSTANT OR VARIABLE REFLUX.

```

```

  DIMENSION XAX(5),YAX(5),X(55),Y(55),XEQ(55),YEQ(55)

```

```

  1 FORMAT( )

```

```

  2 FORMAT(I5)

```

```

  3 FORMAT(2(2X,F6.4))

```

```

  4 FORMAT(F7.4,F10.2)

```

```

  5 FORMAT(5A6)

```

```

  DATA ALENX,ALENY,ASIZ,NCR/10,,10,,0,14,8/

```

```

  DATA XSTRT,YSTRT,XINT,YINT/0.0,0.0,0.1,0.1/

```

```

  C ADD ELEMENT AND READ AXIS TITLES.

```

```

  CALL ERTRAN(6,,@ADD PLOTFILE,AXISMT , )

```

```

  READ(NCR,5) (XAX(J),J=1,5)

```

```

  READ(NCR,5) (YAX(J),J=1,5)

```

```

  READ(NCR,1) NCAX,NCAY

```

```

  NCX=-NCAX

```

```

  NCY=NCAY

```

```

C *PLOT* SETS THE ORIGIN OF THE AXES.

```

```

  CALL PLOT(1,0,1,0,-3)

```

```

C SET UP AXES AND START CONSTRUCTION OF MCCABE-THIELE DIAGRAM.

```

```

  CALL AXIS(0,0,0,0,XAX,NCX,ALENX,0,0,XSTRT,XINT)

```

```

  CALL AXIS(0,0,0,0,YAX,NCY,ALENY,90,0,YSTRT,YINT)

```

```

  CALL PLOT(0,0,ALENY,3)

```

```

  CALL PLOT(ALENX,ALENY,2)

```

```

  CALL PLOT(ALENX,0,0,2)

```

```

  CALL PLOT(0,0,0,3)

```

```

  CALL PLOT(ALENX,ALENY,2)

```

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SUBROUTINE PLOTBT

```

C      ADD DATA FILE AND READ EQUILIBRIUM DATA.
      CALL ERTRAN(6,0,0,BDPLOT,.,.)
      READ(NCR,2) NDAT
      READ(NCR,3) (XEQ(I),YEQ(I),I=1,NDAT)
      XEQ(NDAT+1)=XSTRT
      XEQ(NDAT+2)=XINT
      YEQ(NDAT+1)=YSTRT
      YEQ(NDAT+2)=YINT
C      PLOT EQUILIBRIUM CURVE.
      CALL LINE(XEQ,YEQ,NDAT,1,0,1)
C      READ AND PLOT DATA FOR COMPLETING MCCABE-THIELE DIAGRAM.
      DO 20 J=1,2
      READ(NCR,2) MT
      READ(NCR,3) (X(I),Y(I),I=1,MT)
      READ(NCR,3) YINCPT,XD
      PLOT OPERATING LINE.
      YNCPT=YINCPT*10.
      XDA=XD*10.
      CALL PLOT(0,0,YNCPT,3)
      CALL PLOT(XDA,XDA,2)
      X(MT+1)=XSTRT
      X(MT+2)=XINT
      Y(MT+1)=YSTRT
      Y(MT+2)=YINT
C      INSERT MCCABE-THIELE STEPPED STAGES.
      CALL LINE(X,Y,MT,1,0,1)

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SUBROUTINE PLOTBT

```

C      NUMBER STEPPED STAGES.
      ANO=0.
      MT1=MT-1
      DO 21 I=MT1,2,-2
      XANO=X(I)*10.-0.1
      YANO=Y(I)*10.+0.1
      CALL NUMBER(XANO,YANO,ASIZ,ANO,0.0,-1)
      ANQ=ANO+1.
21    CONTINUE
20    CONTINUE
      IF(NTYPE,NE,2) GO TO 999
      PLOT AVERAGE DISTILLATE COMPOSITION ON DIAGRAM.
      READ(NCR,3) XDAVE,XD
      XDAV=XDAVE*10.
      XDAN=XDAV+0.12
      CALL PLOT(XDAV,0.0,3)
      CALL PLOT(XDAV,XDAV,2)
      YDAN=0.2
      HT=ASIZ
      ANO='XD'
      NCHA=2
      DO 30 I=1,2
      CALL SYMBOL(XDAN,YDAN,HT,ANO,0.0,NCHA)
      XDAN=XDAN+0.28
      YDAN=0.1
      HT=0.07
      ANO='AVE'
      NCHA=3
30    CONTINUE
999   CONTINUE
      RETURN
      END

```

SUBROUTINE PLOTBT

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C	SUBROUTINE PLOTMT(ALENX,NTYPE,YST1,YST2)	0001
C	PLOTTING ROUTINE FOR CONSTRUCTION OF A MCCABE-THIELE OR PONCHON-	
	SAVARIT DIAGRAM FOR A BINARY CONTINUOUS-DISTILLATION SYSTEM,	
	DIMENSION EXAX(5),EYAX(5),HLV(105),XYH(105)	0002
	DIMENSION XAX(5),YAX(5),X(55),Y(55),XEQ(55),YEQ(55)	0003
	DIMENSION XOP(105),YOP(105),XQ(15),YQ(15),ZFLO(15),ZFS(15)	0004
	1 FORMAT( )	0005
	2 FORMAT(15)	0006
	3 FORMAT(2(2X,F6.4))	0007
	4 FORMAT(F7.4,F10.2)	0008
	5 FORMAT(5A6)	0009
	6 FORMAT(2E12.6)	0010
	7 FORMAT(F6.4,E12.6)	0011
	DATA ASTRT,BSVRT,ASIZ,NCR/1.0,1.0,0.14,8/	0012
	IF(NTYPE.EQ.4) GO TO 101	0013
C	CONSTRUCTION OF ENTHALPY-CONCENTRATION DIAGRAM.	
C	ADD ELEMENT AND READ AXIS TITLES.	
C	CALL ERTRAN(6,,@ADD PLOTFILE,AXISPS , )	0014
C	READ(NCR,5) (EXAX(J),J=1,5)	0015
	READ(NCR,5) (EYAX(J),J=1,5)	0016
	READ(NCR,1) NCEAX,NCEAY	0017
	NCEX=-NCEAX	0018
	NCEY=NCEAY	0019
	ALENX=10.	0020
	ALENY=15.	0021

SUBROUTINE PLOTMT

```

C      *PLOT. SETS THE ORIGIN OF THE ENTHALPY DIAGRAM.
C      (IE. LOWER LEFT-HAND CORNER)
CALL PLOT(1.0,13.0,-3)
ASTRT=0.0
BSTRT=-12.0
XSTRT=0.0
XINT=0.1
YSTRT=YST2
YINT=(YST1-YST2)/ALENY
*AXIS. SETS UP THE Y AXIS.
CALL AXIS(0.0,0.0,EYAX,NCEY,ALENY,90.0,YSTRT,YINT)
XST=ABS(YST2/YINT)
*AXIS. SETS UP THE X AXIS.
CALL AXIS(0.0,XST,EXAX,NCEX,ALENX,0.0,XSTRT,XINT)
ADD DATA FILE.
CALL ERTRAN(6,,@ADD BDPLT. . .)
READ COORDINATES OF KEY POINTS ON ENTHALPY DIAGRAM.
C      (IE. *DEL* POINTS AND FEED,BOTTOMS AND DISTILLATE POSITIONS,
READ(NCR,6) DELD,DELB
READ(NCR,7) XD,HD
READ(NCR,7) XB,HB
READ(NCR,7) Z,HF
XD=XD*10.
XB=XB*10.
Z=Z*10.
DELD=DELD/YINT+XST
DELB=DELB/YINT+XST
HD=HD/YINT+XST
HB=HB/YINT+XST
HF=HF/YINT+XST
C      INSERT LINES REPRESENTING OVERALL BALANCES ON ENTHALPY DIAGRAM.
CALL PLOT(XB,HB,3)
CALL PLOT(XB,DELB,2)
CALL PLOT(XD,DELD,2)
CALL PLOT(XD,HD,2)
CALL SYMBOL(Z,HF,ASIZ,1,0.0,-1)

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SUBROUTINE PLOTMT

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DE added=DE added+0.2
XDD=XD+ASIZ
DELXD=DE added+0.1
DE added=DE added-0.2
XBB=XB+ASIZ
DE added=DE added-0.3
HFF=HF+0.2
ZFF=Z-0.2
C MARK KEY POINTS ON ENTHALPY DIAGRAM.
CALL SYMBOL(ZFF,HFF,ASIZ,35,0,0,-1)
CALL SYMBOL(XD,DE added,ASIZ,28,0,0,-1)
CALL SYMBOL(XDD,DE added,ASIZ,33,0,0,-1)
CALL SYMBOL(XB,DE added,ASIZ,28,0,0,-1)
CALL SYMBOL(XBB,DE added,ASIZ,31,0,0,-1)
READ NUMBER OF DATA POINTS AVAILABLE FOR PLOTTING ENTHALPY CURVES.
READ(NCR,2) NENTH
C
C *DO* LOOP II READS DATA AND PLOTS THE ENTHALPY CURVES.
C LIQUID ENTHALPY DATA SUPPLIED FIRST IN DATA FILE BY 'BINARY',
DO II I=1,2
READ(NCR,7) (XYH(J),HLV(J),J=1,NENTH)
XYH(NENTH+1)=XSTRT
XYH(NENTH+2)=XINT
HLV(NENTH+1)=YSTRT
HLV(NENTH+2)=YINT
C PLOT ENTHALPY CURVES.
CALL LINE(XYH,HLV,NENTH,1,0,1)
II CONTINUE
C COMPLETE BOUNDARY LINE.
CALL PLOT(0,0,ALEHY,3)
CALL PLOT(ALENX,ALENY,2)
CALL PLOT(ALENX,0,0,2)
CALL PLOT(0,0,0,0,2)
101 CONTINUE

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SUBROUTINE PLOTMT





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C      PLOT EQUILIBRIUM CURVE.
      CALL LINE(XEQ,YEQ,NDAT,1,0,1)
      IF(NTYPE.EQ.5) GO TO 103
      READ OPERATING LINE DATA FOR MCCABE-THIELE X-Y DIAGRAM.
      READ(NCR,2) NQDAT
      READ(NCR,3) (XQ(I),YQ(I),I=1,NQDAT)
      DO 10 I=1,NQDAT
      XQ(I)=XQ(I)*10.
      YQ(I)=YQ(I)*10.
10      CONTINUE
C      PLOT OPERATING LINES ON MCCABE-THIELE X-Y DIAGRAM.
      YB=YQ(1)
      XB=XQ(1)
      CALL PLOT(XB,YB,3)
      DO 20 I=2,NQDAT
      CALL PLOT(XQ(I),YQ(I),2)
20      CONTINUE
      GO TO 104
103     CONTINUE
C      READ OPERATING LINE DATA FOR PONCHON-SAVARIT X-Y DIAGRAM.
      DO 12 I=1,2
      READ(NCR,2) NPTS
      READ(NCR,3) (XOP(J),YOP(J),J=1,NPTS)
      XOP(NPTS+1)=XSTRT
      XOP(NPTS+2)=XINT
      YOP(NPTS+1)=YSTRT
      YOP(NPTS+2)=YINT
C      PLOT OPERATING LINES ON PONCHON-SAVARIT X-Y DIAGRAM.
      CALL LINE(XOP,YOP,NPTS,1,0,1)
12      CONTINUE
104     CONTINUE
C      READ DATA FOR CONSTRUCTING STEPPED STAGES ON X-Y DIAGRAM.
      READ(NCR,2) MT
      READ(NCR,3) (X(I),Y(I),I=1,MT)
      READ(NCR,2) NN
      READ(NCR,4) (ZFS(I),ZFLO(I),I=2,NN)
      IF(NTYPE.EQ.4) GO TO 105

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SUBROUTINE PLOTMT

C READ COORDINATES OF POINT OF INTERSECTION OF OPERATING LINES  
C FOR PONCHON-SAVARIT X-Y DIAGRAM.

READ(NCR,3) XQ(2),YQ(2)

XQ(2)=XQ(2)\*10,

YQ(2)=YQ(2)\*10,

105 CONTINUE

X(MT+1)=XSTRT

X(MT+2)=XINT

Y(MT+1)=YSTRT

Y(MT+2)=YINT

C INSERT MCCAHE-THIELE STEPPED STAGES,

CALL LINE(X,Y,MT,1,0,1)

C NUMBER STEPPED STAGES,

ANO=0,

MT1=MT-1

DO 21 I=2,MT1,2

XANO=X(I)\*10,-0,1

YANO=Y(I)\*10,+0,1

CALL NUMBER(XANO,YANO,ASIZ,ANO,0,0,-1)

ANO=ANO+1,

21 CONTINUE

YP1=0.21

YP2=0.0

INT=22

C INSERT FEED AND SIDE STREAM Q LINES,

DO 30 I=2,NN

Z=ZFS(I)\*10,

Z1=Z+0.05

IF(ZFLO(I).GT.0.0) NZ=35

IF(ZFLO(I).LT.0.0) NZ=48

CALL SYMBOL(Z1,YP1,ASIZ,NZ,0,0,-1)

CALL PLOT(Z,0,0,3)

CALL PLOT(Z,Z,2)

CALL PLOT(XQ(I),YQ(I),2)

30 CONTINUE

RETURN

END

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SUBROUTINE PLOTMT